

The Mechanistic Significance of the Si-O-Pd Bond in the Palladium Catalyzed Cross-Coupling Reactions of Arylsilanolates

Steven A. Tymonko, Russell C. Smith, Andrea Ambrosi, Michael H. Ober, Hao Wang
and Scott E. Denmark*

*Roger Adams Laboratory, Department of Chemistry, University of Illinois, Urbana, Illinois,
61801*

SUPPORTING INFORMATION

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General Experimental

All reactions were performed in oven (140 °C) and/or flame dried glassware (including NMR tubes for kinetic experiments) under an atmosphere of dry nitrogen or argon, unless noted. Reaction solvents tetrahydrofuran (Fisher, HPLC grade) and diethyl ether (Fisher, BHT stabilized ACS grade) were dried by percolation through two columns packed with neutral alumina under a positive pressure of argon. Reaction solvents hexanes (Fisher, OPTIMA grade) and toluene (Fisher, ACS grade) were dried by percolation through a column packed with neutral alumina and a column packed with Q5 reactant, a supported copper catalyst for scavenging oxygen, under a positive pressure of argon. Reaction solvent dioxane was distilled over sodium prior to use. Benzene (Fisher, ACS grade) was distilled from sodium; benzotrifluoride (Aldrich, ACS grade) was distilled from CaH₂. Reaction solvent DMF was purchased from Acros containing 50 ppm water stored over 4 Å mol. sieves. Reaction solvent acetone (Aldrich) was dried over 4 Å molecular sieves prior to use. Solvents for filtration and chromatography were certified ACS grade. “Brine” refers to a saturated solution of sodium chloride. All reaction temperatures correspond to internal temperatures measured with Teflon coated thermocouples unless otherwise noted.

¹H and ¹³C NMR spectra were recorded on a Varian Unity 500 (500 MHz, ¹H; 126 MHz, ¹³C) or a Varian VXR (500 MHz, 1H; 126 MHz, ¹³C) spectrometer. Spectra are referenced to residual chloroform (δ 7.26 ppm, ¹H; δ 77.0 ppm, ¹³C), residual benzene (δ 7.16 ppm, ¹H; δ 128.0 ppm, ¹³C), residual CD₂Cl₂ (5.32 ppm, ¹H; 53.8 ppm, ¹³C), CFCl₃ (10% in CDCl₃) as an external reference for ¹⁹F NMR (δ 0.00 ppm ¹⁹F), and phosphoric acid (40% in H₂O) as an external reference for ³¹P NMR (δ 0.00 ppm ³¹P). Chemical shifts are reported in ppm, multiplicities are indicated by s (singlet), d (doublet), t (triplet), q (quartet), p (pentet), h (heptet), m (multiplet) and br (broad). Coupling constants, *J*, are reported in Hertz. All assignments are corroborated by 2D experiments (COSY, HETCOR). Spectra are available on request from sdenmark@illinois.edu. Elemental analyses were performed by the University of Illinois Microanalytical Service Laboratory. Mass Spectrometry was performed by the University of Illinois Mass Spectrometer Center. Electron Impact (EI) spectra were performed on a Finnegan-MAT C5 spectrometer. Data are reported in the form of *m/z* (intensity relative to the base peak = 100). Infrared spectra (IR) were recorded on a Perkin Elmer Spectrum BX spectrophotometer in

NaCl cells (film) or as a KBr plate. Peaks are reported in cm^{-1} with indicated relative intensities: s (strong, 67-100%); m (medium, 34-66%), w (weak, 0-33%). Kugelrohr distillations were performed on a Büchi GKR-50 Kugelrohr and boiling points correspond to uncorrected air bath temperatures (ABT). Melting points were obtained in a vacuum-sealed capillary tube using a Thomas Hoover melting point apparatus and are corrected. Analytical thin-layer chromatography was performed on Merck silica gel plates with QF-254 indicator. Visualization was accomplished with UV (254). Column chromatography was performed using 230-400 mesh silica gel purchased from Silicycle.

Analytical gas chromatography (GC) was performed using a Hewlett Packard 5890 Series II Gas Chromatograph fitted with a flame ionization detector (H_2 carrier gas, 1mL/min) Injections were made on a Hewlett-Packard HP-1 (30 meter) capillary column. The injector temperature was 250 °C, the detector temperature was 300 °C, with a split ratio of 100:1. Retention times (t_R) and integrated ratios were obtained using Agilent Chemstation Software.

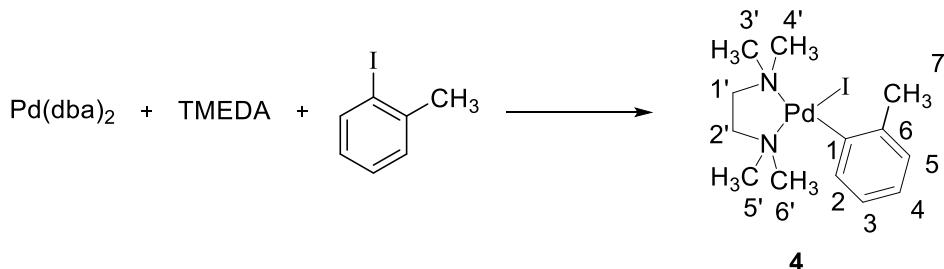
2-Bromotoluene, 2-iodotoluene, 1-bromo-4-fluorobenzene, TMEDA and 2-butanone were purchased from Aldrich and distilled prior to use. Cs^0 (Strem), CuTC (Aldrich), 2-(dimethylamino)ethanethiol hydrochloride (Aldrich), Ph_3P (Aldrich), 1,3-diphenylphosphinopropane (Organometallics), 1,2-diphenylphosphinoethane (Organometallics), 1,2-diphenylphosphinobenzene (Organometallics), 1,1'-bis(diphenylphosphino)ferrocene (Organometallics), $t\text{-Bu}_3\text{P}$ (Strem), $t\text{-Bu}_3\text{P}\cdot\text{HBF}_4$ (Aldrich), and $(t\text{-Bu}_3\text{P})_2\text{Pd}$ (Aldrich)¹ were used as received.

Literature Procedures

Allylpalladium(II)chloride dimer,² $\text{Pd}(\text{dba})_2$,³ 1,3-bis(diphenylphosphino)butane monoxide,⁴ 1,3-bis(diphenylphosphino)butane dioxide,⁵ KTC,⁶ (4-methoxyphenyl)dimethylsilanol **5**,⁷ **8**,⁸ potassium (4-methoxyphenyl)dimethylsilanolate $\text{K}^+\text{5}^-$,⁹ cesium (4-methoxyphenyl)dimethylsilanolate $\text{Cs}^+\text{5}^-$,⁹ 4'-methoxy-2-methylbiphenyl **12**,¹⁰ **13p**,¹¹ (4-methoxyphenyl)dimethylsilane **14**,¹² 4-methoxy-4'-fluorobiphenyl **16**,¹³ potassium (4-*n*-butoxyphenyl)dimethylsilanolate $\text{K}^+\text{21}^-$,¹⁴ 4-*n*-butoxy-4'-fluorobiphenyl **22**.¹⁵

**PREPARATION OF ARYL-DERIVED PALLADIUM(II) SILANOLATE COMPLEXES
(SCHEME 1)**

Preparation of [Tetramethylethylenediamine]iodo(2-methylphenyl)palladium (4**)¹⁶**



In a drybox, a 35 mL, one-neck, round-bottom flask was charged with $\text{Pd}(\text{dba})_2$ (1.15 g, 2.0 mmol), tetramethylethylenediamine (0.39 mL, 2.6 mmol, 1.3 equiv), benzene (10 mL) and 2-iodotoluene (0.36 mL, 2.8 mmol, 1.4 equiv). The flask was sealed with a septum, removed from the drybox, and heated in an oil bath at 50 °C for 2 h. The flask was returned to the drybox and the reaction mixture was filtered through 2 g of diatomaceous earth. All solvents were then removed *in vacuo*. Ether (20 mL) was added to the resulting yellow solids and the mixture was stirred for 5 min. The supernatant was then removed by pipette. This washing process was repeated 3 times and the solids were collected on a glass fritted funnel, and were dried under vacuum (0.1 mm Hg) overnight to afford 362 mg (41%) of **4** as a pale orange powder.

Data for [Tetramethylethylenediamine]iodo(2-methylphenyl)palladium (**4**):

mp: 179-181 °C (dec)

$^1\text{H NMR}$: (500 MHz, C_6H_6)

7.43 (d, $J = 7.5$, 1 H, HC(2)), 7.03 (d, $J = 7.5$, 1 H, HC(5)), 6.91-6.98 (m, 2 H, HC(3 and 4)), 2.92 (s, 3 H, $\text{H}_3\text{C}(3',4',5' \text{ or } 6')$), 2.25 (s, 3 H, $\text{H}_3\text{C}(7)$), 2.16 (s, 3 H, $\text{H}_3\text{C}(3',4',5' \text{ or } 6')$), 1.71 (s, 3 H, $\text{H}_3\text{C}(3',4',5' \text{ or } 6')$), 1.64 (t, $J = 9.5$, 1 H, $\text{H}_2\text{C}(1' \text{ or } 2')$), 1.53 (t, $J = 9.5$, 1 H, $\text{H}_2\text{C}(1' \text{ or } 2')$), 1.45 (s, 3 H, $\text{H}_3\text{C}(3',4',5' \text{ or } 6')$), 1.30 (t, $J = 9.5$, 2 H, $\text{H}_2\text{C}(1' \text{ or } 2')$)

$^{13}\text{C NMR}$: (126 MHz, C_6H_6)

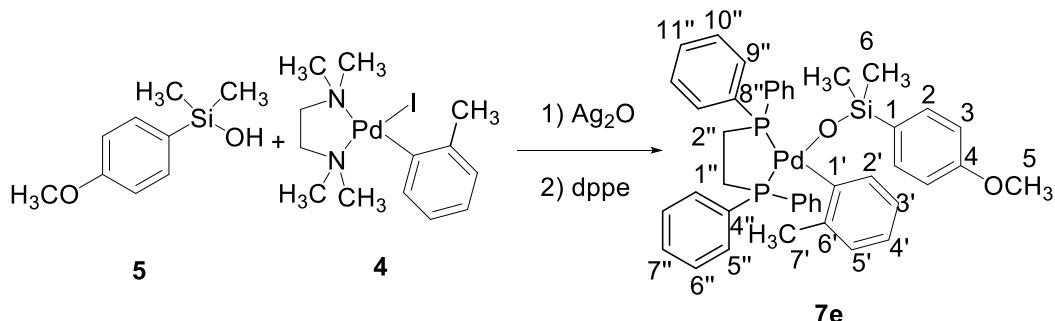
145.7 (C(1)), 142.2 (C(6)), 136.1 (C(4)), 127.3 (C(3)), 123.9 (C(5)), 123.0 (C(2)), 61.2 (C(1' or 2')), 57.4 (C(1' or 2')), 49.9 (C(3',4',5', or 6')), 49.4 (C(3',4',5', or 6')), 48.1 (C(3',4',5', or 6')), 47.9 (C(3',4',5', or 6')), 28.3 (C(7))

Analysis: C₁₇H₂₈IN₂Pd (440.66)

Calcd: C, 35.43; H, 5.26% N, 6.36% I, 28.80%

Found: C, 35.27; H, 5.14% N, 6.22% I, 29.07%

Preparation of [1,2-Bis(diphenylphosphino)ethane](2-methylphenyl)[1-(4-methoxyphenyl)-1,1,-dimethylsilanolato-κ-O]palladium (7e)



In a drybox, a 50-mL, one-necked flask equipped with a magnetic stir bar was charged with **4** (220 mg, 0.50 mmol), (4-methoxyphenyl)dimethylsilanol (100 mg, 0.55 mmol, 1.1 equiv) and THF (10 mL). While stirring, Ag₂O (115 mg, 0.50 mmol, 1.0 equiv) was added and the resulting suspension was stirred in the drybox for 12 h. The solvents were removed *in vacuo* (0.1 mm Hg), then toluene (10 mL) was added, and the mixture was filtered through 2 g of diatomaceous earth. The solvents were removed *in vacuo* (0.1 mm Hg) for 4 h to afford a yellow wax. THF (5 mL) and dppe (199 mg, 0.50 mmol, 1.00 equiv) were added and the resulting solution was stirred for 16 h in the drybox. The solvent was removed *in vacuo* to afford an off-white foam. The foam was dissolved in a minimal amount of benzene (1 mL) and was layered with hexane (5 mL). The solution was allowed to sit at room temperature for 24 h then at -20 °C for 24 h. The resulting solids were collected on a fritted glass funnel, then were washed with hexanes (5 mL), and dried under vacuum (0.1 mm Hg) for 12 h to afford 284 mg (73%) of **7e** as pale yellow needles.

Data for [1,2-bis(diphenylphosphino)ethane](2-methylphenyl)[1-(4-methoxyphenyl)-1,1,-dimethylsilanolato-κ-O]palladium (7e):

¹H NMR: (500 MHz, C₆H₆)

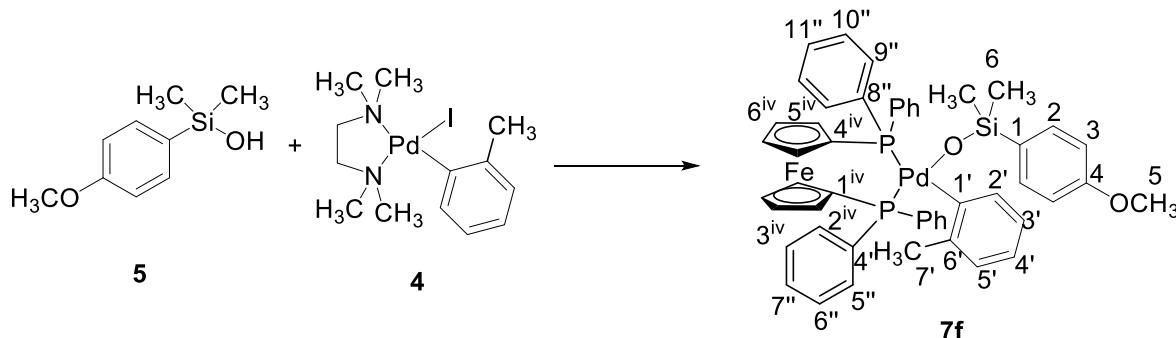
8.33 (t, *J* = 9.5, 2 H, HC(6'' or 10'')), 8.06 (t, *J* = 9.0, 2 H, HC(6'' or 10'')), 7.78 (d, *J* = 8.0, 2 H, HC(2)), 7.60 (dd, *J* = 8.0, 11.5, 2 H, HC(6'' or 10'')), 7.46 (t, *J* = 7.5, 1 H, HC(2')), 6.87-7.15 (m, 14H, CH(5'', 7'', 9'', 11'', and 6'' or 10'')), 6.73

(t, 8.0, 1 H, HC(2' or 5')), 6.72 (t, 8.0, 1 H, HC(2' or 5')), 6.62 (t, J = 9.5, 2 H, HC(3' and 4')), 3.40 (s, 3 H, H₃C(5)), 2.46 (s, 3 H, H₃C(7')), 1.80-2.2 (bm, 3 H, H₂C(2'') and 2 x 1'')), 1.20-1.40 (bs, 1 H, H₂C(2'')), 0.41 (s, 3 H, H₃C(6)), 0.39 (s, 3 H, H₃C(6))

³¹P NMR: (202 MHz, C₆H₆CH₃)
50.8 (d, J = 46.1), 26.0 (d, J = 46.1)

Analysis: C₄₂H₄₄O₂P₂PdSi (777.25)
Calcd: C, 64.90; H, 5.71%
Found: C, 64.78 H, 5.65%

Preparation of [1,1'-Bis(diphenylphosphino)ferrocene](2-methylphenyl)[1-(4-methoxyphenyl)-1,1,-dimethylsilanolato- κ -O]palladium (7f)



In a drybox, a 50-mL, one-necked flask equipped with a magnetic stir bar was charged with **4** (220 mg, 0.50 mmol), (4-methoxyphenyl)dimethylsilanol (100 mg, 0.55 mmol, 1.1 equiv) and THF (10 mL). While stirring, Ag₂O (115 mg, 0.50 mmol, 1.0 equiv) was added and the resulting suspension was stirred in the drybox for 12 h. The solvents were removed *in vacuo* (0.1 mm Hg), then toluene (10 mL) was added, and the mixture was filtered through 2 g of diatomaceous earth. The solvents were removed *in vacuo* (0.1 mm Hg) for 4 h to afford a yellow wax. THF (5 mL) and dppf (249 mg, 0.50 mmol, 1.00 equiv) were added and the resulting solution was stirred for 36 h in the drybox. The solvent was removed *in vacuo* to afford an off-white foam. The foam was dissolved in a minimal amount of benzene (1 mL) and was layered with hexane (5 mL). The solution was allowed to sit at room temperature for 24 h then at -20 °C for 24 h. The resulting solids were collected on a fritted glass funnel, then were washed with hexanes (5 mL), and dried under vacuum (0.1 mm Hg) for 12 h to afford 268 mg (64%) of **7f** as

an orange powder.

Data for [1,1'-bis(diphenylphosphino)ferrocene](2-methylphenyl)[1-(4-methoxyphenyl)-1,1-dimethylsilanolato- κ -O]palladium (7f):

¹H NMR: (500 MHz, C₆H₆)

8.42 (dd, *J* = 8.5, 9.5 8 H, HC(6'' and 10'')), 7.43 (t, *J* = 7.5, 1 H, HC(2'')), 7.30 (d, *J* = 8.0, 2 H, HC(2)), 7.24 (t, *J* = 7.5, 2 H, HC(7'' or 11'')), 7.09-7.18 (m, 2H, CH(7'' or 11'')), 6.90 (d, *J* = 6.5, 1 H, HC(5'')), 6.79 (d, *J* = 8.5, 2 H, HC(3)), 6.74-6.80 (m, 6H, HC(3', 4', and 5'' or 9'')), 6.67 (t, *J* = 7.5, 4 H, HC(5'' or 9'')), 4.80 (s, 1 H, HC(3^{iv} or 6^{iv})), 4.36 (s, 1 H, HC(3^{iv} or 6^{iv})), 4.04 (s, 1 H, HC(3^{iv} or 6^{iv})), 3.96 (s, 1 H, HC(3^{iv} or 6^{iv})), 3.65 (s, 1 H, HC(2^{iv} or 5^{iv})), 3.63 (s, 2 H, HC(2^{iv} or 5^{iv})), 3.56 (s, 1 H, HC(2^{iv} or 5^{iv})), 3.37 (s, 3 H, H₃C(5)), 2.71 (s, 3 H, H₃C(7')), 0.19 (s, 3 H, H₃C(6)), 0.08 (s, 3 H, H₃C(6))

³¹P NMR: (202 MHz, C₆H₆CH₃)

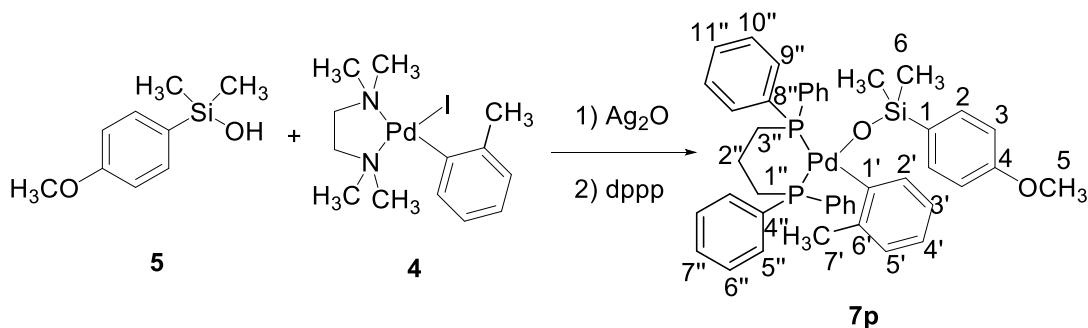
33.4 (d, *J* = 33.8), 9.2 (d, *J* = 33.8)

Analysis: C₅₀H₄₈FeO₂P₂PdSi (933.21)

Calcd: C, 64.35; H, 5.18%

Found: C, 64.32 H, 5.14%

Preparation of [1,3-Bis(diphenylphosphino)propane](2-methylphenyl)[1-(4-methoxyphenyl)-1,1-dimethylsilanolato- κ -O]palladium (7p)



In a drybox, a 50-mL, one-necked flask equipped with a magnetic stir bar was charged with **4** (620 mg, 1.40 mmol), (4-methoxyphenyl)dimethylsilanol (280 mg, 1.54 mmol, 1.1 equiv) and THF (25 mL). While stirring, Ag₂O (323 mg, 1.40 mmol, 1.0 equiv) was added and the resulting suspension was stirred in the drybox for 24 h. The solvents were removed *in vacuo*

(0.1 mm Hg), then toluene (15 mL) was added, and the mixture was filtered through 2 g of diatomaceous earth. The solvents were removed *in vacuo* (0.1 mm Hg) for 4 h to afford a yellow wax. THF (20 mL) and dppp (606 mg, 1.47 mmol, 1.05 equiv) were added and the resulting solution was stirred for 12 h in the drybox. The solvent was removed *in vacuo* to afford an off-white foam. The foam was dissolved in a minimal amount of benzene (2 mL) and was layered with hexane (15 mL). The solution was allowed to sit at room temperature for 24 h then at -20 °C for 24 h. The resulting solids were collected on a fritted glass funnel, then were washed with hexanes (5 mL), and dried under vacuum (0.1 mm Hg) for 12 h to afford 942 mg (74%) of **7p** as off-white needles.

Data for [1,3-bis(diphenylphosphino)propane](2-methylphenyl)[1-(4-methoxyphenyl)-1,1-dimethylsilanolato-κ-O]palladium (7p):

mp: 103-104 °C (dec)

¹H NMR: (500 MHz, C₆H₆)

8.11 (t, *J* = 8.5, 2 H, HC(6'' or 10'')), 7.93 (t, *J* = 8.0, 2 H, HC(6'' or 10'')), 7.58 (t, *J* = 8.5, 2 H, HC(6'' or 10'')), 7.40 (d, *J* = 9.0, 2 H, HC(2)), 7.30 (t, *J* = 7.5, 1 H, HC(2')), 6.92-7.11 (m, 6H, CH(5'', 6'', 9'', or 10'')), 6.90 (d, *J* = 8.0, 1 H, HC(5')), 6.80-6.87 (m, 3 H, HC(5'', 7'', 9'', or 11'')), 6.79 (d, *J* = 8.5, 2 H, HC(3)), 6.73 (t, *J* = 6.5, 2 H, HC(3' and 4')), 6.48-6.68 (m, 5H, HC(5'', 7'', 9'' or 11'')), 3.35 (s, 3 H, H₃C(5)), 2.51 (s, 3 H, H₃C(7')), 2.08 (m, 1 H, H₂C(1'' or 3'')), 1.76 (m, 1 H, H₂C(1'' or 3'')), 1.66 (m, 1 H, H₂C(1'' or 3'')), 1.52 (m, 1 H, H₂C(1'' or 3'')), 1.30-1.40 (bs, 1 H, H₂C(2'')), 1.08 (m, 1 H, H₂C(2'')), 0.21 (s, 3 H, H₃C(6)), 0.13 (s, 3 H, H₃C(6))

¹³C NMR: (126 MHz, d⁸ THF)

162.3 (C(1')), 161.2 (C(4)), 159.7 (C(6')), 144.1 (C(4')), 140.2 (C(1)), 136.9 (C(5')), 136.1, (d, *J* = 12, C(4' or 8')), 135.4 (C(2)), 135.2 (d, *J* = 12, C(4' or 8')), 134.3 (C(7' or 10')), 134.2 (d, *J* = 12, C(4' or 8')), 133.9 (C(7' or 11')), 133.0 (C(7' or 11')), 132.5 (d, *J* = 9, C(4' or 8')), 131.5 (C(7' or 11')), 130.4 (d, *J* = 14, C(5' or 9')), 129.6 (d, *J* = 3, C(6' or 10')), 129.3 (d, *J* = 9, C(5' or 9')), 129.1 (C(6' or 10')), 129.0 (d, *J* = 4, C(6' or 10')), 128.9 (d, *J* = 10, C(5' or 9')), 127.9 (d, *J* = 10, C(5' or 9')), 123.7 (d, *J* = 8, C(6' or 10')), 123.5 (C(3')), 113.9 (C(2')), 112.6 (C(3)), 54.8 (C(5)), 29.7 (d, *J* = 8, C(1'' or 3'')), 29.5 (d, *J* = 8, C(1'' or

3'')), 27.0 (d, $J = 19$, C(2'')), 19.8 (C(7')), 3.3 (C(6))

³¹P NMR: (202 MHz, C₆H₆CH₃)

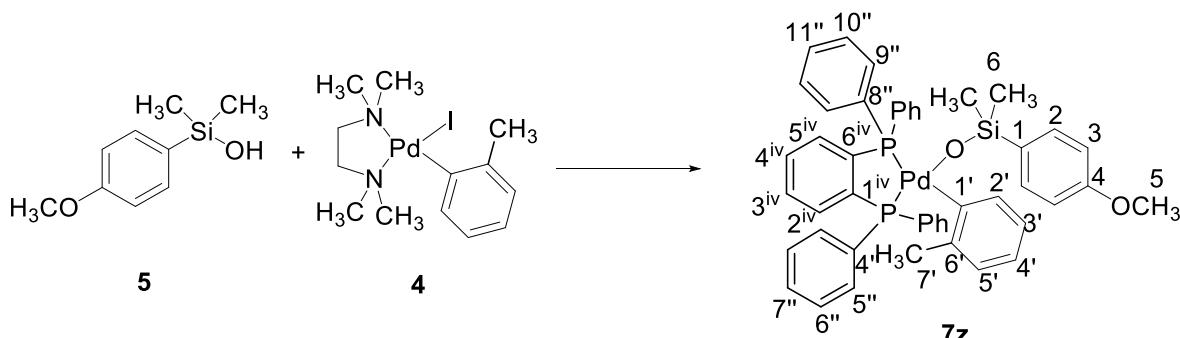
21.1 (d, $J = 48.8$), -10.5 (d, $J = 48.7$)

Analysis: C₄₃H₄₆O₂P₂PdSi · C₆H₆ (869.39)

Calcd: C, 67.69; H, 6.03%

Found: C, 67.52; H, 5.81%

Preparation of [1,2-Bis(diphenylphosphino)benzene](2-methylphenyl)[1-(4-methoxyphenyl)-1,1,-dimethylsilanolato- κ -O]palladium (7z)



In a drybox, a 50-mL, one-necked flask equipped with a magnetic stir bar was charged with **4** (220 mg, 0.50 mmol), (4-methoxyphenyl)dimethylsilanol (100 mg, 0.55 mmol, 1.1 equiv) and THF (10 mL). While stirring, Ag₂O (115 mg, 0.50 mmol, 1.0 equiv) was added and the resulting suspension was stirred in the drybox for 12 h. The solvents were removed *in vacuo* (0.1 mm Hg), then toluene (10 mL) was added, and the mixture was filtered through 2 g of diatomaceous earth. The solvents were removed *in vacuo* (0.1 mm Hg) for 4 h to afford a yellow wax. THF (5 mL) and 1,2-bis(diphenylphosphine)benzene (223 mg, 0.50 mmol, 1.00 equiv) were added and the resulting solution was stirred for 7 d in the drybox. The solvent was removed *in vacuo* to afford an off-white foam. The foam was dissolved in a minimal amount of benzene (1 mL) and was layered with hexane (5 mL). The solution was allowed to sit at room temperature for 24 h then at -20 °C for 24 h. The resulting solids were collected on a fritted glass funnel, then were washed with hexanes (5 mL), and dried under vacuum (0.1 mm Hg) for 12 h to afford 347 mg (84%) of **7z** as pale yellow needles.

Data for [1,2-bis(diphenylphosphino)benzene](2-methylphenyl)[1-(4-methoxyphenyl)-1,1,-dimethylsilanolato- κ -O]palladium (**7z**):

¹H NMR: (500 MHz, C₆H₆)

8.29 (dd, *J* = 7.0, 9.5, 2 H, HC(6'' or 10'')), 7.65-7.74 (m, , 4 H, HC(6'' or 10'')), 7.58 (d, *J* = 8.5, 2 H, HC(2)), 7.47 (t, *J* = 7.5, 1 H, HC(2')), 7.25 (t, *J* = 7.5, 1 H, HC(5')), 6.68-7.20 (m, 22H, CH(3, 3', 4', 5'', 7'', 9'', 11'', 2^{iv}, 3^{iv}, 4^{iv}, 5^{iv} and 6''or 10'')), 3.42 (s, 3 H, H₃C(5)), 2.37 (s, 3 H, H₃C(7')), 0.33 (s, 3 H, H₃C(6)), 0.31 (s, 3 H, H₃C(6))

³¹P NMR: (202 MHz, C₆H₆CH₃)

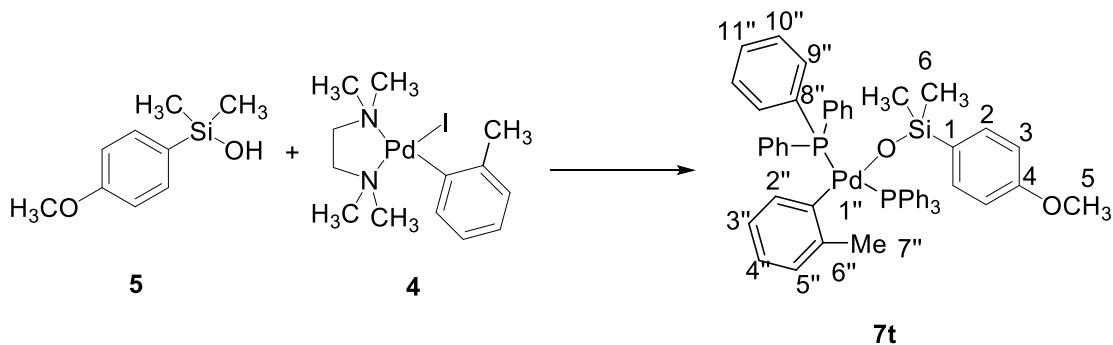
52.2 (d, *J* = 30.5), 35.9 (d, *J* = 30.5)

Analysis: C₄₆H₄₄O₂P₂PdSi (825.29)

Calcd: C, 66.94; H, 5.37%

Found: C, 66.56 H, 5.13%

Preparation of *trans*-Bis(triphenylphosphine)(2-methylphenyl)[1-(4-methoxyphenyl)-1,1-dimethylsilanolato- κ -O]palladium (7t)



In a drybox, a 50-mL, one-necked flask equipped with a magnetic stir bar was charged with **4** (440 mg, 1.00 mmol), (4-methoxyphenyl)dimethylsilanol (200 mg, 1.10 mmol, 1.1 equiv) and THF (10 mL). While stirring, Ag₂O (230 mg, 1.0 mmol, 1.0 equiv) was added and the resulting suspension was stirred in the drybox for 12 h. The solvents were removed *in vacuo* (0.1 mm Hg), then toluene (20 mL) was added, and the mixture was filtered through 2 g of diatomaceous earth. The solvents were removed *in vacuo* (0.1 mm Hg) for 4 h to afford a yellow wax. THF (10 mL) and Ph₃P (524 mg, 2.00 mmol, 2.00 equiv) were added and the resulting solution was stirred for 36 h in the drybox. The solvent was removed *in vacuo* to afford an off-white foam. The foam was dissolved in a minimal amount of THF (2 mL) and was layered with hexane (5 mL). The solution was allowed to sit at room temperature for 24 h then at -20 °C for

24 h. The resulting solids were collected on a fritted glass funnel, then were washed with hexanes (5 mL), and dried under vacuum (0.1 mm Hg) for 12 h to afford 663 mg (73%) of **7t** as pale yellow plates.

Data for *trans*-bis(triphenylphosphine)(2-methylphenyl)[1-(4-methoxyphenyl)-1,1-dimethylsilanolato- κ -O]palladium (**7t**):

¹H NMR: (500 MHz, C₆H₆)

7.69-7.74 (m, 12 H, HC(10')), 7.38 (d, *J* = 8.0, 2 H, HC(2)), 6.98-7.04 (m, 19 H, HC(9', 11' and 2')), 6.84 (d, *J* = 8.0, 1H, HC(5')), 3.45 (s, 3H, H₃C(5)), 2.07 (s, 3H, H₃C(7)), -0.24 (s, 3H, HC(6))

¹³C NMR: (126 MHz, C₆H₆)

159.5 (C(4)), 152.5 (C(6')), 145.3 (C(2')), 142.2 (C(4')), 140.2 (C(5')), 136.2 (C(1)), 135.3 (C(11')), 135.2 (C(10')), 130.2 (t, *J* = 21 C(8')), 129.9 (C(9')), 128.3 (C(2)), 123.8 (C(3')), 122.4 (C(1)), 112.8 (C(3)), 54.5 (C(5)), 26.6 (C(7')), 3.5 (C(6))

³¹P NMR: (202 MHz, C₆H₆CH₃)

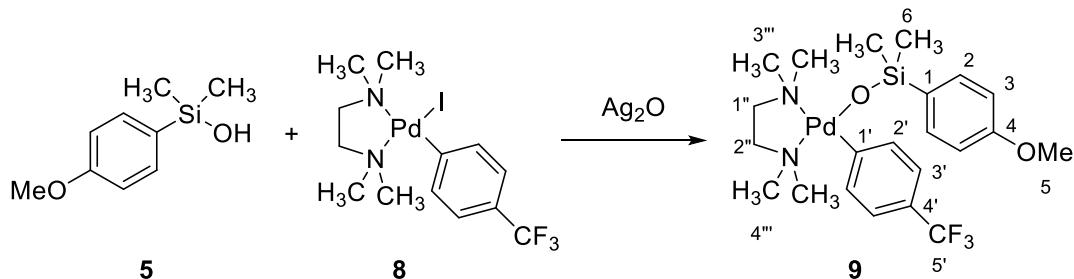
22.6

Analysis: C₅₂H₅₀OP₂PdSi (903.41)

Calcd: C, 69.13; H, 5.58%

Found: C, 69.06; H, 5.52%

Preparation of [TMEDA](4-Trifluoromethylphenyl)[1-(4-methoxyphenyl)-1,1-dimethylsilanolato- κ -O]palladium (**9**)



In a dry box, a 25 mL one-neck round-bottom flask containing a magnetic stir-bar and fitted with a stop-cock gas inlet was charged with **5** (182 mg, 1.0 mmol), THF (25 mL), and Ag₂O (231 mg, 1.0 mmol, 1.0 equiv). To this solution was added **8** (495 mg, 1.0 mmol, 1.0

equiv). The resulting mixture was stirred for 12 h, and then the solvent was removed under vacuum (0.5 mm Hg) and replaced with toluene (25 mL). The suspension was filtered through a medium porosity glass frit covered with a plug of Celite (2 cm) into a pre-weighed one-neck flask fitted with a stop-cock gas adapter. The solvent was removed under vacuum (0.5 mm Hg) leaving 548 mg (99%) of **9** as an off-white solid residue. An analytical sample was obtained by recrystallization of 50 mg from 0.1 mL of THF and 5 mL of pentane. The crystals were isolated as colorless cubes.

Data for [TMEDA](4-Trifluoromethylphenyl)palladium 4-Methoxyphenyl dimethylsilanolate (9):

mp: 95-98 °C (THF/pentane)

¹H NMR: (500 MHz, THF)

7.83 (d, $J = 8.5$, 2 H, HC(2)), 7.63 (d, $J = 7.6$, 2 H, HC(2')), 7.31 (d, $J = 7.6$, 2 H, HC(3')), 6.99 (d, $J = 8.5$, 2 H, HC(3)), 3.42 (s, 3 H, H₃C(5)), 2.13 (s, 6 H, H₃C(3'') or H₃C(4'')), 1.55 (s, 6 H, H₃C(4'') or H₃C(3'')), 1.40 (bs, 4 H, H₂C(1'') and H₂C(2'')), 0.274 (s, 6 H, H₃C(1''))

¹³C NMR: (126 MHz, THF)

163.5 (C(4)), 161.2 (C(1), 140.5 (C(4')), 139.0 (C(2')), 136.5 (C(3')), 136.4 (C(2)), 126.2 (q, $J = 31.3$, F₃C(5')), 122.8 (t, $J = 3.68$, C(1')) 114.0 (C(3), 64.8 (C(5)), 59.2 (C(1'') or C(2'')), 55.9 (C(1'') or C(2'')), 52.2 (C(3'') or C(4'')), 48.4 (C(3'') or C(4'')), 4.7 (C(6))

¹⁹F NMR: (470 MHz, THF)

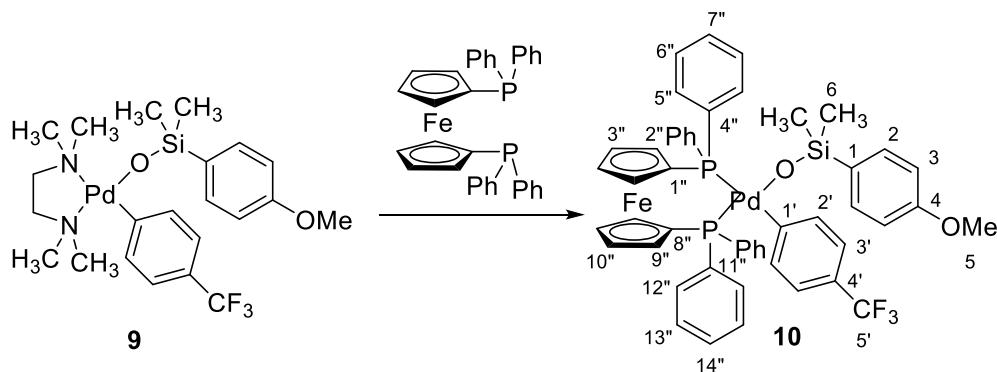
-61.46 (FC(5'))

Analysis: C₂₂H₃₃F₃N₂O₂PdSi (549.01)

Calcd: C, 48.13; H, 6.06; N, 5.10

Found: C, 48.17; H, 6.11; N, 5.16

Preparation of [1,1-Bis(diphenylphosphino)ferrocene](4-trifluoromethylphenyl)[1-(4-methoxyphenyl)-1,1,-dimethylsilanolato- κ -O]palladium (10)



In a dry box, a one-neck round-bottom flask containing magnetic stir-bar and fitted with a stop-cock gas adapter was charged with **9** (275 mg, 0.5 mmol), THF (25 mL), and dppf (277 mg, 0.5 mmol, 1.0 equiv). The resulting mixture was stirred for 1 h and then the solvent was removed under vacuum (0.5 mm Hg) until only ~1 mL remained. To the solution was added pentane (25 mL) and a precipitate was formed over a 30 min period. The suspended solid was filtered off through a fine porosity glass fritte then was dried to afford 363 mg, (73%) of **10** as a yellow/orange powder. An analytical sample was obtained by sacrificial recrystallization of 50 mg from 0.1 mL THF and 5 mL hexane where the initial solids formed were discarded and the remaining material allowed to crystallize. The crystals were isolated as small orange cubes.

Data for [1,1-bis(diphenylphosphino)ferrocene](4-trifluoromethylphenyl)[1-(4-methoxyphenyl)-1,1,-dimethylsilanolato- κ -O]palladium (**10**)

mp: 100-105 C (dec, THF/hexane)

¹H NMR: (500 MHz, d₈-THF)

9.01 (m, 4 H, HC(5'') or HC(12'')), 8.27 (d, $J = 7.3$, 2 H, HC(2)), 8.22-8.11 (m, 8 H, HC(6'') and HC(13'')), 8.11 (t, $J = 6.4$, 2 H, HC(7'') or HC(14'')), 7.96- 7.90 (m, 6 H, , HC(5'') or HC(12''), HC(7'') or HC(14'')), 7.65 (d, $J = 8.4$, 2 H, HC(2')), 7.49 (d, $J = 6.6$, 2 H, HC(3')), 7.31 (d, $J = 8.4$, 2 H, HC(3)), 5.66 (s, 2 H, HC(3'') or HC(10'')), 5.40 (bs, 2 H HC(3'') or HC(10'')), 5.04 (bs, 2 H, HC(2'') or HC(9'')), 4.48 (s, 3 H, H3C(5)), 4.41 (bs, 2 H, HC(2'') or HC(9'')), 0.28 (s, 6 H, H3C(6))

- 13C NMR:** (126 MHz, d₈-THF)
159.8 (C(4)), 139.3 (C(1)), 137.5 (d, *J* = 1.9, C(7‘) or C(14‘)), 136.4 (d, *J* = 12.9, C(4‘) or C(12‘)), 135.1 (C(7‘) or C(14‘)), 135.0 (d, *J* = 12.9, C(4‘) or C(11‘)), 134.8 (C(2)), 134.5 (C(4‘)), 134.4 (C(2‘)), 134.2 (C(3‘)), 131.0 (d, *J* = 1.8 (C(6‘) or C(14‘)), 130.8 (d, *J* = 1.7, (C(6‘) or C(13‘))), 128.2 (d, *J* = 9.2, (C(5‘) or C(12‘)), 128.6 (d, *J* = 11.0, C(5‘) or C(12‘)) 125.1 (q, *J* = 31.3, 282 C(5‘)), 122.8 (q, *J* = 3.7, C(1‘)), 112.6 (C(3)), 77.7 (*J* = 6.4, C(1‘) or C(8‘)), 77.3 (*J* = 6.4, C(1‘) or C(8‘)), 77.2 (*J* = 11.9, C(2‘) or C(9‘)), 75.3 (*J* = 8.3, C(2‘) or C(9‘)), 74.2 (*J* = 7.4, C(3‘) or C(10‘)), 73.0 (*J* = 3.7, C(3‘) or C(10‘)), 54.7 (C(5)), 3.2 (C(6))
- 31P NMR:** (162 MHz, d₈-THF)
34.9 (d, *J* = 32.0), 11.9 (d, *J* = 32.0)
(162 MHz, d₈-toluene)
32.88 (d, *J* = 31.5), 10.71 (d, *J* = 31.7)
- 19F NMR:** (470 MHz, d₈-THF)
-61.07 (FC(5‘))
- Analysis:** C₅₀H₄₅F₃FeO₂P₂PdSi (987.18)
Calcd: C, 60.83; H, 4.59
Found: C, 60.49; H, 4.61

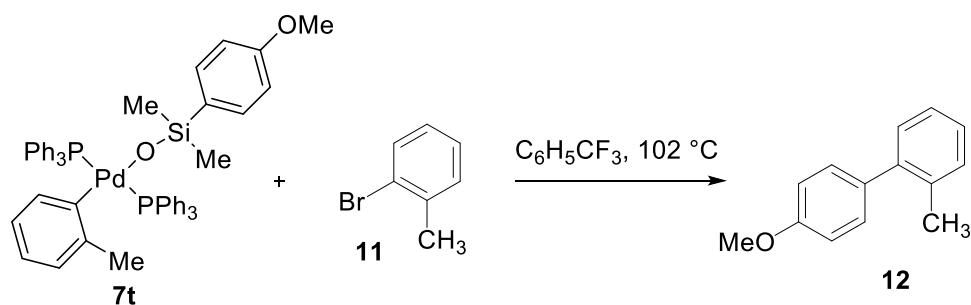
KINETIC STUDIES OF THERMOLYSIS OF ARYLPALLADIUM ARYLSILANOLATE COMPLEXES (TABLE 2)

Response Factors

$$\text{Response factor for } \mathbf{12} = \frac{(\text{mmol } \mathbf{12} * \text{area internal standard})}{(\text{mmol internal standard} * \text{area } \mathbf{12})}$$

naphthalene (mmol)	naphthalene (area)	product 12 (mmol)	product 12 (area)	response factor
0.0998	4157580	0.0974	4184310	1.054
0.1001	5000900	0.0999	4698670	1.062
0.0983	4701810	0.1010	4518390	1.069

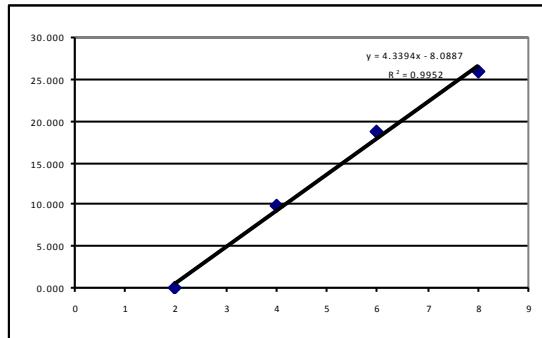
Thermolysis of **7t (Table 2, entry 1)**



In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with naphthalene (15 mg), **7t** (45 mg, 0.05 mmol), 2-bromotoluene (20 μL , 0.167 mmol, 3.3 equiv), and dry benzotrifluoride (4.0 mL). The flask was removed from the drybox, purged with argon, and then was placed in a 105°C oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 30 μL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.5 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC.

Run 1

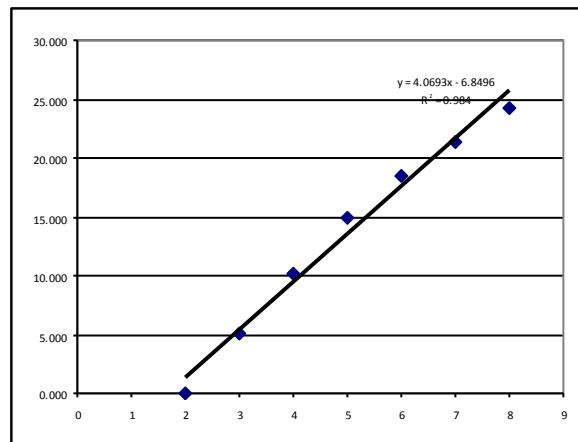
Time	area, std	area, 12	conv, μmol
2	1	0	0.000
2	1	0	0.000
2	1	0	0.000
4	59059	4287	9.614
4	59524	4358	9.697
4	58847	4389	9.878
6	64418	9222	18.961
6	64634	9099	18.646
6	64901	9204	18.783
8	53716	10388	25.614
8	54206	10638	25.993
8	54147	10676	26.115



rate = 0.0180 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
2	1	0	0.000
2	1	0	0.000
2	1	0	0.000
3	46998	1895	5.007
3	46860	1910	5.061
3	46342	1906	5.107
4	48983	4011	10.168
4	48353	3967	10.187
4	48479	3983	10.202
5	51995	6221	14.857
5	51443	6199	14.963
5	52728	6428	15.138
6	44738	6689	18.565
6	45006	6771	18.681
6	45058	6679	18.406
7	44331	7556	21.164
7	44898	7870	21.765
7	44581	7640	21.280
8	48157	9470	24.418
8	48730	9439	24.052
8	48908	9617	24.416

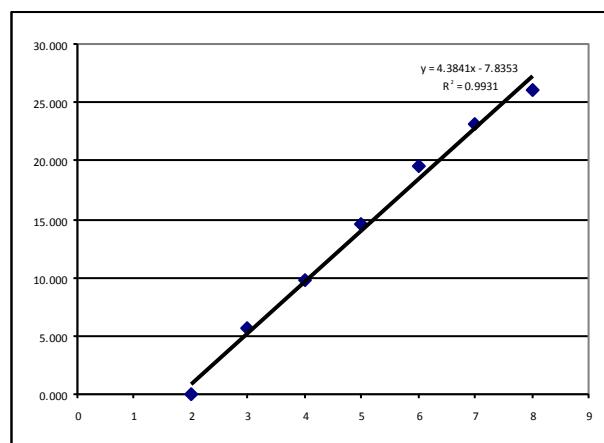


rate = 0.0169 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
2	1	0	0.000
2	1	0	0.000
2	1	0	0.000
3	53875	2420	5.578
3	53537	2444	5.668

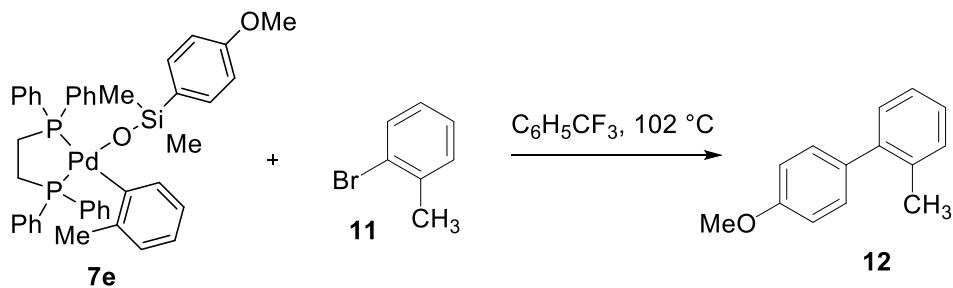
3	53228	2369	5.526
4	57807	4606	9.894
4	58271	4593	9.787
4	57980	4550	9.744
5	56711	6666	14.595
5	56649	6680	14.642
5	56077	6591	14.594
6	61895	9607	19.273
6	61769	9620	19.339
6	61492	9793	19.775
7	79437	14592	22.809
7	79719	14854	23.137
7	80549	15127	23.319
8	49516	10419	26.128
8	50163	10459	25.890
8	50300	10568	26.088



rate = 0.0183 mM/sec

avg. rate = 1.77×10^{-2} mM/s

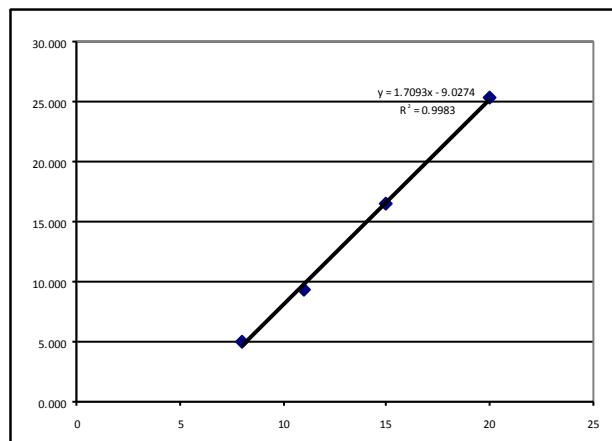
Thermolysis of 7e (Table 2, entry 2)



In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with naphthalene (15 mg), **7e** (39 mg, 0.05 mmol), 2-bromotoluene (20 μL , 0.167 mmol, 3.3 equiv), and dry benzotrifluoride (4.0 mL). The flask was removed from the drybox, purged with argon, and then was placed in a $105\text{ }^\circ\text{C}$ oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 30 μL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.5 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC.

Run 1

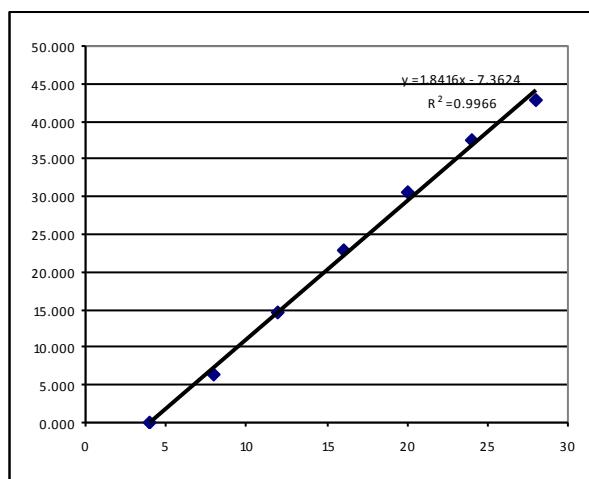
Time	area, std	area, 12	conv, μmol
8	39213	1555	4.924
8	41046	1642	4.967
8	40905	1722	5.227
11	41668	3119	9.295
11	43423	3227	9.228
11	43752	3323	9.431
15	41585	5460	16.303
15	43147	5809	16.717
15	43990	5833	16.465
20	37354	7599	25.260
20	38795	7867	25.180
20	38415	7916	25.587
40	41532	15390	46.013
40	40984	15701	47.570
40	41100	15724	47.505



Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	1	0	0.000
4	1	0	0.000
4	1	0	0.000
8	52787	1987	6.544
8	53276	1969	6.425
8	53855	1983	6.401
12	46810	3943	14.643
12	46852	3939	14.615
12	46814	3935	14.612
16	67557	8900	22.902
16	67821	8940	22.915
16	68719	8971	22.694
20	59846	10421	30.271
20	59939	10425	30.235
20	59658	10682	31.127
24	53035	11474	37.610
24	53361	11451	37.305
24	53357	11570	37.696
28	51488	12644	42.690
28	51522	12772	43.094
28	51681	12604	42.396

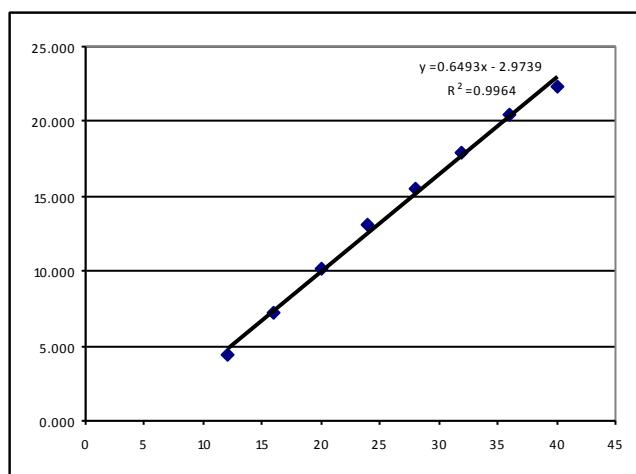


rate = 0.00766 mM/sec

Run 3 (run at half scale vs Run 1 and 2)

Time	area, std	area, 12	conv, μmol
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12	65983	1672	4.405
12	67048	1690	4.382
12	66687	1677	4.372
16	53288	2212	7.216
16	54200	2256	7.236
16	53698	2225	7.203
20	47425	2724	9.985
20	46983	2762	10.220
20	47362	2796	10.263
24	54560	3985	12.697
24	32237	2541	13.702
24	54134	4048	12.999
28	50711	4448	15.248
28	50008	4519	15.709
28	49811	4482	15.642
32	50665	5106	17.519
32	50486	5199	17.902
32	50604	5383	18.492
36	55826	6495	20.225
36	33752	3937	20.277
36	33378	4022	20.947

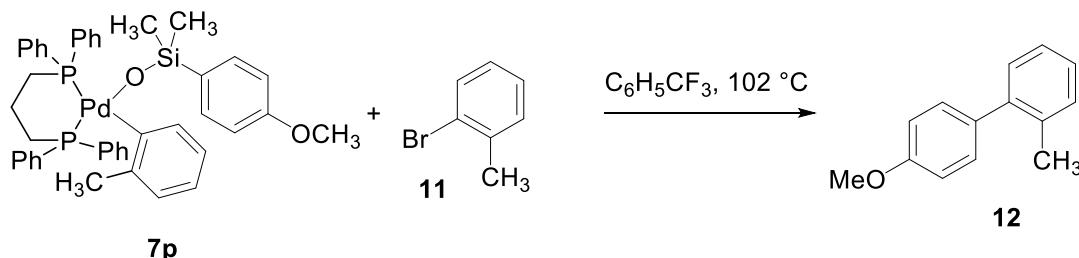


rate = 0.00541 mM/sec

40	51466	6533	22.067
40	51310	6625	22.446
40	50927	6634	22.645

$$\text{avg. rate} = 0.671 \times 10^{-2} \text{ mM/s}$$

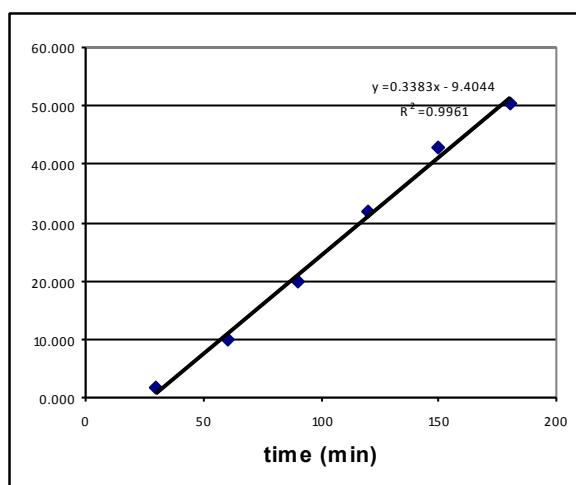
Thermolysis of 7p (Table 2, entry 3)



In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with naphthalene (20 mg), **7p** (45 mg, 0.05 mmol), 2-bromotoluene (20 μL , 0.167 mmol, 3.3 equiv), and dry benzotrifluoride (4.0 mL). The flask was removed from the drybox, purged with argon, and then was placed in a $105\text{ }^{\circ}\text{C}$ oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 30 μL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.5 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC.

Run 1

Time	area, std	area, 12	conv, μmol
30	1	0	0.000
30	128110	1867	2.413
30	101636	1589	2.588
60	84983	5178	10.088
60	84995	5115	9.963
60	84319	5100	10.014
90	100224	12135	20.046
90	108641	12582	19.174
90	106269	12833	19.993
120	63258	12238	32.030
120	65740	12535	31.568
120	108358	20830	31.826
150	81771	21277	43.079
150	85072	22077	42.965
150	68956	17790	42.713
180	50224	15410	50.798



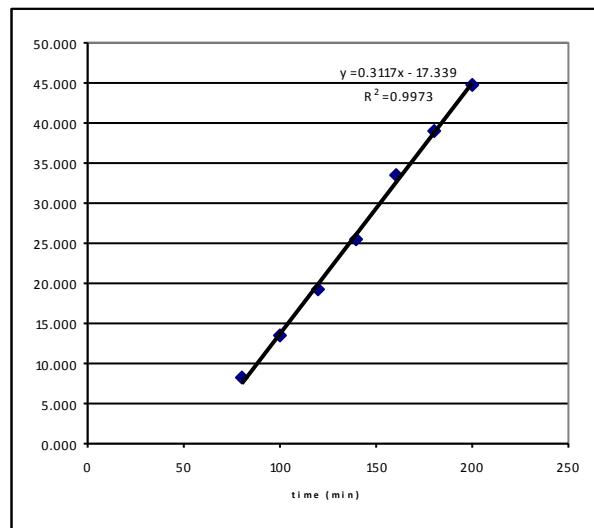
$$\text{rate} = 0.00141 \text{ mM/sec}$$

180	51362	15752	50.775
180	68842	20837	50.112
240	83785	25161	49.719
240	83125	25151	50.094
240	50211	15420	50.845

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

60	66513	1793	4.463
60	67106	1818	4.485
60	66513	1793	4.463
80	67368	3362	8.262
80	66662	3345	8.308
80	66768	3401	8.433
100	64085	5221	13.488
100	64523	5241	13.448
100	64684	5377	13.763
120	97467	11326	19.239
120	98625	11456	19.231
120	97244	11455	19.503
140	107587	16470	25.345
140	107370	16500	25.443
140	107451	16644	25.645
160	81990	16355	33.025
160	83914	16855	33.255
160	47951	9903	34.192
180	71878	16732	38.540
180	72984	17194	39.004
180	42608	10240	39.789
200	78868	21500	45.133
200	78999	21239	44.511
200	80067	21588	44.639



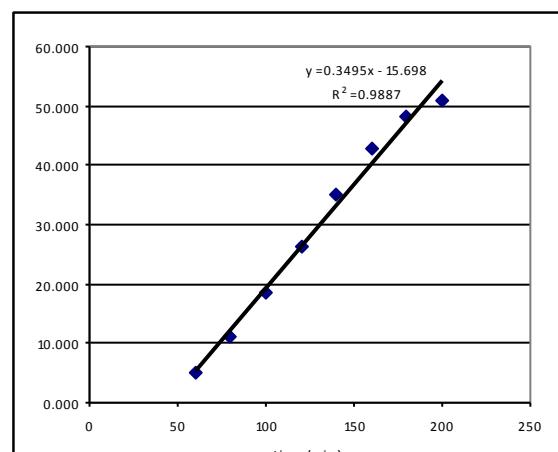
rate = 0.00130 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
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60	95078	2875	5.006
60	94825	2913	5.086
60	95101	2914	5.073
80	85479	5752	11.141
80	84735	5753	11.241
80	84482	5658	11.088
100	79383	8851	18.460
100	79433	8859	18.465
100	47003	5259	18.524
120	88977	14113	26.260
120	89942	14253	26.236

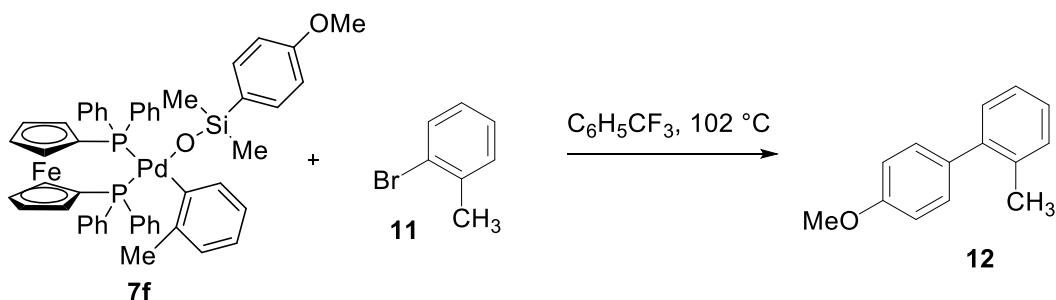
120	89940	14408	26.522
140	80924	17210	35.210
140	80974	17165	35.096
140	48287	10250	35.144
160	89260	22759	42.214
160	52710	13567	42.614
160	52956	13788	43.107
180	89344	25655	47.541
180	88844	25524	47.564
180	52275	15611	49.442
200	85287	25936	50.348
200	84991	26110	50.862
200	84273	26167	51.407



$$\text{rate} = 0.00145 \text{ mM/sec}$$

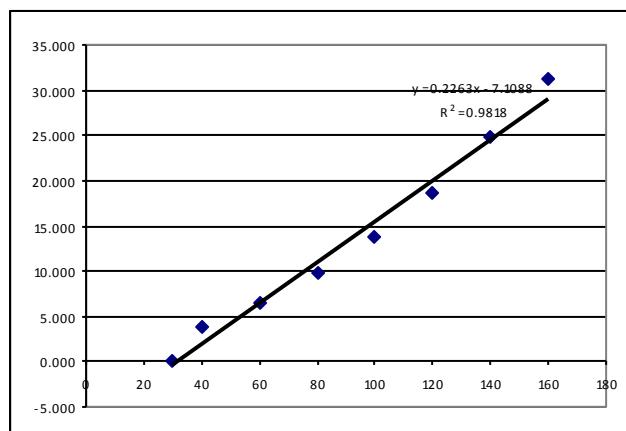
$$\text{avg. rate} = 0.139 \times 10^{-2} \text{ mM/s}$$

Thermolysis of 7f (Table 2, entry 4)



In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with naphthalene (15 mg), **7f** (47 mg, 0.05 mmol), 2-bromotoluene (20 μL , 0.167 mmol, 3.3 equiv), and dry benzotrifluoride (4.0 mL). The flask was removed from the drybox, purged with argon, and then was placed in a 105°C oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 30 μL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.5 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC.

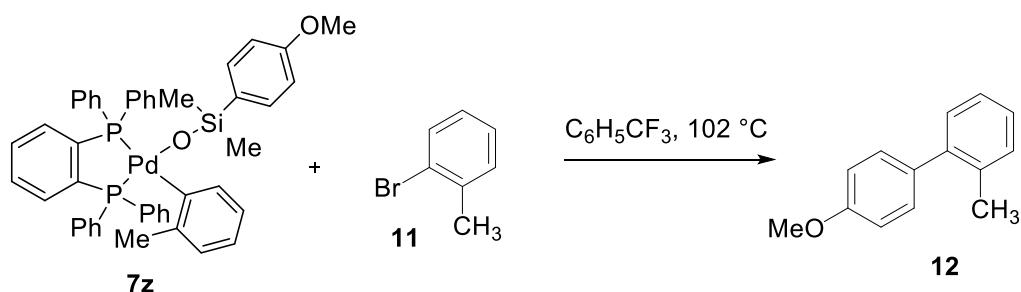
Time	area, std	area, 12	conv, μmol
30	1	0	0.000
30	1	0	0.000
30	1	0	0.000
40	56858	1785	3.898
40	59669	1741	3.623
40	59289	1777	3.722
60	83950	4308	6.372
60	86727	4435	6.350
60	86326	4574	6.579
80	58836	4805	10.141
80	61975	4941	9.900
80	62027	4763	9.535
100	63283	7185	14.098
100	64909	7123	13.626
100	65933	7109	13.388
120	64131	9774	18.925
120	66297	9878	18.501



$$\text{rate} = 9.41 \times 10^{-4} \text{ mM/sec}$$

120	65753	9722	18.359
140	55745	11206	24.961
140	57216	11388	24.714
140	32475	6428	24.578
160	55779	14823	32.998
160	57810	14407	30.945
160	33836	8114	29.777

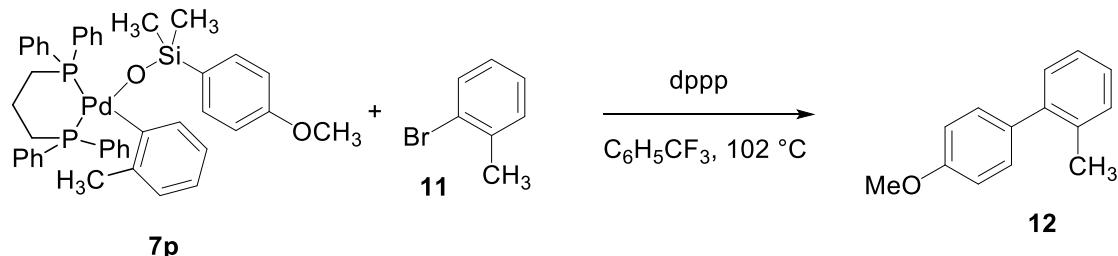
Thermolysis of 7z (Table 2, entry 5)



In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with naphthalene (15 mg), **7z** (41 mg, 0.05 mmol), 2-bromotoluene (20 μL , 0.167 mmol, 3.3 equiv), and dry benzotrifluoride (4.0 mL). The flask was removed from the drybox, purged with argon, and then was placed in a 105°C oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 30 μL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.5 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC. No reaction was observed after 24h. The reaction was repeated twice with identical results.

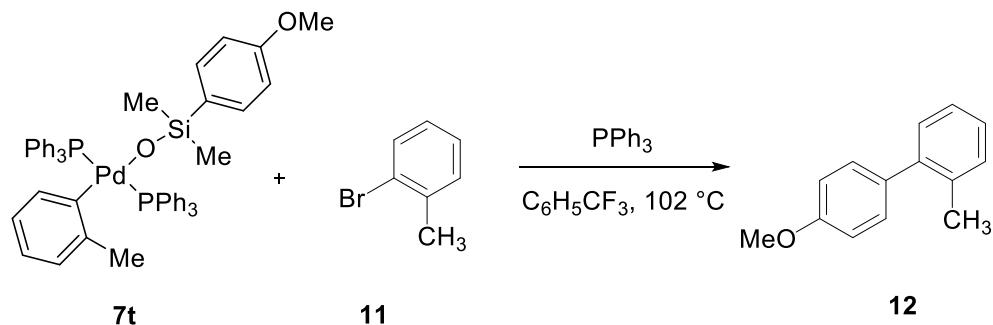
**EFFECT OF ADDED PHOSPHINE ON THE RATE OF THERMAL
TRANSMETALATION OF **7p** AND **7t** (SCHEME 2)**

Thermolysis of **7p in the presence of dppp**



In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with naphthalene (10 mg), **7p** (45 mg, 0.05 mmol), 2-bromotoluene (20 μL , 0.167 mmol, 3.3 equiv), dppp (10.3 mg, 0.025 mmol) and dry benzotrifluoride (4.0 mL). The flask was removed from the drybox, purged with argon, and then was placed in a 105°C oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 30 μL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.5 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC. Cross-coupling product **12** was not observed after 2 h. The procedure was performed twice with identical results.

Thermolysis of **14t** in the presence of Ph_3P

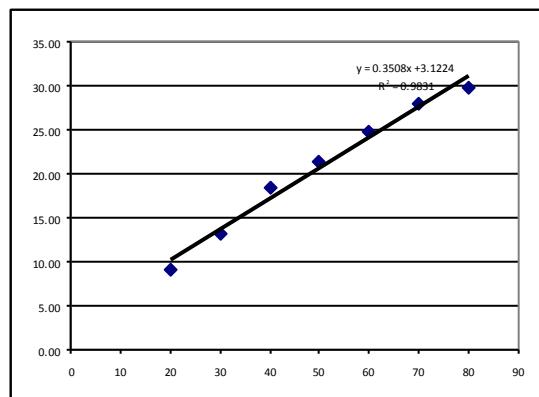


In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with naphthalene (15 mg), **7t** (45 mg, 0.05 mmol), 2-bromotoluene (20 μL , 0.167 mmol, 3.3 equiv), PPh_3 (6.6 mg, 0.025 mmol, 0.5 equiv) and dry benzotrifluoride (4.0 mL). The flask was removed from the drybox, purged with argon, and then was placed in a 105 $^\circ\text{C}$ oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 30 μL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.5 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC.

Run 1

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

20	6887	763	9.17
20	7261	805	9.18
20	7139	762	8.84
30	10084	1578	12.95
30	10191	1630	13.24
30	10358	1659	13.26
40	8377	1833	18.11
40	8440	1873	18.37
40	8517	1935	18.81
50	10938	2823	21.36
50	10912	2787	21.14
50	10664	2814	21.84
60	8212	2432	24.52
60	8118	2456	25.04
60	8232	2461	24.75

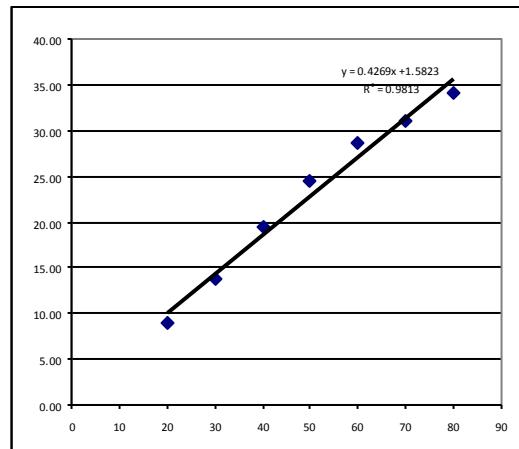


70	10294	3499	28.14	rate = 1.45×10^{-3} mM/sec
70	10790	3629	27.84	
70	10883	3658	27.82	
80	8340	3005	29.83	
80	8031	2874	29.62	
80	8180	2969	30.05	

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

20	9724	1030	8.77
20	10134	1129	9.22
20	10468	1125	8.90
30	10492	1816	14.33
30	10408	1767	14.05
30	10826	1659	12.69
40	9074	2233	20.37
40	10020	2308	19.07
40	9889	2277	19.06
50	11147	3309	24.57
50	11324	3317	24.25
50	11177	3299	24.43
60	8547	2873	27.83
60	8458	2954	28.91
60	8158	2904	29.47
70	11695	4315	30.54
70	11519	4284	30.79
70	11558	4419	31.65
80	10718	4481	34.61
80	11000	4548	34.23
80	10999	4483	33.74



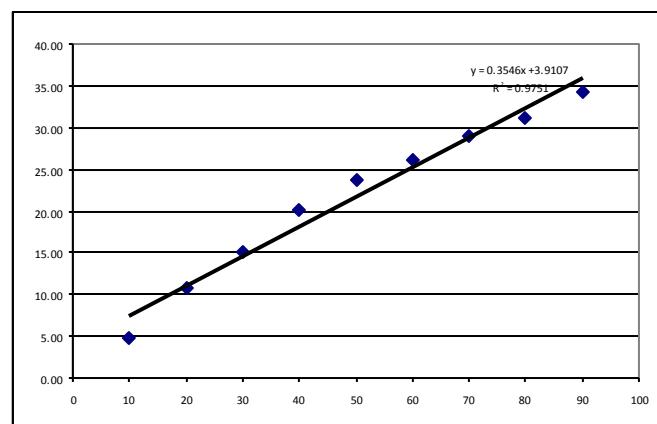
rate = 1.77×10^{-3} mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

10	12385	756	5.05
10	12348	719	4.82
10	12775	741	4.80
20	10110	1230	10.07
20	9666	1290	11.05
20	9384	1236	10.90
30	10790	1919	14.72
30	10395	1948	15.51
30	10739	1952	15.05
40	9490	2276	19.85
40	9168	2371	21.41
40	10069	2284	18.78
50	9099	2679	24.37

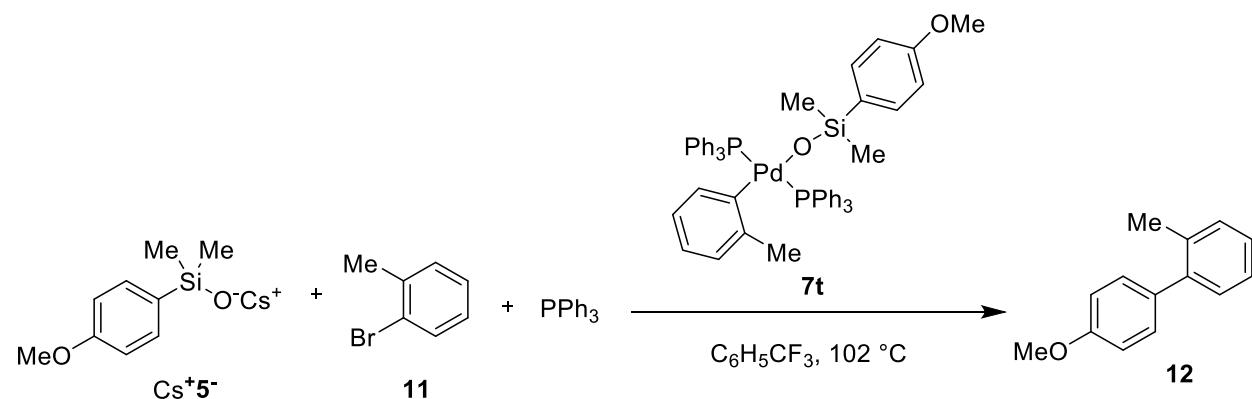
50	10258	2795	22.56
50	8809	2541	23.88
60	7772	2461	26.21
60	8037	2534	26.10
60	5824	1852	26.32
70	7110	2475	28.82
70	7327	2559	28.91
70	7186	2507	28.88
80	7805	2906	30.82
80	8083	3003	30.75
80	7927	3021	31.55
90	8256	3556	35.66
90	7913	3285	34.37
90	8189	3273	33.09



rate = 1.47×10^{-3} mM/sec

avg. rate = 1.56×10^{-2} mM/s

General procedure I: Effect of Added Phosphine on the Rate of the Catalytic Cross-Coupling of Cs⁺5⁻ (Footnote 12)



Order in PPh₃ (3.75 mM)

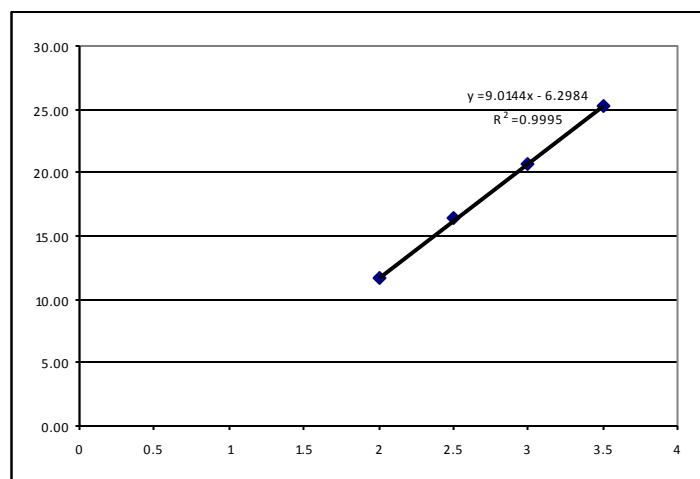
Cs ⁺ 5 ⁻	62.5 mM
2-Bromotoluene	83 mM
7t	7.5 mM
PPh ₃	3.75 mM

In a drybox, a one-necked, 5-mL round-bottomed flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with Cs⁺5⁻ (39 mg, 0.125 mmol, 0.78 equiv), 7t (13.5 mg, 0.015 mmol, 0.090 equiv), naphthalene (12 mg), PPh₃ (2.0 mg, 0.0075 mmol, 0.047 equiv), 2-bromotoluene (20 μL, 0.16 mmol, 1.0 equiv) and dry benzotrifluoride (2.0 mL). The flask was placed in a 105 °C oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 50 μL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.0 mL of ethyl acetate and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC.

Run 1

Time	area, std	area, 12	conv, μmol
2	21798	2560	11.67
2	21145	2449	11.51

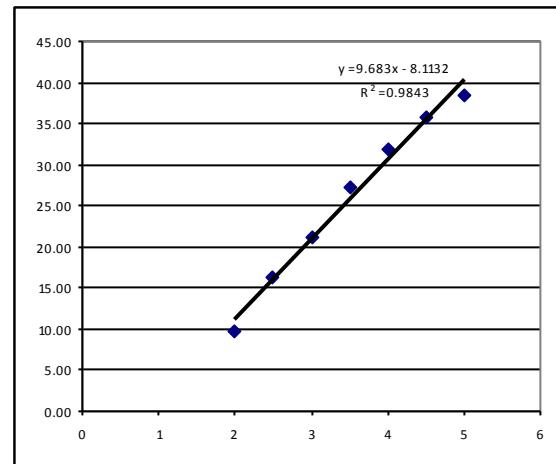
2	21173	2492	11.69
2.5	24839	4093	16.37
2.5	24598	4036	16.30
2.5	24706	4133	16.62
3	20886	4356	20.72
3	20922	4375	20.77
3	22037	4566	20.58
3.5	20344	5120	25.00
3.5	20258	5246	25.72
3.5	21533	5408	24.95
4	18764	5514	29.19
4	19319	5609	28.84
4	19665	5800	29.30



rate = 0.0751 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
2	23965	2276	7.86
2	19465	2443	10.39
2	18725	2406	10.64
2.5	18458	3604	16.16
2.5	18389	3749	16.88
2.5	18733	3587	15.85
3	16781	4406	21.73
3	16373	4251	21.49
3	16573	4128	20.62
3.5	17289	5716	27.37
3.5	17280	5775	27.67
3.5	17513	5721	27.04
4	21898	8338	31.52
4	22659	8798	32.14
4	21473	8231	31.73
4.5	19223	8285	35.68
4.5	18863	8224	36.09
4.5	19703	8409	35.33
5	18849	8818	38.73
5	18611	8567	38.11
5	18743	8671	38.30

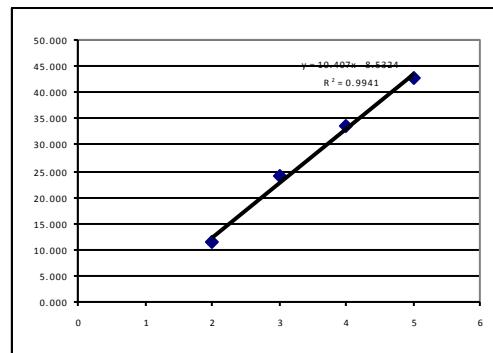


rate = 0.0807 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
2	93311	8576	11.412
2	93699	8455	11.205
2	95136	8663	11.307

3	89097	17178	23.940
3	88721	17109	23.945
3	87342	16726	23.779
4	81946	22610	34.260
4	80985	22046	33.802
4	84529	22312	32.776
5	81422	27452	41.865
5	78398	27549	43.634
5	79687	27446	42.767
6	75606	29791	48.927
6	75841	29538	48.361
6	75127	29876	49.379



rate = **0.0867 mM/sec**

$$\text{avg. rate} = 8.08 \times 10^{-2} \text{ mM/s}$$

Order in PPh₃ (7.5 mM)

Cs⁺**12**⁻ 62.5 mM

2-Bromotoluene 83 mM

7t 7.5 mM

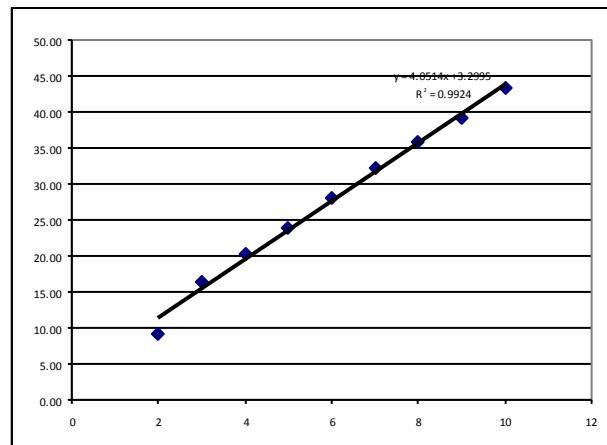
PPh₃ 7.50 mM

Following the General Procedure I, a mixture of silanolate Cs⁺**12**⁻ (39 mg, 0.125 mmol), **7t** (13.5 mg, 0.015 mmol, 0.090 equiv), naphthalene (11 mg), PPh₃ (4.0 mg, 0.015 mmol) and 2-bromotoluene (20 μL, 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μmol
2	15759	1609	9.30
2	16187	1688	9.50
2	8571	850	9.03
3	7799	1414	16.51
3	7613	1384	16.55
3	7799	1414	16.51
4	6764	1553	20.91
4	7309	1582	19.71
4	8347	1865	20.35
5	6960	1837	24.03
5	6453	1683	23.75

5	6960	1837	24.03
6	7151	2193	27.93
6	6493	1994	27.96
6	8180	2540	28.27
7	8376	2933	31.89
7	9165	3221	32.00
7	8134	2896	32.42
8	9487	3687	35.39
8	8729	3452	36.01
8	8961	3555	36.12
9	8388	3630	39.41
9	8943	3720	37.88
9	8663	3808	40.03
10	9524	4508	43.10
10	9777	4638	43.20
10	8728	4182	43.63

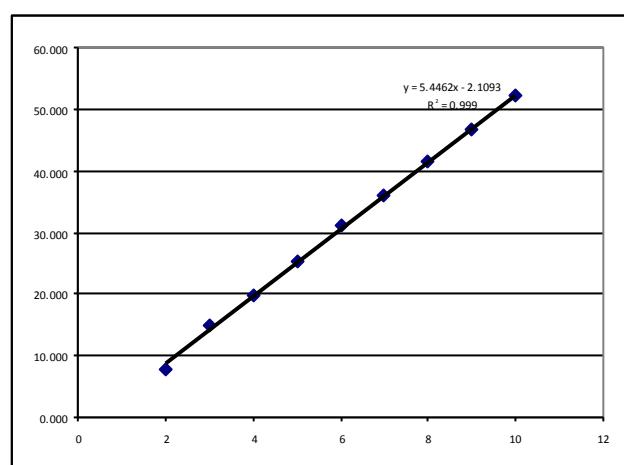


rate = 0.0338 mM/sec

Run 2

Time area, std area, 12 conv, μmol

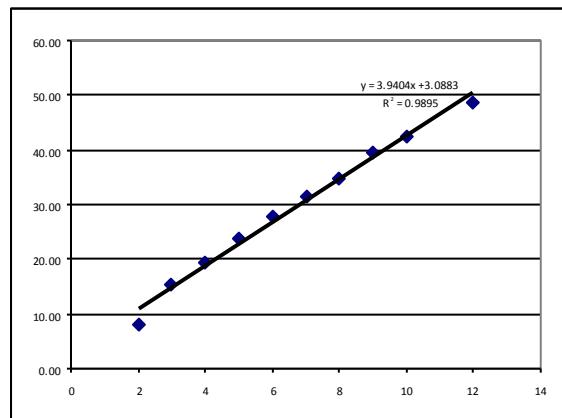
2	68147	4319	7.870
2	93227	5781	7.700
2	43836	2811	7.963
3	66193	8101	15.197
3	44073	5335	15.031
3	46200	5489	14.753
4	19012	2977	19.443
4	37975	5979	19.550
4	73901	11942	20.065
5	79537	16119	25.165
5	79245	16208	25.397
5	79372	16314	25.522
6	36003	9021	31.113
6	35745	8789	30.531
6	72104	18227	31.389
7	39346	11369	35.879
7	80292	23092	35.712
7	38841	11284	36.074
8	77781	26152	41.750
8	40458	13520	41.495
8	80453	26450	40.823
9	34049	13114	47.825
9	74228	27431	45.887
9	36009	13384	46.152
10	33433	14570	54.113
10	72102	30797	53.037
10	36465	14653	49.897



rate = 0.0453 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
2	19898	1764	8.07
2	19543	1791	8.34
2	20124	1785	8.08
3	20826	3455	15.11
3	20842	3601	15.73
3	20698	3451	15.18
4	20173	4288	19.36
4	20741	4490	19.71
4	21314	4456	19.04
5	19625	5121	23.76
5	20327	5259	23.56
5	19782	5241	24.12
6	17618	5453	28.18
6	17240	5174	27.33
6	17584	5373	27.82
7	17394	5891	30.84
7	18553	6381	31.32
7	18198	6380	31.92
8	19562	7487	34.85
8	20464	7770	34.57
8	19970	7657	34.91
9	17502	7590	39.49
9	17946	7846	39.81
9	17601	7507	38.84
10	18206	8394	41.98
10	17628	8310	42.93
10	17847	8329	42.50



rate = 0.0328 mM/sec

Order in PPh₃ (15.0 mM)

Cs⁺**5**⁻ 62.5 mM

2-Bromotoluene 83 mM

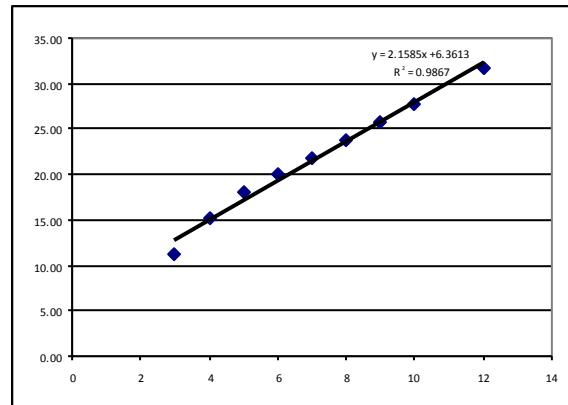
7t 7.5 mM

PPh₃ 15.0 mM

Following the General Procedure I, a mixture of silanolate Cs⁺**5**⁻ (39 mg, 0.125 mmol), **7t** (13.5 mg, 0.015 mmol, 0.090 equiv), naphthalene (11 mg), PPh₃ (8.0 mg, 0.030 mmol) and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μmol
3	19651	2423	11.23
3	19532	2407	11.22
3	19609	2439	11.33
4	19362	3205	15.07
4	19405	3249	15.25
4	19473	3286	15.37
5	15796	3098	17.86
5	15780	3190	18.41
5	15401	3045	18.00
6	24640	5376	19.87
6	24648	5355	19.78
6	24239	5408	20.32
7	18878	4389	21.17
7	18686	4576	22.30
7	18006	4375	22.12
8	18413	4838	23.93
8	18507	4871	23.97
8	18962	4861	23.34
9	17184	4952	26.24
9	17336	4862	25.54
9	17264	4782	25.22
10	19706	6077	28.08
10	19690	5886	27.22
10	19997	6149	28.00
12	22783	8033	32.11
12	23976	8250	31.33
12	23764	8327	31.91

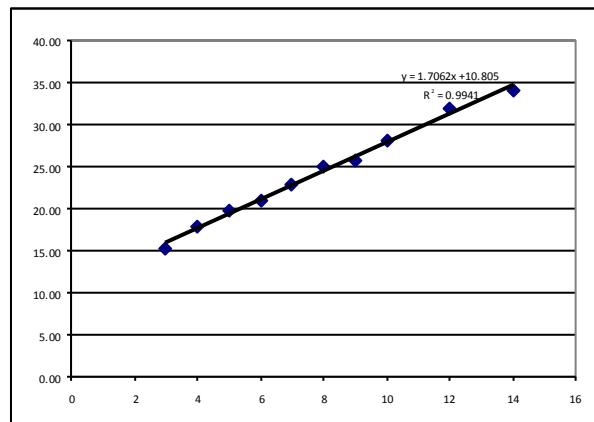


rate = **0.0179 mM/sec**

Run 2

Time	area, std	area, 12	conv, μmol
3	7299	1243	15.51
3	6068	975	14.63
3	6629	1145	15.73
4	7067	1430	18.43
4	7769	1488	17.44
4	7027	1347	17.45
5	8284	1802	19.81
5	7938	1699	19.49
5	8734	1913	19.94
6	8180	1835	20.43
6	8906	2116	21.63
6	9073	2076	20.84
7	8237	2003	22.14
7	6740	1661	22.44

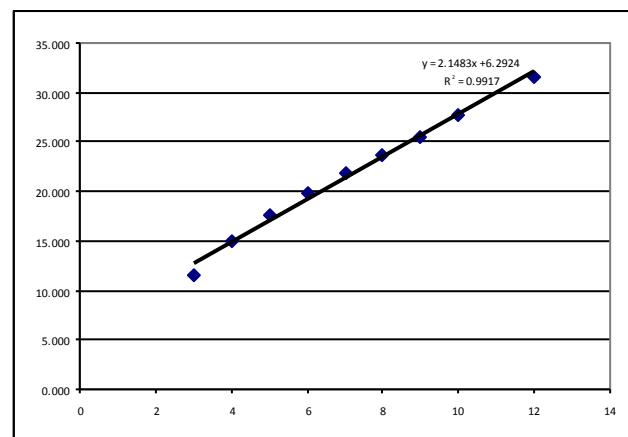
7	9064	2383	23.94
8	6528	1763	24.59
8	6224	1747	25.56
8	6765	1823	24.54
9	9251	2657	26.15
9	8762	2446	25.42
9	7043	1959	25.33
10	6459	2009	28.32
10	5860	1766	27.44
10	6037	1869	28.19
12	6343	2172	31.18
12	7241	2549	32.05
12	6866	2456	32.57
14	7962	2962	33.88
14	7607	2858	34.21
14	7763	2908	34.11



rate = **0.0142 mM/sec**

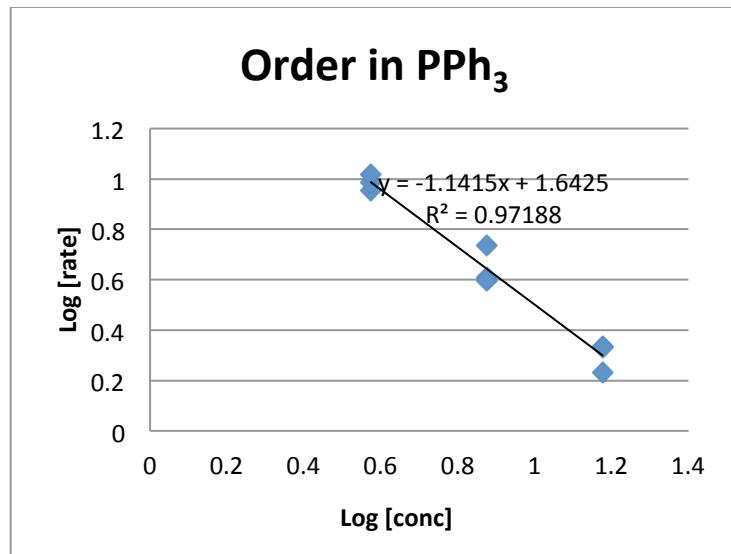
Run 3

Time	area, std	area, 12	conv, μmol
3	84449	7737	11.376
3	84873	7908	11.570
3	84534	7962	11.695
4	80840	9585	14.723
4	80020	9728	15.095
4	81009	9828	15.064
5	82094	11450	17.319
5	81790	11577	17.576
5	82368	11754	17.719
6	73254	11664	19.771
6	73933	11803	19.823
6	74492	11901	19.838
7	77398	13464	21.601
7	78144	13674	21.728
7	56749	10133	22.172
8	75582	14570	23.937
8	37398	7057	23.431
8	76740	14583	23.596
9	78247	16082	25.521
9	59530	12094	25.226
9	79599	16628	25.939
10	79149	17686	27.746
10	78352	17604	27.899
10	60230	13280	27.378
12	79011	20051	31.511
12	81418	20486	31.243
12	78122	20056	31.878



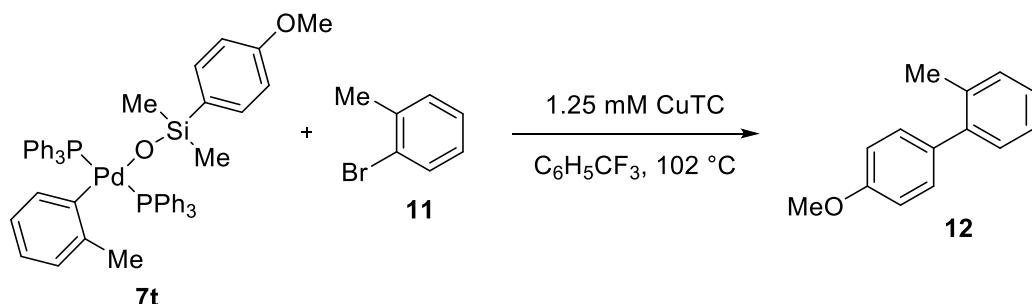
rate = **0.0178 mM/sec**

avg. rate = **$1.66 \times 10^{-2} \text{ mM/s}$**



GENERATION OF LIGANDLESS ARYLPALLADIUM(II) ARYLSILANOLATE SPECIES USING CuTC (Table 3)

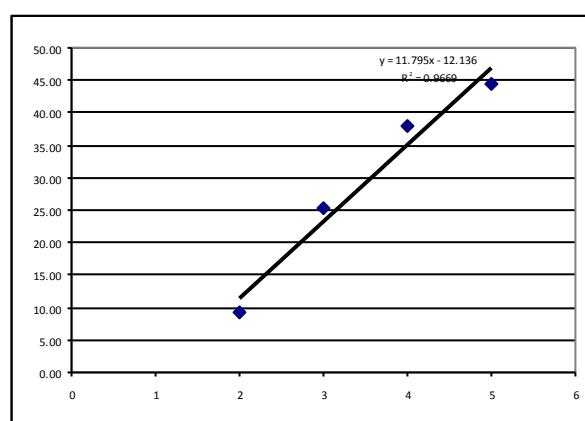
General Procedure II: Thermolysis of **7t in the presence of CuTC (Table 3, entry 1)**



In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with naphthalene (10 mg), **7t** (45 mg, 0.05 mmol), 2-bromotoluene (20 µL, 0.167 mmol, 3.3 equiv), CuTC (0.95 mg, 0.005 mmol, 0.1 equiv) and dry benzotrifluoride (4.0 mL). The flask was removed from the drybox, purged with argon, and then was placed in a 105 °C oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 30 µL aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 µL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.5 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC.

Run 1

Time	area, std	area, 12	conv, µmol
2	9765	1093	9.27
2	9723	1114	9.48
2	10069	1060	8.71
3	13219	4069	25.48
3	13897	4273	25.45
3	14696	4432	24.96
4	10153	4654	37.95
4	10870	5005	38.12
4	10771	4874	37.46
5	11491	6087	43.85



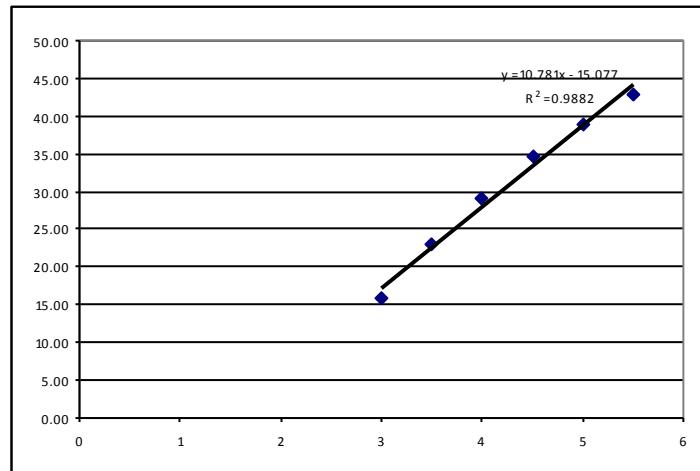
5	11422	6101	44.22
5	11358	6148	44.81
6	9271	5320	47.50
6	9304	5385	47.91
6	9457	5536	48.46

rate 4.92×10^{-2} mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
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3	13778	2209	15.93
3	13323	2122	15.82
3	13872	2245	16.08
3.5	11203	2493	22.11
3.5	10548	2476	23.32
3.5	10698	2514	23.34
4	10143	2939	28.78
4	10389	3028	28.95
4	10124	2983	29.27
4.5	13298	4619	34.50
4.5	13722	4794	34.70
4.5	13571	4762	34.86
5	14896	5883	39.23
5	14263	5571	38.80
5	14624	5736	38.96
5.5	13568	5988	43.84
5.5	14216	6134	42.86
5.5	14346	6061	41.97
6	11228	5222	46.20
6	11481	5483	47.44
6	11703	5417	45.98

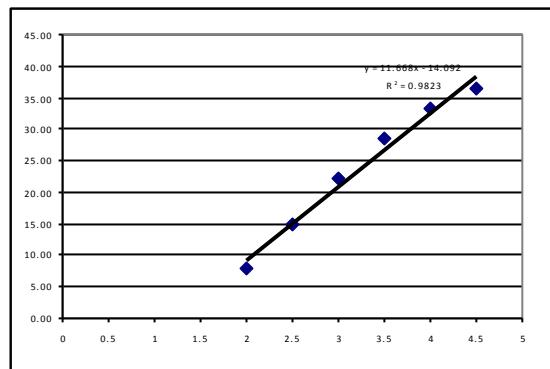


rate 4.49×10^{-2} mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
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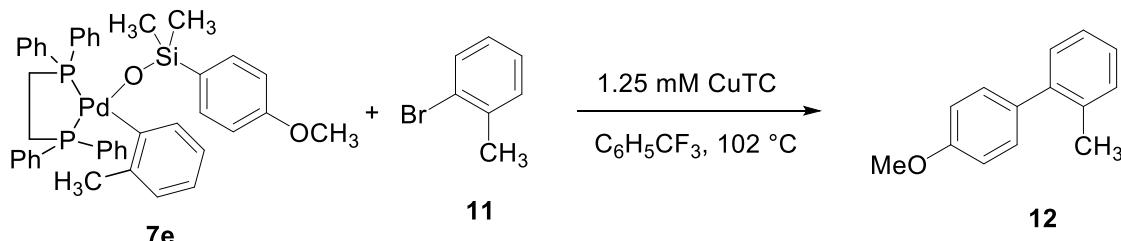
2	10366	1105	7.94
2	10925	1151	7.85
2	10873	1148	7.87
2.5	12027	2403	14.89
2.5	12119	2386	14.67
2.5	12190	2466	15.07
3	9978	2945	21.99
3	10466	3060	21.78
3	10221	3069	22.37
3.5	9593	3653	28.37
3.5	9279	3561	28.59



3.5	9740	3740	28.61
4	10177	4628	33.88
4	10082	4536	33.52
4	10454	4526	32.26
4.5	10556	5126	36.18
4.5	10667	5181	36.19
4.5	10791	5345	36.90
5	9016	4767	39.39
5	9348	4933	39.32
5	9440	5051	39.86

rate 4.87×10^{-2} mM/sec
avg. rate = 4.76×10^{-2} mM/s

Thermolysis of **7e** in the presence of CuTC (Table 3, entry 2)

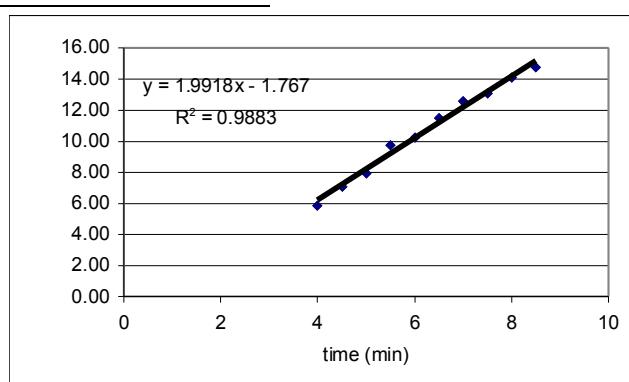


Run 1

Following General Procedure II, a mixture of **7e** (39 mg, 0.05 mmol, 1.0 equiv), naphthalene (10.8 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 μ L, 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis.

time, min	area naph	area 12	mmol 12
1	13982	0	0.00
1	12629	0	0.00
1	13743	0	0.00
1.5	15342	0	0.00
1.5	14956	0	0.00
1.5	15337	0	0.00
2	15752	0	0.00
2	15691	0	0.00
2	16465	0	0.00
2.5	15891	0	0.00
2.5	16530	0	0.00
2.5	15498	0	0.00
3	16561	0	0.00
3	16078	0	0.00

3	15308	0	0.00
3.5	14274	0	0.00
3.5	13965	0	0.00
3.5	13767	0	0.00
4	14218	807	5.92
4	14989	822	5.72
4	14651	829	5.90
4.5	15307	992	6.76
4.5	15554	1147	7.69
4.5	15563	1017	6.82
5	15227	1194	8.18
5	15441	1154	7.80
5	15887	1196	7.85
5.5	14950	1354	9.45
5.5	14832	1377	9.68
5.5	14938	1449	10.12
6	11950	1165	10.17
6	12797	1241	10.11
6	11689	1167	10.41
6.5	13729	1537	11.68
6.5	14829	1582	11.13
6.5	13936	1570	11.75
7	10775	1303	12.61
7	10486	1281	12.74
7	10780	1273	12.32
7.5	16382	1997	12.71
7.5	15474	1958	13.20
7.5	15253	1937	13.25
8	12450	1667	13.97
8	12195	1658	14.18
8	12449	1684	14.11
8.5	16721	2288	14.27
8.5	14744	2175	15.39
8.5	16419	2295	14.58



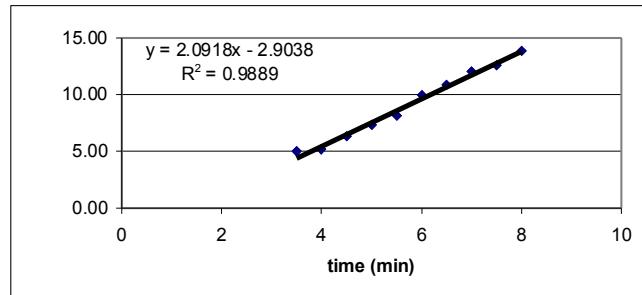
$$\text{rate} = 8.29 \times 10^{-3} \text{ mM/s}$$

Run 2

Following General Procedure II, a mixture of **7e** (39 mg, 0.05 mmol, 1.0 equiv),

naphthalene (9.4 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 μ L, 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis.

time, min	area naph	area 12	umol 12
1	9987	0	0.00
1	9987	0	0.00
1	9987	0	0.00
1.5	10687	0	0.00
1.5	10687	0	0.00
1.5	10687	0	0.00
2	11730	0	0.00
2	11730	0	0.00
2	11730	0	0.00
2.5	11362	0	0.00
2.5	11362	0	0.00
2.5	11362	0	0.00
3	13764	0	0.00
3	13764	0	0.00
3	13764	0	0.00
3.5	12200	786	5.01
3.5	12200	786	5.01
3.5	12200	786	5.01
4	13444	899	5.20
4	13444	899	5.20
4	13444	899	5.20
4.5	13288	1078	6.31
4.5	13288	1078	6.31
4.5	13288	1078	6.31
5	12510	1173	7.30
5	12510	1173	7.30
5	12510	1173	7.30
5.5	15190	1584	8.11
5.5	15190	1584	8.11
5.5	15190	1584	8.11
6	12974	1656	9.93
6	12974	1656	9.93
6	12974	1656	9.93
6.5	14135	1977	10.88
6.5	14135	1977	10.88
6.5	14135	1977	10.88
7	13322	2059	12.03
7	13322	2059	12.03
7	13322	2059	12.03
7.5	13500	2181	12.57
7.5	13500	2181	12.57
7.5	13500	2181	12.57
8	13534	2415	13.89
8	13534	2415	13.89
8	13534	2415	13.89

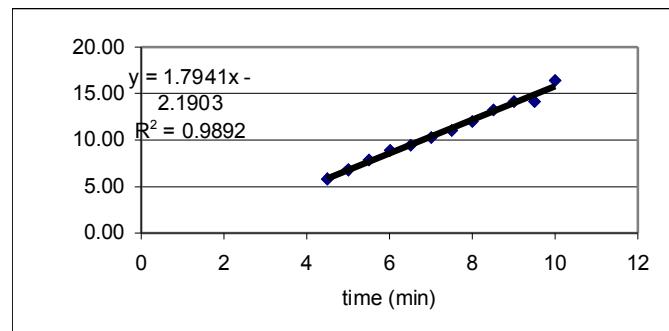


$$\text{rate} = 8.70 \times 10^{-3} \text{ mM/s}$$

Run 3

Following General Procedure II, a mixture of **7e** (39 mg, 0.05 mmol, 1.0 equiv), naphthalene (12.5 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 μ L, 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis.

time	area naph	area 12	$\mu\text{mol } \mathbf{12}$
1	11884	0	0.00
1	11856	0	0.00
1	12754	0	0.00
1.5	14133	0	0.00
1.5	14060	0	0.00
1.5	13438	0	0.00
2	11943	0	0.00
2	12053	0	0.00
2	12700	0	0.00
2.5	14775	0	0.00
2.5	14223	0	0.00
2.5	12835	0	0.00
3	12953	0	0.00
3	12706	0	0.00
3	13666	0	0.00
3.5	14856	0	0.00
3.5	14319	0	0.00
3.5	13971	0	0.00
4	14531	0	0.00
4	13881	0	0.00
4	13108	0	0.00
4.5	14414	778	5.59
4.5	14368	747	5.38
4.5	12801	808	6.53
5	13924	882	6.55
5	13598	903	6.87
5	12839	868	7.00
5.5	11142	855	7.94
5.5	11559	892	7.99
5.5	11698	848	7.50
6	14108	1139	8.35
6	13380	1218	9.42
6	13350	1128	8.74
6.5	13874	1255	9.36
6.5	13281	1213	9.45
6.5	13071	1200	9.50
7	13096	1278	10.10
7	13186	1378	10.81
7	13266	1256	9.80
7.5	13680	1483	11.22
7.5	13744	1450	10.92

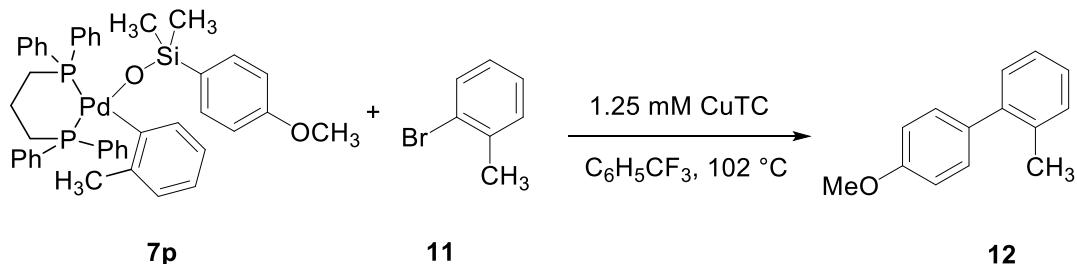


7.5	14110	1474	10.81
8	15048	1652	11.36
8	14406	1665	11.96
8	13130	1606	12.66
8.5	12996	1696	13.50
8.5	12773	1673	13.55
8.5	14610	1793	12.70
9	14361	1929	13.90
9	13733	1890	14.24
9	13576	1859	14.17
9.5	16056	2173	14.00
9.5	15258	2110	14.31
9.5	15439	2092	14.02
10	11778	1772	15.57
10	10675	1714	16.61
10	10264	1689	17.03

$$\text{rate} = 7.46 \times 10^{-3} \text{ mM/s}$$

$$\text{avg. rate} = 8.15 \times 10^{-3} \text{ mM/s}$$

Thermolysis of **14p** in the presence of CuTC (Table 3, entry 3)

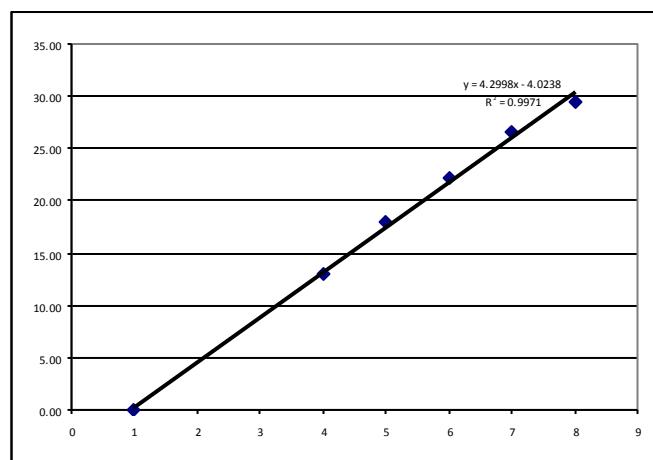


Following General Procedure II, a mixture of **7p** (43 mg, 0.05 mmol, 1.0 equiv), naphthalene (9.4 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 µL, 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, µmol
1	1	0	0.00
1	1	0	0.00
1	1	0	0.00
4	14848	2321	12.94
4	14848	2321	12.94

4	14848	2321	12.94
5	23038	4862	17.47
5	7610	1706	18.56
5	8010	1730	17.88
6	8377	2376	23.48
6	9810	2565	21.64
6	39135	10207	21.59
7	15304	5006	27.08
7	23342	7298	25.88
7	15565	5012	26.66
8	7678	2681	28.91
8	29806	10742	29.83
8	7677	2750	29.65

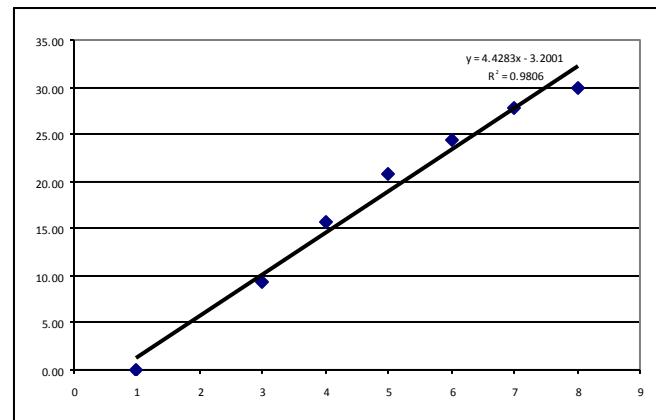


$$\text{rate} = 1.79 \times 10^{-2} \text{ mM/s}$$

Run 2

Time area, std area, 12 conv, μmol

1	1	0	0.00
1	1	0	0.00
1	1	0	0.00
3	42759	5958	9.23
3	42655	5911	9.18
3	43014	6097	9.39
4	41144	9475	15.25
4	37650	9462	16.64
4	41802	9498	15.05
5	37691	11542	20.28
5	37645	11798	20.75
5	36256	11847	21.64
6	36417	13417	24.40
6	37114	13621	24.30
6	36988	13839	24.78
7	39434	16576	27.84
7	39975	16798	27.83
7	41028	17308	27.94
8	39320	17788	29.96
8	29373	13396	30.20
8	39815	17936	29.83



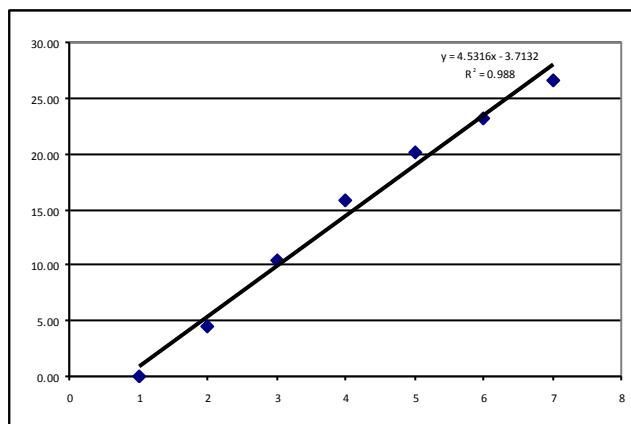
$$\text{rate} = 1.84 \times 10^{-2} \text{ mM/s}$$

Run 3

Time area, std area, 12 conv, μmol

1	1	0	0.00
1	1	0	0.00

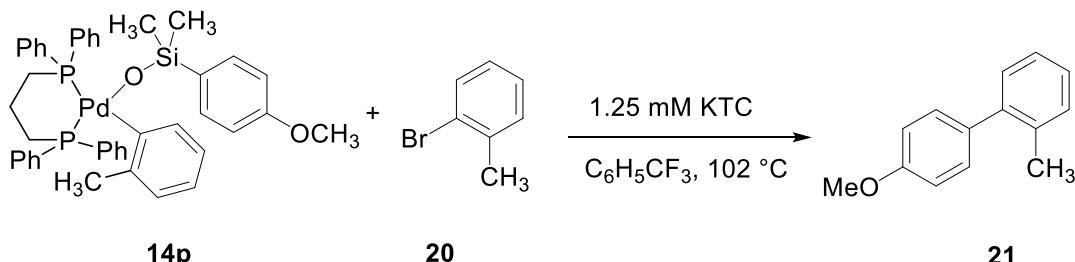
1	1	0	0.00
2	33849	1708	4.59
2	34312	1712	4.54
2	33849	1708	4.59
3	56720	6547	10.51
3	45725	5203	10.36
3	56720	6547	10.51
4	54966	9637	15.97
4	44183	7805	16.09
4	44394	7604	15.60
5	49869	10991	20.07
5	50264	11080	20.07
5	50263	11152	20.20
6	19723	4960	22.90
6	47495	12391	23.76
6	19723	4960	22.90
7	51994	15281	26.76
7	50784	14775	26.49
7	51994	15281	26.76



$$\text{rate} = 1.88 \times 10^{-2} \text{ mM/sec}$$

$$\text{avg. rate} = 1.84 \times 10^{-2} \text{ mM/s}$$

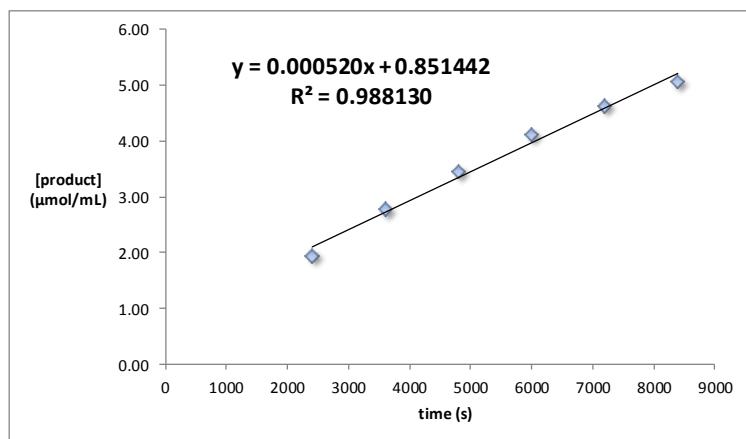
Thermolysis of **7p** in the presence of KTC (Footnote 14)



Run 1

Following General Procedure II, a mixture of **14p** (43 mg, 0.05 mmol, 1.0 equiv), naphthalene (10.0 mg), KTC (0.8 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 μL , 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 $^\circ\text{C}$. Aliquots of the mixture were then taken for GC analysis.

time (s)	naphthalene		product			average mM
	mmol	area	area	mmol	(%)	
2400	0.079	5.82E+03	7.23E+02	7.72E-03	15.4	
	0.079	4.61E+03	5.82E+02	7.84E-03	15.7	1.94
	0.079	6.42E+03	7.92E+02	7.67E-03	15.3	
3600	0.079	5.36E+03	9.37E+02	1.09E-02	21.7	
	0.079	7.99E+03	1.45E+03	1.12E-02	22.5	2.78
	0.079	5.21E+03	9.44E+02	1.13E-02	22.5	
4800	0.079	4.97E+03	9.95E+02	1.24E-02	24.9	
	0.079	7.19E+03	1.70E+03	1.47E-02	29.4	3.45
	0.079	6.61E+03	1.53E+03	1.43E-02	28.7	
6000	0.079	1.08E+04	2.76E+03	1.58E-02	31.6	
	0.079	6.90E+03	1.84E+03	1.65E-02	33.1	4.11
	0.079	9.85E+03	2.69E+03	1.69E-02	33.9	
7200	0.079	8.45E+03	2.43E+03	1.78E-02	35.7	
	0.079	8.48E+03	2.54E+03	1.86E-02	37.2	4.62
	0.079	8.05E+03	2.47E+03	1.90E-02	38.0	
8400	0.079	9.44E+03	3.01E+03	1.98E-02	39.6	
	0.079	7.17E+03	2.37E+03	2.05E-02	41.1	5.07
	0.079	7.71E+03	2.54E+03	2.05E-02	41.0	

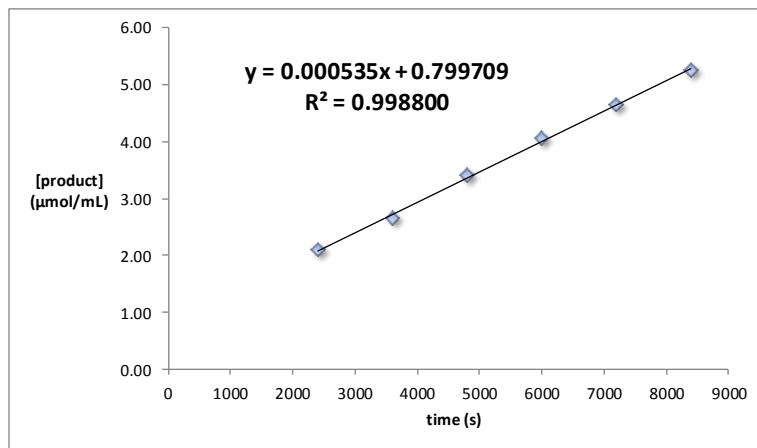


$$\text{rate} = 5.20 \times 10^{-4} \text{ mM/s}$$

Run 2

Following General Procedure II, a mixture of **14p** (43 mg, 0.05 mmol, 1.0 equiv), naphthalene (12.1 mg), KTC (0.8 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 μL , 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis.

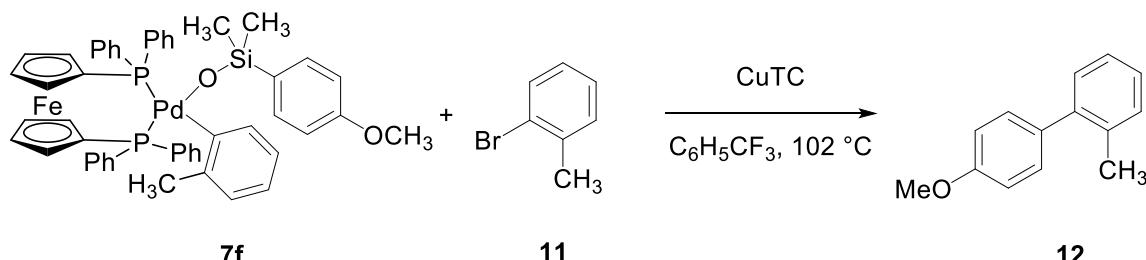
time (s)	naphthalene		product		
	mmol	area	area	mmol	(%)
2400	0.094	1.28E+04	1.40E+03	8.10E-03	16.2
	0.094	8.72E+03	9.94E+02	8.42E-03	16.8
	0.094	1.60E+04	1.85E+03	8.56E-03	17.1
3600	0.094	9.43E+03	1.25E+03	9.81E-03	19.6
	0.094	7.35E+03	1.02E+03	1.03E-02	20.5
	0.094	1.07E+04	1.73E+03	1.19E-02	23.8
4800	0.094	1.48E+04	2.77E+03	1.38E-02	27.7
	0.094	1.14E+04	2.09E+03	1.36E-02	27.2
	0.094	1.17E+04	2.12E+03	1.34E-02	26.8
6000	0.094	1.18E+04	2.54E+03	1.60E-02	31.9
	0.094	1.51E+04	3.33E+03	1.64E-02	32.7
	0.094	1.11E+04	2.46E+03	1.64E-02	32.8
7200	0.094	1.16E+04	2.97E+03	1.88E-02	37.6
	0.094	8.71E+03	2.20E+03	1.87E-02	37.3
	0.094	9.02E+03	2.23E+03	1.83E-02	36.6
8400	0.094	1.25E+04	3.57E+03	2.10E-02	42.0
	0.094	1.22E+04	3.48E+03	2.12E-02	42.3
	0.094	1.21E+04	3.42E+03	2.10E-02	42.0



$$\text{rate} = 5.35 \times 10^{-4} \text{ mM/s}$$

$$\text{avg. rate} = 5.28 \times 10^{-4} \text{ mM/s}$$

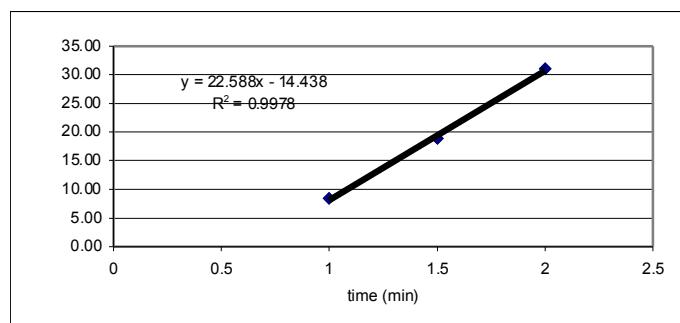
Thermolysis of **7f** in the presence of CuTC (Table 3, entry 4)



Run 1

Following General Procedure II, a mixture of **7f** (47 mg, 0.05 mmol, 1.0 equiv), naphthalene (9.6 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 µL, 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis.

time, min	area naph	area 12	µmol 12
1	8703	901	8.23
1	8804	1005	9.07
1	8621	875	8.07
1.5	12133	2886	18.90
1.5	12211	2898	18.86
1.5	12280	2896	18.74
2	11164	4264	30.35
2	10464	4141	31.45
2	10852	4278	31.33

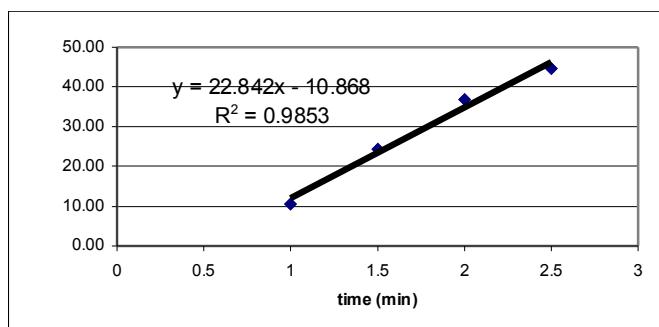


$$\text{rate} = 9.42 \times 10^{-2} \text{ mM/s}$$

Run 2

Following General Procedure II, a mixture of **7f** (47 mg, 0.05 mmol, 1.0 equiv), naphthalene (9.3 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 µL, 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis.

time, min	area naph	area 12	$\mu\text{mol } \mathbf{12}$
1	11121	1504	10.41
1	11049	1671	11.64
1	11176	1442	9.93
1.5	9784	3264	25.68
1.5	11020	3398	23.74
1.5	10511	3216	23.56
2	11524	5580	37.28
2	11378	5439	36.80
2	11581	5504	36.59
2.5	10959	6705	47.10
2.5	11810	6720	43.81
2.5	12496	6935	42.73



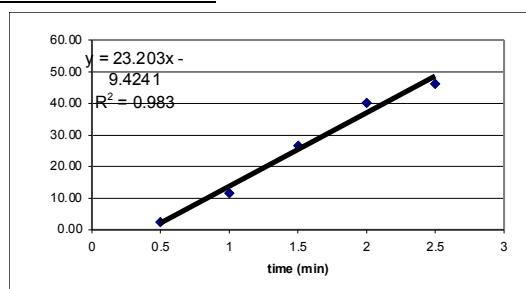
$$\text{rate} = 9.50 \times 10^{-2} \text{ mM/s}$$

Run 3

Following General Procedure II, a mixture of **7f** (47 mg, 0.05 mmol, 1.0 equiv), naphthalene (8.9 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 μL , 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis.

time, min	area naph	area 12	$\mu\text{mol } \mathbf{12}$
0.5	29670	1098	2.73
0.5	28296	890	2.32
0.5	28533	899	2.32
1	36631	5641	11.35
1	35237	5490	11.48
1	35263	5539	11.57
1.5	40480	14620	26.61
1.5	40140	14404	26.44
1.5	39716	14598	27.08
2	28674	15081	38.75
2	28328	15643	40.68
2	28665	15959	41.02
2.5	35983	22214	45.48

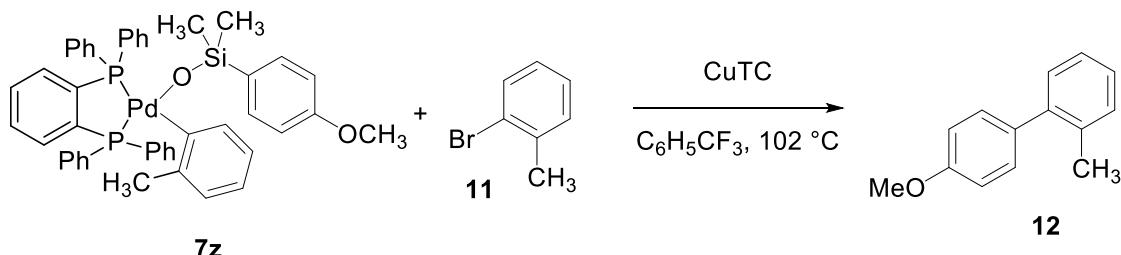
2.5	34904	22078	46.60
2.5	35568	22339	46.27



$$\text{rate} = 9.67 \times 10^{-2} \text{ mM/s}$$

$$\text{avg. rate} = 9.21 \times 10^{-2} \text{ mM/s}$$

Thermolysis of **7z** in the presence of CuTC (Table 3, entry 5)



Run 1

Following General Procedure II, a mixture of **7z** (41 mg, 0.05 mmol, 1.0 equiv), naphthalene (8.7 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 μ L, 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis. No product formation was observed after 3 h of heating.

Run 2

Following General Procedure II, a mixture of **7z** (41 mg, 0.05 mmol, 1.0 equiv), naphthalene (12.8 mg), CuTC (1 mg, 0.005 mmol, 0.1 equiv) and 2-bromotoluene (20 μ L, 0.16 mmol, 3.3 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at 105 °C. Aliquots of the mixture were then taken for GC analysis. No product formation was observed after 3 h of heating.

INDEX OF KINETIC EXPERIMENTS FOR THE CROSS-COUPING OF CESIUM (4-METHOXYPHENYL)DIMETHYLSILANOLATE USING 1,3-BIS(DIPHENYLPHOSPHINO)PROPANE BIS-OXIDE AS A LIGAND (TABLE 4)

	Page Number
Order in Cs ⁺ 5 ⁻ with APC/dPPP(O) ₂	
62.5 mM	S54
125 mM	S56
250 mM	S58
Order in 2-bromotoluene with APC/dPPP(O) ₂	
83 mM	S54
167 mM	S60
333 mM	S62
Order in dPPP(O) ₂ with Cs ⁺ 5 ⁻ at (125 mM)	S64
Order in Cs ⁺ 5 ⁻ with APC/Ph ₃ P(O) (15 mM)	
62.5 mM	S66
125 mM	S68
250 mM	S70
Order in APC with APC/dPPP(O) ₂	
7.5 mM	S54
3.75 mM	S73
11.25 mM	S75

General Procedure III: Analysis and Kinetic Measurements for Cross-Coupling of Cs⁺5⁻
Initial Rates using Phosphine Oxide as the Ligand (Table 4)

Order in Cs⁺5⁻ with APC/dppp(O)₂ (Table 4, entry 1)

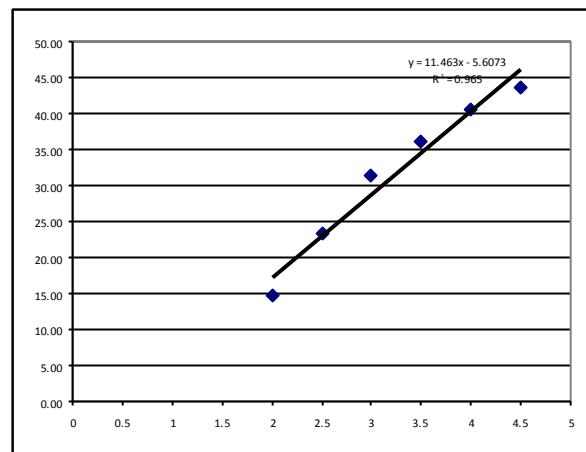
Cs ⁺ 5 ⁻	62.5 mM
2-Bromotoluene	83 mM
APC	3.75 mM
dppp(O) ₂	7.5 mM

In a drybox, a one-necked, 5-mL round-bottomed flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with Cs⁺5⁻ (39 mg, 0.125 mmol, 0.75 equiv), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (10 mg), and 2-bromotoluene (20 μ L, 0.16 mmol, 1.0 equiv) and dry benzotrifluoride (2.0 mL). The flask was placed in a 105 °C oil bath. Reaction progress was monitored by GC analysis. Sampling of the reaction was performed by removing 50 μ L aliquots of the mixture by syringe and quenching the aliquots at regular intervals. The quench was performed as follows: the aliquot was injected into 100 μ L of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 1.0 mL of ethyl acetate and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC.

Run 1

Time	area, std	area, 12	conv, μ mol
------	-----------	----------	-----------------

2	9039	1619	14.83
2	9540	1671	14.50
2	8910	1630	15.14
2.5	9429	2646	23.23
2.5	9426	2683	23.56
2.5	9443	2662	23.34
3	8404	3139	30.92
3	7872	3018	31.74
3	7938	3015	31.44
3.5	8016	3447	35.60
3.5	7943	3487	36.34
3.5	8850	3859	36.10
4	8357	4142	41.03



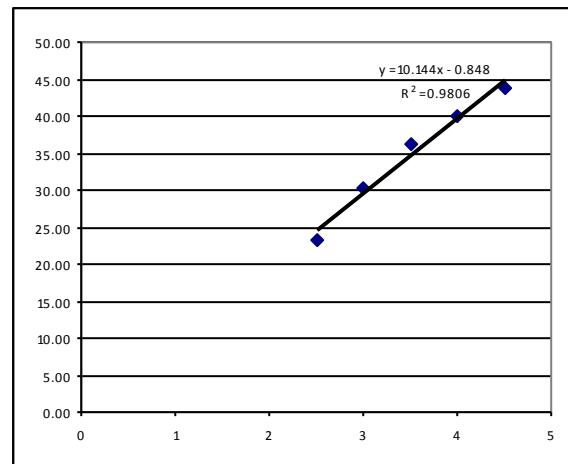
4	9292	4495	40.05
4	8931	4409	40.87
4.5	8012	4199	43.38
4.5	7901	4163	43.62
4.5	7741	4110	43.95

rate = 0.0955 mM/sec

Run 2

Time area, std area, 12 conv, μmol

2.5	7546	2034	24.54
2.5	8486	2134	22.90
2.5	7634	1880	22.42
3	8124	2685	30.10
3	8553	2875	30.61
3	8832	2892	29.82
3.5	7036	2810	36.37
3.5	7602	3010	36.05
3.5	6111	2418	36.03
4	8071	3544	39.98
4	8696	3837	40.18
4	9391	4069	39.45
4.5	7370	3559	43.97
4.5	7862	3751	43.44
4.5	7519	3632	43.99

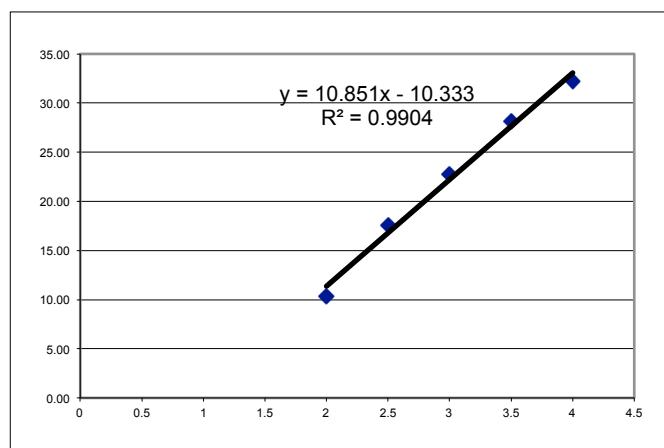


rate = 0.0845 mM/sec

Run 3

Time area, std area, 12 conv, μmol

2	17401	2190	10.42
2	17037	2154	10.47
2	17456	2163	10.26
2.5	18257	3875	17.57
2.5	18260	3832	17.37
2.5	18509	3978	17.79
3	21819	6022	22.85
3	21978	6055	22.81
3	22469	6148	22.65
3.5	18401	6292	28.31
3.5	17909	6047	27.95
3.5	18800	6400	28.18
4	18107	7005	32.03
4	18038	6988	32.07
4	17794	7002	32.57
4.5	18828	7990	35.13
4.5	18644	7951	35.30
4.5	17899	7639	35.33



rate = 0.0905 mM/sec

avg. rate = 9.02×10^{-2} mM/s

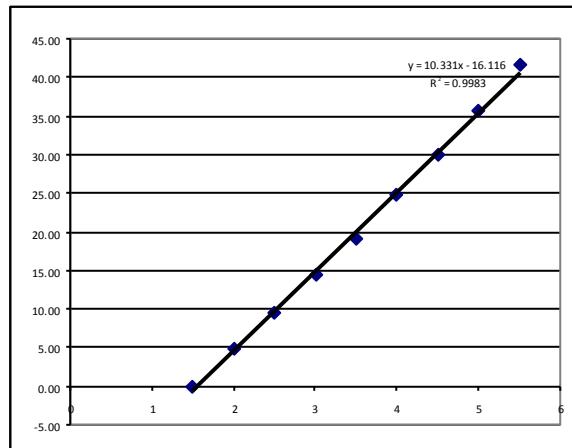
Table 4, entry 2

$\text{Cs}^+ \mathbf{5}^-$	125 mM
2-Bromotoluene	83 mM
APC	3.75 mM
dppp(O) ₂	7.5 mM

Following the General Procedure III, a mixture of silanolate $\text{Cs}^+ \mathbf{5}^-$ (78 mg, 0.250 mmol, 1.5 equiv), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (10 mg), and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μmol
1.5	1	0	0.00
1.5	1	0	0.00
1.5	1	0	0.00
2	57759	3613	5.18
2	43602	2383	4.52
2	57759	3613	5.18
2.5	14242	1540	8.95
2.5	58344	6140	8.71
2.5	58941	7854	11.03
3	59989	11733	16.19
3	44769	7437	13.75
3	59805	9906	13.71
3.5	60961	14217	19.31
3.5	61510	14113	18.99
3.5	61459	14151	19.06
4	57100	17013	24.66
4	56481	17114	25.08
4	56573	16924	24.76
4.5	16301	5838	29.65
4.5	16248	6013	30.64
4.5	46728	16785	29.74
5	15976	6695	34.69
5	15420	6739	36.18
5	15745	6891	36.23
5.5	14179	7072	41.29
5.5	13234	6769	42.34
5.5	57590	28712	41.27

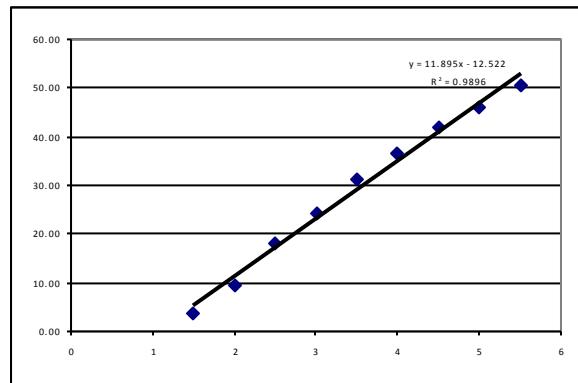


rate = 0.0861 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

1.5	51699	2283	3.66
1.5	50502	2206	3.62
1.5	51699	2283	3.66
2	66484	7738	9.63
2	66489	7708	9.60
2	66278	7721	9.64
2.5	63901	14117	18.29
2.5	63363	13740	17.95
2.5	64165	13784	17.78
3	61905	18343	24.53
3	16797	4924	24.27
3	61630	18054	24.25
3.5	55118	20823	31.27
3.5	56083	20976	30.96
3.5	27825	10534	31.34
4	61868	27253	36.47
4	31213	13896	36.85
4	31627	14020	36.70
4.5	15611	7982	42.33
4.5	61972	31319	41.84
4.5	60808	30697	41.79
5	63762	35032	45.48
5	64313	36053	46.41
5	48599	27245	46.41
5.5	59569	36007	50.04
5.5	59196	36449	50.97
5.5	15862	9636	50.29

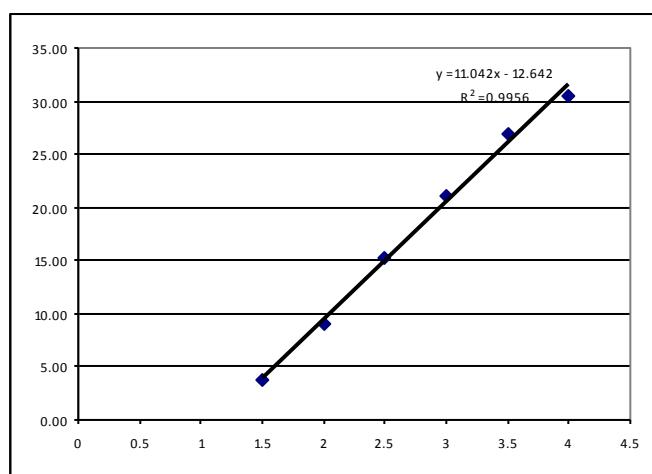


rate = 0.0991 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

1.5	71276	2937	3.75
1.5	70946	2902	3.72
1.5	71276	2937	3.75
2	53080	5215	8.95
2	35434	3506	9.01
2	18085	1814	9.13
2.5	16559	2760	15.18
2.5	60532	10107	15.20
2.5	45349	7512	15.08
3	46905	10720	20.81
3	47953	11230	21.32
3	47469	10897	20.90
3.5	15274	4364	26.02
3.5	13887	4159	27.27



3.5	15899	4783	27.39
4	16387	5556	30.87
4	65024	21616	30.27
4	16575	5525	30.35

rate = 0.0920 mM/sec

avg. rate = 9.24×10^{-2} mM/s

Table 4, entry 3

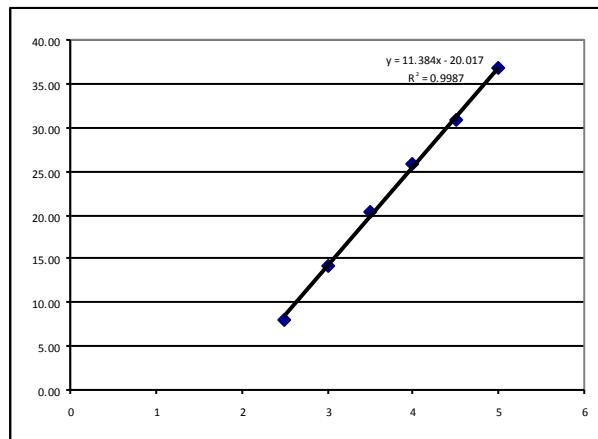
Cs ⁺ 5 ⁻	250 mM
2-Bromotoluene	83 mM
APC	3.75 mM
dppp(O) ₂	7.5 mM

Following the General Procedure III, a mixture of silanolate Cs⁺**5**⁻ (157 mg, 0.500 mmol, 3.0 equiv), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (9 mg), and 2-bromotoluene (20 μ L, 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time area, std area, **12** conv, μ mol

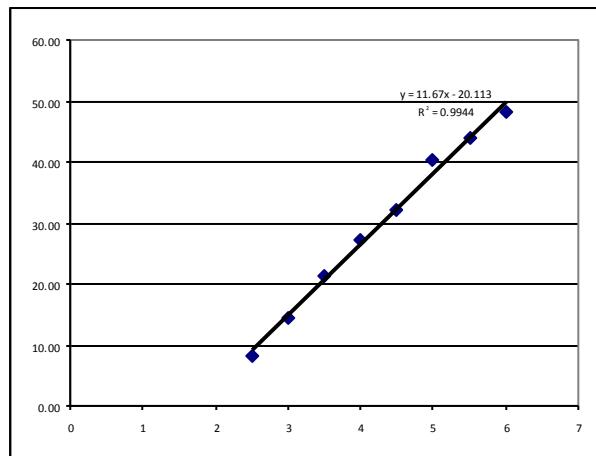
2.5	16088	1743	8.07
2.5	16948	1824	8.02
2.5	16033	1752	8.14
3	17246	3240	14.00
3	16725	3208	14.29
3	16670	3110	13.90
3.5	16455	4444	20.12
3.5	16984	4660	20.44
3.5	17086	4714	20.56
4	19326	6710	25.87
4	18784	6480	25.70
4	19608	6903	26.23
4.5	19637	8107	30.76
4.5	19422	7983	30.62
4.5	18927	7950	31.29
5	18692	9275	36.97
5	18472	8993	36.27
5	18802	9312	36.90



rate = 0.0948 mM/sec

Run 2

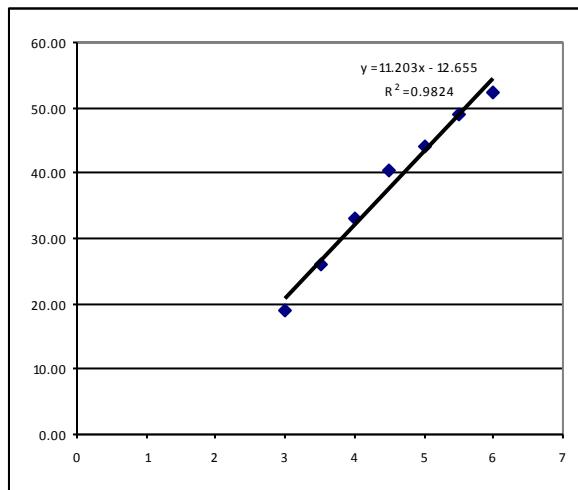
Time	area, std	area, 12	conv, μmol
2.5	18229	1992	8.14
2.5	17968	2000	8.29
2.5	17914	1972	8.20
3	10307	1975	14.28
3	9872	1984	14.97
3	10004	1932	14.39
3.5	8187	2400	21.84
3.5	7964	2270	21.24
3.5	8287	2279	20.49
4	8195	3020	27.46
4	8060	2893	26.74
4	8247	3015	27.24
4.5	8909	3843	32.14
4.5	8678	3788	32.52
4.5	8550	3675	32.02
5	6660	3671	41.07
5	7124	3761	39.33
5	6863	3712	40.30
5.5	8127	4781	43.83
5.5	8933	5372	44.80
5.5	8756	5121	43.57
6	8786	5605	47.53
6	9000	5819	48.17
6	9489	6250	49.07



rate = 0.0973 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
3	8662	1976	18.88
3	8277	1904	19.04
3	8807	2062	19.38
3.5	7600	2404	26.18
3.5	8265	2623	26.27
3.5	8637	2716	26.03
4	7389	2955	33.11
4	6726	2752	33.87
4	7092	2780	32.45
4.5	8330	3981	39.56
4.5	7711	3797	40.76
4.5	8170	3997	40.50
5	9891	5277	44.16
5	9619	5138	44.22
5	9874	5256	44.06
5.5	8961	5367	49.58



5.5	8652	5039	48.21
5.5	9743	5789	49.19
6	8516	5424	52.72
6	8970	5682	52.44
6	9565	6046	52.33

$$\text{rate} = 0.0933 \text{ mM/sec}$$

$$\text{avg. rate} = 9.51 \times 10^{-2} \text{ mM/sec}$$

Order in 2-bromotoluene with APC/dPPP(O)₂ (Table 4, entry 4)

Cs⁺**5**⁻ 125 mM

2-Bromotoluene 167 mM

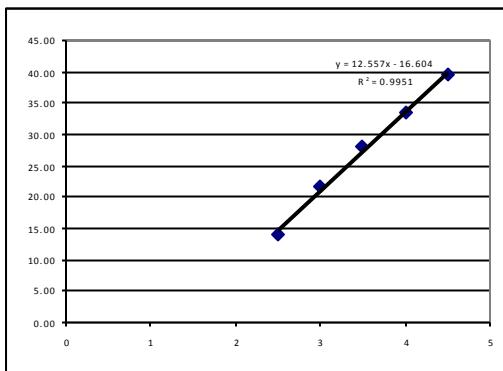
APC 3.75 mM

dPPP(O)₂ 7.5 mM

Following the General Procedure III, a mixture of silanolate Cs⁺**5**⁻ (78 mg, 0.250 mmol, 1.5 equiv), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), dPPP(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (10 mg), and 2-bromotoluene (40 µL, 0.33 mmol, 2.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

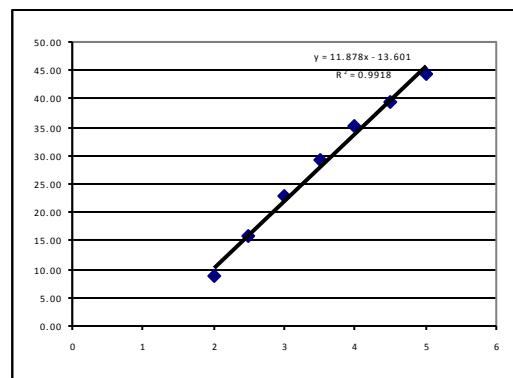
Time	area, std	area, 12	conv, µmol
2.5	5856	971	13.73
2.5	4338	734	14.01
2.5	5812	994	14.16
3	6153	1573	21.16
3	5917	1573	22.01
3	5769	1527	21.91
3.5	3978	1383	28.78
3.5	4491	1519	28.00
3.5	4093	1368	27.67
4	5885	2367	33.30
4	6151	2479	33.36
4	5123	2080	33.61
4.5	5519	2585	38.77
4.5	4334	2135	40.78
4.5	5677	2669	38.92



$$\text{rate} = 0.105 \text{ mM/sec}$$

Run 2

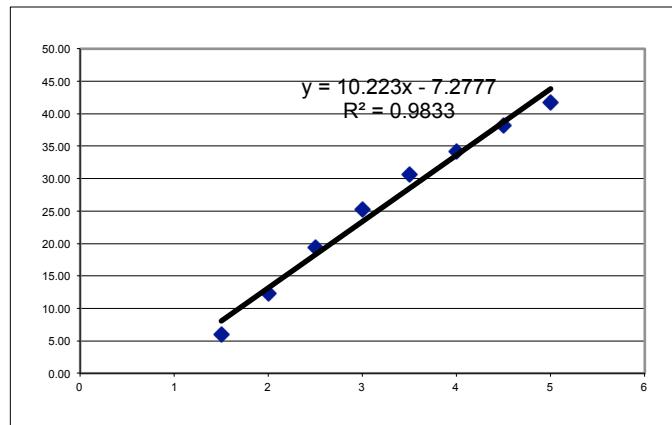
Time	area, std	area, 12	conv, μmol
2	7909	910	8.57
2	7775	937	8.98
2	12272	1403	8.52
2.5	12923	2750	15.85
2.5	13321	2844	15.91
2.5	12785	2747	16.01
3	12888	4012	23.19
3	12461	3869	23.13
3	15864	4843	22.74
3.5	11297	4480	29.55
3.5	11158	4367	29.16
3.5	4109	1589	28.81
4	8806	4184	35.40
4	9374	4444	35.32
4	9270	4296	34.53
4.5	9063	4893	40.22
4.5	5181	2749	39.53
4.5	5426	2838	38.97
5	9824	5827	44.19
5	9107	5490	44.91
5	7990	4707	43.89



rate = 0.0989 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
1.5	12622	730	5.75
1.5	12850	802	6.20
1.5	13783	879	6.34
2	13021	1511	11.53
2	13733	1748	12.64
2	13715	1733	12.55
2.5	16499	3247	19.55
2.5	17074	3305	19.23
2.5	16639	3210	19.16
3	14770	3843	25.85
3	13184	3323	25.04
3	14902	3680	24.53
3.5	15748	4864	30.68
3.5	15225	4748	30.98
3.5	15315	4705	30.52
4	15500	5290	33.90
4	15341	5320	34.45
4	15694	5375	34.02
4.5	12380	4785	38.39



4.5	14826	5704	38.22
4.5	13154	5024	37.94

rate = 0.0852 mM/sec
avg. rate = 9.63 x 10⁻² mM/s

Table 4, entry 5

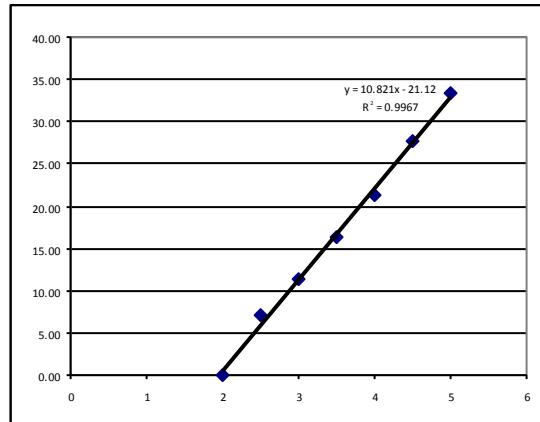
Cs ⁺ 5 ⁻	125 mM
2-Bromotoluene	333 mM
APC	3.75 mM
dppp(O) ₂	7.5 mM

Following the General Procedure III, a mixture of silanolate Cs⁺**5**⁻ (78 mg, 0.250 mmol, 1.5 equiv), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (11 mg), and 2-bromotoluene (80 μ L, 0.66 mmol, 4.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μ mol
------	-----------	-----------------	-----------------

2	1	0	0.00
2	1	0	0.00
2	1	0	0.00
2.5	18321	1647	8.19
2.5	17823	1476	7.54
2.5	18708	1124	5.47
3	15745	2126	12.30
3	17707	2152	11.07
3	33522	4052	11.01
3.5	16388	2901	16.12
3.5	31045	5507	16.15
3.5	16215	2918	16.39
4	33510	7808	21.22
4	48626	11423	21.39
4	16795	3965	21.50
4.5	14022	4292	27.87
4.5	14330	4325	27.48
4.5	13947	4277	27.92
5	14843	5192	31.85
5	13256	5126	35.21



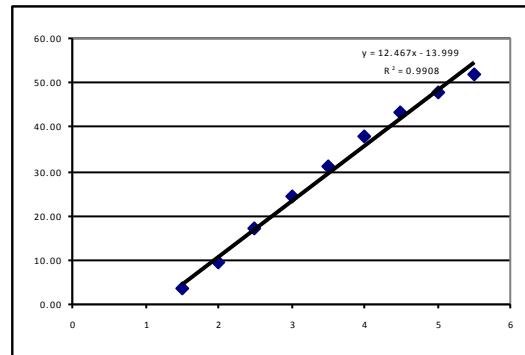
5	14350	5220	33.12
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$$\text{rate} = 0.0902 \text{ mM/sec}$$

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

1.5	14973	748	3.72
1.5	14901	756	3.78
1.5	15322	736	3.58
2	14736	1783	9.01
2	13658	1770	9.66
2	14449	1795	9.26
2.5	14037	3154	16.74
2.5	17421	3984	17.04
2.5	54933	13125	17.80
3	17666	5840	24.63
3	17810	5842	24.44
3	17994	5811	24.06
3.5	17848	7509	31.34
3.5	18147	7540	30.96
3.5	17988	7444	30.83
4	16425	8504	38.57
4	16087	8143	37.71
4	16568	8410	37.82
4.5	18166	10513	43.12
4.5	18796	10730	42.53
4.5	17972	10826	44.88
5	18711	11987	47.73
5	18748	12093	48.06
5	19122	12270	47.81
5.5	16140	11212	51.75
5.5	15923	11011	51.52
5.5	16073	11187	51.85

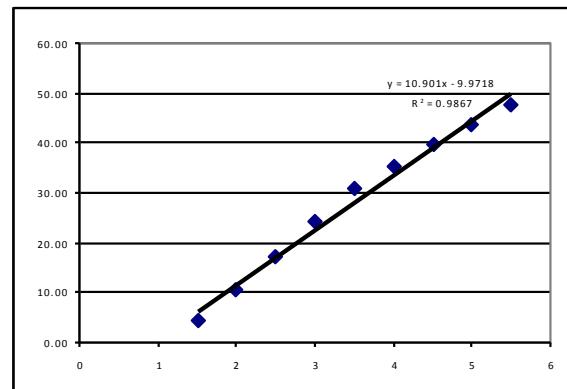


$$\text{rate} = 0.104 \text{ mM/sec}$$

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

1.5	16906	970	4.75
1.5	17389	943	4.49
1.5	17355	963	4.59
2	19442	2420	10.30
2	19250	2505	10.77
2	19545	2418	10.24
2.5	21061	4408	17.33
2.5	20495	4264	17.22
2.5	20381	4364	17.73
3	19134	5662	24.50
3	19466	5640	23.98



3	19645	5721	24.11
3.5	20188	7598	31.16
3.5	19867	7337	30.57
3.5	20619	7548	30.30
4	16117	6801	34.93
4	15537	6713	35.77
4	15639	6722	35.58
4.5	21154	10223	40.01
4.5	21384	10162	39.34
4.5	22000	10565	39.75
5	19076	10065	43.68
5	18841	10092	44.34
5	19045	9892	43.00
5.5	17949	10321	47.60
5.5	17550	10100	47.64
5.5	18103	10336	47.26

$$\text{rate} = 0.0908 \text{ mM/sec}$$

$$\text{avg. rate} = 9.50 \times 10^{-2} \text{ mM/s}$$

Order in dPPP(O)₂ with Cs⁺5⁻ at (125 mM) (Table 4, entry 6)

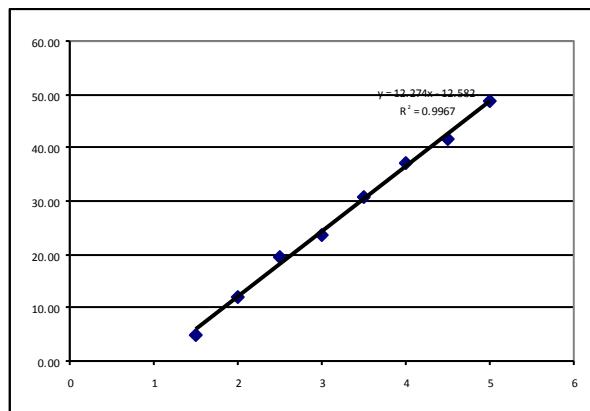
Cs ⁺ 5 ⁻	125 mM
2-Bromotoluene	83 mM
APC	3.75 mM
dPPP(O) ₂	15 mM

Following the General Procedure III, a mixture of silanolate Cs⁺5⁻ (78 mg, 0.250 mmol, 1.5 equiv), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), dPPP(O)₂ (13.2 mg, 0.030 mmol, 0.180 equiv), naphthalene (10 mg), and 2-bromotoluene (20 μ L, 0.167 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μmol
1.5	19493	1138	4.83
1.5	19167	1118	4.83
1.5	19287	1165	5.00

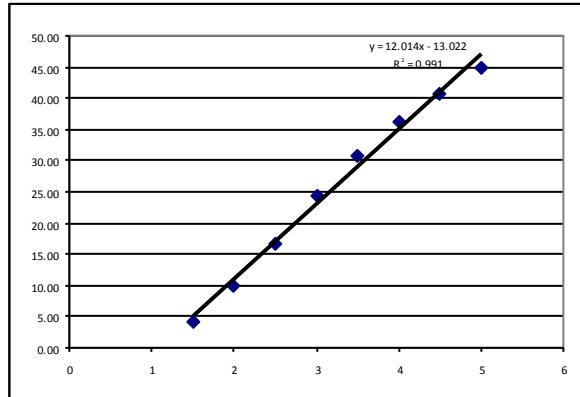
2	16823	2442	12.02
2	17199	2507	12.07
2	17521	2517	11.89
2.5	20646	5018	20.12
2.5	21106	5219	20.47
2.5	21815	4855	18.42
3	16178	4693	24.01
3	16148	4266	21.87
3	16426	4859	24.49
3.5	18989	7270	31.69
3.5	19609	7217	30.47
3.5	19959	7286	30.22
4	17076	7935	38.47
4	17248	7541	36.19
4	17194	7656	36.86
4.5	26131	13187	41.78
4.5	25624	12962	41.87
4.5	25718	12976	41.77
5	18615	11127	49.48
5	19403	11412	48.69
5	19410	11228	47.89



rate = 0.102 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
1.5	19405	834	4.63
1.5	19056	702	3.96
1.5	19338	732	4.07
2	23733	2225	10.09
2	25013	2254	9.70
2	25287	2265	9.64
2.5	20103	3288	17.60
2.5	20421	3015	15.89
2.5	20717	3254	16.90
3	22455	5083	24.36
3	21414	4910	24.67
3	22365	5096	24.52
3.5	25151	7247	31.01
3.5	25051	7178	30.84
3.5	25210	7218	30.81
4	21658	7288	36.21
4	21681	7364	36.55
4	21517	7207	36.05
4.5	23738	8997	40.79
4.5	23577	8931	40.76
4.5	23538	8887	40.63
5	25446	10546	44.60
5	25073	10460	44.89
5	25087	10584	45.40

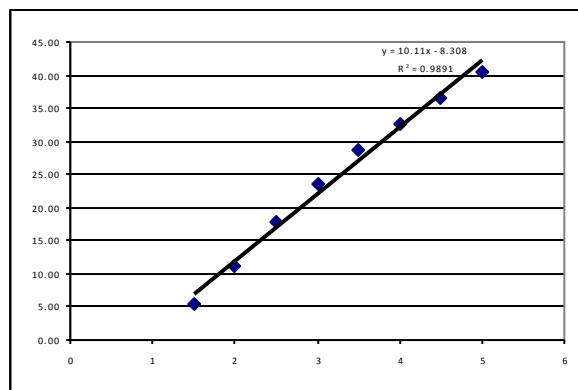


rate = 0.100 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

1.5	20061	1329	5.48
1.5	19989	1266	5.24
1.5	20464	1306	5.28
2	22297	3048	11.32
2	22875	3039	11.00
2	22725	3079	11.22
2.5	26268	5658	17.83
2.5	26051	5496	17.46
2.5	27117	5807	17.73
3	26012	7183	22.86
3	24908	7108	23.62
3	26016	7466	23.76
3.5	18417	6323	28.42
3.5	18223	6351	28.85
3.5	18161	6403	29.19
4	22283	8855	32.90
4	22693	8918	32.53
4	22875	9078	32.85
4.5	20533	8915	35.94
4.5	20902	9362	37.08
4.5	20522	9188	37.06
5	17621	8771	41.20
5	18462	9031	40.49
5	19003	9161	39.91



$$\text{rate} = 0.0843 \text{ mM/sec}$$

$$\text{avg. rate} = 9.54 \times 10^{-2} \text{ mM/s}$$

Order in $\text{Cs}^+ \mathbf{5}^-$ with APC/Ph₃P(O) (15 mM) (Table 4, entry 7)

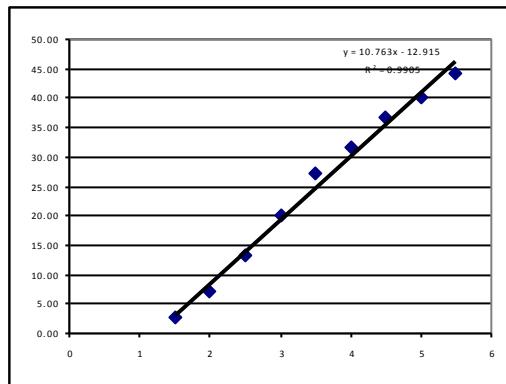
$\text{Cs}^+ \mathbf{5}^-$	62.5 mM
2-Bromotoluene	83 mM
APC	3.75 mM
Ph ₃ P(O)	15.0 mM

Following the General Procedure III, a mixture of silanolate $\text{Cs}^+ \mathbf{5}^-$ (39 mg, 0.125 mmol, 0.75 equiv), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), Ph₃P(O) (8.3 mg, 0.030 mmol, 0.18 equiv), naphthalene (10 mg), and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the

mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μmol
1.5	105410	3236	2.54
1.5	109484	3472	2.63
1.5	110949	3605	2.69
2	59688	5262	7.30
2	80114	7009	7.24
2	79382	6932	7.23
2.5	69362	11158	13.32
2.5	36779	6023	13.56
2.5	52724	8459	13.28
3	15597	3667	19.46
3	43519	10684	20.32
3	43546	10596	20.14
3.5	31772	10117	26.36
3.5	62370	20912	27.76
3.5	31866	10404	27.03
4	41975	15736	31.03
4	55700	21238	31.56
4	42029	16316	32.14
4.5	78358	35582	37.59
4.5	55944	24662	36.49
4.5	38791	16799	35.85
5	20176	9827	40.32
5	23415	11079	39.17
5	64196	31854	41.08
5.5	23730	12428	43.35
5.5	21388	11363	43.98
5.5	22715	12350	45.01



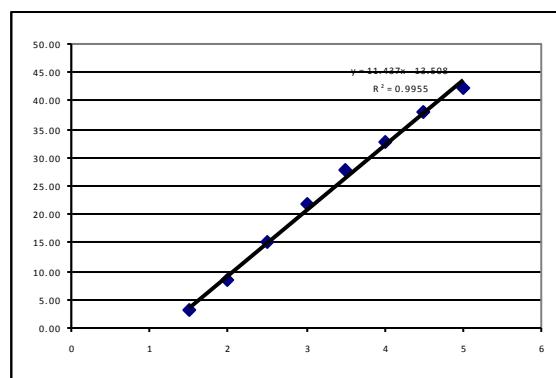
rate = 0.0897 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
1.5	54408	2006	3.36
1.5	73503	2559	3.17
1.5	73532	2514	3.11
2	67041	6084	8.26
2	65872	6174	8.53
2	67179	6115	8.29
2.5	70837	11730	15.08
2.5	67516	11157	15.05
2.5	71189	12086	15.46
3	68116	16222	21.69
3	68485	16142	21.46
3	67694	16301	21.93

Time	area, std	area, 12	conv, μmol
1.5	54408	2006	3.36
1.5	73503	2559	3.17
1.5	73532	2514	3.11
2	67041	6084	8.26
2	65872	6174	8.53
2	67179	6115	8.29
2.5	70837	11730	15.08
2.5	67516	11157	15.05
2.5	71189	12086	15.46
3	68116	16222	21.69
3	68485	16142	21.46
3	67694	16301	21.93

3.5	79310	23809	27.34
3.5	58461	17935	27.94
3.5	81849	24986	27.80
4	55807	20953	34.19
4	76660	26763	31.79
4	74217	26602	32.64
4.5	67046	27741	37.68
4.5	66874	27492	37.43
4.5	65015	28094	39.35
5	62238	28922	42.32
5	62444	28870	42.10
5	61915	28532	41.96
5.5	70446	35070	45.33
5.5	70756	34909	44.93
5.5	70895	35066	45.04

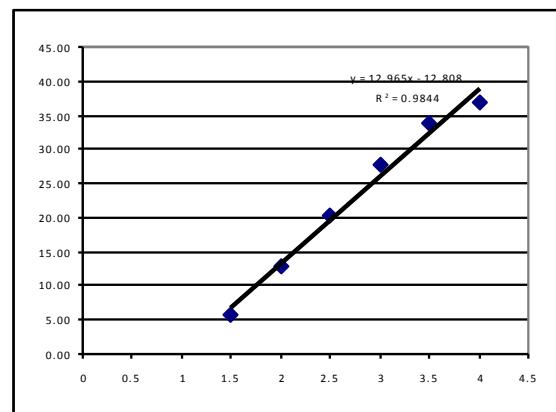


rate = 0.0953 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

1.5	66027	4372	5.48
1.5	66912	4544	5.62
1.5	65412	4477	5.67
2	57943	9067	12.95
2	55534	8545	12.74
2	54735	8291	12.54
2.5	59683	14676	20.36
2.5	15454	3703	19.84
2.5	15354	3820	20.60
3	64778	21539	27.52
3	64205	21435	27.64
3	65032	22238	28.31
3.5	51457	21059	33.88
3.5	52292	21289	33.70
3.5	52594	21661	34.09
4	62573	27695	36.64
4	61985	27454	36.66
4	61169	27339	37.00



rate = 0.108 mM/sec

avg. rate = 9.77×10^{-2} mM/s

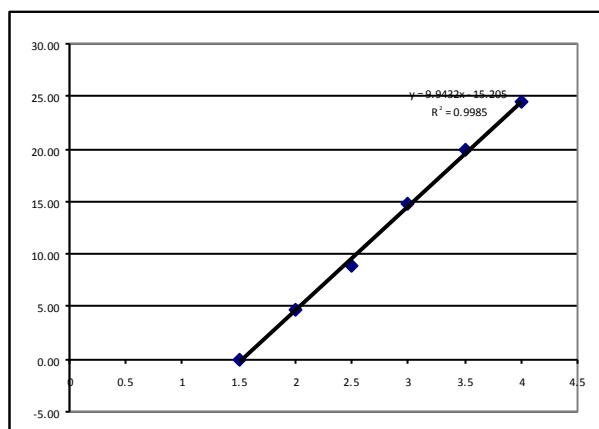
Table 4, entry 8

Cs ⁺ 5 ⁻	125 mM
2-Bromotoluene	83 mM
APC	3.75 mM
Ph ₃ P(O)	15.0 mM

Following the General Procedure III, a mixture of silanolate $\text{Cs}^+\text{5}^-$ (78 mg, 0.25 mmol), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), $\text{Ph}_3\text{P}(\text{O})$ (8.3 mg, 0.030 mmol, 0.18 equiv), naphthalene (10 mg), and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

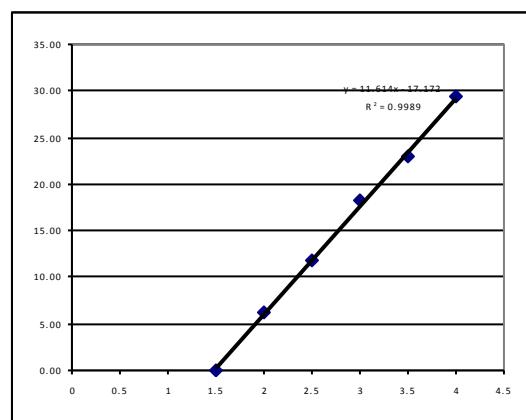
Time	area, std	area, 12	conv, μmol
1.5	1	0	0.00
1.5	1	0	0.00
1.5	1	0	0.00
2	79559	3945	4.52
2	27890	1500	4.90
2	79559	3945	4.52
2.5	20989	2069	8.98
2.5	21455	2118	8.99
2.5	20989	2069	8.98
3	25142	3989	14.45
3	24340	3930	14.70
3	23100	3914	15.43
3.5	27294	5868	19.58
3.5	26257	5937	20.59
3.5	28979	6168	19.38
4	34248	8984	23.89
4	32323	8924	25.14
4	33158	8913	24.48
4.5	33652	10794	29.21
4.5	35470	11077	28.44
4.5	37301	11512	28.10



$$\text{rate} = 0.0828 \text{ mM/sec}$$

Run 2

Time	area, std	area, 12	conv, μmol
1.5	1	0	0.00
1.5	1	0	0.00
1.5	1	0	0.00
2	29684	2264	6.31
2	29884	2249	6.23
2	30183	2306	6.32
2.5	25362	3603	11.76
2.5	24184	3509	12.01
2.5	26073	3648	11.58
3	16529	3567	17.86



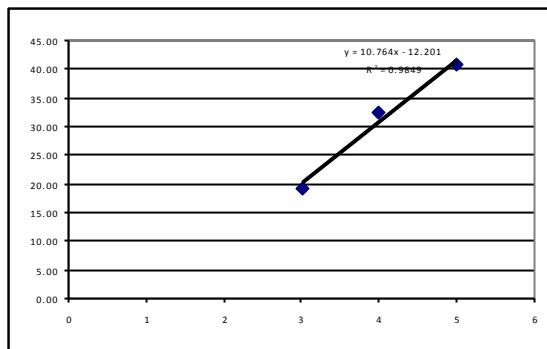
3	16055	3526	18.18
3	16289	3639	18.49
3.5	16848	4750	23.34
3.5	17485	4782	22.64
3.5	18200	5051	22.97
4	14734	5350	30.06
4	16846	5921	29.10
4	16547	5777	28.90
4.5	12763	5718	37.09
4.5	12719	5764	37.51
4.5	14570	6253	35.53

rate = 0.0968 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

3	16154	3749	19.21
3	16447	3831	19.28
3	16073	3781	19.47
4	16039	6059	31.27
4	13742	5617	33.84
4	47781	18511	32.07
5	12763	6310	40.93
5	13252	6447	40.27
5	39732	19848	41.35
6	12583	7090	46.64
6	53789	31032	47.76
6	12434	7102	47.28



rate = 0.0897 mM/sec

avg. rate = 8.98×10^{-2} mM/s

Table 4, entry 9

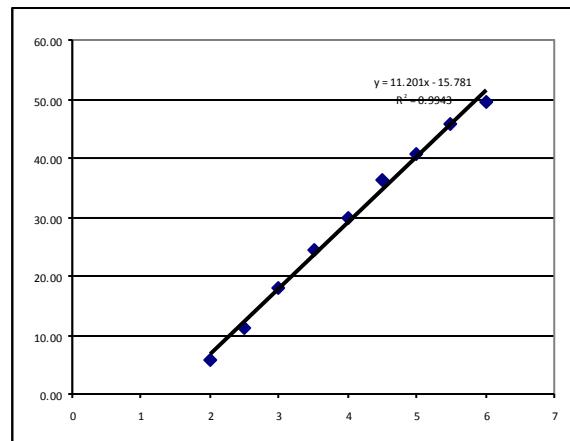
$\text{Cs}^+ \text{5}^-$	250 mM
2-Bromotoluene	83 mM
APC	3.75 mM
$\text{Ph}_3\text{P}(\text{O})$	15.0 mM

Following the General Procedure III, a mixture of silanolate $\text{Cs}^+ \text{5}^-$ (157 mg, 0.50 mmol), APC (2.7 mg, 0.0075 mmol, 0.045 equiv), $\text{Ph}_3\text{P}(\text{O})$ (8.3 mg, 0.030 mmol, 0.18 equiv), naphthalene (10 mg), and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

2	77524	5265	5.62
2	40009	2729	5.65
2	77524	5265	5.62
2.5	33247	4591	11.43
2.5	31590	4227	11.08
2.5	30299	4175	11.41
3	23363	5005	17.73
3	22495	5061	18.62
3	24141	5052	17.32
3.5	17884	5205	24.09
3.5	17613	5154	24.22
3.5	17639	5313	24.93
4	37263	13202	29.33
4	37162	13171	29.34
4	37578	13888	30.59
4.5	28395	12353	36.01
4.5	27569	12179	36.57
4.5	27878	12259	36.40
5	41637	20346	40.45
5	43205	21200	40.62
5	42773	21154	40.94
5.5	26892	14832	45.66
5.5	26887	15077	46.42
5.5	27348	15020	45.46
6	33779	20315	49.79
6	33799	20082	49.18
6	34110	20242	49.12

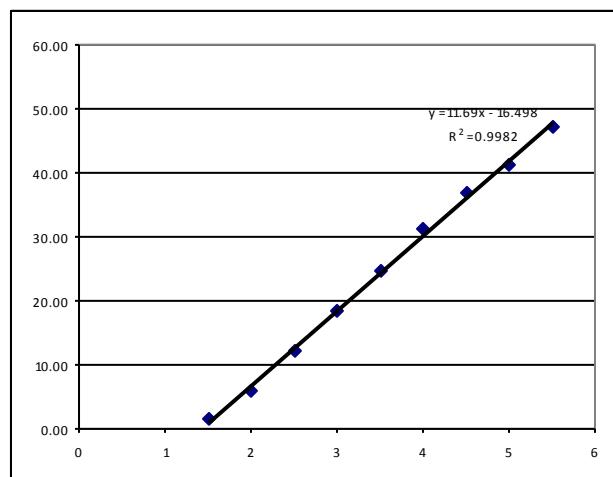


rate = 0.0974 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

1.5	72888	2292	2.34
1.5	97031	3216	2.47
1.5	1	0	0.00
2	83816	6642	5.90
2	84393	6633	5.86
2	22194	1797	6.03
2.5	24682	4093	12.35
2.5	69704	11178	11.95
2.5	23648	3950	12.44
3	19824	5032	18.91
3	20396	5021	18.34
3	40382	9917	18.30
3.5	24112	7897	24.40
3.5	24839	8305	24.91
3.5	23942	8043	25.03



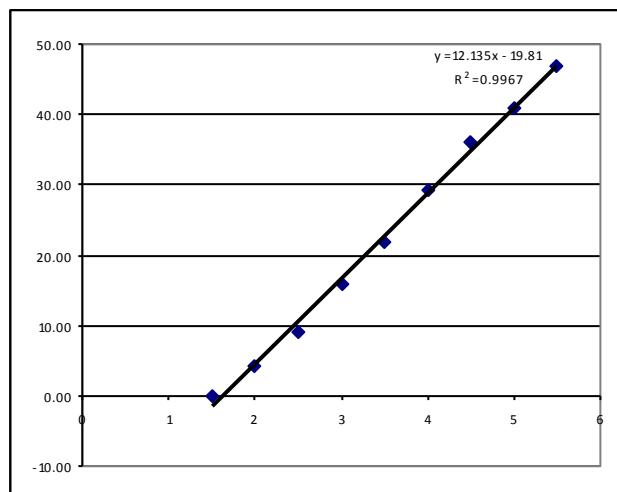
4	23375	9764	31.12
4	24278	10016	30.74
4	23086	9899	31.95
4.5	20600	10298	37.24
4.5	21464	10400	36.10
4.5	39720	19695	36.94
5	16690	9191	41.03
5	16573	9232	41.50
5	16406	9175	41.67
5.5	16469	10898	49.30
5.5	18577	11406	45.74
5.5	17264	10810	46.65

rate = 0.0933 mM/sec

Run 3

Time area, std area, **12** conv, μmol

1.5	1	0	0.00
1.5	1	0	0.00
1.5	1	0	0.00
2	41893	2352	4.18
2	42152	2360	4.17
2	40580	2408	4.42
2.5	21090	2582	9.12
2.5	20880	2600	9.28
2.5	21868	2580	8.79
3	19609	4262	16.19
3	21364	4559	15.90
3	17483	3654	15.57
3.5	19130	5446	21.21
3.5	67894	20401	22.39
3.5	17952	5374	22.30
4	21986	8476	28.72
4	84417	33195	29.30
4	86865	34398	29.50
4.5	56701	28222	37.08
4.5	15402	7221	34.93
4.5	58066	27910	35.81
5	77488	43122	41.46
5	78275	42802	40.74
5	78001	42494	40.59
5.5	67725	42013	46.22
5.5	16899	10375	45.74
5.5	48791	31627	48.29



rate = 0.101mM/sec

avg. rate = $9.72 \times 10^{-2} \text{ mM/s}$

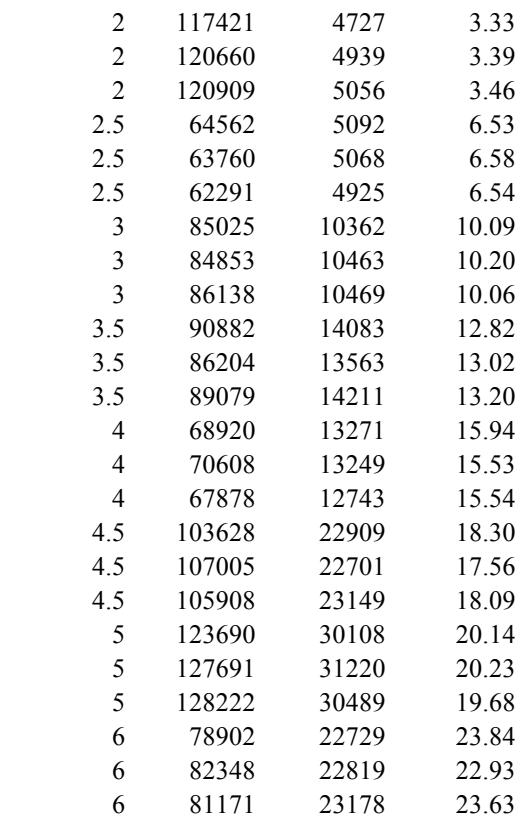
Order in Pd with APC/dPPP(O)₂ (Table 4, entry 10)

Cs ⁺ 5 ⁻	125 mM
2-Bromotoluene	83 mM
APC	1.875 mM
dPPP(O) ₂	7.5 mM

Run 1

Following the General Procedure III, a mixture of silanolate Cs⁺**5**⁻ (78 mg, 0.250 mmol, 1.5 equiv), APC (1.3 mg, 0.00375 mmol, 0.0225 equiv), dPPP(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (10 mg), and 2-bromotoluene (20 μ L, 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Time	area, std	area, 12	conv, μ mol
2	117421	4727	3.33
2	120660	4939	3.39
2	120909	5056	3.46
2.5	64562	5092	6.53
2.5	63760	5068	6.58
2.5	62291	4925	6.54
3	85025	10362	10.09
3	84853	10463	10.20
3	86138	10469	10.06
3.5	90882	14083	12.82
3.5	86204	13563	13.02
3.5	89079	14211	13.20
4	68920	13271	15.94
4	70608	13249	15.53
4	67878	12743	15.54
4.5	103628	22909	18.30
4.5	107005	22701	17.56
4.5	105908	23149	18.09
5	123690	30108	20.14
5	127691	31220	20.23
5	128222	30489	19.68
6	78902	22729	23.84
6	82348	22819	22.93
6	81171	23178	23.63



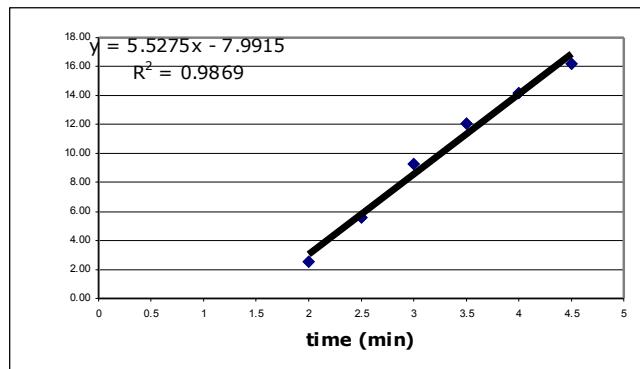
$$\text{rate} = 0.0424 \text{ mM/sec}$$

Run 2

Following the General Procedure III, a mixture of silanolate $\text{Cs}^+\text{5}^-$ (78 mg, 0.250 mmol, 1.5 equiv), APC (1.3 mg, 0.00375 mmol, 0.0225 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (8 mg), and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Time	area, std	area, 12	conv, μmol
2	84351	3219	2.53
2	84023	3252	2.56
2	83269	3193	2.54
2.5	60335	5106	5.60
2.5	59429	5101	5.68
2.5	60250	5019	5.52
3	59870	8416	9.31
3	59399	8340	9.30
3	59680	8302	9.21
3.5	84691	15580	12.18
3.5	84422	15303	12.00
3.5	82710	15021	12.02
4	154782	33515	14.34
4	157587	33048	13.88
4	158258	34137	14.28
4.5	74491	18459	16.41
4.5	74158	18352	16.38
4.5	74430	17738	15.78
5	57445	14862	17.13
5	57343	14749	17.03
5	58165	15021	17.10

Time	area, std	area, 12	conv, μmol
2	84351	3219	2.53
2	84023	3252	2.56
2	83269	3193	2.54
2.5	60335	5106	5.60
2.5	59429	5101	5.68
2.5	60250	5019	5.52
3	59870	8416	9.31
3	59399	8340	9.30
3	59680	8302	9.21
3.5	84691	15580	12.18
3.5	84422	15303	12.00
3.5	82710	15021	12.02
4	154782	33515	14.34
4	157587	33048	13.88
4	158258	34137	14.28
4.5	74491	18459	16.41
4.5	74158	18352	16.38
4.5	74430	17738	15.78
5	57445	14862	17.13
5	57343	14749	17.03
5	58165	15021	17.10



$$\text{rate} = 0.0460 \text{ mM/sec}$$

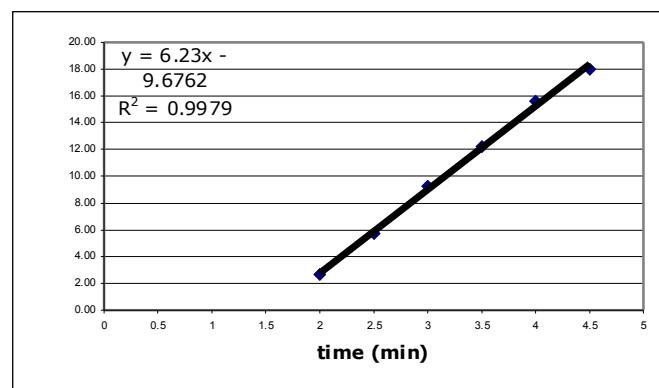
Run 3

Following the General Procedure III, a mixture of silanolate $\text{Cs}^+\text{5}^-$ (78 mg, 0.250 mmol, 1.5 equiv), APC (1.3 mg, 0.00375 mmol, 0.0225 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (8 mg), and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

2	21995	898	2.70
2	31435	1235	2.60

2	30740	1241	2.67
2.5	42915	3732	5.76
2.5	47683	4194	5.82
2.5	45901	3951	5.70
3	36456	5128	9.31
3	21505	3068	9.45
3	37688	5125	9.00
3.5	31164	5708	12.13
3.5	31931	5917	12.27
3.5	24672	4532	12.16
4	23559	5446	15.30
4	25441	6088	15.84
4	15749	3717	15.63
4.5	20739	5604	17.89
4.5	22753	6174	17.97
4.5	13837	3779	18.08
5	23279	7080	20.14
5	28262	8573	20.08
5	27780	8452	20.14



$$\text{rate} = 0.0519 \text{ mM/sec}$$

$$\text{avg. rate} = 4.68 \times 10^{-2} \text{ mM/s}$$

Table 4, entry 11

Cs ⁺ 5 ⁻	125 mM
2-Bromotoluene	83 mM
APC	5.625 mM
dppp(O) ₂	7.5 mM

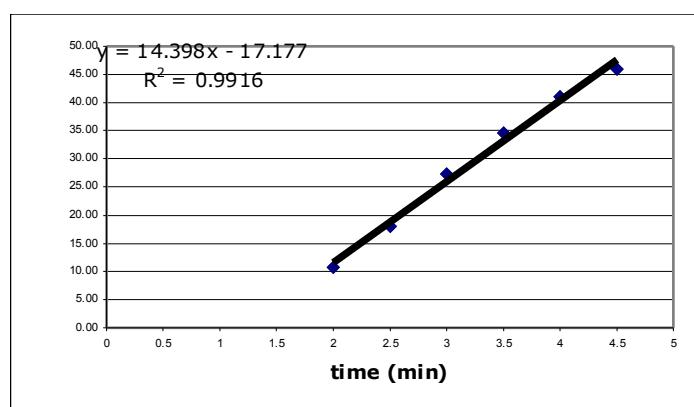
Run 1

Following the General Procedure III, a mixture of silanolate Cs⁺**5**⁻ (78 mg, 0.250 mmol, 1.5 equiv), APC (4.0 mg, 0.01125 mmol, 0.0675 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (13 mg), and 2-bromotoluene (20 μ L, 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

2	29900	3012	10.79
2	31930	3207	10.76

2	20250	2027	10.73
2.5	24513	4125	18.03
2.5	17661	2932	17.79
2.5	20462	3533	18.50
3	33106	8415	27.24
3	31366	7928	27.09
3	31094	8003	27.58
3.5	32722	10585	34.66
3.5	32286	10394	34.50
3.5	20392	6572	34.54
4	29701	11342	40.92
4	29485	11303	41.08
4	18248	6968	40.92
4.5	36174	15585	46.17
4.5	36867	15698	45.63
4.5	36392	15672	46.15
5	23419	10715	49.03
5	13523	6097	48.31
5	13198	5966	48.44



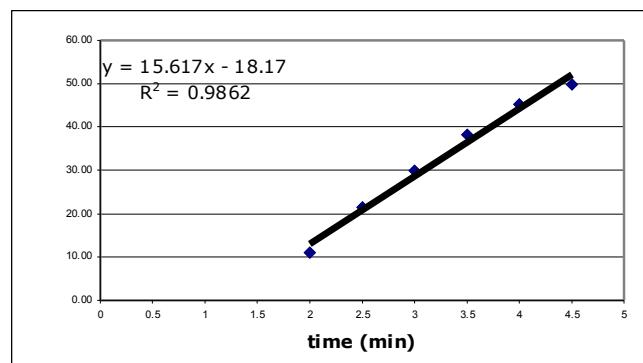
rate = 0.120 mM/sec

Run 2

Following the General Procedure III, a mixture of silanolate $\text{Cs}^+ \text{5}^-$ (78 mg, 0.250 mmol, 1.5 equiv), APC (4.0 mg, 0.01125 mmol, 0.0675 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (13 mg), and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Time area, std area, **12** conv, μmol

2	16137	1695	11.30
2	15140	1572	11.17
2	27750	2748	10.66
2.5	32610	6618	21.84
2.5	20900	4153	21.38
2.5	35367	7076	21.53
3	19796	5459	29.68
3	20488	5745	30.18
3	17231	4834	30.19
3.5	13030	4595	37.95
3.5	11177	3970	38.22
3.5	21990	7932	38.82
4	13173	5515	45.05
4	10985	4608	45.14
4	12600	5364	45.81
4.5	13008	6023	49.83
4.5	11263	5297	50.61
4.5	12709	5860	49.62



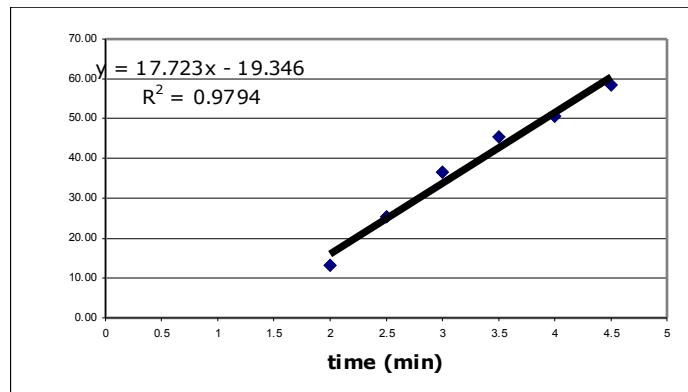
5	9175	4438	52.05
5	16610	8409	54.48
5	9464	4841	55.05

rate = 0.130 mM/sec

Run 3

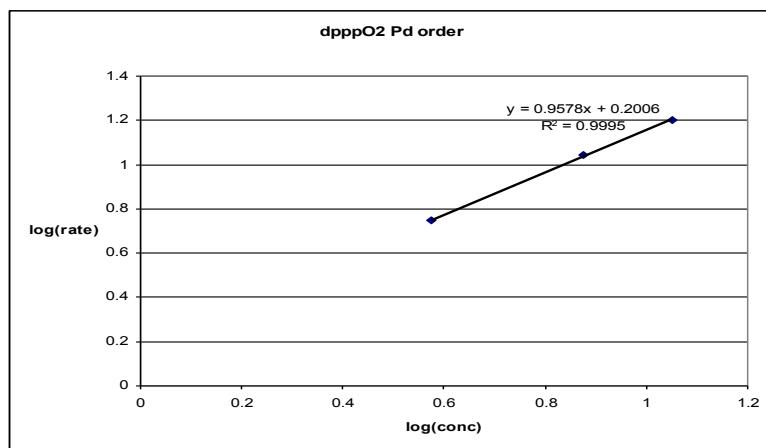
Following the General Procedure III, a mixture of silanolate $\text{Cs}^+ \text{5}^-$ (78 mg, 0.250 mmol, 1.5 equiv), APC (4.0 mg, 0.01125 mmol, 0.0675 equiv), dppp(O)₂ (6.6 mg, 0.015 mmol, 0.090 equiv), naphthalene (13 mg), and 2-bromotoluene (20 μL , 0.16 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Time	area, std	area, 12	conv, μmol
2	23509	2970	13.54
2	21619	2627	13.02
2	19721	2380	12.93
2.5	17261	4058	25.19
2.5	11917	2874	25.84
2.5	11538	2740	25.45
3	11917	4030	36.24
3	19322	6536	36.25
3	18870	6510	36.97
3.5	10893	4598	45.23
3.5	17178	7156	44.64
3.5	10170	4414	46.51
4	8982	4273	50.98
4	8524	3990	50.16
4	10521	4961	50.53
4.5	9121	4916	57.76
4.5	14204	7789	58.76
4.5	8402	4592	58.57
5	15002	9033	64.52
5	7814	4659	63.89
5	6775	3835	60.66



rate = 0.148 mM/sec

avg. rate = 13.27×10^{-2} mM/s

Determination of Rate Order for Pd

KINETICS STUDY FOR CATALYTIC CROSS-COUPLING EMPLOYING **13p AS CATALYST (Table 5)**

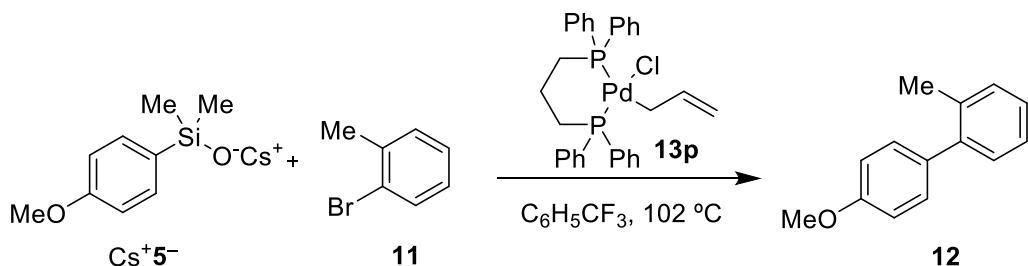


Table 5, entry 1

Cs⁺5⁻ 125 mM

2-Bromotoluene 83 mM

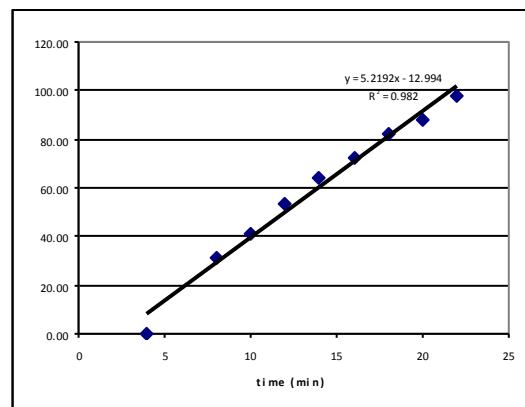
13p 7.5 mM

Following General Procedure III, a mixture of silanolate $\text{Cs}^+ \text{5}^-$ (157 mg, 0.5 mmol, 1.5 equiv), **13p** (17.8 mg, 0.030 mmol, 0.090 equiv), naphthalene (30 mg), and 2-bromotoluene (40 μL , 0.33 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12 conv, μmol
4	1	0.00
4	1	0.00
4	1	0.00
8	18490	2337
8	17630	2294
8	18265	2292
10	21638	3550
10	21256	3596
10	21326	3551
12	18452	3891
12	17113	3682
12	18246	3951
14	17672	4550
14	18105	4647
14	18233	4670
16	16226	4765
16	16497	4732
		72.93
		71.23

16	16086	4781	73.81
18	19848	6483	81.12
18	19963	6720	83.60
18	20677	6806	81.74
20	19753	6832	85.89
20	20282	7015	85.89
20	18993	7029	91.91
22	16392	6721	101.82
22	18035	6868	94.57
22	17602	6828	96.33

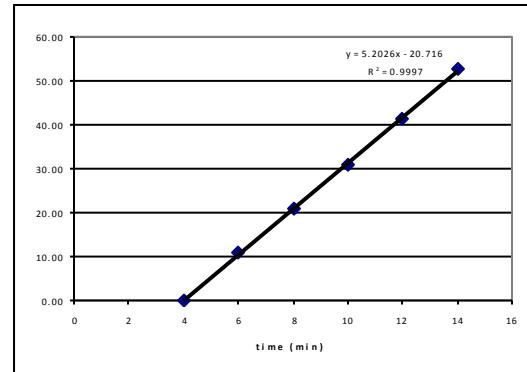


rate = 0.0217 mM/s

Run 2

Time area, std area, **12** conv, μmol

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
6	59105	2598	10.92
6	59105	2598	10.92
6	59105	2598	10.92
8	21886	1852	21.01
8	22582	1838	20.21
8	21493	1839	21.25
10	60569	7368	30.21
10	20146	2525	31.13
10	20492	2610	31.63
12	20602	3458	41.68
12	60751	10179	41.61
12	21260	3504	40.93
14	22089	4647	52.25
14	21337	4430	51.56
14	20757	4490	53.72



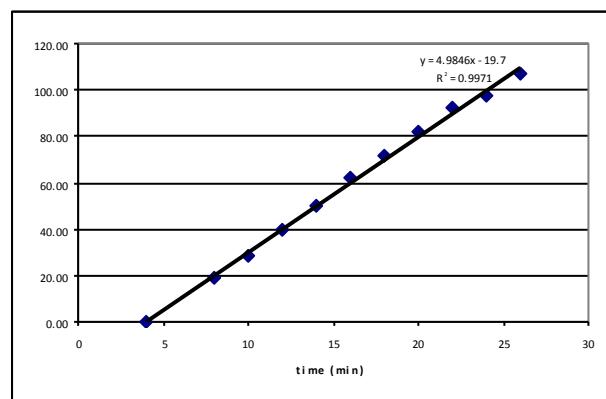
rate = 0.0217 mM/s

Run 3

Time area, std area, **12** conv, μmol

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
8	17584	1216	18.32
8	18051	1264	18.55
8	17510	1247	18.87
10	17553	1898	28.64
10	17649	1914	28.73

10	17838	1951	28.97
12	19626	3021	40.78
12	19266	2886	39.68
12	20431	3064	39.73
14	20130	3861	50.81
14	21517	4203	51.74
14	21894	4036	48.83
16	17698	4165	62.34
16	18026	4291	63.06
16	18328	4278	61.83
18	19332	5300	72.62
18	18904	5129	71.87
18	20284	5368	70.10
20	19169	6107	84.39
20	19076	5848	81.21
20	20324	6204	80.86
22	19894	7008	93.31
22	20200	6740	88.39
22	19071	6789	94.30
24	21531	8266	101.70
24	22155	8100	96.85
24	22367	7947	94.12
26	18611	7624	108.52
26	17982	7392	108.89
26	20187	7912	103.82



rate = 0.0208 mM/s

avg. rate = 2.14×10^{-2} mM/s

Table 5, entry 2

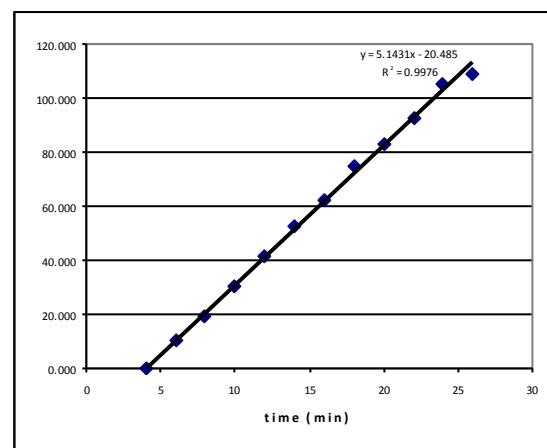
Cs ⁺ 5 ⁻	250 mM
2-Bromotoluene	83 mM
13p	7.5 mM

Following the General Procedure III, a mixture of silanolate Cs⁺**5**⁻ (157 mg, 0.5 mmol, 3.0 equiv), **13p** (8.9 mg, 0.015 mmol, 0.09 equiv), naphthalene (20 mg), and 2-bromotoluene (20 μ L, 0.165 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μ mol
4	1	0	0.000
4	1	0	0.000

4	1	0	0.000
6	71792	4368	10.073
6	23811	1513	10.520
6	71792	4368	10.073
8	21593	2485	19.053
8	41199	4800	19.289
8	20477	2280	18.434
10	20261	3933	32.138
10	22013	4017	30.212
10	22731	3977	28.967
12	21634	5350	40.943
12	21301	5323	41.373
12	20941	5267	41.641
14	20378	6436	52.289
14	20084	6434	53.038
14	20448	6342	51.349
16	20730	7625	60.898
16	19719	7579	63.634
16	60491	22648	61.987
18	19098	8753	75.880
18	20003	8896	73.631
18	19848	8964	74.773
20	22157	10669	79.721
20	20860	10270	81.511
20	19605	10463	88.359
22	19430	10461	89.137
22	18468	10198	91.423
22	17190	10130	97.565
24	19248	11421	98.238
24	16403	11240	113.449
24	18173	11398	103.839
26	24665	15513	104.129
26	22319	14762	109.504
26	21984	14971	112.747

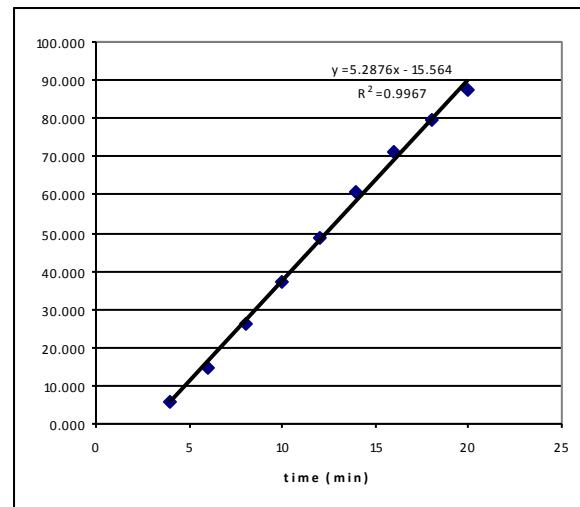


rate = 0.0428mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
4	69333	2524	6.027
4	68716	2316	5.580
4	69019	2365	5.673
6	80187	7056	14.568
6	80633	7082	14.541
6	81031	7097	14.500
8	86604	13659	26.112
8	86955	13711	26.106
8	86936	13715	26.119
10	84746	19112	37.338
10	84239	19101	37.541

10	85286	19143	37.161
12	66306	19254	48.076
12	65443	19332	48.907
12	65808	19267	48.472
14	71468	26065	60.382
14	72004	26270	60.404
14	70381	26147	61.507
16	85199	36446	70.823
16	85120	36863	71.700
16	85462	36644	70.989
18	70584	33964	79.666
18	69372	33486	79.917
18	70209	33585	79.198
20	69459	36570	87.168
20	69805	36640	86.902
20	68859	36419	87.564

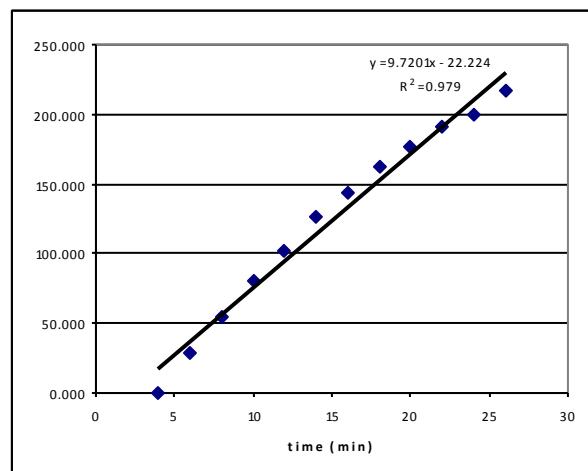


$$\text{rate} = 0.0440 \text{ mM/sec}$$

Run 3

The following experiment was run on 4-mL scale.

Time	area, std	area, 12	conv, μmol
4	1	0	0.0000
4	1	0	0.0000
4	1	0	0.0000
6	23725	2780	29.0997
6	26188	2897	27.4724
6	49079	5442	27.5368
8	80720	17163	52.8035
8	27036	5725	52.5876
8	24727	5769	57.9401
10	21969	7324	82.7920
10	23559	7348	77.4573
10	21893	7323	83.0680
12	24522	10116	102.4479
12	73730	30164	101.6003
12	48306	20032	102.9849
14	49763	25673	128.1210
14	25630	13108	127.0100
14	26264	13157	124.4074
16	24301	13941	142.4689
16	23826	13625	142.0155
16	46992	27652	146.1345
18	22282	14850	165.5094
18	22819	14869	161.8212
18	68825	44438	160.3462
20	24707	17578	176.6850
20	24882	17659	176.2508
20	48088	34487	178.1021



22	21408	16163	187.4978
22	39795	31589	197.1322
22	63752	48737	189.8520
24	24484	19503	197.8196
24	23741	19351	202.4206
24	24119	19271	198.4245
26	17978	16149	223.0769
26	18905	16418	215.6720
26	19257	16304	210.2596

rate = 0.0405mM/sec

avg. rate = 4.24×10^{-2} mM/s

Table 5, entry 3

Cs⁺5⁻ 500 mM

2-Bromotoluene 83 mM

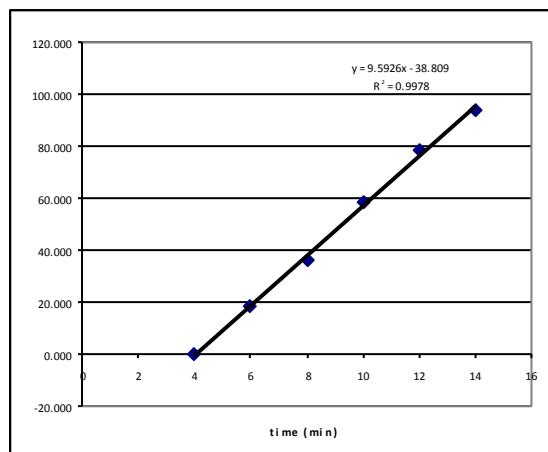
13p 7.5 mM

Following the General Procedure III, a mixture of silanolate Cs⁺5⁻ (314 mg, 1.00 mmol, 6.0 equiv), **13p** (8.9 mg, 0.015 mmol, 0.09 equiv), naphthalene (20 mg), and 2-bromotoluene (20 μ L, 0.165 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μ mol
------	-----------	-----------------	-----------------

4	1	0	0.000
4	1	0	0.000
4	1	0	0.000
6	27000	2973	18.230
6	27208	3008	18.304
6	83854	8999	17.768
8	25679	5652	36.440
8	25113	5500	36.260
8	25358	5526	36.079
10	21088	7494	58.835
10	44260	15460	57.831
10	21770	7833	59.570
12	25339	11923	77.903
12	25298	12059	78.919
12	25123	11915	78.520
14	24430	13402	90.825
14	22134	13453	100.628



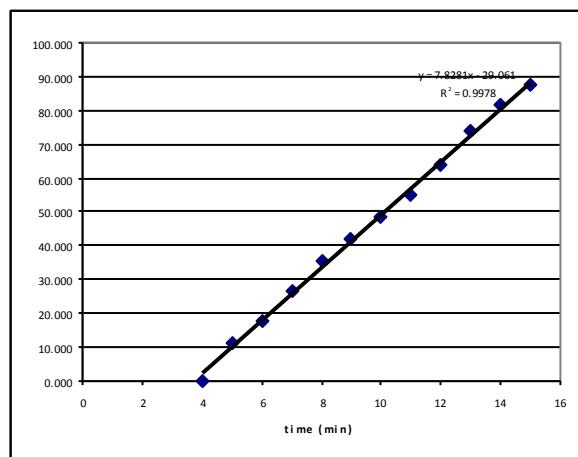
14	25183	13586	89.319
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rate = 0.0799mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	1	0	0.000
4	1	0	0.000
4	1	0	0.000
5	28589	1847	10.696
5	27608	1890	11.334
5	27011	1870	11.462
6	24997	2666	17.658
6	25433	2705	17.609
6	23937	2641	18.267
7	26362	4428	27.809
7	28165	4436	26.076
7	29537	4536	25.425
8	23370	5224	37.009
8	24602	5059	34.045
8	23599	5166	36.243
9	110358	28212	42.324
9	37001	9382	41.980
9	37888	9504	41.530
10	89093	26423	49.102
10	89093	26423	49.102
10	6464	1870	47.896
11	70699	24069	56.364
11	70442	23350	54.880
11	106538	34983	54.364
12	81883	31463	63.616
12	82641	32033	64.174
12	81324	31414	63.953
13	76733	33901	73.146
13	77632	34516	73.610
13	51955	23394	74.548
14	83445	40740	80.831
14	82683	40938	81.973
14	84233	41408	81.388
15	81707	43132	87.398
15	81325	43097	87.737
15	81967	43296	87.452



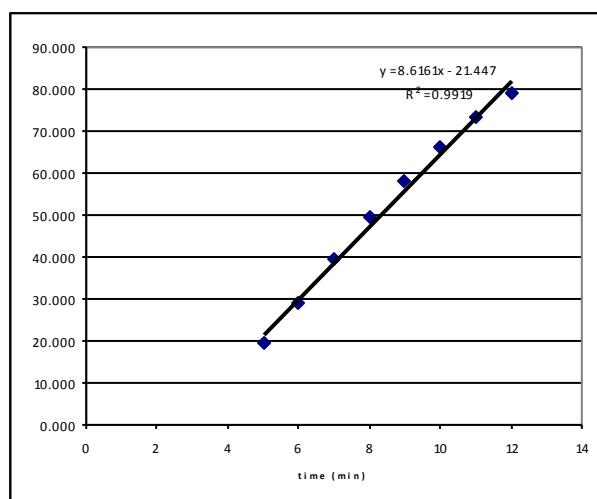
rate = 0.0652mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

5	81116	9674	19.745
5	80889	9595	19.639
5	27440	3264	19.694

6	62869	11063	29.134
6	65573	11402	28.788
6	64751	11462	29.307
7	62006	14973	39.979
7	61578	14844	39.910
7	21474	5019	38.696
8	62712	18955	50.042
8	63437	18775	49.000
8	62921	18928	49.804
9	66144	23004	57.580
9	64996	22999	58.584
9	66433	23232	57.898
10	79458	31635	65.916
10	78068	31443	66.682
10	78489	31438	66.314
11	64921	28745	73.305
11	65823	29373	73.880
11	66337	28962	72.282
12	76937	36014	77.499
12	76405	36659	79.436
12	75792	36548	79.836



rate = 0.0718 mM/sec

avg. rate = 7.23×10^{-2} mM/s

Table 5, entry 4

Cs⁺**5**⁻ 250 mM

2-Bromotoluene 42 mM

13p 7.5 mM

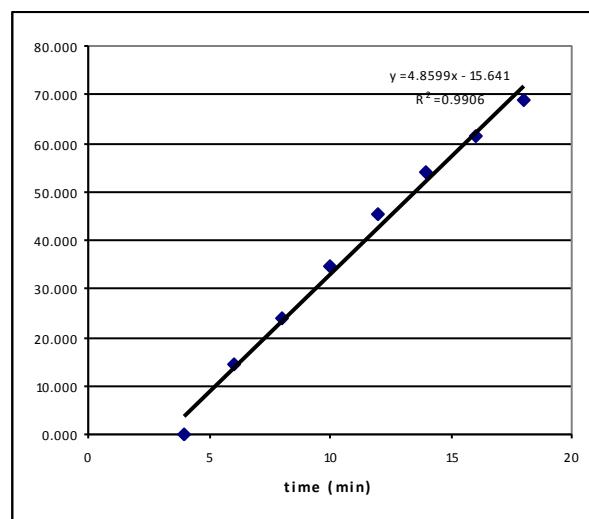
Following the General Procedure III, a mixture of silanolate Cs⁺**5**⁻ (157 mg, 0.5 mmol, 6.0 equiv), **13p** (8.9 mg, 0.015 mmol, 0.18 equiv), naphthalene (20 mg), and 2-bromotoluene (10 μ L, 0.083 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time area, std area, **12** conv, μ mol

4	1	0	0.000
4	1	0	0.000
4	1	0	0.000
6	20804	1824	14.516
6	21622	1767	13.530
6	19219	1784	15.368

8	19327	2883	24.697
8	20651	2951	23.658
8	20651	2951	23.658
10	18777	3815	33.638
10	17560	3808	35.903
10	18331	3843	34.709
12	20748	5711	45.572
12	21150	5850	45.794
12	21725	5797	44.178
14	22320	7291	54.082
14	22759	7353	53.490
14	22161	7224	53.969
16	19266	7155	61.486
16	19320	7142	61.203
16	60491	22648	61.987
18	24273	10059	68.610
18	23272	9671	68.801
18	23441	9735	68.757

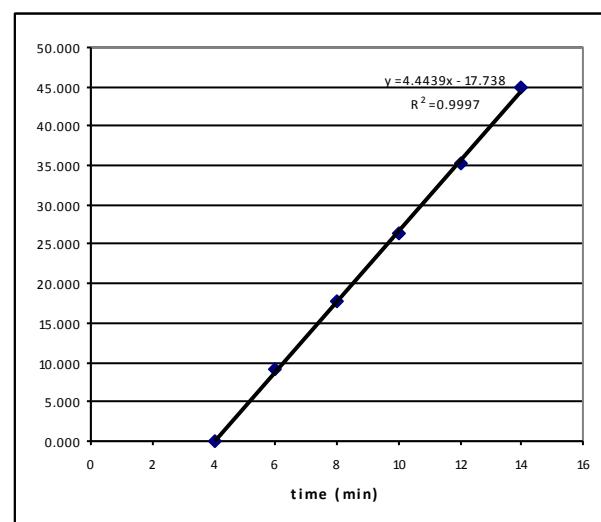


rate = 0.0404mM/sec

Run 2

Time	area, std	area, 22	conv, μmol
------	-----------	----------	-----------------------

4	1	0	0.000
4	1	0	0.000
4	1	0	0.000
6	28488	1598	9.287
6	29122	1601	9.102
6	29436	1633	9.185
8	23457	2681	18.923
8	26700	2787	17.282
8	26638	2766	17.191
10	18932	3004	26.270
10	19140	3177	27.481
10	59077	9153	25.651
12	26697	5592	34.679
12	26428	5849	36.642
12	28159	5845	34.366
14	23842	6719	46.657
14	25507	6656	43.203
14	24744	6683	44.716



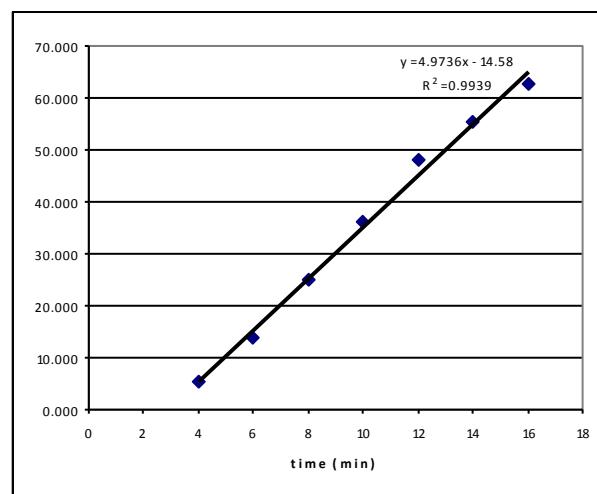
rate = 0.0370mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	109341	3403	5.153
4	110412	3459	5.187
4	109974	3487	5.250

6	81721	6621	13.414
6	27468	2262	13.634
6	25362	2163	14.120
8	55925	8424	24.939
8	54871	8265	24.938
8	52712	8142	25.573
10	85763	18688	36.076
10	29334	6354	35.862
10	86663	18912	36.130
12	52735	14896	46.766
12	25539	7523	48.769
12	25175	7375	48.501
14	81858	26918	54.443
14	80594	26759	54.970
14	51746	17632	56.414
16	24660	9262	62.183
16	75462	28746	63.068
16	76149	28926	62.890



$$\text{rate} = 0.0414 \text{ mM/sec}$$

$$\text{avg. rate} = 3.96 \times 10^{-2} \text{ mM/s}$$

Table 5, entry 5

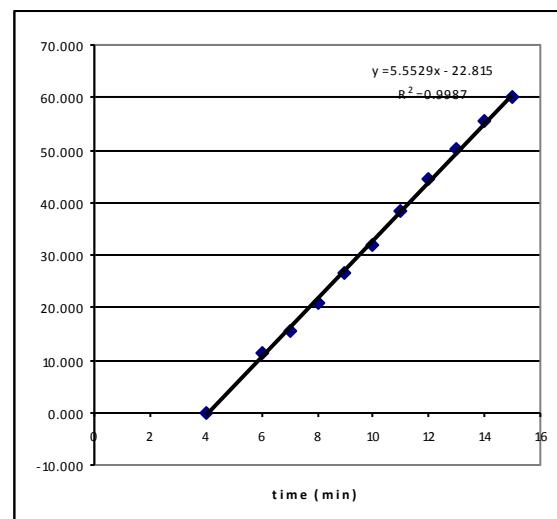
Cs ⁺ 5 ⁻	250 mM
2-Bromotoluene	167 mM
13p	7.5 mM

Following the General Procedure III, a mixture of silanolate Cs⁺5⁻ (157 mg, 0.5 mmol, 1.5 equiv), **13p** (8.9 mg, 0.015 mmol, 0.045 equiv), naphthalene (20 mg), and 2-bromotoluene (30 μ L, 0.33 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (2 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μmol
4	1	0	0.000
4	1	0	0.000
4	1	0	0.000
6	23052	1562	11.218
6	41547	2892	11.524
6	23052	1562	11.218

7	32819	3149	15.886
7	33630	3110	15.311
7	33668	3104	15.264
8	22289	2825	20.984
8	22709	2811	20.494
8	21979	2778	20.926
9	29192	4530	25.692
9	29366	4544	25.618
9	26565	4567	28.463
10	24633	4796	32.234
10	25102	4843	31.942
10	25776	4842	31.101
11	23817	5512	38.316
11	25099	5567	36.722
11	22710	5498	40.082
12	21768	5776	43.931
12	21317	5757	44.713
12	22010	5924	44.561
13	22132	6722	50.285
13	21882	6686	50.587
13	21957	6615	49.879
14	22268	7618	56.639
14	23367	7705	54.592
14	23605	7843	55.009
15	32159	11377	58.571
15	31871	11783	61.210
15	32005	11588	59.945

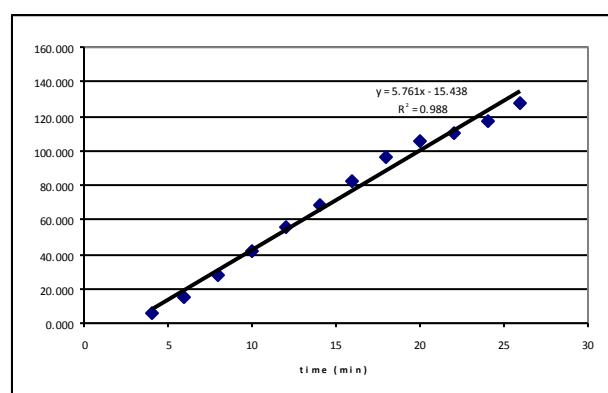


rate = 0.0463mM/sec

Run 2

Time	area, std	area, 12	conv, umol
------	-----------	----------	------------

4	114908	4146	5.974
4	117524	4183	5.893
4	69019	2365	5.673
6	31447	2709	14.262
6	60359	5292	14.516
6	57104	5137	14.894
8	50912	8641	28.100
8	27162	4599	28.032
8	52381	8644	27.321
10	27656	7053	42.222
10	87542	21709	41.057
10	28407	7266	42.348
12	71771	23913	55.163
12	70050	23776	56.194
12	108610	35714	54.441
14	57591	23306	67.000
14	37642	15553	68.407
14	38130	15822	68.699



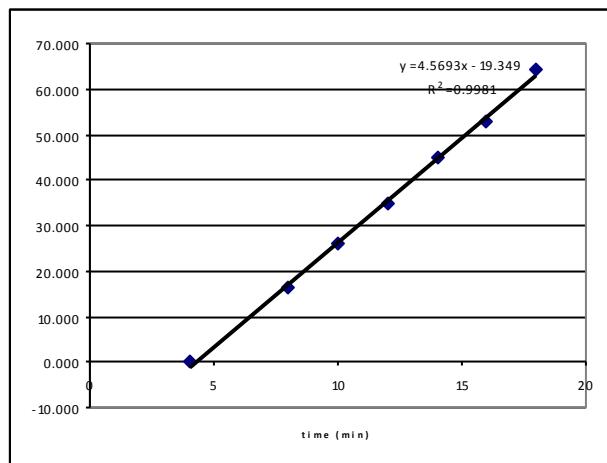
16	55365	27306	81.655
16	83031	40298	80.353
16	53149	27034	84.212
18	36762	21743	97.922
18	112486	63249	93.092
18	35780	20957	96.972
20	80587	50163	103.057
20	51165	33181	107.368
20	52487	33274	104.957
22	110745	71923	107.523
22	72892	48804	110.850
22	75554	50402	110.446
24	73933	51790	115.976
24	72631	51624	117.676
24	25090	18022	118.922
26	106379	80124	124.700
26	34658	28083	134.153
26	107205	81015	125.115

rate = 0.0480mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	1	0	0.000
4	1	0	0.000
4	1	0	0.000
8	31563	3057	16.035
8	30150	3087	16.952
8	30435	2948	16.037
10	26857	4217	25.996
10	25500	4225	27.431
10	28972	4388	25.075
12	27659	5788	34.646
12	28349	5852	34.176
12	26804	5818	35.936
14	28425	7646	44.534
14	26536	7499	46.787
14	29275	7599	42.975
16	27074	8667	53.000
16	27871	8849	52.565
16	27522	8716	52.432
18	29336	11855	66.905
18	30154	11293	62.005
18	30230	11724	64.209



rate = 0.0381mM/sec

avg. rate = 4.41×10^{-2} mM/s

Table 5, entry 6

Cs⁺5⁻ 125 mM

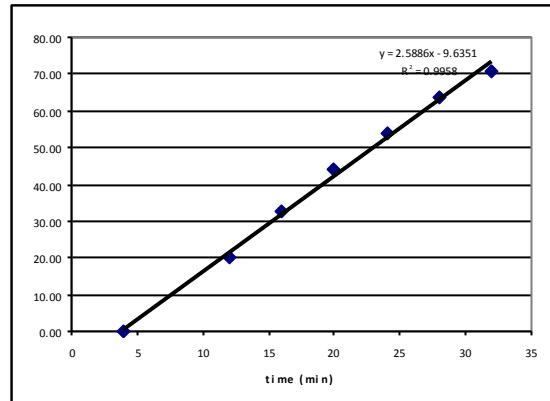
2-Bromotoluene 83 mM

13p 3.75 mM

Following the General Procedure III, a mixture of silanolate Cs⁺5⁻ (157 mg, 0.5 mmol, 1.5 equiv), **13p** (8.9 mg, 0.015 mmol, 0.045 equiv), naphthalene (30 mg), and 2-bromotoluene (40 μ L, 0.33 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μ mol
4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
12	66000	5223	19.65
12	23443	1933	20.48
12	66700	5298	19.73
16	35606	4693	32.73
16	35095	4638	32.82
16	69727	9063	32.28
20	23098	4117	44.26
20	22163	3940	44.15
20	69286	12224	43.81
24	69279	14998	53.76
24	68733	14942	53.99
24	45583	9865	53.75
28	57839	14370	61.70
28	19275	4944	63.70
28	19041	4979	64.94
32	18547	5320	71.23
32	19432	5486	70.11
32	36814	10482	70.71



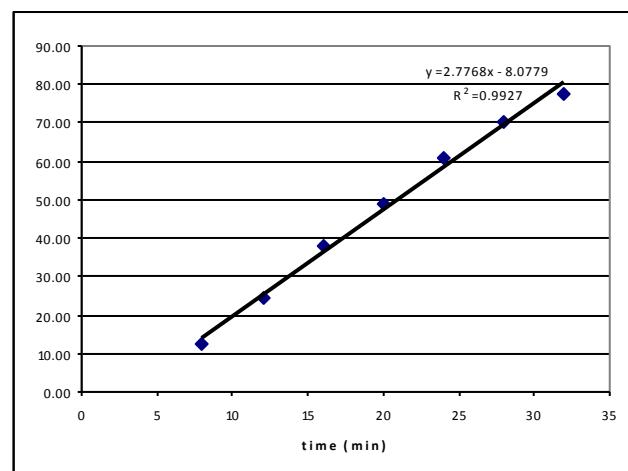
$$\text{rate} = 0.0108 \text{ mM/sec}$$

Run 2

Time	area, std	area, 12	conv, μ mol
------	-----------	-----------------	-----------------

8	47352	2334	12.24
8	47352	2334	12.24
8	47352	2334	12.24
12	45702	4240	23.04
12	15935	1592	24.81
12	16058	1629	25.19

16	56991	8473	36.92
16	38113	5912	38.52
16	57556	8802	37.98
20	18184	3441	46.99
20	17173	3428	49.57
20	16875	3419	50.32
24	16397	3971	60.14
24	15371	3873	62.57
24	16185	3941	60.47
28	16984	4914	71.85
28	18505	5162	69.28
28	17864	4995	69.44
32	15392	4926	79.48
32	16258	5031	76.85
32	16390	5046	76.46

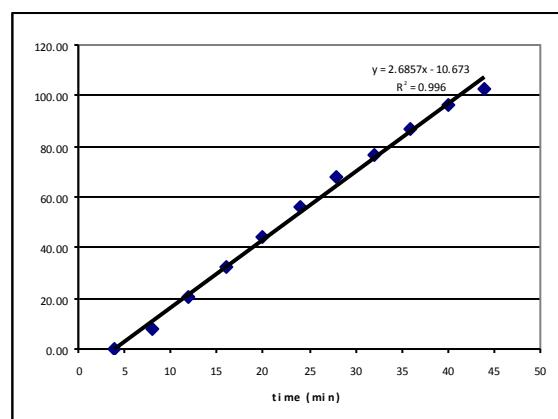


rate = 0.0116 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
8	26260	837	7.92
8	27112	833	7.63
8	28326	899	7.88
12	22079	1800	20.25
12	21845	1863	21.18
12	21764	1766	20.15
16	27837	3496	31.19
16	26248	3586	33.93
16	27872	3527	31.43
20	20852	3687	43.91
20	21060	3713	43.78
20	19427	3496	44.69
24	21226	4741	55.47
24	20013	4641	57.59
24	20628	4526	54.49
28	22412	6327	70.11
28	23740	6301	65.91
28	22789	6208	67.65
32	19803	5979	74.98
32	18854	5848	77.03
32	18288	5780	78.49
36	24216	8616	88.36
36	23980	8411	87.11
36	24873	8613	86.00
40	19193	7095	91.80
40	17719	7241	101.49
40	18765	7212	95.45



rate = 0.0112 mM/sec

avg. rate = 1.12×10^{-2} mM/s

Table 5, entry 7

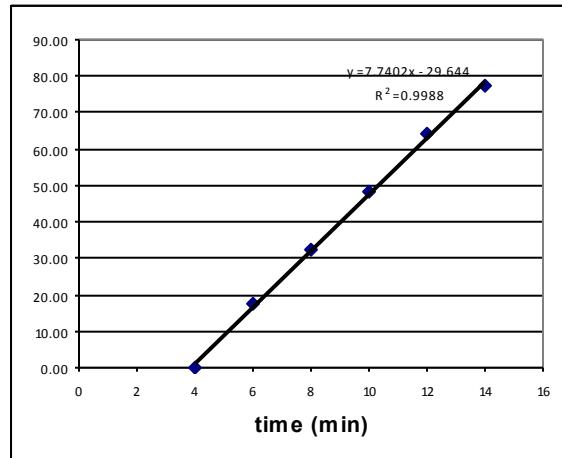
Cs⁺5⁻	125 mM
2-Bromotoluene	83 mM
13p	11.25 mM

Following the General Procedure III, a mixture of silanolate Cs⁺5⁻ (157 mg, 0.5 mmol, 1.5 equiv), **13p** (26.7 mg, 0.045 mmol, 0.135 equiv), naphthalene (30 mg), and 2-bromotoluene (40 μ L, 0.33 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μ mol
------	-----------	-----------------	-----------------

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
6	37067	2630	17.62
6	54886	3955	17.90
6	54886	3955	17.90
8	51368	6725	32.51
8	52487	6856	32.44
8	18021	2372	32.69
10	15923	3207	50.02
10	50250	9443	46.67
10	50719	9830	48.13
12	19392	5136	65.77
12	61708	15552	62.59
12	21074	5390	63.52
14	57702	17758	76.43
14	57886	17797	76.35
14	37153	11937	79.79



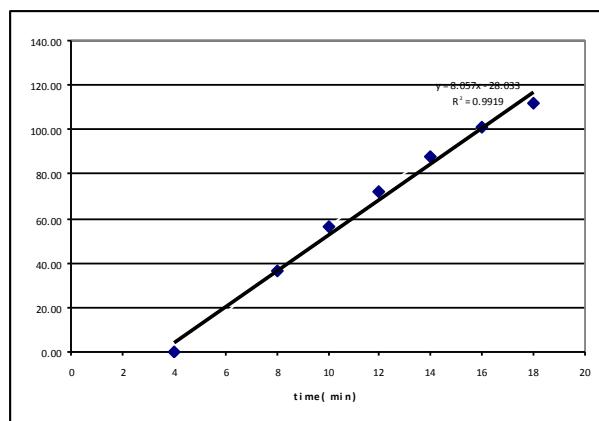
$$\text{rate} = 0.0323 \text{ mM/sec}$$

Run 2

Time	area, std	area, 12	conv, μ mol
------	-----------	-----------------	-----------------

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
8	15755	2307	36.36
8	16058	2320	35.88
8	29975	4511	37.37
10	15102	3578	58.84
10	51299	10912	52.83

10	16213	3699	56.66
12	16990	4939	72.19
12	52319	15117	71.76
12	18162	5204	71.16
14	18237	6322	86.09
14	17769	6305	88.12
14	17089	6082	88.39
16	17139	6918	100.24
16	16810	6947	102.63
16	17785	7107	99.24
18	54511	24372	111.03
18	18866	8622	113.50
18	19631	8779	111.06

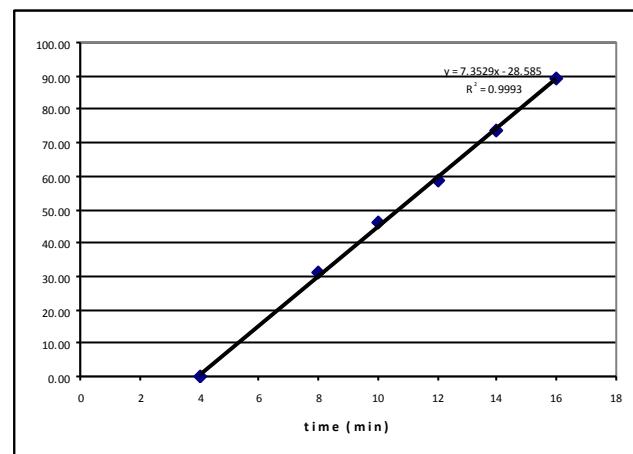


rate = 0.0336 mM/sec

Run 3

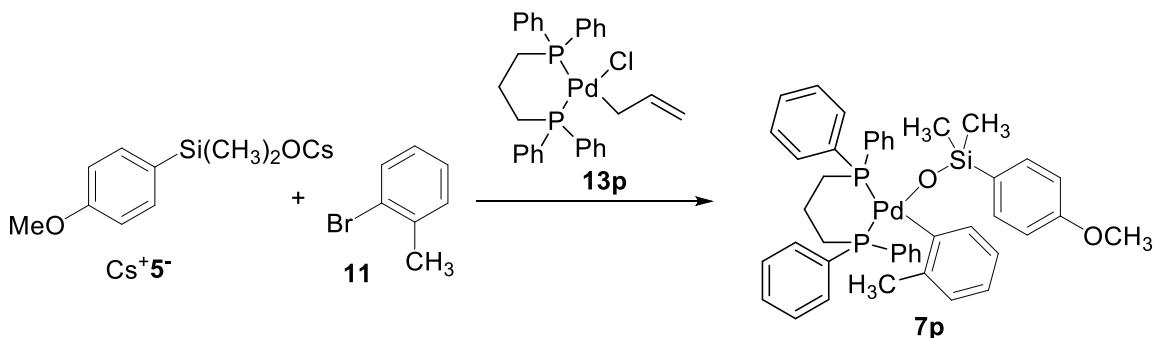
Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
8	17233	2180	31.42
8	16536	2027	30.44
8	15126	1905	31.28
10	16225	3026	46.32
10	16745	3091	45.84
10	16102	3009	46.41
12	17341	4164	59.63
12	18159	4221	57.73
12	35062	8411	59.57
14	17471	5302	75.37
14	17399	5278	75.33
14	17406	4936	70.42
16	15316	5470	88.69
16	15668	5643	89.44
16	15869	5709	89.34



rate = 0.0307 mM/sec

avg. rate = 3.22×10^{-2} mM/s

***In situ* detection of **7p** by ^{31}P NMR (Scheme 3)**

In a drybox, naphthalene (5 mg), **13p** (4.5 mg, 0.0075 mmol, 0.09 equiv), and 2-bromotoluene (10 μL , 0.082 mmol, 1.0 equiv) were dissolved in dry toluene-d⁸ (1.0 mL) at room temperature in a 5mm NMR tube. To this solution was added **Cs⁺5-** (40 mg, 0.125 mmol, 1.5 equiv) and the tube sealed with a septum. The tube was removed from the drybox, and was placed in a 90 °C oil bath for 30 min. The tube was removed from the bath and analyzed by ^{31}P NMR spectroscopy. A 100 μL aliquot of the solution was injected into 500 μL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride. This yellow solution was diluted with 3.0 mL of ether and the organic phase was filtered through a 0.5 x 1.0 cm plug of silica gel. The filtrate was analyzed by GC to determine reaction conversion.

 ^{31}P NMR (202 MHz, $\text{C}_6\text{H}_5\text{CH}_3$):

31.9 (s, dppp(O)₂), 21.6 (d, $J = 48.6$), -9.9 (d, $J = 48.8$)

GC analysis

area, std	area, 12	conv, μmol	conversion = 37 %
238889	173007	29.98	

KINETICS STUDY FOR CATALYTIC CROSS-COUPLING EMPLOYING ARYLSILANOLATE COMPLEX **7p AS CATALYST (Table 6)**

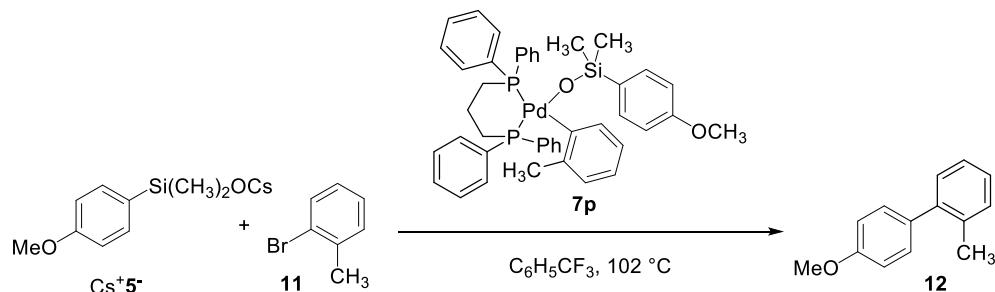


Table 6, entry 1

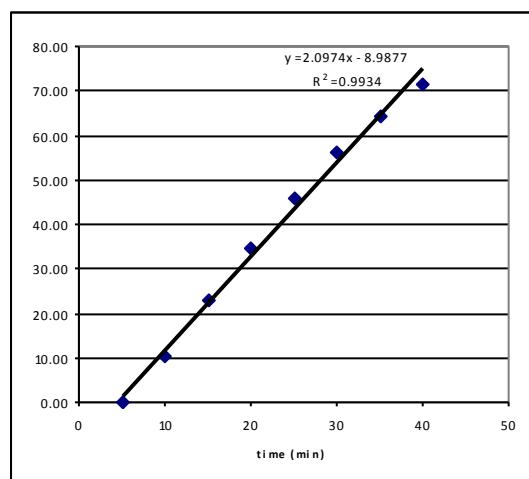
Cs⁺5⁻	125 mM
2-Bromotoluene	83 mM
7p	1.875 mM

Following General Procedure III, a mixture of silanolate **Cs⁺5⁻** (157 mg, 0.5 mmol, 1.5 equiv), **7p** (6.8 mg, 0.0075 mmol, 0.0225 equiv), naphthalene (30 mg), and 2-bromotoluene (40 μL, 0.33 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time area, std area, **12** conv, μmol

5	1	0	0.00
5	1	0	0.00
5	1	0	0.00
10	26457	1108	10.40
10	26842	1072	9.92
10	26226	1138	10.78
15	23287	2118	22.59
15	23089	2107	22.66
15	23808	2250	23.47
20	30436	4163	33.97
20	29422	4279	36.12
20	32368	4351	33.38
25	26012	4814	45.96
25	25628	4716	45.70
25	25354	4745	46.48
30	23179	5272	56.48
30	22009	4835	54.56
30	22622	5214	57.24



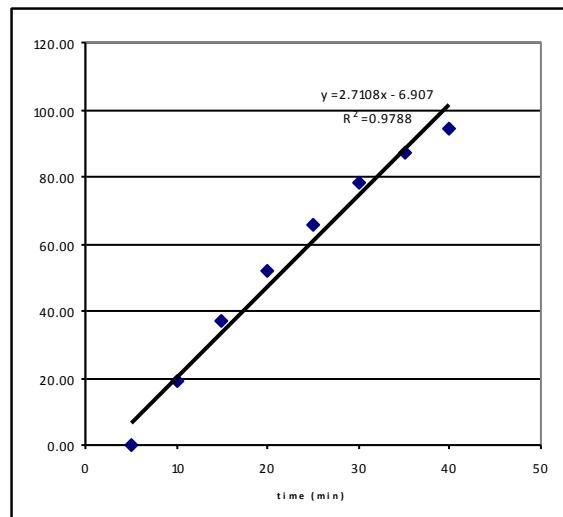
35	23295	6174	65.82
35	23414	6001	63.65
35	23176	5907	63.30
40	25007	7020	69.71
40	23514	6835	72.19
40	24481	7150	72.53

rate = 0.00875 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

5	1	0	0.00
5	1	0	0.00
5	1	0	0.00
10	22356	1510	16.77
10	29319	2771	23.47
10	22356	1510	16.77
15	25341	3642	35.69
15	23808	3576	37.30
15	22213	3331	37.24
20	25891	5444	52.22
20	26445	5452	51.20
20	27607	5762	51.83
25	23405	6255	66.37
25	24373	6280	63.99
25	23263	6179	65.96
30	20073	6552	81.06
30	19399	6076	77.78
30	20253	6222	76.29
35	20635	7353	88.49
35	21396	7539	87.50
35	21711	7509	85.89
40	20621	7792	93.84
40	20786	7788	93.05
40	20569	7895	95.32



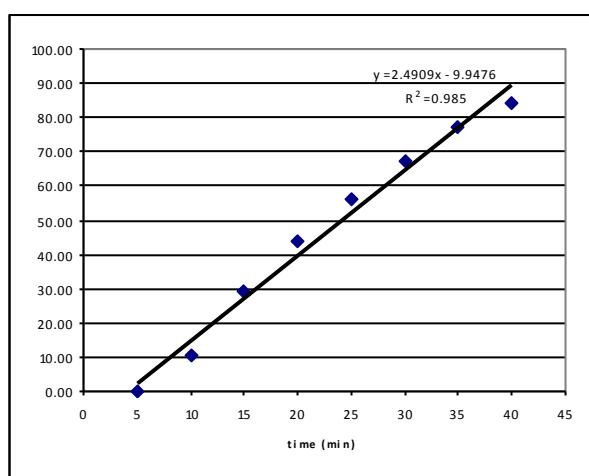
rate = 0.0113 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

5	1	0	0.00
5	1	0	0.00
5	1	0	0.00
10	26188	1079	10.23
10	25415	1172	11.45
10	26164	1102	10.46
15	21601	2581	29.67
15	20954	2506	29.70
15	79336	9334	29.22
20	24680	4297	43.24

20	22026	4028	45.42
20	22437	3834	42.44
25	18221	4139	56.41
25	18967	4269	55.90
25	38078	8663	56.50
30	60207	16123	66.50
30	19447	5281	67.44
30	18978	5235	68.50
35	48932	14805	75.14
35	31510	10064	79.32
35	15721	4876	77.03
40	43882	14534	82.25
40	43969	14607	82.50
40	20162	7067	87.05



$$\text{rate} = 0.0103 \text{ mM/sec}$$

$$\text{avg. rate} = 1.01 \times 10^{-2} \text{ mM/s}$$

Table 6, entry 2

$\text{Cs}^+ \text{5}^-$ 125 mM

2-Bromotoluene 83 mM

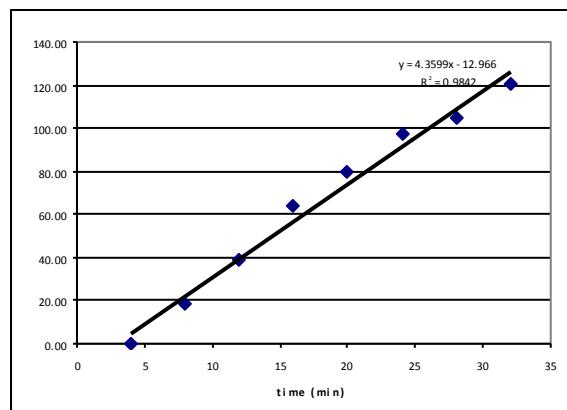
7p 3.75 mM

Following the General Procedure III, a mixture of silanolate $\text{Cs}^+ \text{5}^-$ (157 mg, 0.5 mmol, 1.5 equiv), **7p** (13.6 mg, 0.015 mmol, 0.045 equiv), naphthalene (30 mg), and 2-bromotoluene (40 μL , 0.33 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time area, std area, **12** conv, μmol

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
8	84481	6156	18.10
8	82451	6278	18.91
8	82449	6202	18.68
12	78409	11978	37.94
12	78709	12796	40.37
12	77810	12006	38.32
16	22784	5942	64.77
16	68141	17483	63.72
16	69287	18052	64.70
20	73123	23133	78.56



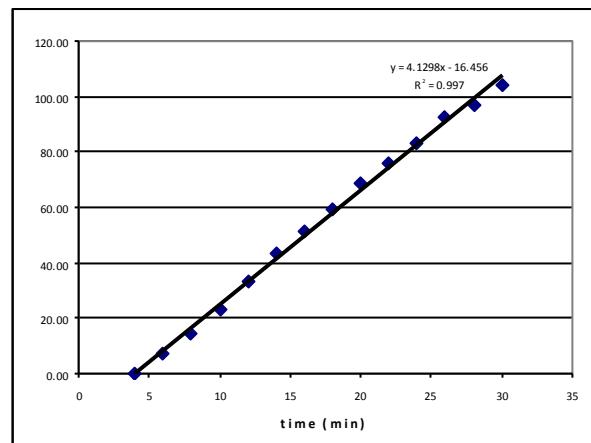
20	23719	7653	80.13
20	71795	23187	80.20
24	78012	30444	96.91
24	78796	30630	96.54
24	26487	10470	98.17
28	21998	9257	104.51
28	21699	9315	106.61
28	21557	9073	104.52
32	24013	11947	123.56
32	24971	12157	120.90
32	75187	35167	116.16

rate = 0.0182 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
6	62765	1753	6.94
6	62765	1753	6.94
6	62765	1753	6.94
8	57239	3373	14.63
8	57765	3351	14.41
8	57239	3373	14.63
10	62854	5721	22.60
10	66335	6097	22.83
10	44372	4192	23.46
12	22673	2986	32.71
12	22128	3055	34.29
12	23650	3173	33.32
14	59999	10347	42.83
14	23664	4030	42.29
14	44158	7862	44.22
16	22827	4645	50.53
16	25979	5215	49.85
16	49047	10472	53.02
18	20945	4970	58.93
18	20765	4883	58.40
18	20896	5168	61.42
20	16616	4559	68.14
20	16879	4757	69.99
20	50798	13949	68.19
22	64207	19579	75.73
22	23487	7015	74.17
22	23444	7315	77.49
24	20232	6671	81.88
24	56581	19460	85.41
24	59401	19630	82.07
26	60974	22494	91.62



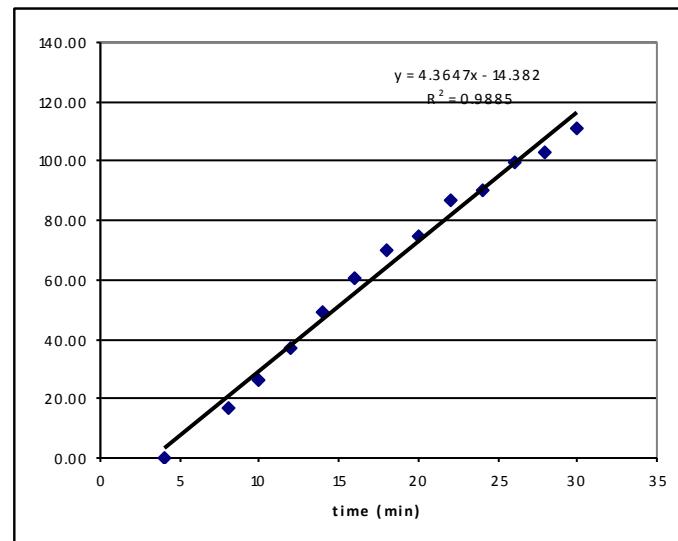
26	21777	8134	92.76
26	21159	7898	92.70
28	60919	23812	97.07
28	26499	10202	95.61
28	26085	10254	97.62
30	55231	22835	102.68
30	18831	8090	106.69
30	18914	7806	102.49

rate = 0.0173 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
4	1	0	0.00
8	23965	1639	16.98
8	24264	1661	17.00
8	24770	1702	17.06
10	68886	7423	26.76
10	23791	2564	26.76
10	24239	2560	26.23
12	24093	3543	36.52
12	25329	3741	36.68
12	23647	3669	38.53
14	66762	13291	49.44
14	21827	4256	48.42
14	22919	4559	49.40
16	63840	15466	60.16
16	22403	5509	61.07
16	22693	5504	60.23
18	21135	5958	70.01
18	22835	6473	70.40
18	22799	6334	68.99
20	61798	18603	74.76
20	25603	7591	73.63
20	25502	7749	75.46
22	61418	21000	84.91
22	21473	7430	85.93
22	21266	7595	88.69
24	69078	25888	93.07
24	28347	9946	87.13
24	26831	9798	90.69
26	25305	10240	100.49
26	47311	19020	99.84
26	24615	9852	99.40
28	28164	11861	104.59



28	28350	11813	103.48
28	30328	12273	100.50
30	24289	10710	109.50
30	23132	10397	111.62
30	23774	10778	112.59

$$\text{rate} = \mathbf{0.0182 \text{ mM/sec}}$$

$$\text{avg. rate} = \mathbf{1.79 \times 10^{-2} \text{ mM/s}}$$

Table 6, entry 3

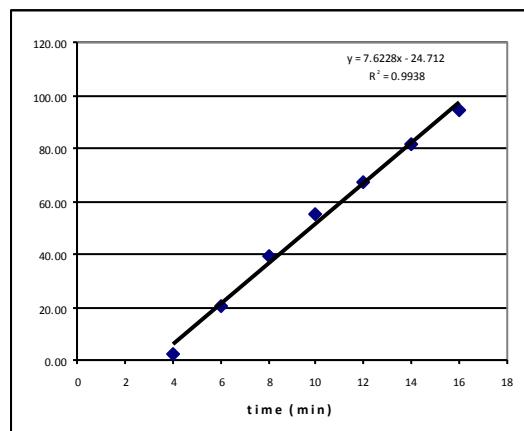
Cs ⁺ 5 ⁻	125 mM
2-Bromotoluene	83 mM
7p	7.5 mM

Following the General Procedure III, a mixture of silanolate Cs⁺**5**⁻ (157 mg, 0.5 mmol, 1.5 equiv), **7p** (27.2 mg, 0.030 mmol, 0.090 equiv), naphthalene (30 mg), and 2-bromotoluene (40 μ L, 0.33 mmol, 1.0 equiv) were dissolved in dry benzotrifluoride (4 mL) at room temperature followed by stirring at reflux. Aliquots of the mixture were then taken for GC analysis.

Run 1

Time	area, std	area, 12	conv, μmol
------	-----------	-----------------	-----------------------

4	1	0	0.00
4	74870	2173	7.21
4	1	0	0.00
6	20479	1779	21.57
6	63256	5081	19.95
6	63713	5143	20.05
8	60733	9249	37.82
8	63201	10081	39.61
8	62700	10017	39.68
10	59937	13322	55.20
10	61625	13699	55.21
10	61795	13703	55.07
12	62045	16811	67.29
12	65347	17787	67.60
12	66678	18072	67.31
14	64882	21519	82.37
14	69649	22249	79.33
14	23571	7866	82.88
16	57777	22014	94.62
16	21611	8017	92.13



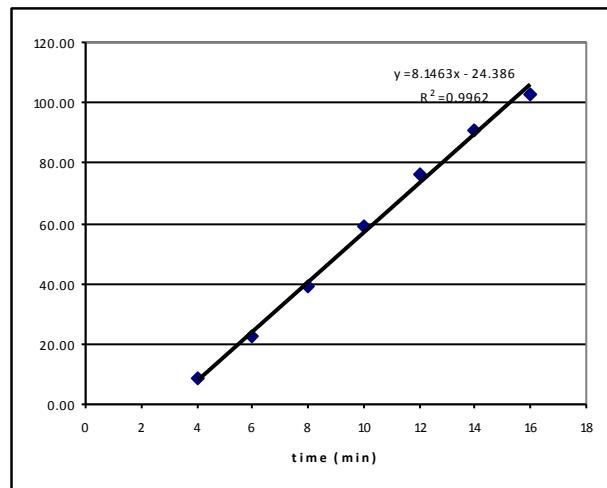
16	21182	8271	96.97
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rate = 0.0318 mM/sec

Run 2

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	49106	1746	8.83
4	49010	1697	8.60
4	49106	1746	8.83
6	67304	5980	22.07
6	70360	6359	22.44
6	72526	6672	22.85
8	22753	3613	39.43
8	69791	10884	38.73
8	70580	11326	39.85
10	26128	6257	59.47
10	81619	19277	58.65
10	81979	19461	58.95
12	65626	19880	75.23
12	69139	21486	77.18
12	68168	21013	76.55
14	62485	22745	90.40
14	69789	25418	90.45
14	69841	25558	90.88
16	63600	26083	101.85
16	67426	27886	102.71
16	66095	27857	104.67

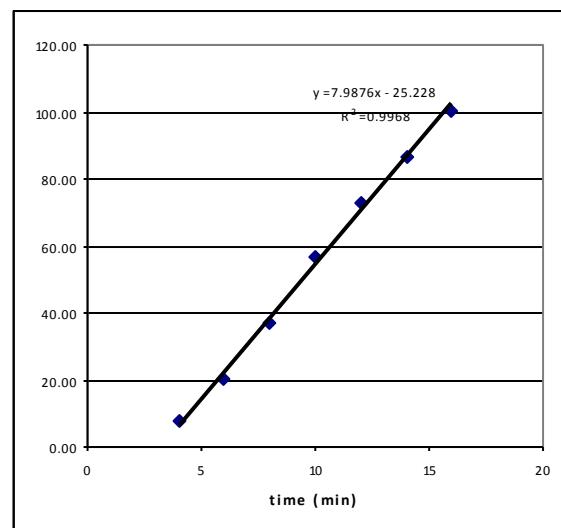


rate = 0.0339 mM/sec

Run 3

Time	area, std	area, 12	conv, μmol
------	-----------	----------	-----------------------

4	101820	3118	7.60
4	98708	3167	7.97
4	103016	3143	7.58
6	97098	7836	20.04
6	96104	7873	20.34
6	95316	7970	20.77
8	72175	10978	37.77
8	73373	10812	36.59
8	71716	10765	37.28
10	76428	17651	57.35
10	77172	17729	57.05
10	79722	17987	56.03
12	78683	22973	72.51
12	78710	23128	72.97
12	78865	23308	73.40
14	69195	24388	87.53

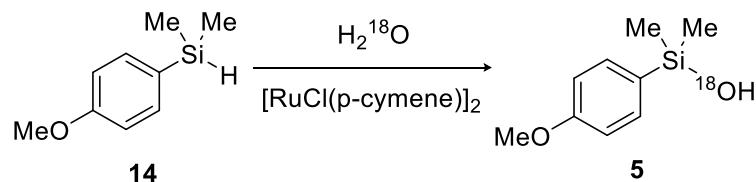


14	70832	24252	85.03
14	69242	24543	88.03
16	88670	35352	99.01
16	88169	36258	102.13
16	88822	35988	100.62

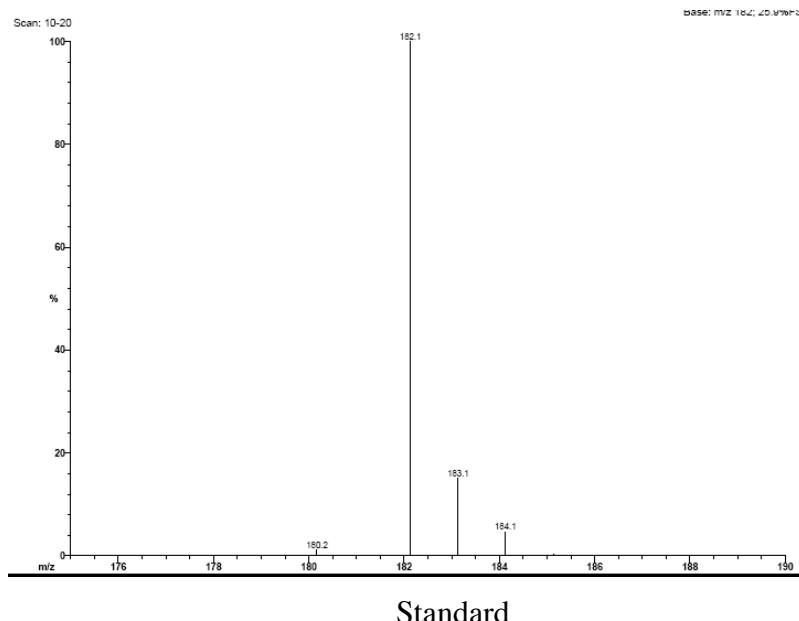
rate = 0. 0333 mM/sec

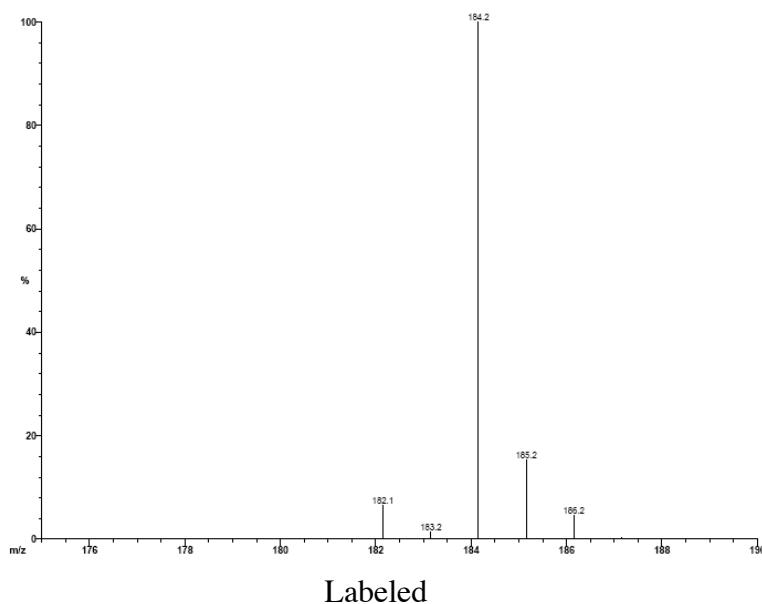
avg. rate = 3.30×10^{-2} mM/s

Preparation of ^{18}O -labeled (4-methoxyphenyl)dimethylsilanol (Scheme 4)

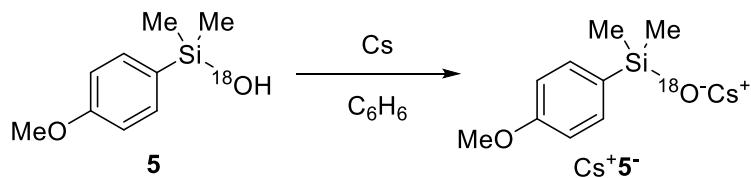


To a 50-mL, round-bottom flask equipped with a magnetic stir bar was added (4-methoxyphenyl)dimethylsilane **14** (831 mg, 5.0 mmol), 15 mL of acetonitrile, and 95% ^{18}O water (200 μL , 10.0 mmol, 2.0 equiv). To the resulting solution was added $[\text{RuCl}_2(\text{p-cymene})]_2$ (61 mg, 0.10 mmol, 0.02 equiv). The solution was stirred at room temperature open to the atmosphere for 10 min. The volatile solvents were then removed in vacuo to provide a red oil. The oil was taken up in 20 mL of hexanes and then was purified by silica gel chromatography (30 mm x 30 cm) using hexanes/ethyl acetate, 9/1 to afford 847 mg (92%) of **5** as a colorless oil. Label incorporation was determined by mass spectrometry (FI). Unlabeled standard- M/Z = 182.1. Labeled M/Z = 184.2. Incorporation = 94.4 %



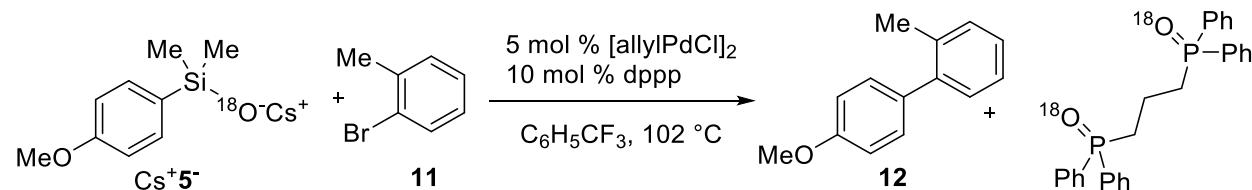


Preparation of ^{18}O -labeled Cesium (4-methoxyphenyl)dimethylsilanolate (Scheme 4)

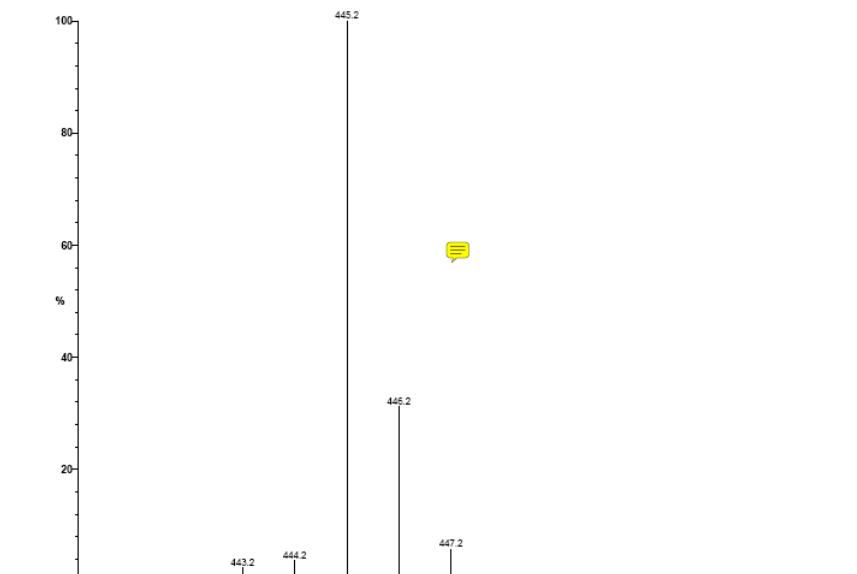


In the drybox, a 100-mL, one-necked flask equipped with a magnetic stir bar was charged with Cs (531 mg, 4.00 mmol, 1.0 equiv), and dry, degassed benzene (40 mL). While stirring, ^{18}O -labeled (4-methoxyphenyl)dimethylsilanol **5** (737 mg, 4.0 mmol, 1.0 equiv) was added and the resulting suspension was stirred in the drybox for 20 min. The resulting solution was filtered through a medium glass frit. The solvents were removed *in vacuo* (0.1 mm Hg) for 4 h to afford a sticky white solid. Hexanes (25 mL) were added and the resulting suspension was stirred for 1 h in the drybox. The resulting solids were collected on a fritted glass funnel, then were washed with hexanes (5 mL), and dried under vacuum (0.1 mm Hg) for 12 h to afford 982 mg (79%) of $\text{Cs}^+ \mathbf{5}^-$ as a white powder.

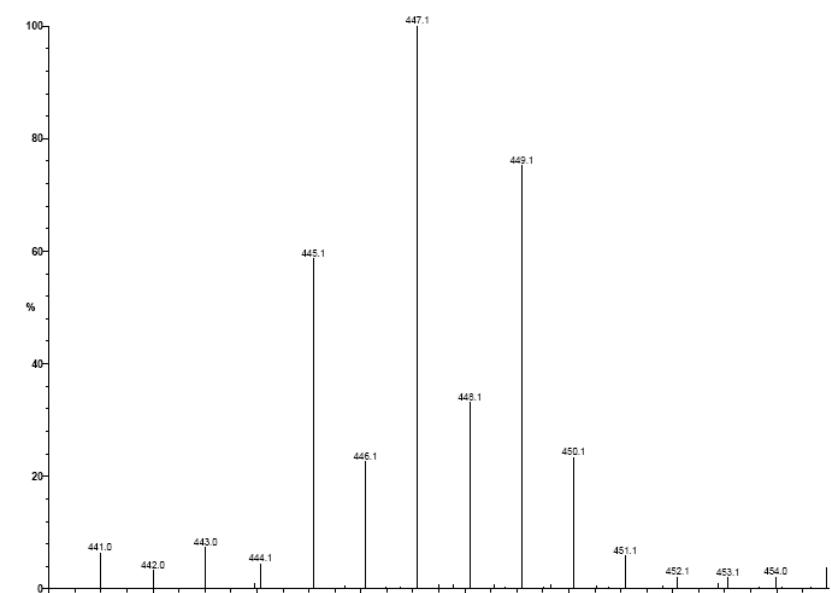
Isolation of ^{18}O labeled 1,3-bis(diphenylphosphino)propane dioxide (Scheme 4)



In a drybox, a one-neck, 10-mL round-bottom flask with a magnetic stir bar and fitted with a reflux condenser and three-way stop-cock (fitted with an argon inlet and a rubber septum) was charged with APC (9.1 mg, 0.025 mmol, 0.05 equiv), dppp (21.0 mg, 0.05 mmol, 0.1 equiv) 2-bromotoluene (63 μL , 0.50 mmol, 1.0 equiv), and dry benzotrifluoride (4.0 mL). To this solution was added $\text{Cs}^+ \text{5}^-$ (190 mg, 0.60 mmol, 1.2 equiv). The flask was removed from the drybox, the stopcock purged with argon, and was placed in a 105 °C oil bath for 3 h. The reaction was quenched by the addition of 2 mL of a 10% aqueous solution of 2-dimethylaminoethanethiol hydrochloride followed by stirring at rt for 30 min. The organics were separated and purified by chromatography (silica gel, 30 mm x 30 cm, CH_2Cl_2) to afford 8 mg (30%) of dppp(O)₂ as a yellow oil and 140 mg of impure **12** as a colorless oil. Bulb-to-bulb distillation (120°C, 0.5 mm Hg) provided 76 mg (67%) of **12** as a colorless oil. Label incorporation in the sample of dppp(O)₂ was determined by mass spectrometry (FAB). Unlabeled standard- M/Z = 445.1. Monolabeled M/Z = 447.1. Bislabeled M/Z = 449.1. Incorporation = 52.3 %



Standard



Labeled Sample

INDEX OF KINETIC EXPERIMENTS FOR CATALYTIC CROSS-COUPING OF ARYLSILANOLATES EMPLOYING (*t*-Bu₃P)₂Pd (Scheme 6)

	Page
Order in K ⁺ 5 ⁻	
75 mM	<i>S113</i>
150 mM	<i>S109</i>
300 mM	<i>S117</i>
Order in 4-Bromofluorobenzene 15	
100 mM	<i>S109</i>
200 mM	<i>S122</i>
400 mM	<i>S125</i>
Order in Palladium	
5.0 mM	<i>S109</i>
10.0 mM	<i>S130</i>
15.0 mM	<i>S134</i>
20.0 mM	<i>S137</i>

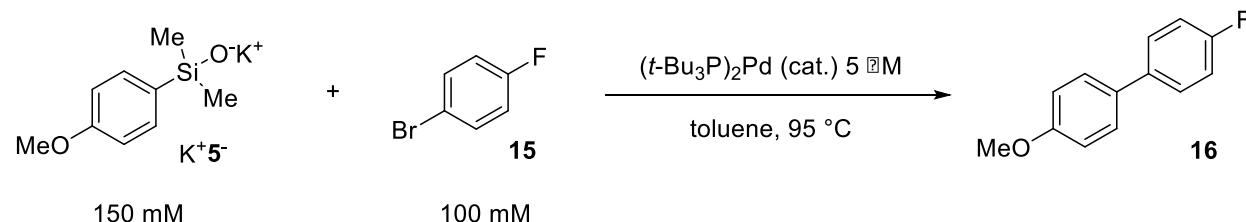
General Procedure IV: Kinetic Measurements for Catalytic Cross-Coupling of Arylsilanolates Employing (*t*-Bu₃P)₂Pd (Scheme 11)

Order in K⁺5⁻

K⁺5⁻ 150 mM

15 100 mM

Pd cat. 5 mM



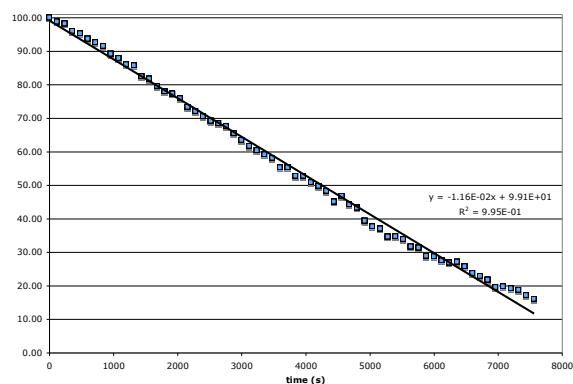
In a drybox, an oven-dried, 2.5-mL vial equipped with a magnetic stir bar was charged with (*t*-Bu₃P)₂Pd (2.5 mg, 0.0049 mmol), toluene (0.8 mL), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) resulting in a colorless solution. K⁺5⁻ (33 mg, 0.15 mmol) was added as a solid and the sides of the vial were rinsed with additional toluene (0.2 mL). The mixture was stirred for 3 min to ensure complete dissolution was observed. The resulting solution was transferred into an oven-dried, 5-mm NMR tube. The tube was sealed with a septum and wrapped with Parafilm to exclude any oxygen or moisture. The NMR tube was removed from the drybox and inserted into a preheated 95 °C NMR probe. The temperature of the reaction solution was allowed to equilibrate for 3 min prior to data collection.¹⁷ The reaction progress was monitored by the disappearance of starting aryl halide (¹⁹F NMR, -115.9 ppm) as compared to an internal reference (PhCF₃, ¹⁹F NMR, -63.2 ppm) via ¹⁹F NMR spectroscopy using the following parameters: at = 0.328, d1 = 1, pw90 = 18.25, pw=pw90/2, nt=24, sampling every 120 s.¹⁸ The rate of the reaction was determined by the slope of the linear plot¹⁹: **rate = 1.16 x 10⁻² mM s⁻¹**.

Run 1

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalization (mM)
0	1.086	1.435	100.03
120	1.076	1.417	98.77
240	1.098	1.409	98.20
360	1.095	1.377	95.99
480	1.083	1.368	95.30
600	1.058	1.345	93.70
720	1.077	1.329	92.61
840	1.056	1.313	91.51
960	1.099	1.280	89.20

1080	1.079	1.261	87.91
1200	1.067	1.235	86.08
1320	1.089	1.231	85.77
1440	1.042	1.183	82.47
1560	1.039	1.173	81.74
1680	1.053	1.143	79.63
1800	1.055	1.119	77.97
1920	1.040	1.109	77.31
2040	1.045	1.089	75.91
2160	1.059	1.052	73.32
2280	1.069	1.034	72.07
2400	1.062	1.012	70.55
2520	1.037	0.993	69.19
2640	1.045	0.982	68.44
2760	1.055	0.970	67.59
2880	1.027	0.940	65.52
3000	1.017	0.911	63.50
3120	1.031	0.886	61.71
3240	1.031	0.867	60.38
3360	1.040	0.849	59.14
3480	1.042	0.834	58.12
3600	1.003	0.793	55.29
3720	1.048	0.794	55.32
3840	1.022	0.757	52.75
3960	1.012	0.758	52.80
4080	1.002	0.730	50.86
4200	1.006	0.714	49.75
4320	1.052	0.693	48.29
4440	1.009	0.647	45.09
4560	1.029	0.672	46.83
4680	1.005	0.636	44.31
4800	1.024	0.621	43.29
4920	0.982	0.565	39.36
5040	0.999	0.541	37.72
5160	1.012	0.532	37.04
5280	0.990	0.497	34.65
5400	1.015	0.499	34.74
5520	0.978	0.486	33.86
5640	0.980	0.454	31.66
5760	0.976	0.450	31.37
5880	0.966	0.414	28.88
6000	0.973	0.412	28.70
6120	0.978	0.396	27.57
6240	0.970	0.385	26.86
6360	1.001	0.390	27.15
6480	0.985	0.370	25.80
6600	0.974	0.341	23.77
6720	0.985	0.328	22.86
6840	1.000	0.313	21.79
6960	0.965	0.281	19.57
7080	0.974	0.284	19.81
7200	0.971	0.275	19.18
7320	0.964	0.269	18.76
7440	1.001	0.246	17.15
7560	0.966	0.230	16.05
7680	0.964	0.232	16.17

7800	0.979	0.210	14.61
7920	0.988	0.208	14.49
8040	0.980	0.203	14.14
8160	1.001	0.190	13.22
8280	0.982	0.168	11.68
8400	0.978	0.161	11.18
8520	0.979	0.160	11.13
8640	0.924	0.151	10.55
8760	0.936	0.141	9.84
8880	0.952	0.137	9.51
9000	0.973	0.108	7.55
9120	0.957	0.128	8.91
9240	0.962	0.114	7.97



K⁺5⁻ 150 mM
15 100 mM
 Pd cat. 5 mM

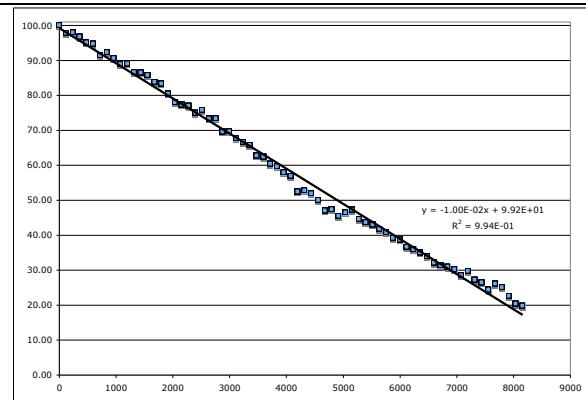
Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (2.7 mg, 0.0053 mmol), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 1.00 x 10⁻² s⁻¹**.

Run 2

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalization (mM)
0	0.968	1.255	99.98
120	0.935	1.228	97.81
240	0.970	1.229	97.91
360	0.964	1.214	96.71
480	0.959	1.193	95.06
600	0.942	1.188	94.66
720	0.932	1.147	91.36
840	0.951	1.158	92.30

960	0.967	1.136	90.55
1080	0.962	1.115	88.88
1200	0.951	1.117	89.01
1320	0.964	1.087	86.58
1440	0.934	1.085	86.46
1560	0.958	1.076	85.75
1680	0.939	1.052	83.80
1800	0.980	1.045	83.30
1920	0.977	1.009	80.42
2040	0.945	0.978	77.94
2160	0.958	0.970	77.29
2280	0.972	0.966	76.95
2400	0.945	0.941	74.96
2520	0.917	0.951	75.75
2640	0.965	0.920	73.27
2760	0.926	0.920	73.33
2880	0.960	0.874	69.60
3000	0.973	0.874	69.63
3120	0.925	0.848	67.61
3240	0.929	0.836	66.64
3360	0.927	0.825	65.72
3480	0.932	0.787	62.72
3600	0.932	0.783	62.37
3720	0.964	0.758	60.40
3840	0.941	0.748	59.63
3960	0.961	0.727	57.89
4080	0.923	0.714	56.89
4200	0.958	0.658	52.42
4320	0.952	0.662	52.75
4440	0.922	0.651	51.91
4560	0.904	0.627	49.98
4680	0.946	0.590	47.01
4800	0.943	0.594	47.36
4920	0.946	0.570	45.45
5040	0.944	0.583	46.47
5160	0.963	0.593	47.25
5280	0.957	0.560	44.58
5400	0.946	0.547	43.61
5520	0.958	0.539	42.96
5640	0.956	0.524	41.74
5760	0.929	0.512	40.77
5880	0.942	0.492	39.16
6000	0.941	0.485	38.66
6120	0.931	0.459	36.59
6240	0.969	0.450	35.88
6360	0.965	0.439	34.95
6480	0.943	0.426	33.91
6600	0.959	0.403	32.10
6720	0.973	0.393	31.28
6840	0.955	0.388	30.90
6960	0.937	0.379	30.18
7080	0.970	0.357	28.46
7200	0.956	0.373	29.69
7320	0.978	0.341	27.17
7440	0.963	0.331	26.41
7560	0.961	0.306	24.37

7680	0.939	0.327	26.08
7800	0.952	0.314	25.04
7920	0.955	0.284	22.60
8040	0.924	0.256	20.36
8160	0.926	0.248	19.74
8280	0.919	0.224	17.89
8400	0.905	0.216	17.24
8520	0.914	0.229	18.29
8640	0.921	0.214	17.06
8760	0.952	0.197	15.71
8880	0.947	0.186	14.82
9000	0.936	0.166	13.23
9120	0.948	0.169	13.44
9240	0.934	0.162	12.90



$$\text{avg. rate} = 1.08 \times 10^{-2} \text{ s}^{-1} \pm 1.1 \times 10^{-3}$$

K⁺5⁻ 75 mM

15 100 mM

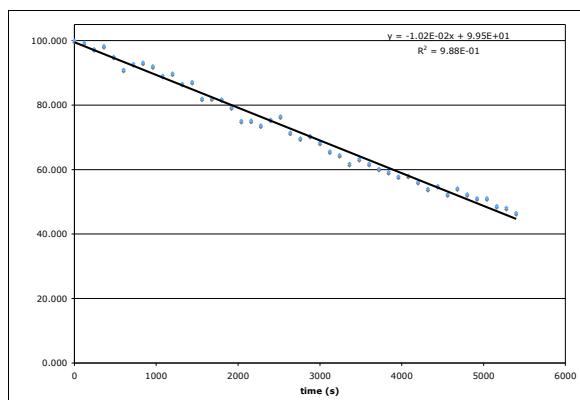
Pd cat. 5 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (2.6 mg, 0.0050 mmol), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (17 mg, 0.075 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 1.02 x 10⁻² mMs⁻¹**.

Run 1

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	1.335	1.744	99.99
120	1.320	1.712	98.18
240	1.342	1.707	97.88
360	1.305	1.676	96.11
480	1.314	1.630	93.47

600	1.336	1.587	91.01
720	1.292	1.566	89.77
840	1.275	1.553	89.04
960	1.264	1.520	87.14
1080	1.282	1.492	85.54
1200	1.244	1.459	83.66
1320	1.261	1.427	81.82
1440	1.240	1.413	81.00
1560	1.267	1.357	77.82
1680	1.243	1.332	76.40
1800	1.229	1.312	75.25
1920	1.235	1.279	73.34
2040	1.257	1.233	70.72
2160	1.240	1.218	69.82
2280	1.245	1.199	68.78
2400	1.201	1.184	67.92
2520	1.164	1.163	66.71
2640	1.204	1.124	64.46
2760	1.193	1.087	62.30
2880	1.172	1.079	61.85
3000	1.186	1.059	60.71
3120	1.209	1.036	59.42
3240	1.200	1.012	58.02
3360	1.210	0.977	56.03
3480	1.183	0.978	56.06
3600	1.154	0.932	53.44
3720	1.199	0.943	54.08
3840	1.168	0.904	51.85
3960	1.181	0.893	51.19
4080	1.164	0.883	50.66
4200	1.175	0.862	49.45
4320	1.165	0.823	47.16
4440	1.173	0.842	48.26
4560	1.169	0.800	45.85
4680	1.137	0.806	46.20
4800	1.154	0.789	45.25
4920	1.160	0.775	44.45
5040	1.135	0.758	43.48
5160	1.144	0.727	41.71
5280	1.154	0.726	41.61
5400	1.164	0.708	40.61
5520	1.150	0.706	40.47
5640	1.128	0.674	38.64
5760	1.169	0.693	39.74
5880	1.117	0.679	38.96
6000	1.150	0.668	38.31
6120	1.118	0.656	37.62
6240	1.139	0.637	36.52



Run 2

K⁺5⁻ 75 mM

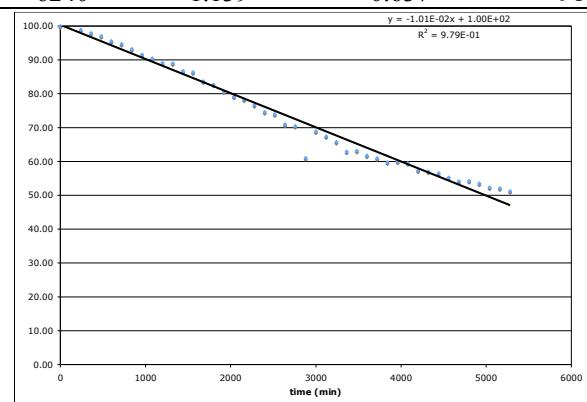
15 100 mM

Pd cat. 5 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (2.6 mg, 0.0050 mmol), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (17 mg, 0.075 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 1.01 x 10⁻² mM s⁻¹**.

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	1.335	1.744	100.00
120	1.320	1.712	100.74
240	1.342	1.707	98.93
360	1.305	1.676	97.93
480	1.314	1.630	97.06
600	1.336	1.587	95.57
720	1.292	1.566	94.67
840	1.275	1.553	93.23
960	1.264	1.520	91.58
1080	1.282	1.492	90.46
1200	1.244	1.459	89.07
1320	1.261	1.427	88.98
1440	1.240	1.413	86.75
1560	1.267	1.357	86.35
1680	1.243	1.332	83.68
1800	1.229	1.312	82.71
1920	1.235	1.279	80.72
2040	1.257	1.233	79.18
2160	1.240	1.218	78.30
2280	1.245	1.199	76.65

2400	1.201	1.184	74.62
2520	1.164	1.163	73.96
2640	1.204	1.124	70.94
2760	1.193	1.087	70.44
2880	1.172	1.079	61.01
3000	1.186	1.059	68.85
3120	1.209	1.036	67.44
3240	1.200	1.012	65.76
3360	1.210	0.977	62.92
3480	1.183	0.978	63.13
3600	1.154	0.932	61.73
3720	1.199	0.943	60.97
3840	1.168	0.904	59.72
3960	1.181	0.893	60.01
4080	1.164	0.883	59.48
4200	1.175	0.862	57.37
4320	1.165	0.823	57.02
4440	1.173	0.842	56.52
4560	1.169	0.800	55.26
4680	1.137	0.806	54.17
4800	1.154	0.789	54.28
4920	1.160	0.775	53.46
5040	1.135	0.758	52.34
5160	1.144	0.727	52.04
5280	1.154	0.726	51.24
5400	1.164	0.708	51.39
5520	1.150	0.706	48.93
5640	1.128	0.674	48.85
5760	1.169	0.693	47.38
5880	1.117	0.679	48.22
6000	1.150	0.668	100.00
6120	1.118	0.656	100.74
6240	1.139	0.637	98.93



$$\text{avg. rate} = 1.02 \times 10^{-2} \text{ mM s}^{-1} \pm 7.1 \times 10^{-5}$$

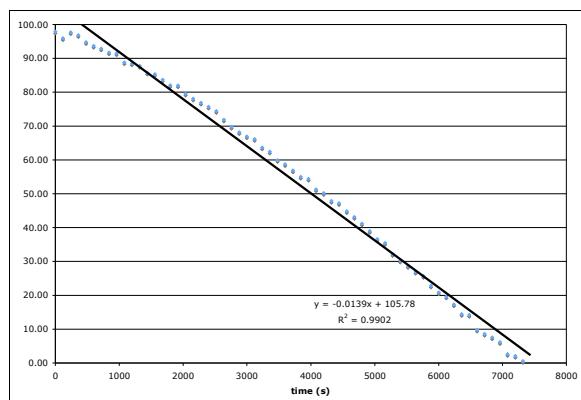
K^+5^-	300 mM
15	100 mM
Pd cat.	5 mM

Following General Procedure IV, the following modifications were required for experiments involving >150 mM in K^+5^- .²⁰ In a drybox, an oven-dried, 2.5-mL vial equipped with a magnetic stir bar was charged with (*t*-Bu₃P)₂Pd (2.5 mg, 0.0049 mmol), toluene (0.8 mL), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) resulting in a colorless solution. Silanolate K+21⁻ (33 mg, 0.15 mmol) was added as a solid and the sides of the vial were rinsed with toluene (0.2 mL). The mixture was stirred for 3 min to ensure complete dissolution. The remaining K^+5^- (33 mg, 0.15 mmol) was added into an oven-dried, 5-mm NMR tube. The reaction solution was transferred into the NMR tube containing the excess silanolate and was shaken vigorously. The tube was sealed with a septum and wrapped with Parafilm to exclude any oxygen or moisture. The NMR tube was removed from the drybox and inserted into a preheated 95 °C NMR probe. The temperature of the reaction solution was allowed to equilibrate for 3 min at which point the NMR tube was ejected and shaken to ensure complete homogeneity. The tube was returned to the heated probe and the reaction progress was monitored by the disappearance of starting aryl halide (¹⁹F NMR, -115.9 ppm) as compared to an internal reference (PhCF₃, ¹⁹F NMR, -63.2 ppm)) via ¹⁹F NMR spectroscopy using the following parameters: at = 0.328, d1 = 1, pw90 = 18.25, pw=pw90/2, nt=24, sampling every 120 s. **rate** = **1.3 x 10⁻² mM s⁻¹**.

Run 1

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	2.465	1.992	97.86
120	2.486	1.982	95.89
240	2.478	1.982	97.70
360	2.501	1.967	96.85
480	2.489	1.954	94.75
600	2.486	1.960	93.69
720	2.494	1.978	92.89
840	2.528	1.931	91.72
960	2.516	1.927	91.44
1080	2.521	1.895	88.78
1200	2.550	1.915	88.50
1320	2.534	1.898	87.76
1440	2.501	1.872	85.77
1560	2.511	1.859	85.35
1680	2.534	1.822	83.67
1800	2.551	1.856	82.04

1920	2.528	1.840	81.97
2040	2.540	1.800	79.53
2160	2.538	1.780	78.11
2280	2.524	1.765	76.84
2400	2.531	1.743	75.70
2520	2.548	1.737	74.44
2640	2.536	1.687	71.88
2760	2.554	1.682	69.76
2880	2.529	1.667	68.16
3000	2.518	1.630	66.96
3120	2.559	1.622	66.16
3240	2.526	1.590	63.65
3360	2.548	1.559	62.45
3480	2.541	1.557	59.92
3600	2.554	1.511	58.68
3720	2.559	1.484	56.90
3840	2.579	1.460	54.99
3960	2.552	1.438	54.40
4080	2.548	1.414	51.29
4200	2.552	1.366	50.20
4320	2.551	1.325	47.89
4440	2.581	1.295	47.19
4560	2.554	1.272	44.82
4680	2.549	1.257	43.12
4800	2.549	1.209	41.19
4920	2.564	1.187	38.97
5040	2.579	1.138	36.60
5160	2.567	1.115	35.45
5280	2.548	1.081	32.16
5400	2.548	1.045	30.21
5520	2.555	1.034	28.60
5640	2.574	0.975	26.94
5760	2.553	0.954	25.73
5880	2.559	0.910	22.82
6000	2.557	0.897	20.83
6120	2.550	0.852	19.59
6240	2.569	0.819	17.27
6360	2.541	0.783	14.43
6480	2.577	0.740	14.17
6600	2.582	0.695	9.80
6720	2.575	0.674	8.61
6840	2.584	0.611	7.56
6960	2.589	0.574	6.14
7080	2.597	0.543	2.60
7200	2.558	0.512	2.07
7320	2.581	0.489	0.57



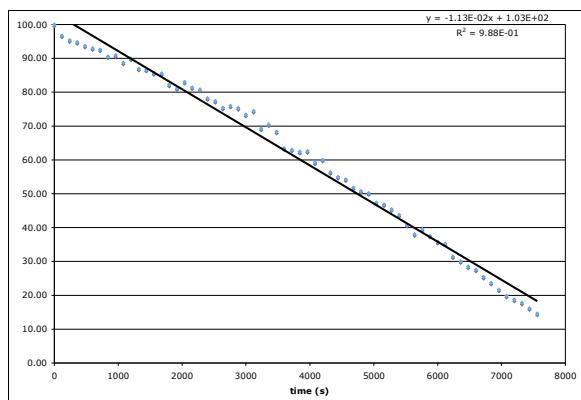
$\text{K}^+ \text{5}^-$ 300 mM
15 100 mM
 Pd cat. 5 mM

Following General Procedure IV, a mixture of $(t\text{-Bu}_3\text{P})_2\text{Pd}$ (2.6 mg, 0.0050 mmol), benzotrifluoride (5 μL) and **15** (11 μL , 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by $\text{K}^+ \text{5}^-$ (66 mg, 0.30 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ^{19}F NMR spectroscopy. **rate = $1.13 \times 10^{-2} \text{ mM s}^{-1}$** .

Run 2

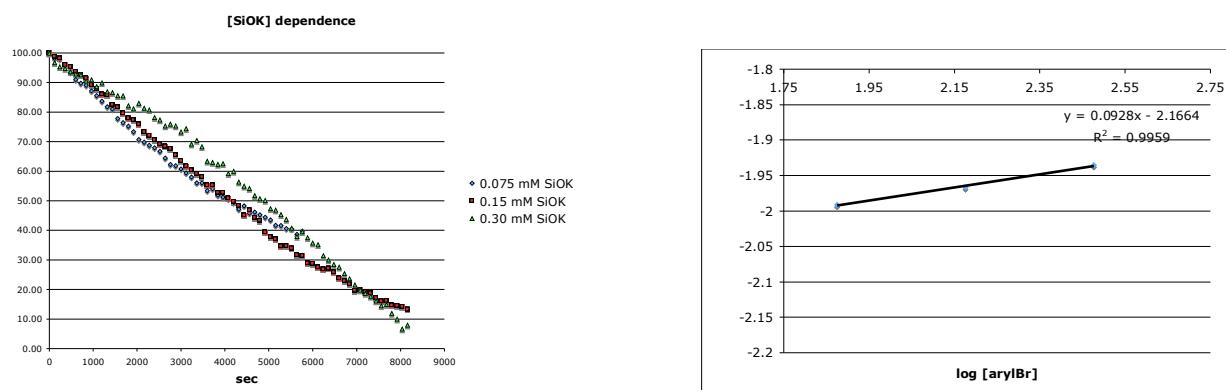
Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	0.997	1.135	100.00
120	0.959	1.254	96.78
240	0.961	1.214	95.40
360	0.966	1.196	94.81
480	0.966	1.189	93.75
600	0.948	1.176	92.99
720	0.943	1.166	92.66
840	0.987	1.162	90.59
960	0.997	1.136	91.00
1080	0.961	1.141	88.71
1200	0.974	1.112	89.92
1320	0.980	1.128	86.96
1440	0.954	1.091	86.66
1560	0.942	1.087	85.64
1680	0.951	1.074	85.56
1800	0.957	1.073	82.24
1920	0.959	1.031	81.19
2040	0.958	1.018	83.08
2160	0.973	1.042	81.40
2280	0.935	1.021	80.79
2400	0.939	1.013	78.26

2520	0.963	0.981	77.42
2640	0.960	0.971	75.37
2760	0.963	0.945	76.00
2880	0.943	0.953	75.35
3000	0.954	0.945	73.37
3120	0.954	0.920	74.50
3240	0.954	0.934	69.22
3360	0.928	0.868	70.49
3480	0.962	0.884	68.36
3600	0.955	0.857	63.45
3720	0.968	0.796	63.01
3840	0.985	0.790	62.40
3960	0.939	0.782	62.59
4080	0.973	0.785	59.29
4200	0.926	0.743	60.03
4320	0.976	0.753	56.39
4440	0.947	0.707	54.99
4560	0.967	0.690	54.26
4680	0.949	0.680	51.83
4800	0.932	0.650	50.80
4920	0.971	0.637	50.19
5040	0.979	0.629	47.39
5160	0.946	0.594	46.92
5280	0.945	0.588	45.42
5400	0.921	0.570	43.77
5520	0.951	0.549	40.83
5640	0.958	0.512	38.04
5760	0.952	0.477	39.58
5880	0.963	0.496	37.59
6000	0.937	0.471	35.78
6120	0.936	0.449	35.25
6240	0.944	0.442	31.50
6360	0.936	0.395	30.06
6480	0.949	0.377	28.50
6600	0.970	0.357	27.58
6720	0.958	0.346	25.46
6840	0.970	0.319	23.72
6960	0.944	0.297	21.65
7080	0.935	0.272	19.84
7200	0.969	0.249	18.77
7320	0.935	0.235	17.78
7440	0.954	0.223	16.20
7560	0.942	0.203	14.60
7680	0.965	0.183	14.96
7800	0.917	0.188	12.02
7920	0.953	0.151	10.06
8040	0.982	0.126	6.70
8160	0.966	0.084	8.03
8280	0.934	0.101	7.37



$$\text{avg. rate} = 1.22 \times 10^{-2} \text{ mM s}^{-1} \pm 1.2 \times 10^{-3}$$

Order in K⁺5⁻:



Order in 15K⁺5⁻ 150 mM

15 200 mM

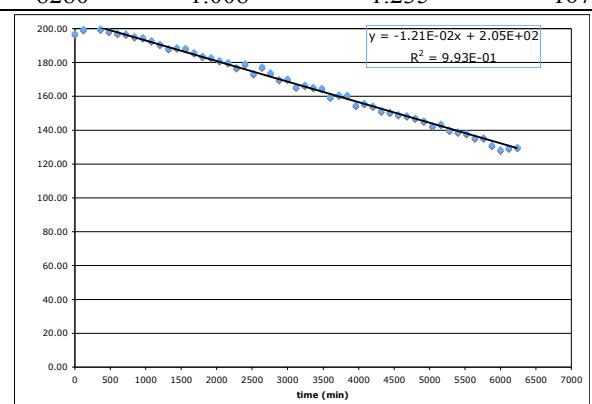
Pd cat. 5 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (2.6 mg, 0.0050 mmol), benzotrifluoride (5 μ L) and **15** (22 μ L, 35 mg, 0.20 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 1.12 x 10⁻² mMs⁻¹**.

Run 1

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	0.984	2.266	197.08
120	0.996	2.295	199.59
240	1.009	2.309	200.78
360	1.028	2.298	199.86
480	1.025	2.283	198.56
600	1.007	2.269	197.33
720	0.998	2.265	196.98
840	1.028	2.247	195.43
960	1.035	2.240	194.81
1080	1.029	2.219	192.97
1200	0.995	2.195	190.86
1320	0.982	2.166	188.33
1440	1.034	2.171	188.81
1560	1.017	2.167	188.39
1680	1.051	2.137	185.82
1800	1.030	2.113	183.78
1920	1.022	2.104	182.94
2040	1.008	2.084	181.19
2160	1.028	2.070	180.01
2280	1.035	2.037	177.09
2400	1.003	2.062	179.31
2520	1.023	1.996	173.59
2640	0.989	2.041	177.46
2760	1.034	2.000	173.92
2880	1.020	1.955	170.01
3000	1.001	1.959	170.32
3120	1.026	1.905	165.64
3240	1.009	1.917	166.66
3360	1.023	1.902	165.42
3480	1.004	1.896	164.83
3600	1.025	1.836	159.66
3720	0.991	1.850	160.87
3840	1.018	1.847	160.64
3960	1.040	1.780	154.82
4080	1.019	1.793	155.94

4200	0.999	1.775	154.36
4320	1.026	1.742	151.50
4440	1.017	1.732	150.63
4560	1.009	1.718	149.36
4680	0.998	1.709	148.64
4800	1.026	1.694	147.29
4920	1.019	1.674	145.54
5040	1.019	1.640	142.62
5160	0.994	1.651	143.57
5280	1.012	1.610	140.03
5400	1.039	1.599	139.08
5520	0.998	1.589	138.16
5640	0.995	1.559	135.53
5760	0.985	1.559	135.60
5880	1.004	1.508	131.13
6000	0.983	1.478	128.49
6120	1.007	1.490	129.54
6240	1.020	1.494	129.94
6360	1.033	1.463	127.23
6480	0.984	1.437	124.99
6600	0.991	1.379	119.89
6720	1.025	1.360	118.28
6840	1.050	1.345	116.99
6960	1.009	1.340	116.52
7080	1.032	1.320	114.79
7200	1.005	1.355	117.87
7320	1.003	1.329	115.59
7440	1.040	1.316	114.41
7560	0.993	1.314	114.26
7680	1.010	1.291	112.29
7800	1.012	1.264	109.91
7920	1.020	1.272	110.57
8040	1.019	1.269	110.34
8160	1.030	1.273	110.68
8280	1.008	1.235	107.43



K^+ 5⁻ 150 mM

15 200 mM

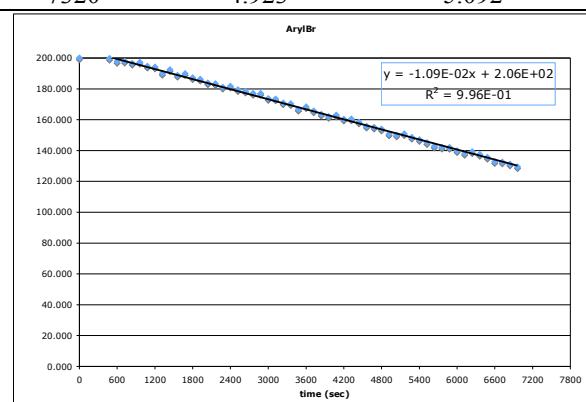
Pd cat. 5 mM

Following General Procedure IV, a mixture of $(t\text{-Bu}_3\text{P})_2\text{Pd}$ (2.6 mg, 0.0050 mmol), benzotrifluoride (5 μL) and **15** (22 μL , 35 mg, 0.20 mmol) were dissolved in toluene (0.8 mL), followed by $\text{K}^+\text{5}^-$ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ^{19}F NMR spectroscopy. **rate = 1.09 x 10⁻² mMs⁻¹**.

Run 2

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	4.745	7.762	100.00
120	4.728	7.791	100.37
240	4.714	7.745	99.78
360	4.669	7.762	100.00
480	4.657	7.610	98.04
600	4.672	7.555	97.34
720	4.726	7.648	98.53
840	4.689	7.541	97.15
960	4.715	7.605	97.98
1080	4.697	7.477	96.33
1200	4.715	7.487	96.46
1320	4.727	7.345	94.63
1440	4.730	7.441	95.87
1560	4.719	7.287	93.88
1680	4.756	7.388	95.18
1800	4.868	7.451	96.00
1920	4.810	7.327	94.40
2040	4.810	7.233	93.19
2160	4.847	7.265	93.60
2280	4.818	7.132	91.88
2400	4.774	7.097	91.43
2520	4.784	7.020	90.44
2640	4.844	7.073	91.13
2760	4.811	6.966	89.74
2880	4.758	6.890	88.77
3000	4.856	6.896	88.84
3120	4.811	6.827	87.95
3240	4.842	6.767	87.18
3360	4.832	6.732	86.73
3480	4.872	6.645	85.61
3600	4.821	6.638	85.52
3720	4.840	6.558	84.49
3840	4.870	6.512	83.90
3960	4.868	6.453	83.14
4080	4.848	6.463	83.27
4200	4.946	6.487	83.57
4320	4.871	6.391	82.33
4440	4.904	6.357	81.90
4560	4.950	6.307	81.26
4680	4.907	6.224	80.19
4800	4.922	6.195	79.81
4920	4.930	6.078	78.31
5040	4.979	6.109	78.70

5160	4.926	6.075	78.27
5280	4.948	6.012	77.46
5400	4.933	5.934	76.45
5520	4.906	5.823	75.02
5640	4.949	5.782	74.49
5760	4.992	5.804	74.78
5880	4.903	5.695	73.37
6000	4.896	5.601	72.16
6120	4.907	5.546	71.45
6240	4.869	5.541	71.38
6360	4.871	5.478	70.58
6480	4.894	5.425	69.89
6600	4.940	5.367	69.15
6720	4.971	5.387	69.40
6840	4.980	5.343	68.83
6960	4.926	5.215	67.19
7080	4.929	5.154	66.40
7200	4.913	5.177	66.70
7320	4.923	5.092	65.60



$$\text{avg. rate} = 1.15 \times 10^{-2} \text{ s}^{-1} \pm 8.5 \times 10^{-4}$$

K⁺5⁻ 150 mM

15 400 mM

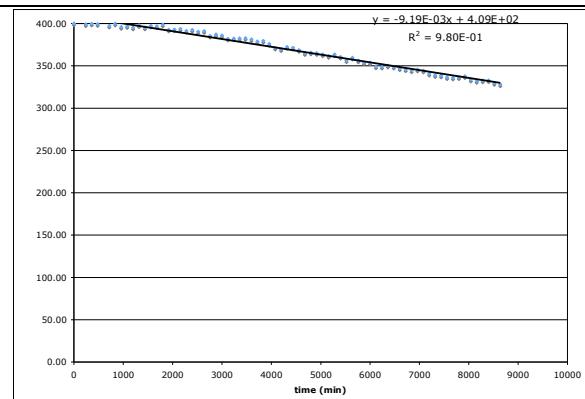
Pd cat. 5 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (2.6 mg, 0.0050 mmol), benzotrifluoride (5 μ L) and **15** (44 μ L, 70 mg, 0.20 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 9.19 x 10⁻³ mM s⁻¹**.

Run 1

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	2.424	8.506	400.02
120	2.445	8.559	402.47
240	2.466	8.484	398.96
360	2.464	8.500	399.70
480	2.491	8.492	399.35
600	2.486	8.511	400.23
720	2.502	8.450	397.37
840	2.483	8.501	399.77
960	2.511	8.420	395.95
1080	2.498	8.436	396.73
1200	2.506	8.405	395.24
1320	2.576	8.460	397.82
1440	2.556	8.403	395.15
1560	2.543	8.446	397.20
1680	2.535	8.440	396.87
1800	2.554	8.478	398.70
1920	2.555	8.350	392.68
2040	2.574	8.355	392.88
2160	2.558	8.369	393.55
2280	2.574	8.322	391.33
2400	2.575	8.346	392.50
2520	2.546	8.297	390.17
2640	2.578	8.315	391.00
2760	2.576	8.194	385.34
2880	2.584	8.234	387.21
3000	2.603	8.207	385.93
3120	2.583	8.124	382.03
3240	2.618	8.129	382.28
3360	2.631	8.131	382.35
3480	2.566	8.137	382.65
3600	2.610	8.107	381.25
3720	2.622	8.057	378.90
3840	2.619	8.069	379.44
3960	2.602	7.992	375.82
4080	2.626	7.896	371.33
4200	2.574	7.860	369.61
4320	2.634	7.918	372.34
4440	2.643	7.893	371.18
4560	2.639	7.836	368.51
4680	2.923	7.762	365.02
4800	2.606	7.778	365.77
4920	2.624	7.761	364.95
5040	2.620	7.730	363.51
5160	2.605	7.688	361.51
5280	2.618	7.728	363.40
5400	2.639	7.668	360.61
5520	2.628	7.580	356.46
5640	2.634	7.646	359.57
5760	2.631	7.576	356.26
5880	2.638	7.529	354.05
6000	2.657	7.522	353.75
6120	2.620	7.422	349.04

6240	2.623	7.419	348.87
6360	2.625	7.444	350.05
6480	2.634	7.420	348.93
6600	2.646	7.379	346.99
6720	2.637	7.351	345.68
6840	2.636	7.325	344.47
6960	2.646	7.347	345.52
7080	2.645	7.319	344.17
7200	2.650	7.242	340.58
7320	2.638	7.204	338.79
7440	2.680	7.198	338.47
7560	2.626	7.152	336.34
7680	2.611	7.144	335.93
7800	2.663	7.156	336.51
7920	2.671	7.178	337.56
8040	2.631	7.092	333.50
8160	2.645	7.062	332.09
8280	2.633	7.071	332.51
8400	2.631	7.078	332.85
8520	2.634	7.009	329.60
8640	2.647	6.973	327.90



Run 3

K⁺5⁻ 150 mM

15 400 mM

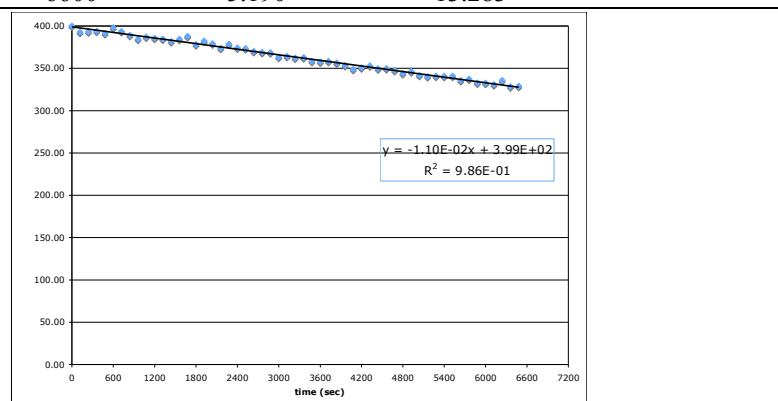
Pd cat. 5 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (2.6 mg, 0.0050 mmol), benzotrifluoride (5 μ L) and **15** (44 μ L, 70 mg, 0.20 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 1.10 x 10⁻² mM s⁻¹**.

Run 2

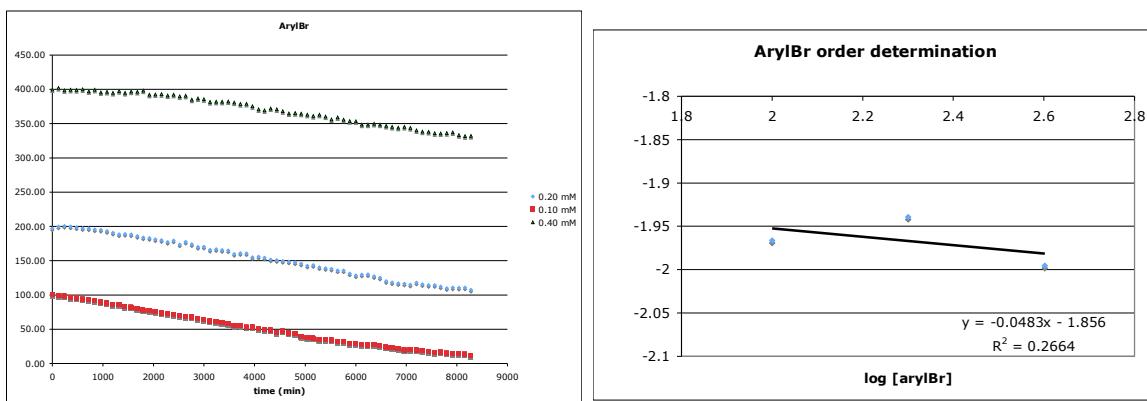
Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	5.088	16.116	400.01
120	5.087	15.823	392.72
240	5.061	15.839	393.12
360	5.124	15.868	393.85
480	5.136	15.762	391.22
600	5.129	16.033	397.93
720	5.114	15.862	393.69
840	5.119	15.671	388.97
960	5.122	15.492	384.52
1080	5.063	15.600	387.18
1200	5.072	15.535	385.59
1320	5.067	15.494	384.55
1440	5.102	15.384	381.82
1560	5.103	15.489	384.43
1680	5.092	15.630	387.93
1800	5.102	15.229	377.98
1920	5.108	15.395	382.12
2040	5.125	15.268	378.96
2160	5.129	15.058	373.73
2280	5.105	15.255	378.63
2400	5.125	15.073	374.11
2520	5.076	15.043	373.36
2640	5.117	14.913	370.15
2760	5.091	14.861	368.85
2880	5.144	14.846	368.48
3000	5.085	14.637	363.29
3120	5.105	14.670	364.12
3240	5.142	14.592	362.16
3360	5.106	14.611	362.66
3480	5.142	14.441	358.43
3600	5.098	14.412	357.71
3720	5.131	14.434	358.25
3840	5.113	14.359	356.39
3960	5.091	14.243	353.52
4080	5.087	14.059	348.94
4200	5.177	14.129	350.69
4320	5.171	14.225	353.07
4440	5.219	14.085	349.60
4560	5.177	14.099	349.93
4680	5.250	14.010	347.74
4800	5.132	13.857	343.93
4920	5.230	13.947	346.16
5040	5.190	13.766	341.67
5160	5.196	13.709	340.26
5280	5.208	13.726	340.69
5400	5.207	13.719	340.50
5520	5.235	13.729	340.76
5640	5.190	13.528	335.77
5760	5.186	13.597	337.47
5880	5.187	13.411	332.86
6000	5.161	13.400	332.59
6120	5.198	13.326	330.76

6240	5.255	13.530	335.81
6360	5.116	13.226	328.27
6480	5.179	13.245	328.75
6600	5.190	13.283	329.67



$$\text{avg. rate} = 1.01 \times 10^{-2} \text{ mM s}^{-1} \pm 1.3 \times 10^{-3}$$

Order in 15 with $(t\text{-Bu}_3\text{P})_2\text{Pd}$:



Order in Pd with (*t*-Bu₃P)₂PdK⁺**5**⁻ 150 mM**15** 100 mM

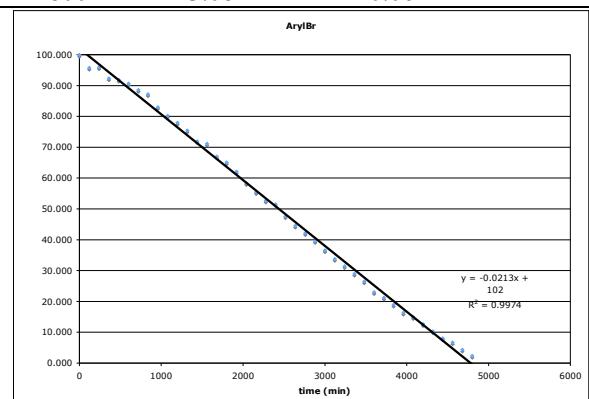
Pd cat. 10 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (5.1 mg, 0.010 mmol), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺**5**⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 2.13 x 10⁻² mM s⁻¹**.

Run 1

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	2.881	2.434	99.959
120	2.925	2.366	95.725
240	2.930	2.375	95.921
360	2.925	2.282	92.340
480	2.941	2.282	91.820
600	2.927	2.241	90.621
720	2.946	2.205	88.589
840	2.927	2.157	87.199
960	2.948	2.066	82.945
1080	2.976	2.017	80.188
1200	2.970	1.956	77.924
1320	2.999	1.911	75.425
1440	3.014	1.832	71.905
1560	2.974	1.787	71.121
1680	3.016	1.707	66.966
1800	3.028	1.665	65.063
1920	3.015	1.582	62.103
2040	3.045	1.501	58.331
2160	3.062	1.434	55.410
2280	3.033	1.350	52.661
2400	3.040	1.321	51.444
2520	3.055	1.228	47.562
2640	3.056	1.149	44.488
2760	3.069	1.092	42.122
2880	3.054	1.021	39.575
3000	3.042	0.940	36.580
3120	3.045	0.869	33.776
3240	3.034	0.805	31.398
3360	3.053	0.747	28.939
3480	3.055	0.684	26.504
3600	3.069	0.596	22.992
3720	3.029	0.542	21.182
3840	3.079	0.492	18.913
3960	3.080	0.425	16.335
4080	3.074	0.388	14.931

4200	3.093	0.331	12.666
4320	3.036	0.264	10.283
4440	3.103	0.211	8.037
4560	3.085	0.174	6.677
4680	3.092	0.113	4.324
4800	3.084	0.061	2.327



K⁺5⁻ 150 mM

15 100 mM

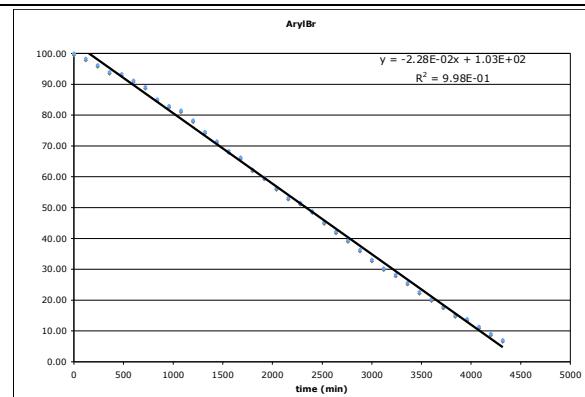
Pd cat. 10 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (5.1 mg, 0.010 mmol), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 2.28 x 10⁻² mM s⁻¹**.

Run 2

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	3.160	2.534	100.01
120	3.175	2.491	98.31
240	3.149	2.439	96.26
360	3.161	2.382	94.01
480	3.139	2.366	93.37
600	3.150	2.312	91.22
720	3.158	2.260	89.19
840	3.145	2.157	85.13
960	3.171	2.103	83.00
1080	3.176	2.067	81.56
1200	3.154	1.985	78.32
1320	3.143	1.891	74.62
1440	3.155	1.813	71.55
1560	3.157	1.733	68.39
1680	3.132	1.680	66.28

1800	3.196	1.580	62.37
1920	3.165	1.514	59.76
2040	3.195	1.428	56.34
2160	3.169	1.348	53.21
2280	3.182	1.307	51.57
2400	3.176	1.239	48.88
2520	3.154	1.148	45.32
2640	3.172	1.071	42.28
2760	3.166	1.001	39.52
2880	3.124	0.922	36.39
3000	3.207	0.840	33.16
3120	3.150	0.770	30.37
3240	3.196	0.717	28.29
3360	3.195	0.650	25.65
3480	3.171	0.575	22.68
3600	3.191	0.515	20.32
3720	3.194	0.454	17.90
3840	3.178	0.385	15.18
3960	3.190	0.352	13.88
4080	3.192	0.290	11.44
4200	3.164	0.232	9.15
4320	3.138	0.179	7.06



K⁺5⁻ 150 mM

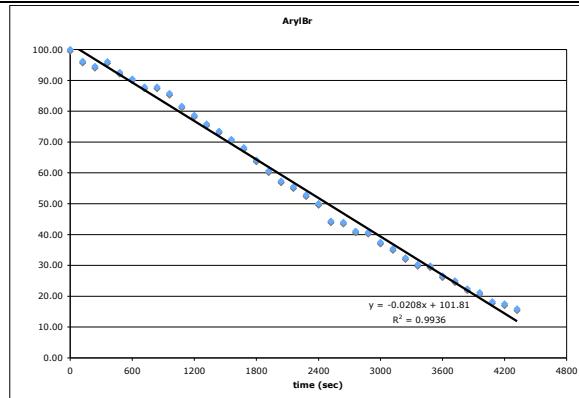
15 100 mM

Pd cat. 10 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (5.1 mg, 0.010 mmol), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 2.08 x 10⁻² mM s⁻¹**.

Run 3

Time (s)	Integral IS (-63.2 ppm)	Integral 25 (-115.9 ppm)	Normalized (mM)
0	3.160	2.534	99.99
120	3.175	2.491	96.21
240	3.149	2.439	94.49
360	3.161	2.382	96.07
480	3.139	2.366	92.59
600	3.150	2.312	90.44
720	3.158	2.260	87.80
840	3.145	2.157	87.83
960	3.171	2.103	85.77
1080	3.176	2.067	81.60
1200	3.154	1.985	78.69
1320	3.143	1.891	75.84
1440	3.155	1.813	73.53
1560	3.157	1.733	70.86
1680	3.132	1.680	68.19
1800	3.196	1.580	64.16
1920	3.165	1.514	60.65
2040	3.195	1.428	57.37
2160	3.169	1.348	55.50
2280	3.182	1.307	52.85
2400	3.176	1.239	50.09
2520	3.154	1.148	44.38
2640	3.172	1.071	43.98
2760	3.166	1.001	41.12
2880	3.124	0.922	40.69
3000	3.207	0.840	37.52
3120	3.150	0.770	35.41
3240	3.196	0.717	32.49
3360	3.195	0.650	30.31
3480	3.171	0.575	29.83
3600	3.191	0.515	26.60
3720	3.194	0.454	25.01
3840	3.178	0.385	22.41
3960	3.190	0.352	21.17
4080	3.192	0.290	18.22
4200	3.164	0.232	17.55
4320	3.138	0.179	15.89



avg. rate = $2.16 \times 10^{-2} \text{ mM s}^{-1} \pm 1.0 \times 10^{-3}$

K⁺5⁻ 150 mM

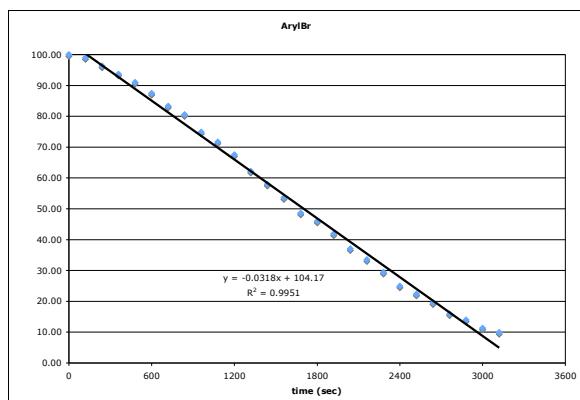
15 100 mM

Pd cat. 15 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (7.6 mg, 0.015 mmol), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 3.18 x 10⁻² mM s⁻¹**.

Run 1

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	2.415	3.030	99.99
120	2.364	3.000	99.00
240	2.377	2.919	96.33
360	2.392	2.837	93.63
480	2.364	2.757	90.98
600	2.399	2.648	87.40
720	2.359	2.524	83.30
840	2.324	2.441	80.56
960	2.391	2.269	74.88
1080	2.323	2.171	71.65
1200	2.359	2.045	67.50
1320	2.384	1.883	62.14
1440	2.398	1.757	57.98
1560	2.372	1.625	53.63
1680	2.359	1.472	48.57
1800	2.381	1.392	45.96
1920	2.344	1.267	41.82
2040	2.347	1.122	37.04
2160	2.372	1.015	33.50
2280	2.346	0.891	29.39
2400	2.374	0.755	24.92
2520	2.305	0.676	22.31
2640	2.313	0.592	19.53
2760	2.314	0.482	15.90
2880	2.301	0.423	13.95
3000	2.304	0.343	11.31
3120	2.340	0.300	9.90
3240	2.347	0.257	8.47
3360	2.329	0.226	7.44



$\text{K}^+ \text{5}^-$ 150 mM

15 100 mM

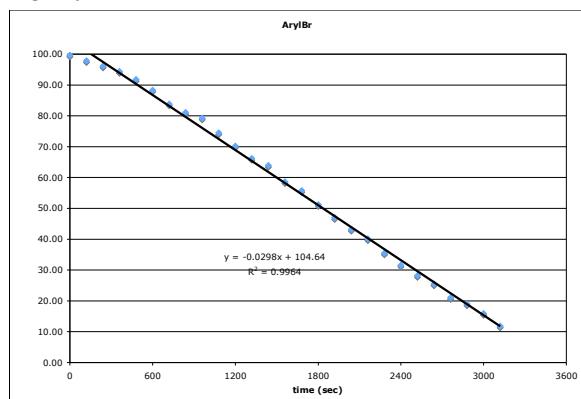
Pd cat. 15 mM

Following General Procedure IV, a mixture of $(t\text{-Bu}_3\text{P})_2\text{Pd}$ (7.6 mg, 0.015 mmol), benzotrifluoride (5 μL) and **24** (11 μL , 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by $\text{K}^+ \text{12}^-$ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ^{19}F NMR spectroscopy. **rate = $2.98 \times 10^{-2} \text{ mM s}^{-1}$** .

Run 2

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	5.023	4.397	99.64
120	5.012	4.319	97.88
240	5.042	4.242	96.12
360	5.109	4.166	94.40
480	5.117	4.050	91.77
600	5.148	3.896	88.29
720	5.049	3.698	83.80
840	5.028	3.578	81.08
960	5.023	3.498	79.26
1080	5.006	3.286	74.47
1200	5.050	3.098	70.19
1320	5.041	2.920	66.17
1440	5.522	2.618	63.92
1560	5.045	2.587	58.63
1680	5.038	2.462	55.78
1800	5.081	2.259	51.19
1920	5.073	2.073	46.99
2040	5.080	1.906	43.20
2160	5.096	1.769	40.08
2280	5.090	1.566	35.47
2400	5.069	1.392	31.55

2520	5.053	1.244	28.19
2640	5.091	1.123	25.44
2760	5.097	0.931	21.09
2880	5.113	0.839	19.01
3000	5.101	0.703	15.93
3120	5.090	0.523	11.85
3240	5.060	0.471	10.67



$\text{K}^+ \text{5}^-$ 150 mM

15 100 mM

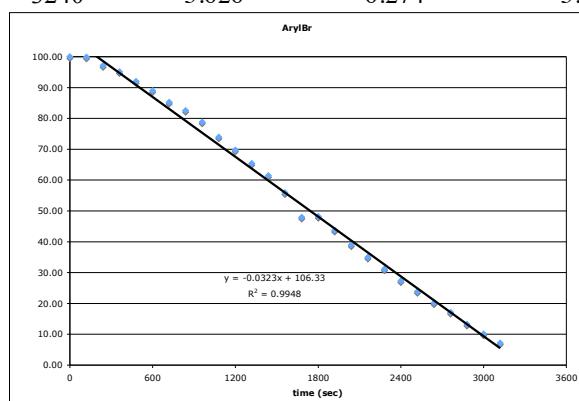
Pd cat. 15 mM

Following General Procedure IV, a mixture of $(t\text{-Bu}_3\text{P})_2\text{Pd}$ (7.6 mg, 0.015 mmol), benzotrifluoride (5 μL) and **15** (11 μL , 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by $\text{K}^+ \text{5}^-$ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ^{19}F NMR spectroscopy. **rate = $3.23 \times 10^{-2} \text{ mM s}^{-1}$** .

Run 3

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	5.038	4.680	100.00
120	5.020	4.674	99.86
240	5.021	4.545	97.12
360	5.027	4.451	95.10
480	5.022	4.304	91.96
600	4.993	4.163	88.96
720	5.005	3.987	85.20
840	5.019	3.863	82.55
960	5.000	3.688	78.81
1080	5.085	3.461	73.95
1200	5.061	3.262	69.69
1320	5.040	3.061	65.40
1440	4.993	2.874	61.41
1560	4.999	2.614	55.85

1680	5.612	2.245	47.96
1800	4.983	2.254	48.17
1920	4.993	2.046	43.73
2040	5.017	1.823	38.96
2160	5.031	1.638	34.99
2280	5.062	1.458	31.15
2400	5.024	1.278	27.31
2520	5.016	1.116	23.86
2640	5.027	0.946	20.21
2760	5.009	0.801	17.11
2880	5.010	0.620	13.24
3000	5.055	0.472	10.09
3120	5.059	0.336	7.18
3240	5.020	0.274	5.85



$$\text{avg. rate} = 3.13 \times 10^{-2} \text{ mM s}^{-1} \pm 1.3 \times 10^{-3}$$

K⁺5⁻ 150 mM

15 100 mM

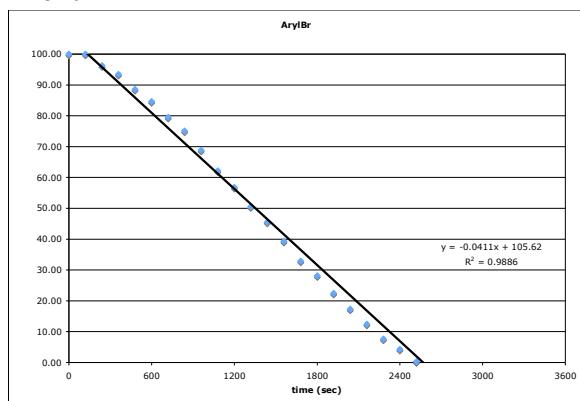
Pd cat. 20 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (10.2 mg, 0.020 mmol), benzotrifluoride (5 µL) and **15** (11 µL, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 4.11 × 10⁻² mM s⁻¹**.

Run 1

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	2.813	2.571	100.00
120	2.757	2.570	99.98
240	2.806	2.474	96.23
360	2.818	2.403	93.45
480	2.800	2.277	88.55

600	2.790	2.176	84.62
720	2.788	2.044	79.51
840	2.780	1.929	75.04
960	2.799	1.770	68.86
1080	2.809	1.599	62.19
1200	2.777	1.460	56.81
1320	2.814	1.299	50.53
1440	2.771	1.171	45.54
1560	2.789	1.011	39.33
1680	2.811	0.845	32.87
1800	2.797	0.724	28.17
1920	2.759	0.577	22.43
2040	2.777	0.445	17.31
2160	2.765	0.319	12.40
2280	2.797	0.196	7.63
2400	2.821	0.109	4.25
2520	2.766	0.008	0.30



K⁺5⁻ 150 mM

15 100 mM

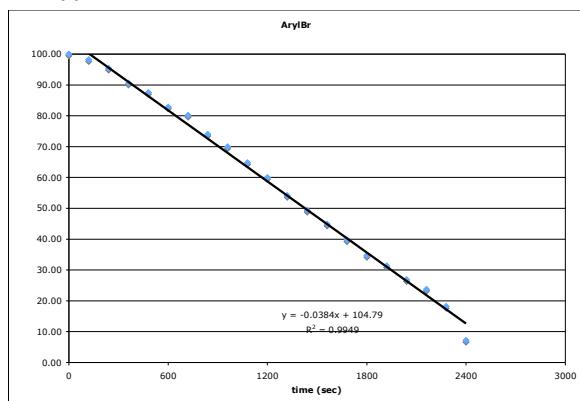
Pd cat. 20 mM

Following General Procedure IV, a mixture of (*t*-Bu₃P)₂Pd (10.2 mg, 0.020 mmol), benzotrifluoride (5 μ L) and **15** (11 μ L, 17.5 mg, 0.10 mmol) were dissolved in toluene (0.8 mL), followed by K⁺5⁻ (33 mg, 0.15 mmol) and toluene (0.2 mL) to afford a clear, colorless solution. The tube was placed into a preheated 95 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 3.84 x 10⁻² mMs⁻¹**.

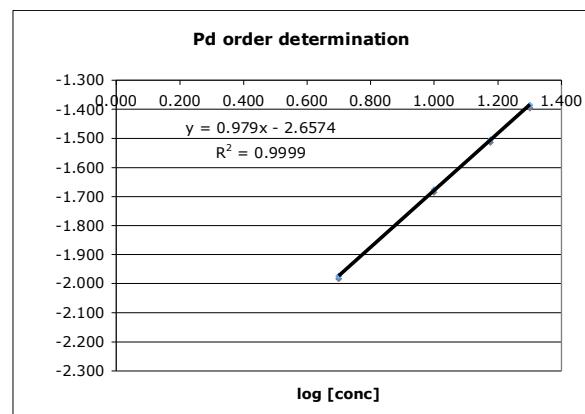
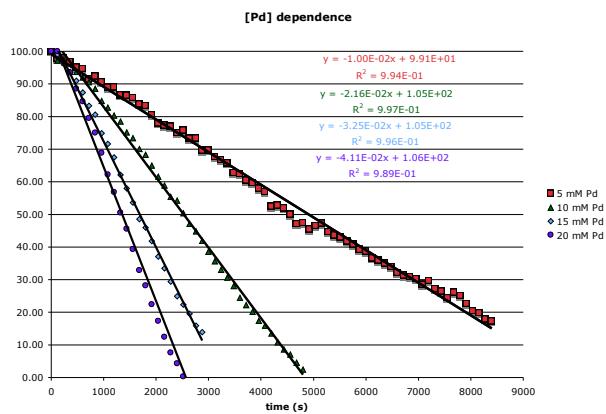
Run 2

Time (s)	Integral IS (-63.2 ppm)	Integral 15 (-115.9 ppm)	Normalized (mM)
0	2.661	3.132	100.00
120	2.633	3.076	98.21
240	2.610	2.988	95.40
360	2.569	2.837	90.59

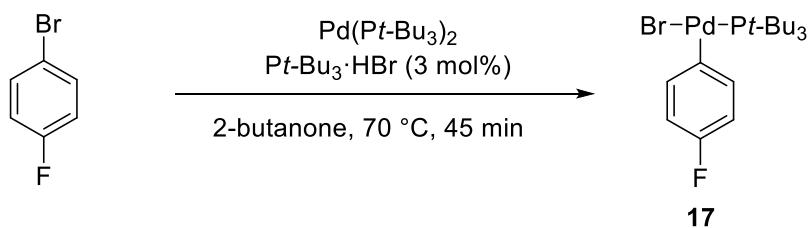
480	2.591	2.741	87.52
600	2.537	2.594	82.82
720	2.509	2.509	80.12
840	2.508	2.317	73.97
960	2.492	2.189	69.90
1080	2.548	2.028	64.74
1200	2.462	1.878	59.96
1320	2.491	1.695	54.13
1440	2.517	1.544	49.31
1560	2.451	1.405	44.87
1680	2.446	1.246	39.77
1800	2.413	1.087	34.70
1920	2.435	0.981	31.34
2040	2.446	0.842	26.90
2160	2.390	0.742	23.70
2280	2.406	0.569	18.18
2400	2.397	0.223	7.12



Order in Pd with $(t\text{-Bu}_3\text{P})_2\text{Pd}$:



Preparation of Bromo(4-fluorophenyl)(tri-*tert*-butylphosphine)palladium (17**) (Scheme 7)**



In a drybox, an oven-dried, 100-mL Schlenk flask was charged with $\text{Pd}(\text{P}t\text{-Bu}_3)_2$ (600 mg, 1.17 mmol, 1.0 equiv) and $t\text{-Bu}_3\text{P}\bullet\text{HBr}$ (10.0 mg, 0.035 mmol, 0.03 equiv) as solids. Degassed 2-butanone (24 mL) was then added, followed by 4-fluorobromobenzene (5.2 mL, 46.9 mmol, 40 equiv). The flask was sealed and removed from the drybox. The flask was then purged with Ar and submerged into a preheated 70 °C oil bath. After 45 min, the flask was removed from the oil bath and the reaction mixture was cooled to room temperature. The volatiles were removed *in vacuo* (0.1 mmHg) to afford an orange solid. The flask was again sealed and returned to a drybox. Toluene (10 mL) was added to the solid residue, resulting in a heterogeneous mixture. The mixture was added to vigorously stirring pentane (60 mL), which resulted in the precipitation of a yellow solid. The solid was collected on an oven-dried M porosity fritted funnel and washed with additional pentane (10 mL). The solid was dried *in vacuo* (0.1 mmHg) for 6 h to afford 465 mg (82%) of **17** as a yellow powder. The data for **17** matched those reported in the literature.²¹

Data for **17:**

¹H NMR: (500 MHz, C₆D₆)

7.25 (m, 2H), 6.65 (t, $J = 8.9$ Hz, 2H), 1.05 (d, $J = 12.5$ Hz, 27H)

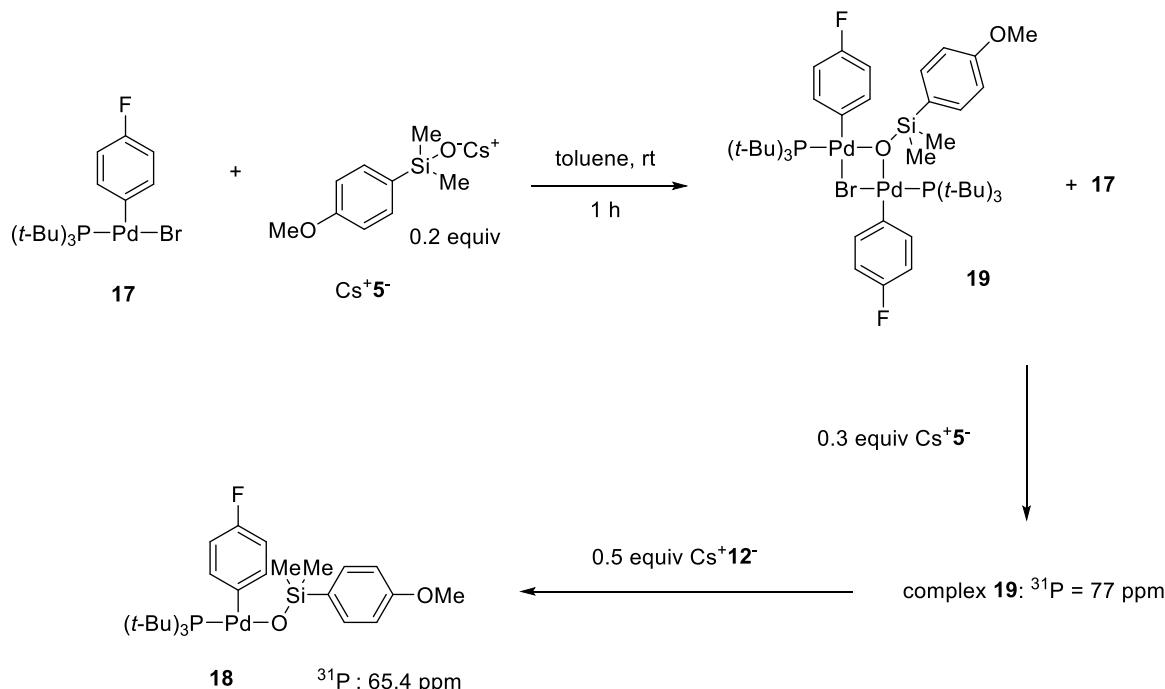
¹⁹F NMR: (470 MHz, C₆D₆)

−122.4

³¹P NMR: (202 MHz, C₆D₆)

66.5

Study of Displacement Reaction: Arylpalladium(II) Bromide **17** with $\text{Cs}^+\text{5}^-$ (Scheme 8)

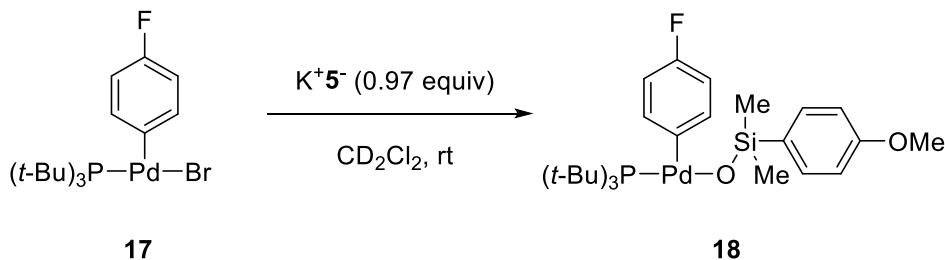


In a drybox, an oven-dried, 3-mL vial equipped with a magnetic stir bar was charged with **17** (5.2 mg, 0.011 mmol) and toluene (0.6 mL) resulting in an orange solution. Then, $\text{Cs}^+\text{5}^-$ (0.8 mg, 0.002 mmol, 0.2 equiv) was added as a solid and the mixture was stirred until the mixture was completely homogeneous. The resulting solution was transferred into an oven-dried, 5-mm NMR tube using a syringe and the tube was sealed with a septum and Parafilm. The sample was analyzed by ^{31}P NMR spectroscopy. The ^{31}P NMR spectrum contained a resonance at 77.0 ppm in addition to that for **17** (64.5 ppm). The NMR tube was returned to the drybox and an additional 0.3 equiv of $\text{Cs}^+\text{5}^-$ (0.9 mg, 0.003 mmol, 0.3 equiv) was added. The mixture was shaken to ensure homogeneity and the sample again analyzed by ^{31}P NMR spectroscopy. The signal for **17** completely vanished leaving a species (77.0 ppm) visible in the ^{31}P NMR as the major constituent, with a visible signal at 65.4 ppm.²² The stoichiometry of this experiment suggested that displacement occurs rapidly and the mutual coexistence of the bromide complex **17** and silanolate resulted in a dimerization (**19**). Subsequent addition of 0.5 equiv of $\text{Cs}^+\text{5}^-$ (1.5 mg, 0.006 mmol, 0.5 equiv) caused disappearance of the signal at 77.0 ppm with concomitant formation of a new signal at 65.4 ppm. This new species has been established as the expected palladium(II) silanolate complex **18**.

Interestingly, the chemical environments of the ^{31}P nuclei in palladium(II) bromide **17**

and palladium(II) silanolate **18** were sufficiently similar to cause the coincidence of the ^{31}P resonances. This result was initially confusing because it appeared that no displacement took place even after 30 min. However, the titration study described above eliminated this possibility and allowed for the subsequent studies to be undertaken.

Spectroscopic Identification and Crystallization of (4-Fluorophenyl)[1-(4-methoxyphenyl)-1,1,-dimethylsilanolato- κ -O](tris-*tert*-butylphosphine)palladium (18**)**



In a drybox, an oven-dried, 3-mL vial was charged with **17** (12.2 mg, 0.0252 mmol) and CD_2Cl_2 (0.4 mL) resulting in an orange solution upon vigorous mixing using a magnetic stir bar. Then, $\text{K}^+\text{5}^-$ (5.4 mg, 0.0244 mmol, 0.97 equiv) was added as a solid and the sides of the vial were rinsed with CD_2Cl_2 (0.4 mL). The mixture was vigorously mixed (Vortex) for ca. 5 min to ensure complete dissolution. As the mixture stirred, a precipitate formed on the sides of the vial. The supernatant solution was transferred into an oven-dried, 5 mM NMR tube using a 25 G needle to ensure none of the precipitate was transferred. The tube was sealed with a septum and wrapped with Parafilm to exclude any oxygen or moisture. The NMR tube was removed from the drybox and placed into a $\text{CO}_2/\text{i-PrOH}$ bath. The tube was then inserted into a cooled -70 °C NMR probe and the product was analyzed by NMR spectroscopy.

Data for **18**:

$^1\text{H NMR}$: (500 MHz, CD_2Cl_2)

7.51 (d, 2 H, $J = 7.3$ Hz), 7.07 (br m, 2 H), 6.83 (d, 2 H, $J = 7.6$ Hz), 6.65 (t, 2 H, $J = 8.5$ Hz), 3.75 (s, 3 H), 1.24 (d, 9 H, $J_{\text{PH}} = 12.5$ Hz), -0.16 (s, 6 H)

$^{13}\text{C NMR}$: (126 MHz, CD_2Cl_2)

160.2 (d, $J_{\text{CF}} = 235$ Hz), 159.0, 136.2, 134.4, 134.2, 126.7, 113.0 (d, $J_{\text{CF}} = 19.8$ Hz), 112.7, 112.3, 54.7, 39.1 (d, $J_{\text{CP}} = 13$ Hz), 31.0, 1.6

¹⁹F NMR: (470 MHz, CD₂Cl₂)

-123.5

³¹P NMR: (202 MHz, CD₂Cl₂)

61.4

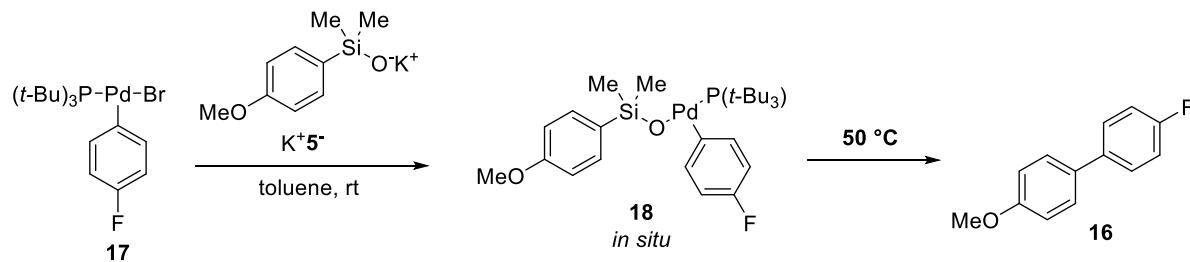
X-ray quality crystals were obtained following an identical procedure as described above with the following modifications. After the reaction mixture was stirred for ca. 5 min, the supernatant solution was transferred to a separate oven-dried vial using a 25 G needle to ensure none of the inorganic salts were transferred. The volatiles were removed in vacuo (2 mm Hg) to afford a brown semi-solid. The residue was dissolved in a minimal amount of pentanes (1 mL) and the vial was sealed with a cap. The vial was thoroughly wrapped in Parafilm to exclude any oxygen or moisture and the vial removed from the drybox. The vial was then placed into a fitted Styrofoam container and inserted into a -80 °C freezer. Small yellow needles formed and were analyzed by X-ray diffraction.

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General Procedure V: Kinetic Measurements for Transmetalation Employing K⁺5⁻ at 50 °C (Table 7).

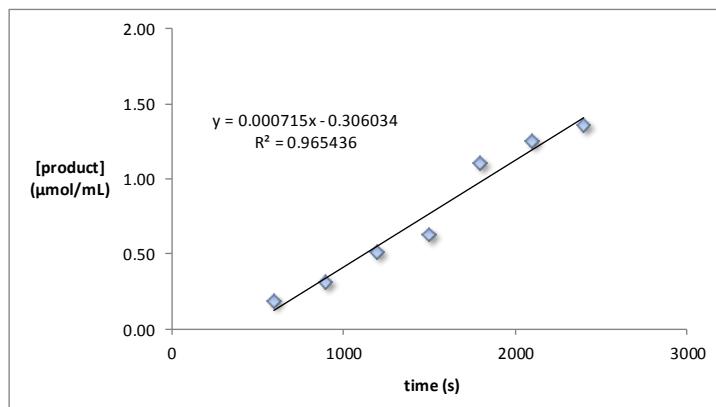


In a drybox, an oven-dried, 4-mL vial was charged with **17** (8.5 mg, 0.0175 mmol) and toluene, resulting in an orange solution. Then, a stock solution of K⁺5⁻ and 1,4-difluorobenzene in toluene was added to a total volume of 0.7 mL (0.025 M in **17**) and the mixture was vigorously mixed. The resulting solution was transferred via syringe into an oven-dried, 5-mm NMR tube. The tube was sealed with a cap and wrapped with Parafilm to exclude any oxygen or moisture. The NMR tube was removed from the drybox and inserted into a preheated 50 °C NMR probe. The temperature of the reaction solution was allowed to equilibrate for ca. 60 seconds prior to data collection. The reaction progress was monitored via ¹⁹F NMR spectroscopy by the appearance of the product **16** ($\delta = -117.2$ ppm) as compared to an internal reference (1,4-difluorobenzene, $\delta = -119.9$ ppm) using the following parameters: at = 0.5 s, d1 = 1 s, pw90 = 15 μ s, pw = pw90/2, sw = -115 to -130 ppm, nt and sampling as specified in each individual experiment. Experiments were repeated in triplicate.

0.5 equivalents (Table 7, entry 1)

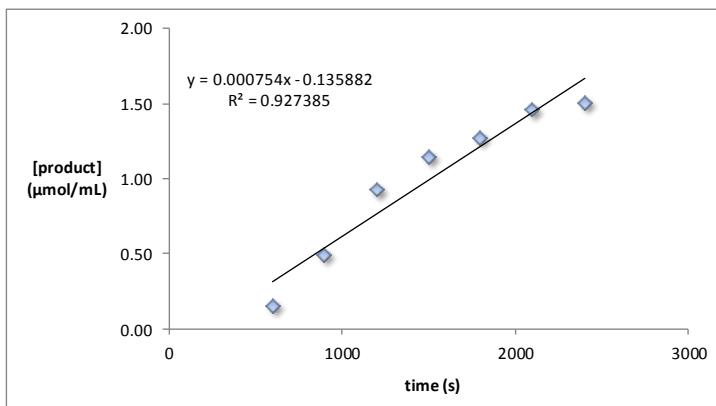
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.55 mL). A stock solution of K⁺5⁻ (0.00875 mmol, 0.5 equiv) and 1,4-difluorobenzene (0.00875 mmol, 0.5 equiv) in toluene (0.15 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = 7.15 · 10⁻⁴ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
600	59.0	7899.0	0.19
900	99.7	7966.5	0.31
1200	160.9	7829.2	0.51
1500	198.7	7933.3	0.63
1800	347.3	7859.8	1.10
2100	397.5	7946.9	1.25
2400	431.9	7906.8	1.37



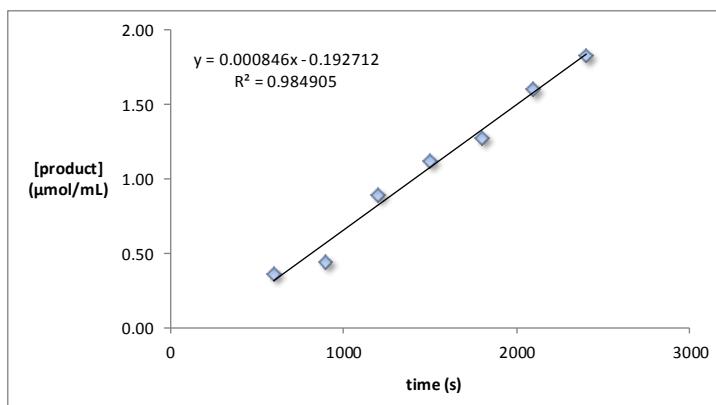
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.58 mL). A stock solution of $\text{K}^+ \mathbf{5}^-$ (0.00875 mmol, 0.5 equiv) and 1,4-difluorobenzene (0.00730 mmol, 0.42 equiv) in toluene (0.12 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $7.54 \cdot 10^{-4}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
600	102.6	14177.6	0.15
900	334.1	14094.4	0.49
1200	631.4	14110.1	0.93
1500	788.8	14293.8	1.15
1800	867.0	14161.5	1.28
2100	1000.3	14293.6	1.46
2400	1041.1	14417.1	1.50



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.58 mL). A stock solution of $\text{K}^+ \mathbf{5}^-$ (0.00875 mmol, 0.5 equiv) and 1,4-difluorobenzene (0.00730 mmol, 0.42 equiv) in toluene (0.12 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $8.46 \cdot 10^{-4}$ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
600	255.5	14532.6	0.37
900	314.8	14764.6	0.44
1200	617.6	14458.0	0.89
1500	783.0	14555.5	1.12
1800	895.5	14591.5	1.28
2100	1109.3	14424.9	1.60
2400	1268.6	14409.6	1.83

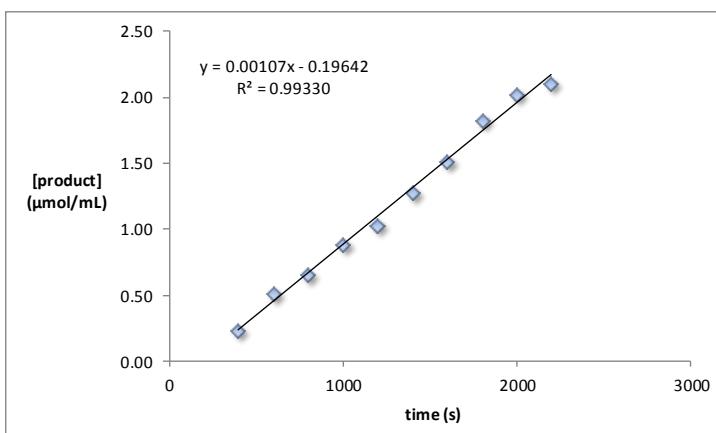


$$\text{avg. rate} = 7.72 \cdot 10^{-4} \pm 6.73 \cdot 10^{-5} \text{ mM/s}$$

0.6 equivalents (Table 7, entry 2)

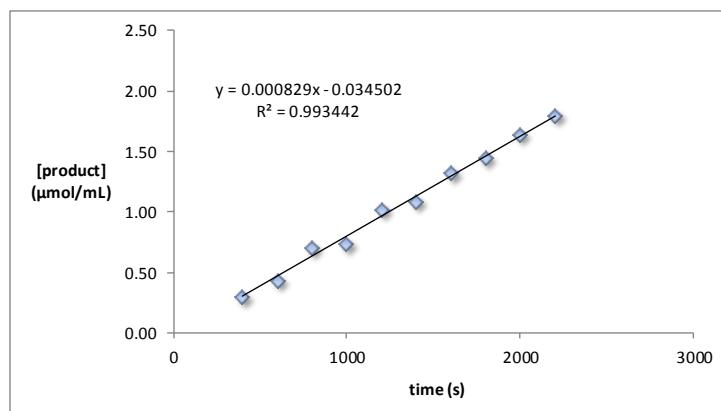
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of K⁺**5**⁻ (0.0105 mmol, 0.6 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 64, sampling every 200 s. **rate = 1.07 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
400	182.7	40901.4	0.22
600	418.3	41623.2	0.50
800	548.5	41909.7	0.65
1000	744.9	42188.0	0.88
1200	860.8	42290.1	1.02
1400	1082.6	42423.6	1.28
1600	1280.4	42381.0	1.51
1800	1543.4	42453.8	1.82
2000	1725.8	42716.9	2.02
2200	1809.6	43114.8	2.10



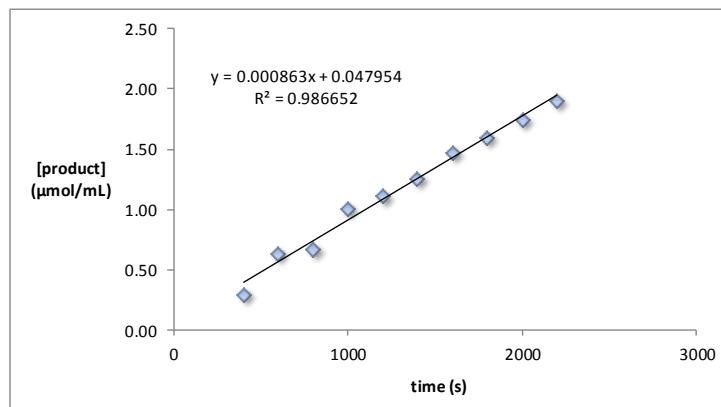
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of K⁺**5**⁻ (0.0105 mmol, 0.6 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 64, sampling every 200 s. **rate = 8.29 · 10⁻⁴ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
400	219.5	37101.6	0.30
600	322.0	37290.4	0.43
800	517.8	37115.9	0.70
1000	549.7	37343.0	0.74
1200	756.1	37306.8	1.01
1400	801.4	37049.9	1.08
1600	988.8	37495.5	1.32
1800	1090.7	37843.8	1.44
2000	1211.6	37135.3	1.63
2200	1333.0	37258.7	1.79



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of $\text{K}^+\text{5}^-$ (0.0105 mmol, 0.6 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 200 s. **rate = $8.63 \cdot 10^{-4}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
400	221.7	37633.7	0.29
600	477.6	37596.4	0.64
800	505.9	37811.5	0.67
1000	755.0	37680.4	1.00
1200	838.6	37654.6	1.11
1400	943.2	37459.2	1.26
1600	1105.5	37456.8	1.48
1800	1198.2	37590.1	1.59
2000	1315.7	37703.2	1.74
2200	1443.6	37855.6	1.91

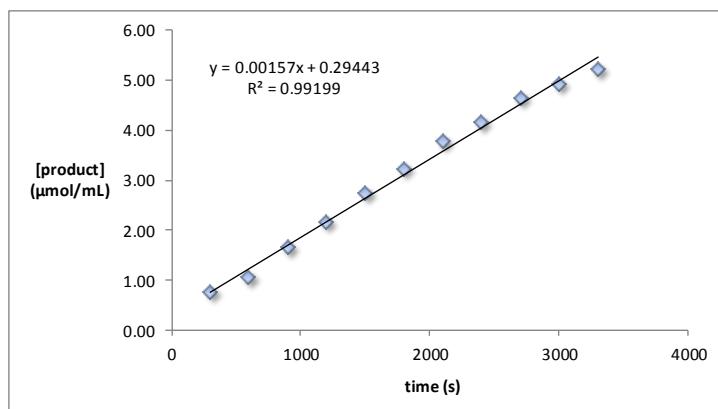


$$\text{avg. rate} = 9.17 \cdot 10^{-4} \pm 1.25 \cdot 10^{-4} \text{ mM/s}$$

0.75 equivalents (Table 7, entry 3)

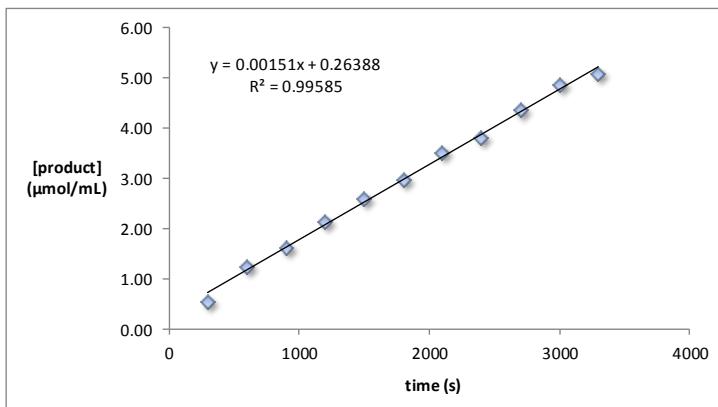
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of $\text{K}^+\text{5}^-$ (0.0131 mmol, 0.75 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $1.57 \cdot 10^{-3}$ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
300	506.2	33125.3	0.76
600	714.0	34233.5	1.04
900	1183.2	35837.1	1.65
1200	1589.1	36782.7	2.16
1500	2037.4	37108.5	2.75
1800	2391.3	37110.4	3.22
2100	2813.9	37213.5	3.78
2400	3118.2	37483.1	4.16
2700	3481.4	37378.4	4.66
3000	3650.7	37185.8	4.91
3300	3889.8	37279.0	5.22



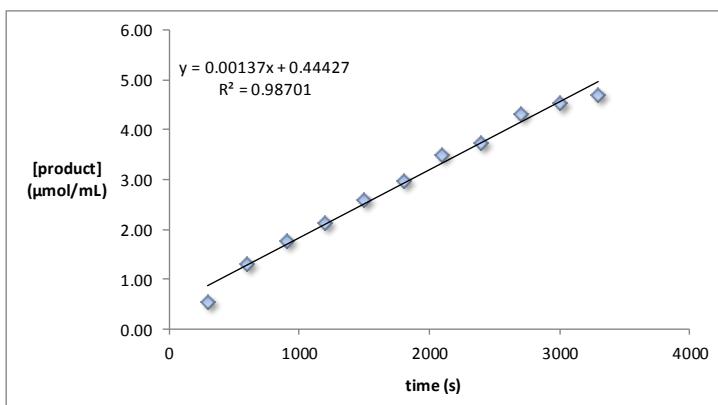
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of $\text{K}^+\text{5}^-$ (0.0131 mmol, 0.75 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $1.51 \cdot 10^{-3}$ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
300	340.9	32395.1	0.53
600	788.6	31889.8	1.24
900	1030.4	31632.8	1.63
1200	1387.2	32326.3	2.15
1500	1710.1	32808.3	2.61
1800	2004.6	33804.8	2.96
2100	2476.3	35157.7	3.52
2400	2764.0	36349.2	3.80
2700	3214.9	36949.6	4.35
3000	3582.7	36881.0	4.86
3300	3756.5	36866.4	5.09



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of $\text{K}^+\text{5}^-$ (0.0131 mmol, 0.75 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $1.37 \cdot 10^{-3}$ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
300	392.3	35774.8	0.55
600	950.6	36069.4	1.32
900	1287.5	36078.9	1.78
1200	1502.2	35472.6	2.12
1500	1860.4	35895.8	2.59
1800	2101.8	35346.7	2.97
2100	2495.8	35719.4	3.49
2400	2624.2	35181.5	3.73
2700	3065.9	35528.3	4.31
3000	3177.2	35063.6	4.53
3300	3355.3	35631.0	4.71

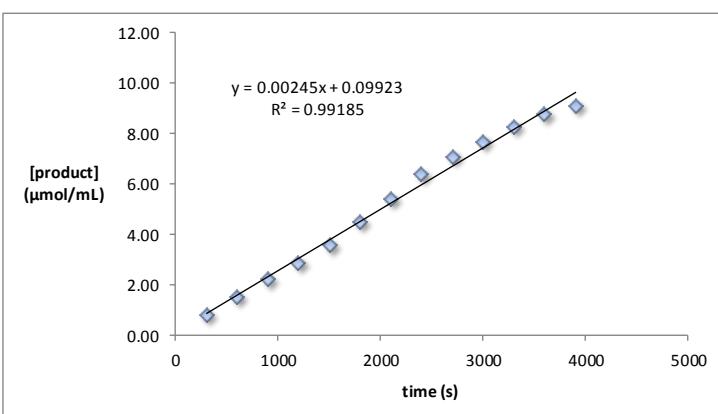


$$\text{avg. rate} = 1.48 \cdot 10^{-3} \pm 1.03 \cdot 10^{-4} \text{ mM/s}$$

1 equivalent (Table 7, entry 4)

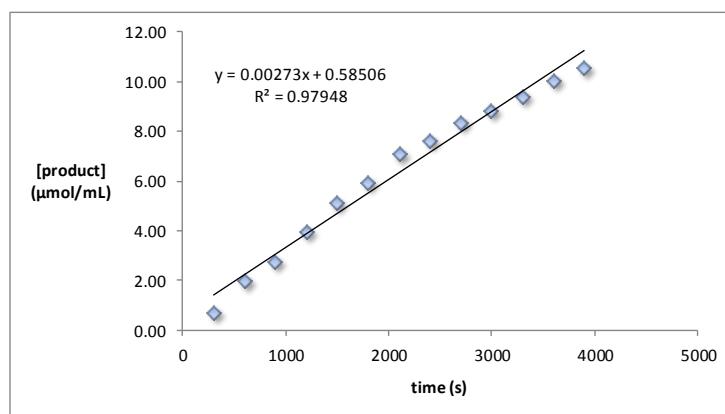
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of K⁺**5**⁻ (0.0175 mmol, 1 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = 2.45 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
300	585.0	35268.7	0.83
600	1093.2	36554.3	1.50
900	1611.3	36673.9	2.20
1200	2085.5	36313.3	2.87
1500	2563.1	35911.3	3.57
1800	3319.3	37187.5	4.46
2100	4004.4	37041.9	5.41
2400	4721.6	37079.1	6.37
2700	5247.0	37091.5	7.07
3000	5672.8	36970.9	7.67
3300	6110.2	36865.8	8.29
3600	6472.0	36842.4	8.78
3900	6648.4	36624.4	9.08



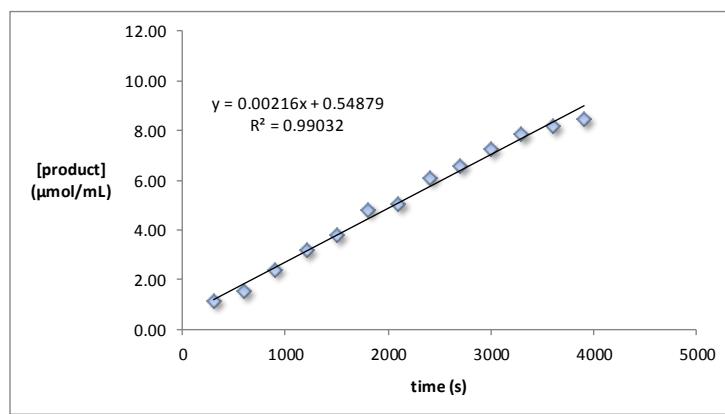
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of K⁺**5**⁻ (0.0175 mmol, 1 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = 2.73 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
300	223.3	15332.2	0.73
600	596.4	15308.3	1.95
900	838.9	15433.4	2.72
1200	1209.7	15441.8	3.92
1500	1530.2	15086.7	5.07
1800	1825.7	15454.1	5.91
2100	2181.0	15451.2	7.06
2400	2378.2	15581.3	7.63
2700	2594.3	15555.2	8.34
3000	2735.6	15472.1	8.84
3300	2885.3	15381.3	9.38
3600	3065.5	15301.2	10.02
3900	3184.2	15075.6	10.56



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of $\text{K}^+\text{5}^-$ (0.0175 mmol, 1 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $2.16 \cdot 10^{-3}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
300	342.2	15288.3	1.12
600	480.0	15539.0	1.54
900	734.7	15436.4	2.38
1200	965.7	15267.6	3.16
1500	1163.2	15411.7	3.77
1800	1483.5	15411.8	4.81
2100	1532.4	15142.0	5.06
2400	1840.3	15175.7	6.06
2700	2006.5	15249.5	6.58
3000	2200.6	15222.3	7.23
3300	2397.9	15295.6	7.84
3600	2508.9	15311.0	8.19
3900	2561.6	15148.1	8.46

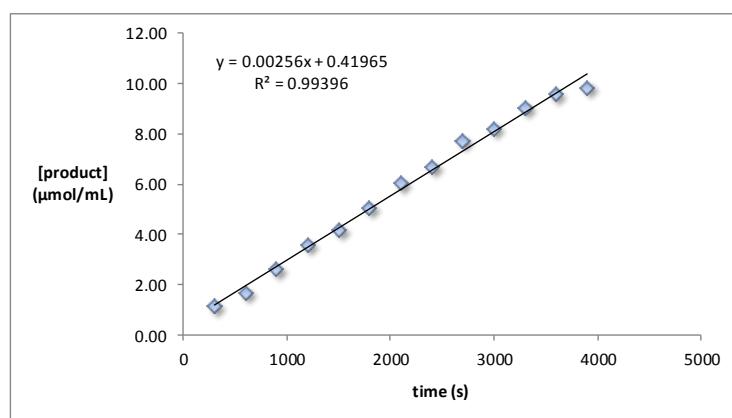


$$\text{avg. rate} = 2.45 \cdot 10^{-3} \pm 2.85 \cdot 10^{-4} \text{ mM/s}$$

1 equivalent - with added $\text{P}(t\text{-Bu})_3$ (2 equivalents)

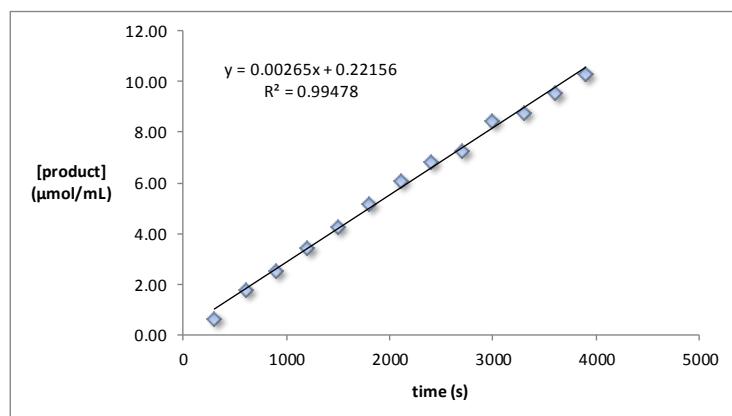
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of $\text{K}^+\text{5}^-$ (0.0175 mmol, 1 equiv), $\text{P}(t\text{-Bu})_3$ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0289 mmol, 1.65 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $2.56 \cdot 10^{-3}$ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
300	513.5	36786.5	1.15
600	753.7	36919.8	1.69
900	1167.3	37076.9	2.60
1200	1613.4	36981.7	3.60
1500	1880.8	37044.1	4.19
1800	2285.7	37294.1	5.06
2100	2746.7	37602.4	6.03
2400	3045.3	37733.2	6.66
2700	3519.5	37649.9	7.72
3000	3774.8	37994.7	8.20
3300	4125.6	37766.7	9.02
3600	4393.8	37928.2	9.57
3900	4538.5	37990.3	9.86



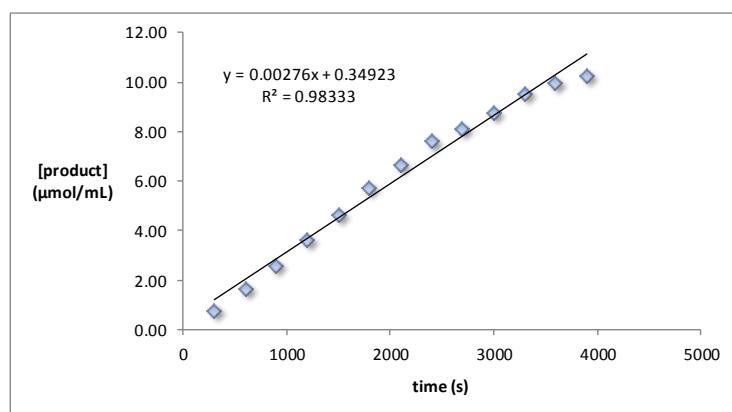
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of $\text{K}^+ \text{5}^-$ (0.0175 mmol, 1 equiv), $\text{P}(t\text{-Bu})_3$ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0289 mmol, 1.65 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $2.65 \cdot 10^{-3}$ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
300	307.4	39198.0	0.65
600	829.8	39050.9	1.75
900	1204.0	39172.5	2.54
1200	1621.8	38923.6	3.44
1500	2021.6	39023.1	4.28
1800	2446.0	39035.5	5.17
2100	2864.8	38972.4	6.07
2400	3231.8	38993.9	6.84
2700	3413.2	38672.1	7.29
3000	3932.7	38299.5	8.48
3300	4113.0	38703.3	8.77
3600	4431.6	38440.2	9.52
3900	4773.1	38300.6	10.29



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.4 mL). A stock solution of $\text{K}^+ \text{5}^-$ (0.0175 mmol, 1 equiv), $\text{P}(t\text{-Bu})_3$ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0289 mmol, 1.65 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = $2.76 \cdot 10^{-3}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
300	454.4	49073.0	0.76
600	988.2	49325.2	1.65
900	1563.8	49560.7	2.61
1200	2160.1	49479.6	3.60
1500	2771.6	49579.2	4.62
1800	3443.7	49698.7	5.72
2100	4028.4	49826.9	6.68
2400	4550.2	49548.2	7.58
2700	4880.3	49615.2	8.12
3000	5243.7	49421.8	8.76
3300	5704.9	49420.1	9.53
3600	5973.9	49424.6	9.98
3900	6143.3	49403.9	10.27

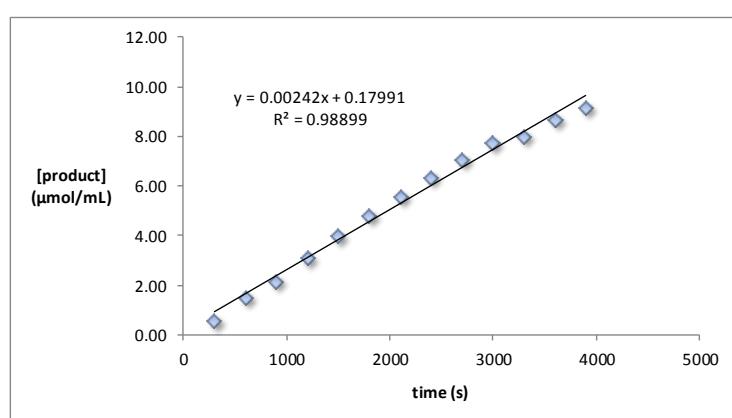


$$\text{avg. rate} = 2.66 \cdot 10^{-3} \pm 1.00 \cdot 10^{-4} \text{ mM/s}$$

1 equivalent - with added P(t-Bu)₃ (5 equivalents)

Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.3 mL). A stock solution of K⁺**5**⁻ (0.0175 mmol, 1 equiv), P(t-Bu)₃ (0.0875 mmol, 5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.3 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 64, sampling every 300 s. **rate = 2.42 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
300	416.0	37751.5	0.55
600	1113.1	37573.6	1.48
900	1622.2	38065.5	2.13
1200	2334.0	37724.4	3.09
1500	2989.6	37602.8	3.98
1800	3620.4	37713.6	4.80
2100	4203.9	37678.7	5.58
2400	4799.8	37731.4	6.36
2700	5345.0	37809.6	7.07
3000	5810.3	37592.0	7.73
3300	6047.7	37741.1	8.01
3600	6499.5	37651.4	8.63
3900	6871.9	37630.5	9.13

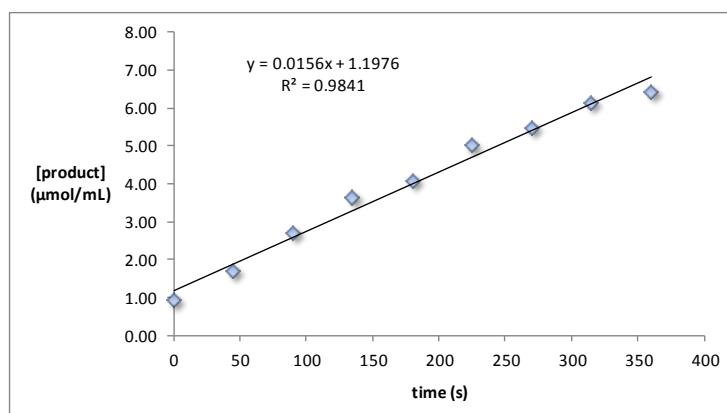


1.5 equivalents (Table 7, entry 5)

Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of K⁺**5**⁻ (0.0263 mmol, 1.5 equiv) and 1,4-difluorobenzene (0.0145 mmol, 0.83 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 16, sampling

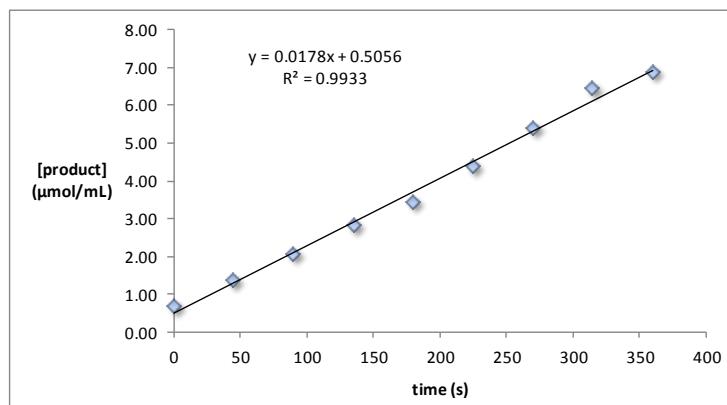
every 45 s. **rate = $1.56 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
0	108.2	4783.5	0.94
45	202.9	4998.0	1.69
90	313.6	4862.9	2.69
135	421.8	4844.8	3.63
180	472.8	4844.9	4.07
225	578.8	4831.3	5.00
270	640.2	4864.7	5.49
315	710.6	4844.6	6.12
360	749.8	4877.9	6.41



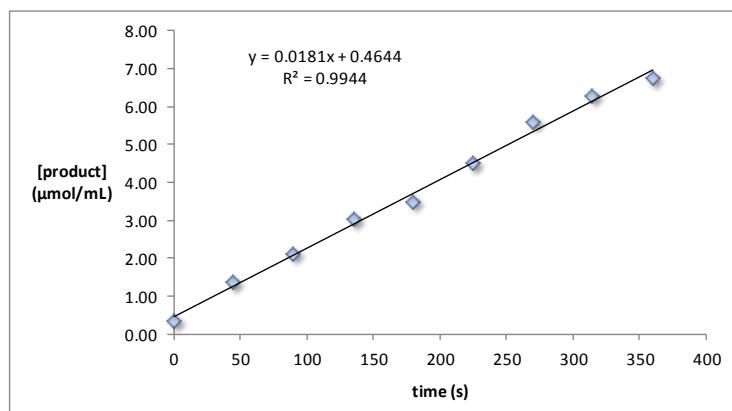
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of $\text{K}^+ \text{5}^-$ (0.0263 mmol, 1.5 equiv) and 1,4-difluorobenzene (0.0145 mmol, 0.83 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = $1.78 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
0	85.3	5150.9	0.69
45	171.8	5200.6	1.38
90	250.5	5103.2	2.05
135	340.8	5041.9	2.82
180	416.8	5071.8	3.43
225	533.6	5085.0	4.38
270	660.2	5097.1	5.40
315	790.3	5124.6	6.43
360	840.2	5098.4	6.87



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of $\text{K}^+ \text{5}^-$ (0.0263 mmol, 1.5 equiv) and 1,4-difluorobenzene (0.0145 mmol, 0.83 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = $1.81 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
0	39.6	4813.9	0.34
45	164.8	5052.9	1.36
90	252.6	4977.8	2.12
135	356.3	4912.6	3.02
180	410.5	4913.9	3.48
225	533.2	4927.4	4.51
270	664.3	4946.8	5.60
315	744.2	4942.9	6.28
360	798.4	4939.9	6.74

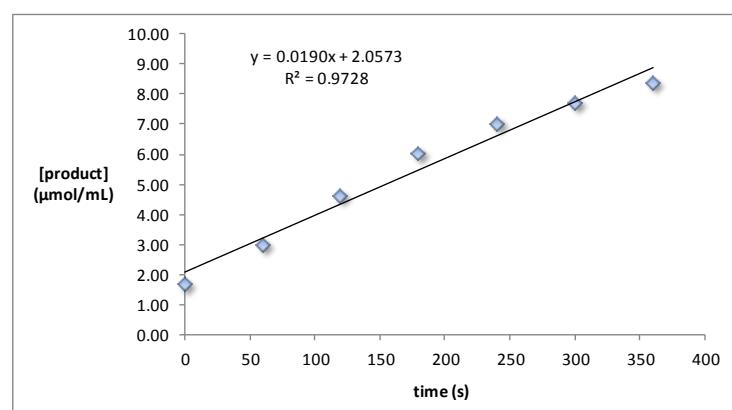


$$\text{avg. rate} = 1.72 \cdot 10^{-2} \pm 1.37 \cdot 10^{-3} \text{ mM/s}$$

2 equivalents (Table 7, entry 6)

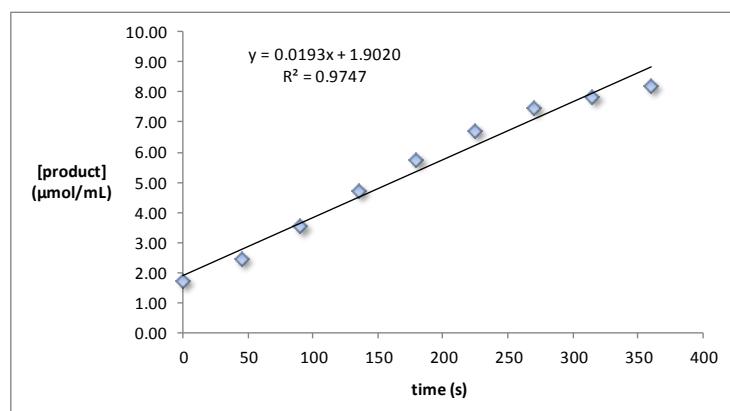
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of K⁺**5**⁻ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0292 mmol, 1.67 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 16, sampling every 60 s. **rate** = $1.90 \cdot 10^{-2}$ mM/s

time (s)	product integral	standard integral	[product] (μmol/mL)
0	380.9	19209.5	1.65
60	665.0	18546.6	2.99
120	1008.2	18249.1	4.61
180	1309.5	18146.7	6.02
240	1528.4	18204.2	7.00
300	1677.0	18211.8	7.68
360	1818.6	18147.7	8.36



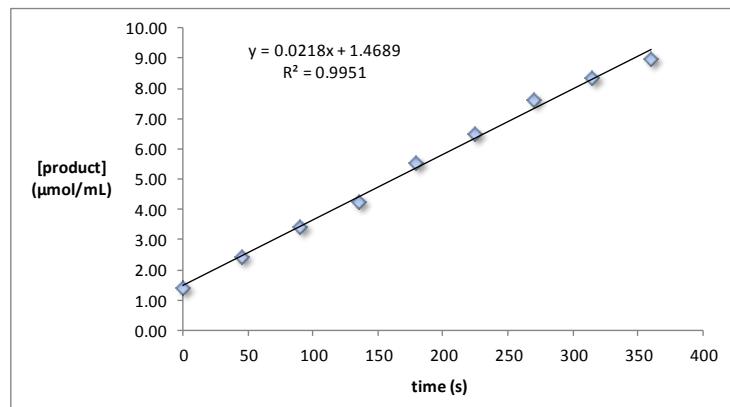
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of K⁺**5**⁻ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0292 mmol, 1.67 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 16, sampling every 45 s. **rate** = $1.93 \cdot 10^{-2}$ mM/s

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
0	388.0	19101.6	1.69
45	523.9	17880.9	2.44
90	738.9	17534.6	3.52
135	997.6	17633.7	4.72
180	1203.2	17413.4	5.76
225	1409.3	17513.9	6.71
270	1573.5	17551.0	7.48
315	1653.1	17633.1	7.82
360	1732.2	17636.5	8.19



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of $\text{K}^+ \mathbf{5}^-$ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0292 mmol, 1.67 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = $2.18 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
0	328.8	19678.2	1.39
45	543.5	18834.1	2.41
90	753.2	18260.6	3.44
135	934.5	18427.2	4.23
180	1217.1	18244.6	5.57
225	1444.4	18600.6	6.48
270	1701.9	18602.9	7.63
315	1879.1	18754.0	8.36
360	1994.1	18568.2	8.96

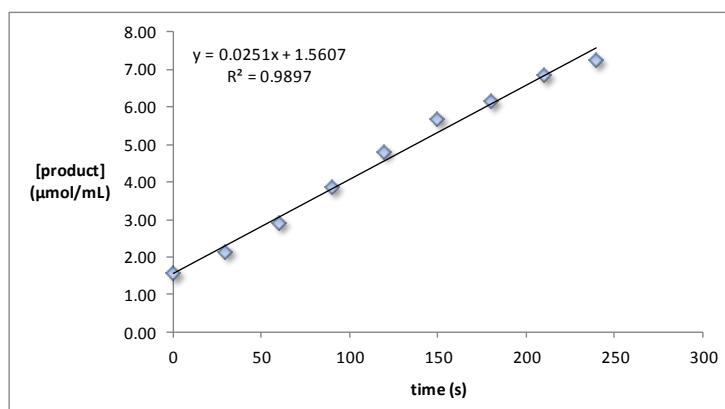


$$\text{avg. rate} = 2.00 \cdot 10^{-2} \pm 1.54 \cdot 10^{-3} \text{ mM/s}$$

3 equivalents (Table 7, entry 7)

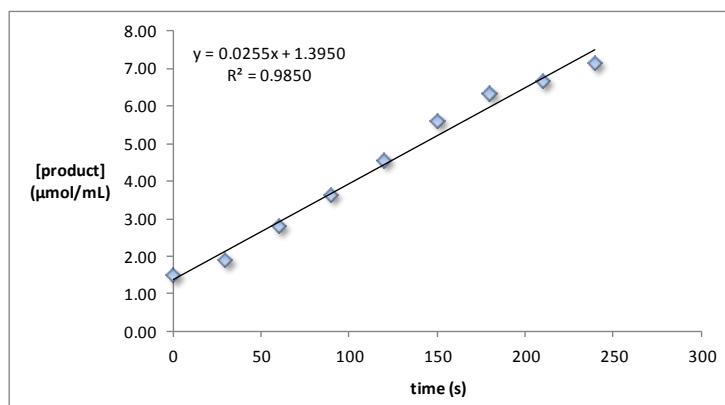
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of $\text{K}^+ \mathbf{5}^-$ (0.0524 mmol, 3 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = $2.51 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
0	216.5	6977.0	1.55
30	289.6	6727.1	2.15
60	374.3	6472.9	2.89
90	488.1	6344.0	3.85
120	612.2	6359.1	4.81
150	726.7	6410.5	5.67
180	787.9	6387.3	6.17
210	865.7	6329.2	6.84
240	919.8	6344.9	7.25



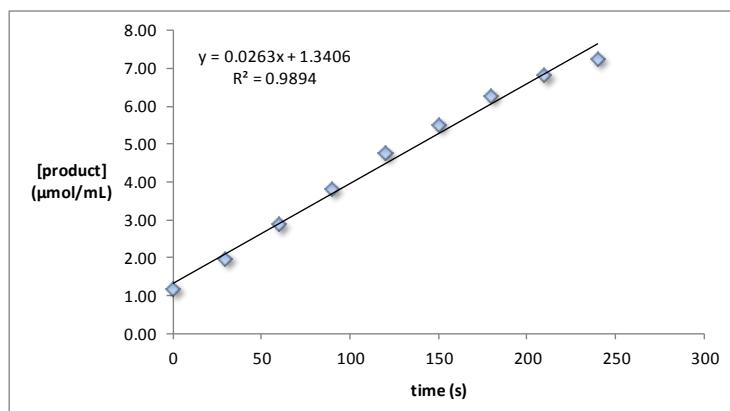
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of $\text{K}^+\text{5}^-$ (0.0524 mmol, 3 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s.
rate = $2.55 \cdot 10^{-2}$ mM/s

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
0	201.4	6757.7	1.49
30	250.5	6612.9	1.89
60	353.5	6351.5	2.78
90	462.3	6324.7	3.65
120	571.6	6274.4	4.56
150	699.0	6249.8	5.59
180	788.0	6207.7	6.35
210	832.2	6258.4	6.65
240	896.0	6279.3	7.13



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of $\text{K}^+\text{5}^-$ (0.0524 mmol, 3 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s.
rate = $2.63 \cdot 10^{-2}$ mM/s

time (s)	product integral	standard integral	[product] (μmol/mL)
0	155.0	6616.8	1.17
30	255.4	6477.6	1.97
60	361.8	6272.3	2.88
90	473.5	6199.7	3.82
120	587.2	6182.7	4.75
150	692.2	6254.7	5.53
180	780.5	6211.8	6.28
210	851.3	6255.9	6.80
240	899.8	6202.7	7.25



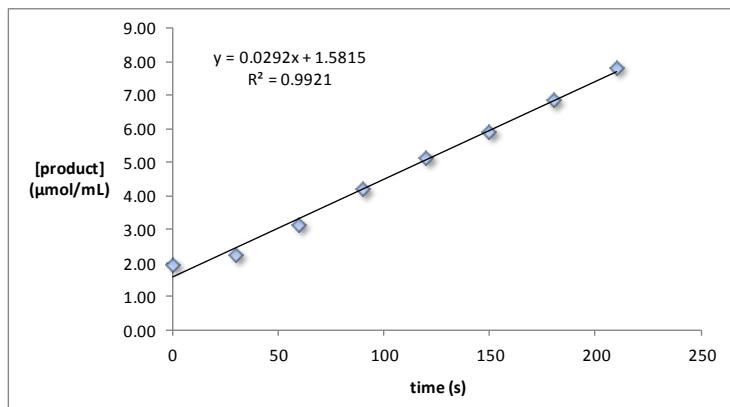
$$\text{avg. rate} = 2.56 \cdot 10^{-2} \pm 6.11 \cdot 10^{-4} \text{ mM/s}$$

5 equivalents (Table 7, entry 8)

Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of K⁺**5**⁻ (0.0875 mmol, 5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 8, sampling every 30 s.

$$\text{rate} = 2.92 \cdot 10^{-2} \text{ mM/s}$$

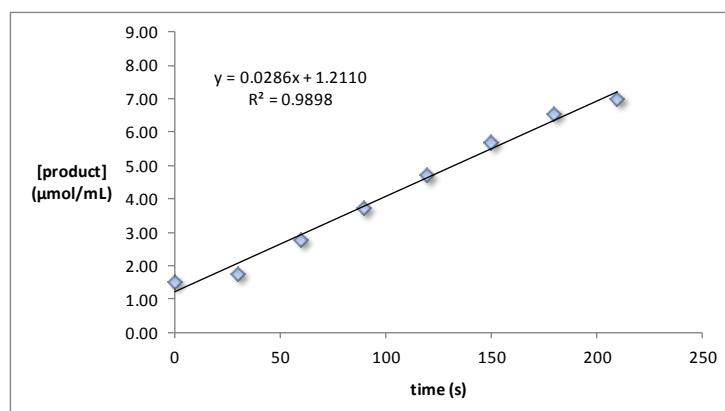
time (s)	product integral	standard integral	[product] (μmol/mL)
0	264.8	6784.2	1.95
30	297.1	6694.8	2.22
60	401.6	6439.2	3.12
90	527.9	6311.3	4.18
120	648.3	6305.7	5.14
150	749.2	6328.0	5.92
180	856.3	6236.8	6.86
210	979.7	6259.7	7.83



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of K⁺**5**⁻ (0.0875 mmol, 5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 8, sampling every 30 s.

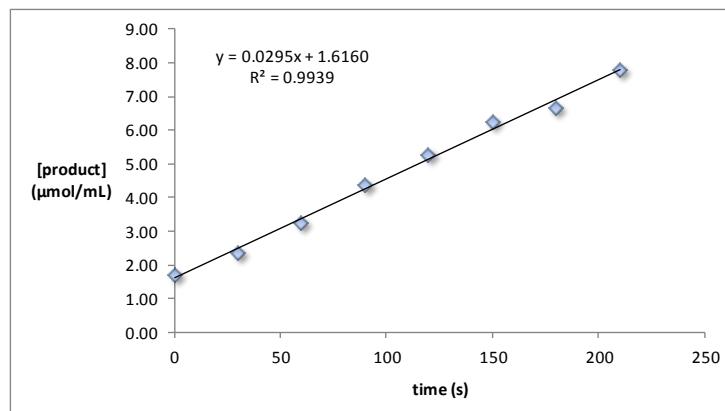
$$\text{rate} = 2.86 \cdot 10^{-2} \text{ mM/s}$$

time (s)	product integral	standard integral	[product] (μmol/mL)
0	184.4	6143.2	1.50
30	231.6	6586.5	1.76
60	353.4	6324.4	2.79
90	466.8	6249.2	3.73
120	584.6	6207.2	4.71
150	712.6	6258.1	5.69
180	802.0	6127.8	6.54
210	877.1	6255.3	7.01



Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of K⁺**5**⁻ (0.0875 mmol, 5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = 2.95 · 10⁻² mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
0	231.2	6761.3	1.71
30	313.3	6710.2	2.33
60	418.8	6482.2	3.23
90	565.1	6426.5	4.40
120	678.5	6438.0	5.27
150	809.4	6460.6	6.26
180	862.0	6456.4	6.68
210	991.3	6365.0	7.79

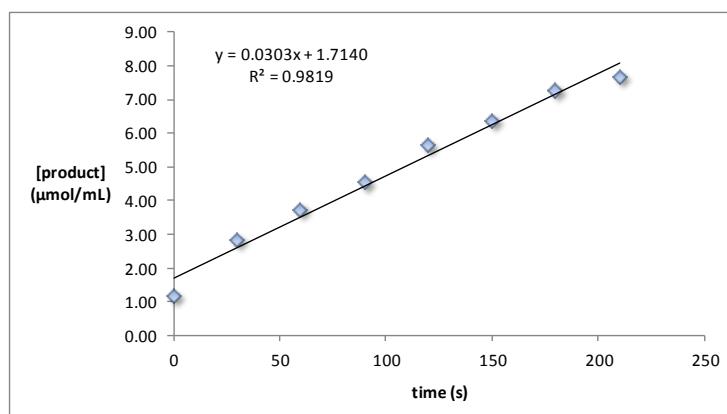


$$\text{avg. rate} = 2.91 \cdot 10^{-2} \pm 4.58 \cdot 10^{-4} \text{ mM/s}$$

7.5 equivalents (Table 7, entry 9)

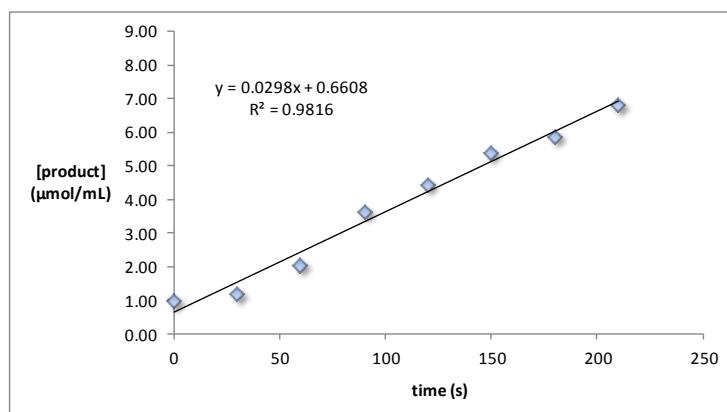
Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of K⁺**5**⁻ (0.131 mmol, 7.5 equiv) and 1,4-difluorobenzene (0.021 mmol, 1.2 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = 3.03 · 10⁻² mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
0	153.1	7712.0	1.19
30	349.0	7431.6	2.82
60	452.2	7255.1	3.74
90	542.9	7190.1	4.53
120	688.0	7333.6	5.63
150	773.1	7279.6	6.37
180	878.7	7264.8	7.26
210	934.0	7321.8	7.65

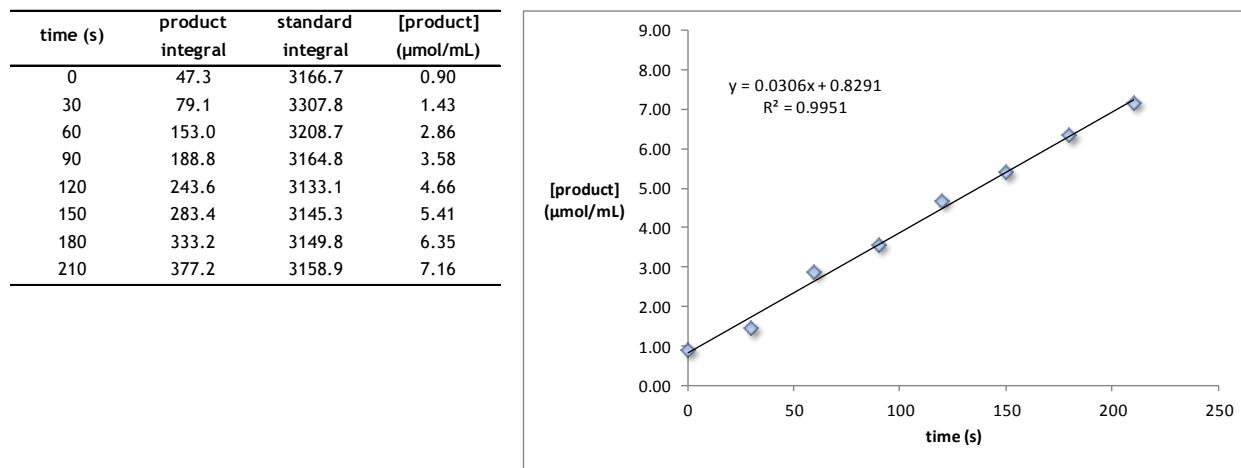


Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{K}^+ \text{5}^-$ (0.131 mmol, 7.5 equiv) and 1,4-difluorobenzene (0.021 mmol, 1.2 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = $2.98 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
0	51.2	3167.2	0.97
30	64.2	3276.1	1.18
60	108.2	3172.3	2.05
90	191.5	3156.0	3.64
120	234.0	3154.4	4.45
150	283.1	3154.1	5.39
180	306.4	3121.6	5.89
210	354.1	3124.0	6.80

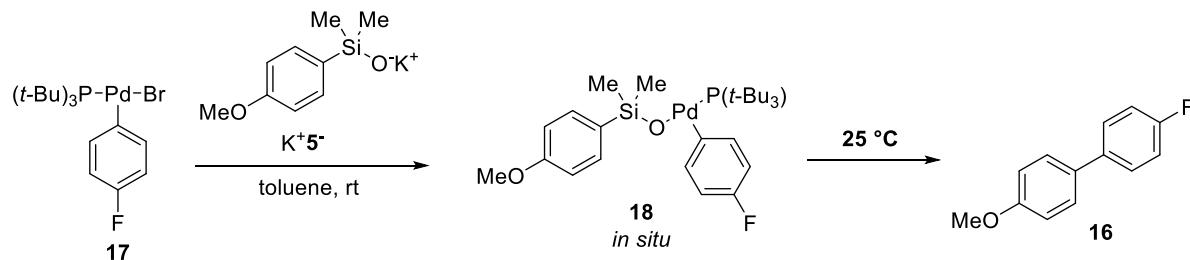


Following General Procedure V, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{K}^+ \text{5}^-$ (0.131 mmol, 7.5 equiv) and 1,4-difluorobenzene (0.021 mmol, 1.2 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 50 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = $3.06 \cdot 10^{-2}$ mM/s**



$$\text{avg. rate} = 3.02 \cdot 10^{-2} \pm 4.04 \cdot 10^{-4} \text{ mM/s}$$

General Procedure VI: Kinetic Measurements for Transmetalation of **18 at 25 °C Employing 7.5 Equiv of K^+5^-**

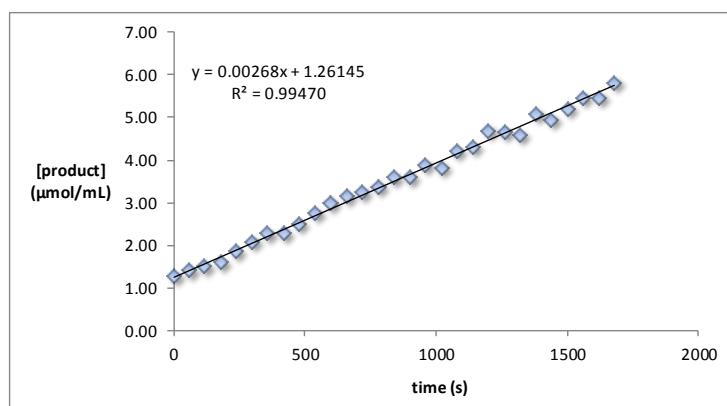


In a drybox, an oven-dried, 4-mL vial was charged with **17** (8.5 mg, 0.0175 mmol) and toluene, resulting in an orange solution. Then, a stock solution of K^+5^- and 1,4-difluorobenzene in toluene was added to a total volume of 0.7 mL (0.025 M in **17**) and the mixture was vigorously mixed. The resulting solution was transferred via syringe into an oven-dried, 5-mm NMR tube. The tube was sealed with a cap and wrapped with Parafilm to exclude any oxygen or moisture. The NMR tube was removed from the drybox and inserted into a pre-warmed 25 °C NMR probe. The temperature of the reaction solution was allowed to equilibrate for ca. 60 seconds prior to data collection. The reaction progress was monitored via ^{19}F NMR spectroscopy by the appearance of the product **16** ($\delta = -117.2$ ppm) as compared to an internal reference (1,4-difluorobenzene, $\delta = -119.9$ ppm) using the following parameters: $\text{at} = 0.5$ s, $\text{d}1 = 1$ s, $\text{pw90} = 15$ μs , $\text{pw} = \text{pw90}/2$, $\text{sw} = -115$ to -130 ppm, nt and sampling as specified in each individual experiment. Experiments were repeated in triplicate.

7.5 equivalents

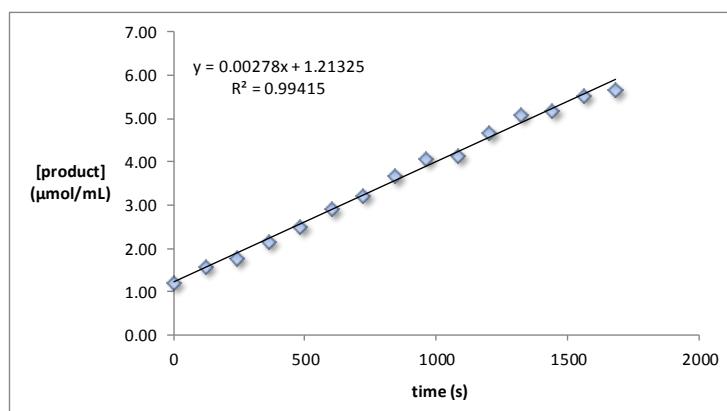
Following General Procedure VI, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of K⁺**5**⁻ (0.131 mmol, 7.5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 16, sampling every 60 s. **rate = 2.68 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
0	345.7	13627.8	1.27
60	381.3	13510.1	1.41
120	408.6	13489.6	1.51
180	436.1	13471.7	1.62
240	500.6	13449.0	1.86
300	564.1	13485.2	2.09
360	619.4	13503.4	2.29
420	615.5	13521.9	2.28
480	672.1	13414.4	2.51
540	747.2	13473.8	2.77
600	802.4	13451.7	2.98
660	852.7	13494.8	3.16
720	868.1	13399.5	3.24
780	907.5	13461.8	3.37
840	961.7	13412.9	3.59
900	973.5	13489.6	3.61
960	1044.0	13416.4	3.89
1020	1034.8	13576.8	3.81
1080	1139.5	13504.2	4.22
1140	1158.4	13464.4	4.30
1200	1252.3	13387.3	4.68
1260	1253.7	13418.4	4.67
1320	1236.0	13426.3	4.60
1380	1370.8	13469.0	5.09
1440	1331.0	13436.2	4.95
1500	1408.6	13532.9	5.20
1560	1467.2	13414.8	5.47
1620	1463.7	13360.0	5.48
1680	1569.9	13469.5	5.83



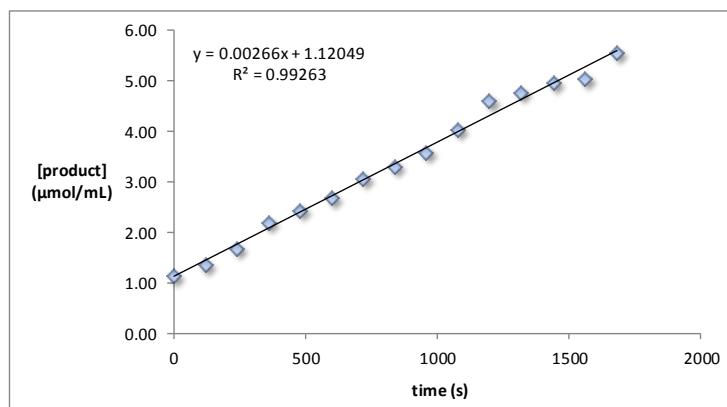
Following General Procedure VI, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of K⁺**5**⁻ (0.131 mmol, 7.5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 16, sampling every 120 s. **rate = 2.78 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
0	318.2	13324.9	1.19
120	444.2	14137.5	1.57
240	513.0	14417.8	1.78
360	610.7	14149.4	2.16
480	713.3	14354.6	2.48
600	822.3	14165.6	2.90
720	905.2	14102.3	3.21
840	1023.8	14024.5	3.65
960	1143.8	14035.7	4.07
1080	1156.7	14004.5	4.13
1200	1310.9	13997.0	4.68
1320	1424.2	14023.6	5.08
1440	1453.5	14068.4	5.17
1560	1544.6	13974.1	5.53
1680	1607.9	14211.4	5.66



Following General Procedure VI, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of K⁺**5**⁻ (0.131 mmol, 7.5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 16, sampling every 120 s. **rate = 2.66 · 10⁻³ mM/s**

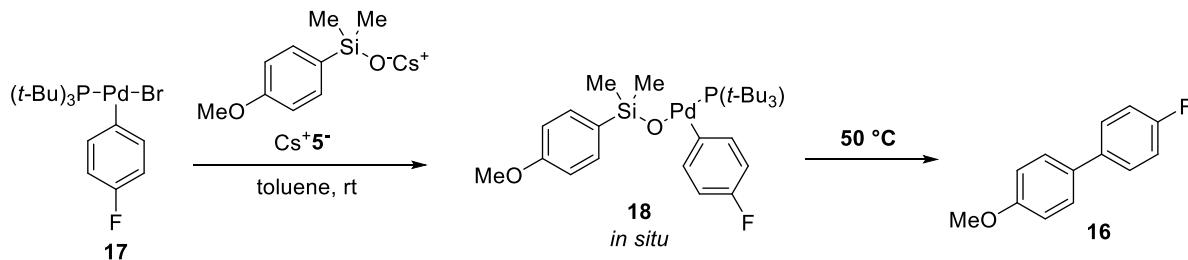
time (s)	product integral	standard integral	[product] (μmol/mL)
0	318.5	13897.9	1.15
120	377.0	14101.8	1.34
240	475.7	14134.2	1.68
360	623.5	14187.6	2.20
480	687.8	14128.9	2.43
600	750.0	14049.8	2.67
720	866.4	14183.1	3.05
840	926.3	14054.8	3.30
960	1012.8	14174.1	3.57
1080	1138.6	14085.6	4.04
1200	1294.2	14073.0	4.60
1320	1342.0	14050.4	4.78
1440	1394.9	14088.7	4.95
1560	1413.6	14048.3	5.03
1680	1562.1	14067.2	5.55



$$\text{avg. rate} = 2.71 \cdot 10^{-3} \pm 6.43 \cdot 10^{-5} \text{ mM/s}$$

KINETIC EXPERIMENTS FOR CROSS-COUPLING OF Cs⁺**5**⁻ WITH **17**

General Procedure VII: Kinetic Measurements for Transmetalation of **18** at 50 °C Starting with Cs⁺**5**⁻ (Table 8)

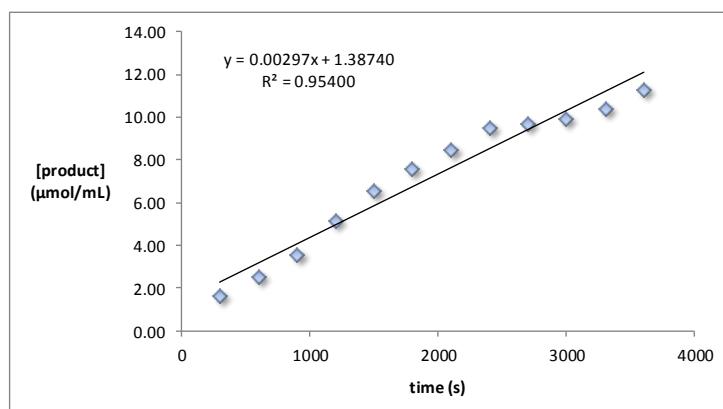


In a drybox, an oven-dried, 5-mm NMR tube was charged with **17** (9.4 mg, 0.0194 mmol), 1,4-difluorobenzene (1 μ L, 0.00974 mmol, 0.5 equiv) and toluene (0.4 mL) resulting in an orange solution upon vigorous mixing. Then, Cs⁺**5**⁻ (6.0 mg, 0.0192 mmol, 0.98 equiv) was added as a solid and the sides of the tube were rinsed with toluene (0.4 mL). The mixture was vigorously mixed using a vortex mixer for ca. 2 min to ensure complete dissolution. The tube was sealed with a septum and wrapped with Parafilm to exclude any oxygen or moisture. The NMR tube was removed from the drybox and inserted into a preheated 50 °C NMR probe. The temperature of the reaction solution was allowed to equilibrate for ca. 90 s prior to data collection. The reaction progress was monitored via ¹⁹F NMR spectroscopy by the appearance of the product **16** ($\delta = -117.2$ ppm) as compared to an internal reference (1,4-difluorobenzene, $\delta = -119.9$ ppm) using the following parameters: at = 0.328, d1 = 0, pw90 = 8.85, pw = pw90/2, nt = 128, sampling every 300 s. Experiments were repeated in triplicate.

1 equivalent (Table 8, entry 1)

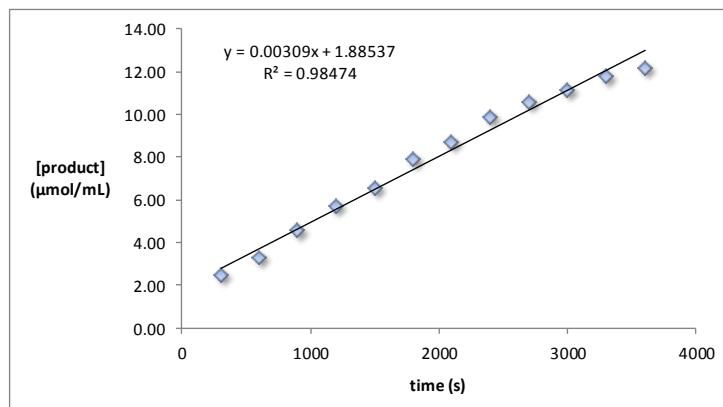
Following General Procedure VII, a mixture of **17** (9.4 mg, 0.0194 mmol) and 1,4-difluorobenzene (1 μ L) was dissolved in toluene (0.4 mL), followed by the addition of Cs⁺**5**⁻ (6.2 mg, 0.0192 mmol, 0.98 equiv) and toluene (0.4 mL) to afford an orange solution. The tube was placed into a preheated 50 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 2.97 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
300	493.4	7397.3	1.62
600	717.4	6934.6	2.51
900	981.3	6733.7	3.53
1200	1471.5	6950.8	5.13
1500	1876.7	6963.7	6.54
1800	2201.5	7049.4	7.57
2100	2406.6	6901.6	8.46
2400	2749.4	7003.2	9.52
2700	2848.2	7131.6	9.68
3000	2925.1	7136.3	9.94
3300	3062.9	7132.8	10.41
3600	3215.2	6900.0	11.30

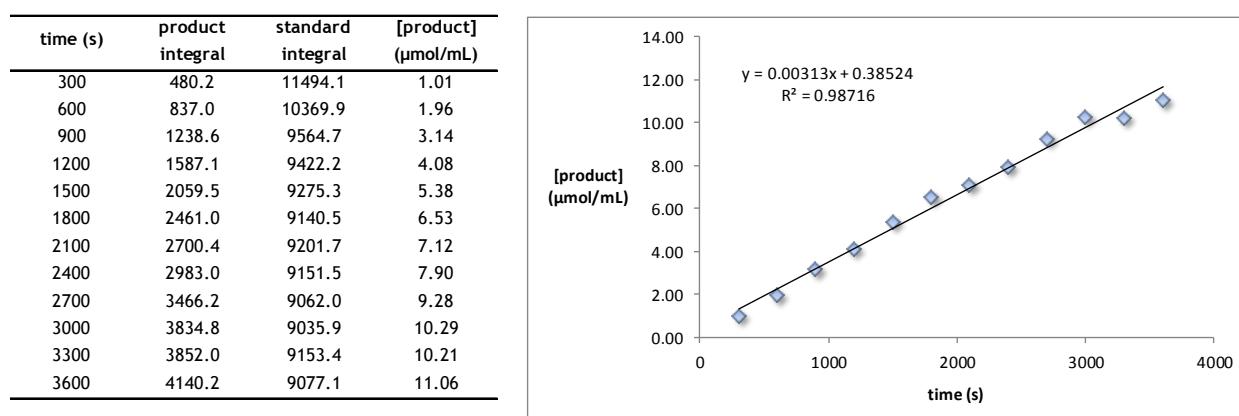


Following General Procedure VII, a mixture of **17** (9.4 mg, 0.0194 mmol) and 1,4-difluorobenzene (1 μL) was dissolved in toluene (0.4 mL), followed by the addition of Cs⁺**5**⁻ (6.2 mg, 0.0192 mmol, 0.98 equiv) and toluene (0.4 mL) to afford an orange solution. The tube was placed into a preheated 50 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 3.09 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
300	805.2	7863.9	2.48
600	1220.5	8997.9	3.29
900	1774.2	9456.3	4.55
1200	2171.5	9228.6	5.71
1500	2447.8	9020.8	6.58
1800	2832.9	8731.2	7.87
2100	3042.2	8431.4	8.75
2400	3334.1	8181.4	9.88
2700	3480.4	7965.8	10.60
3000	3656.3	7938.6	11.17
3300	3799.0	7802.5	11.81
3600	3872.4	7712.3	12.18

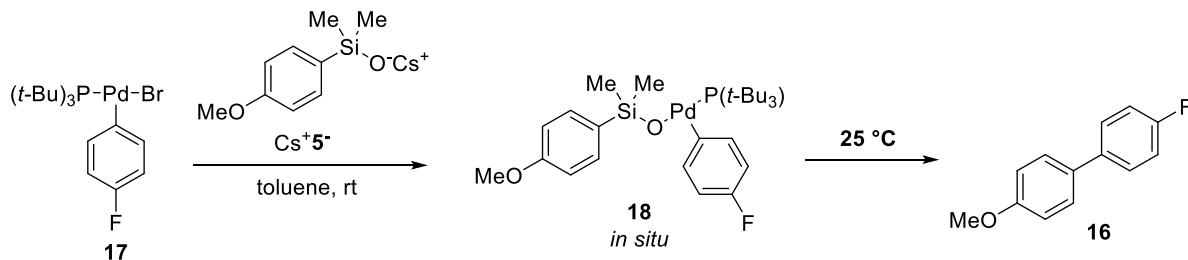


Following General Procedure VIII, a mixture of **17** (9.4 mg, 0.0194 mmol) and 1,4-difluorobenzene (1 μL) was dissolved in toluene (0.4 mL), followed by the addition of Cs⁺**5**⁻ (6.2 mg, 0.0192 mmol, 0.98 equiv) and toluene (0.4 mL) to afford an orange solution. The tube was placed into a preheated 50 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy. **rate = 3.13 · 10⁻³ mM/s**



$$\text{avg. rate} = 3.06 \cdot 10^{-3} \pm 8.24 \cdot 10^{-5} \text{ mM/s}$$

General Procedure VIII: Kinetic Measurements for Transmetalation Employing Cs^+5^- at 25 °C (Table 8)

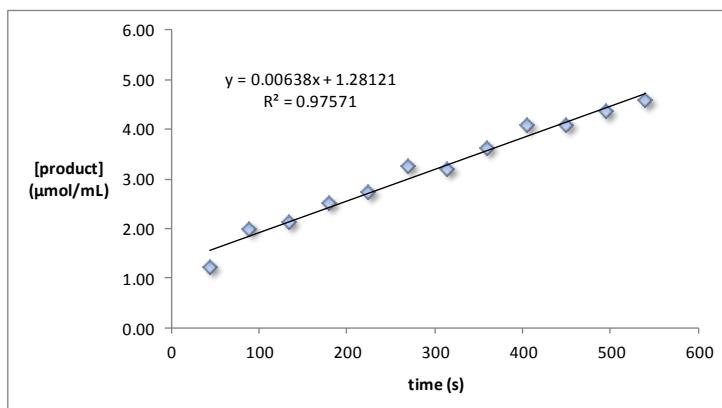


In a drybox, an oven-dried, 4-mL vial was charged with **17** (8.5 mg, 0.0175 mmol) and toluene, resulting in an orange solution. Then, a stock solution of Cs^+5^- and 1,4-difluorobenzene in toluene was added to a total volume of 0.7 mL (0.025 M in **17**) and the mixture was vigorously mixed. The resulting solution was transferred via syringe into an oven-dried, 5-mm NMR tube. The tube was sealed with a cap and wrapped with Parafilm to exclude any oxygen or moisture. The NMR tube was removed from the drybox and immediately inserted into a pre-warmed 25 °C NMR probe. The reaction progress was monitored via ^{19}F NMR spectroscopy by the appearance of the product **16** ($\delta = -117.2$ ppm) as compared to an internal reference (1,4-difluorobenzene, $\delta = -119.9$ ppm) using the following parameters: $\text{at} = 0.5$ s, $\text{d1} = 1$ s, $\text{pw90} = 15$ μs , $\text{pw} = \text{pw90}/2$, $\text{sw} = -115$ to -130 ppm, nt and sampling as specified in each individual experiment. Experiments were repeated in triplicate.

1.5 equivalents (Table 8, entry 2)

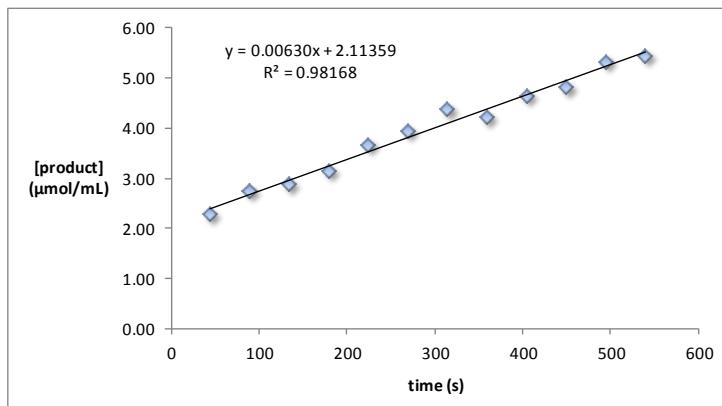
Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of Cs⁺**5**⁻ (0.0263 mmol, 1.5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = 6.29 · 10⁻³ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
45	133.6	5445.5	1.23
90	215.1	5399.7	1.99
135	231.7	5411.0	2.14
180	271.1	5378.6	2.52
225	295.0	5401.2	2.73
270	351.0	5397.2	3.25
315	346.5	5409.8	3.20
360	389.8	5393.0	3.61
405	439.4	5386.7	4.08
450	438.8	5367.7	4.09
495	466.4	5346.6	4.36
540	492.7	5388.5	4.57



Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of Cs⁺**5**⁻ (0.0263 mmol, 1.5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = 6.30 · 10⁻³ mM/s**

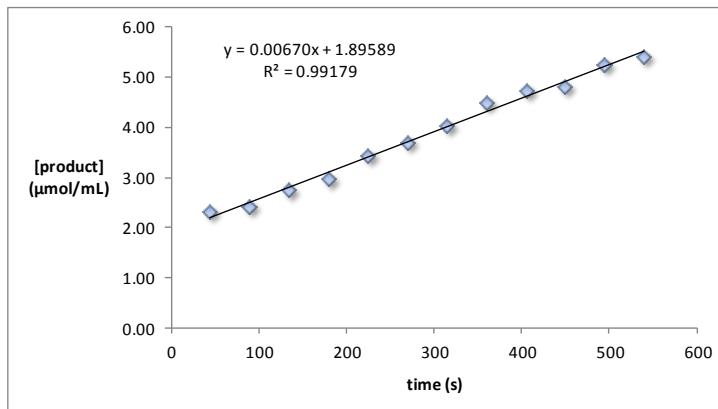
time (s)	product integral	standard integral	[product] (μmol/mL)
45	257.3	5642.6	2.28
90	308.5	5596.6	2.76
135	320.3	5578.2	2.87
180	350.0	5554.6	3.15
225	408.2	5581.1	3.66
270	438.6	5562.4	3.94
315	487.2	5561.2	4.38
360	464.9	5533.2	4.20
405	516.7	5556.6	4.65
450	537.1	5572.1	4.82
495	595.2	5585.6	5.33
540	605.4	5567.9	5.44



Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.2 mL). A stock solution of Cs⁺**5**⁻ (0.0263 mmol, 1.5 equiv) and 1,4-difluorobenzene (0.0175

mmol, 1 equiv) in toluene (0.5 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = $6.70 \cdot 10^{-3}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
45	254.7	5490.0	2.32
90	264.9	5469.0	2.42
135	299.0	5442.5	2.75
180	323.8	5469.4	2.96
225	372.4	5444.6	3.42
270	402.3	5423.0	3.71
315	438.6	5458.0	4.02
360	483.4	5405.9	4.47
405	514.0	5432.3	4.73
450	523.8	5445.3	4.81
495	574.5	5470.1	5.25
540	590.1	5451.3	5.41

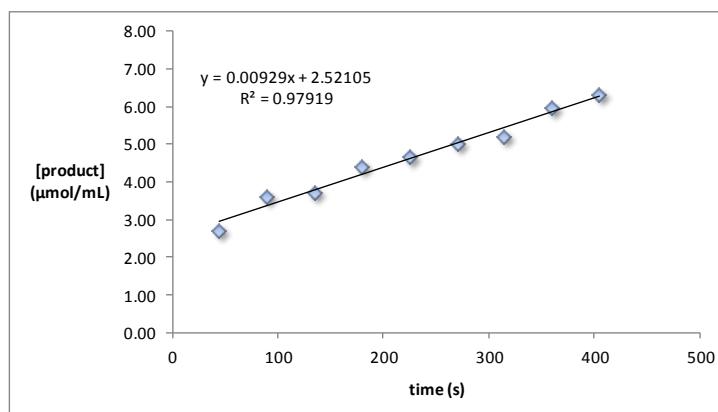


$$\text{avg. rate} = 6.46 \cdot 10^{-3} \pm 2.12 \cdot 10^{-4} \text{ mM/s}$$

2 equivalents (Table 8, entry 3)

Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{Cs}^+ \text{5}^-$ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = $9.29 \cdot 10^{-3}$ mM/s**

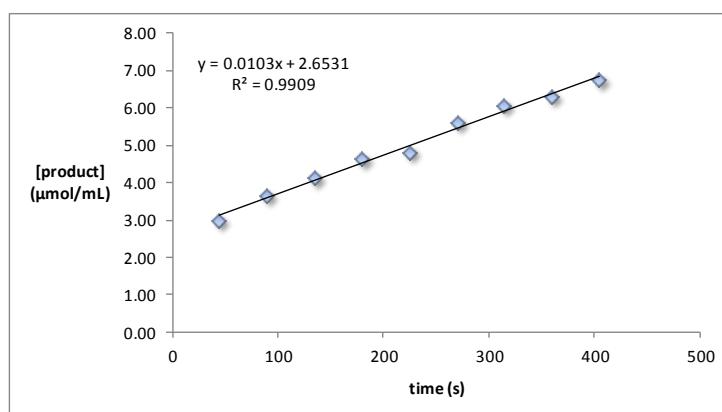
time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
45	302.3	5576.7	2.71
90	392.3	5452.6	3.60
135	402.8	5442.1	3.70
180	479.1	5459.9	4.39
225	504.5	5432.7	4.64
270	542.1	5422.8	5.00
315	567.6	5449.6	5.21
360	651.2	5457.5	5.97
405	686.6	5448.7	6.30



Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{Cs}^+ \text{5}^-$ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C

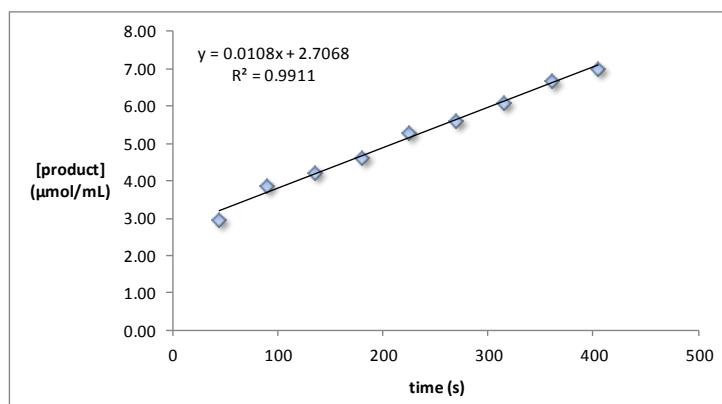
NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = $1.03 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
45	328.7	5516.4	2.98
90	394.7	5445.7	3.62
135	446.5	5452.1	4.10
180	505.0	5451.8	4.63
225	525.9	5457.0	4.82
270	610.0	5452.5	5.59
315	649.6	5374.8	6.04
360	680.3	5403.4	6.29
405	726.9	5390.9	6.74



Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{Cs}^+\text{5}^-$ (0.035 mmol, 2 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 16, sampling every 45 s. **rate = $1.08 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
45	327.6	5545.0	2.95
90	419.5	5449.4	3.85
135	462.0	5457.6	4.23
180	504.0	5435.8	4.64
225	573.9	5439.9	5.27
270	608.3	5426.0	5.61
315	662.0	5451.8	6.07
360	717.7	5380.3	6.67
405	760.6	5447.3	6.98



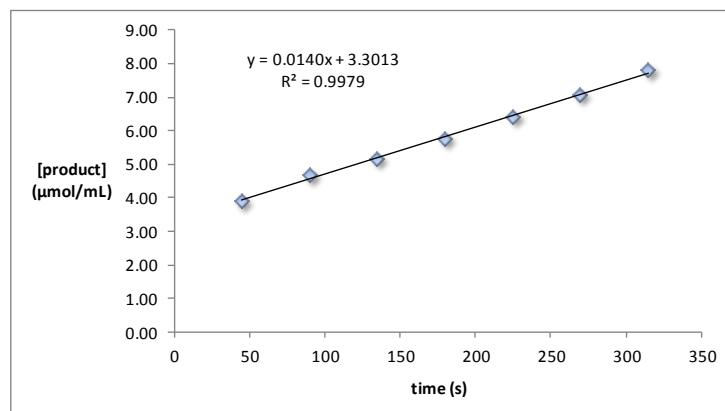
$$\text{avg. rate} = 1.01 \cdot 10^{-2} \pm 7.69 \cdot 10^{-4} \text{ mM/s}$$

3 equivalents (Table 8, entry 4)

Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{Cs}^+\text{5}^-$ (0.0525 mmol, 3 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 16,

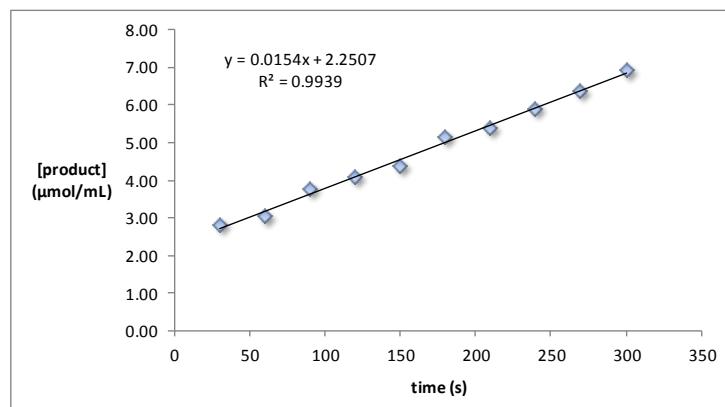
sampling every 45 s. **rate = $1.40 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
45	1033.4	13229.2	3.91
90	1210.3	12968.6	4.67
135	1337.3	12954.6	5.16
180	1511.5	13071.3	5.78
225	1665.9	13010.7	6.40
270	1842.5	13080.9	7.04
315	2031.2	13041.4	7.79



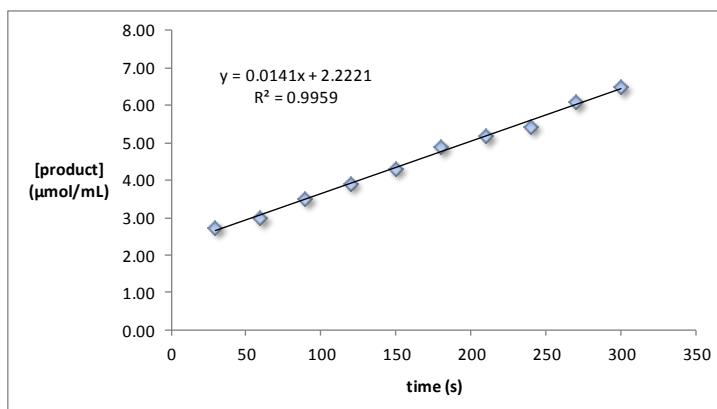
Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{Cs}^+ \text{5}^-$ (0.0525 mmol, 3 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = $1.54 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
30	441.6	7866.9	2.81
60	467.1	7706.8	3.03
90	575.0	7620.2	3.77
120	619.2	7590.5	4.08
150	665.3	7582.8	4.39
180	778.7	7561.5	5.15
210	827.8	7654.4	5.41
240	895.0	7573.2	5.91
270	970.2	7606.7	6.38
300	1051.7	7576.6	6.94



Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{Cs}^+ \text{5}^-$ (0.0525 mmol, 3 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = $1.41 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
30	424.6	7804.1	2.72
60	460.9	7703.9	2.99
90	529.7	7578.0	3.49
120	590.1	7604.8	3.88
150	651.6	7564.9	4.31
180	740.0	7576.9	4.88
210	786.5	7593.1	5.18
240	821.4	7549.2	5.44
270	919.8	7551.4	6.09
300	976.1	7530.7	6.48

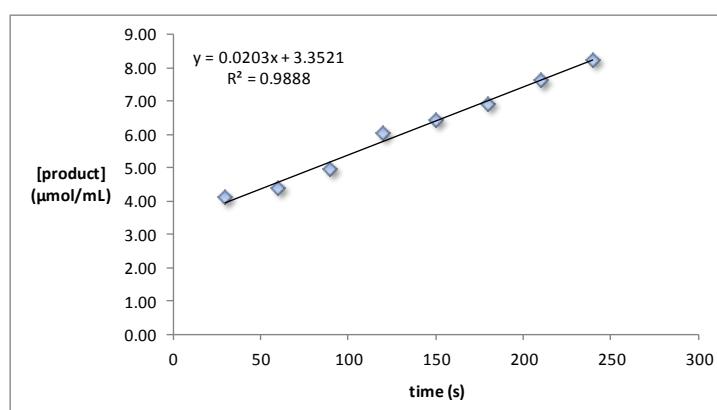


$$\text{avg. rate} = 1.45 \cdot 10^{-2} \pm 7.81 \cdot 10^{-4} \text{ mM/s}$$

5 equivalents (Table 8, entry 5)

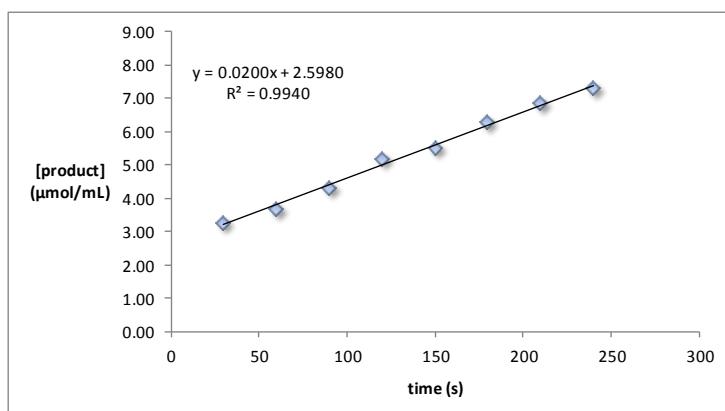
Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{Cs}^+ \mathbf{5}^-$ (0.0875 mmol, 5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = $2.03 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] ($\mu\text{mol/mL}$)
30	623.0	7574.0	4.11
60	654.1	7433.0	4.40
90	733.0	7375.4	4.97
120	888.9	7344.1	6.05
150	950.4	7377.7	6.44
180	1013.2	7324.2	6.92
210	1127.1	7374.0	7.64
240	1209.2	7356.0	8.22



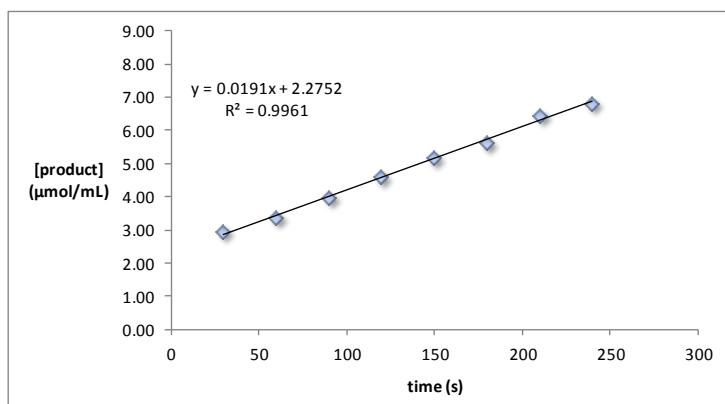
Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of $\text{Cs}^+ \mathbf{5}^-$ (0.0875 mmol, 5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ^{19}F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = $2.00 \cdot 10^{-2}$ mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
30	496.9	7587.7	3.27
60	550.8	7511.7	3.67
90	629.1	7295.0	4.31
120	752.6	7274.9	5.17
150	804.3	7319.7	5.49
180	920.1	7307.0	6.30
210	1002.0	7332.1	6.83
240	1069.9	7322.2	7.31



Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of Cs⁺**5**⁻ (0.0875 mmol, 5 equiv) and 1,4-difluorobenzene (0.0175 mmol, 1 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = 1.91 · 10⁻² mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
30	445.3	7608.6	2.93
60	499.0	7467.8	3.34
90	592.8	7449.7	3.98
120	676.0	7398.8	4.57
150	765.5	7397.4	5.17
180	832.7	7438.7	5.60
210	957.2	7428.2	6.44
240	1000.3	7344.0	6.81

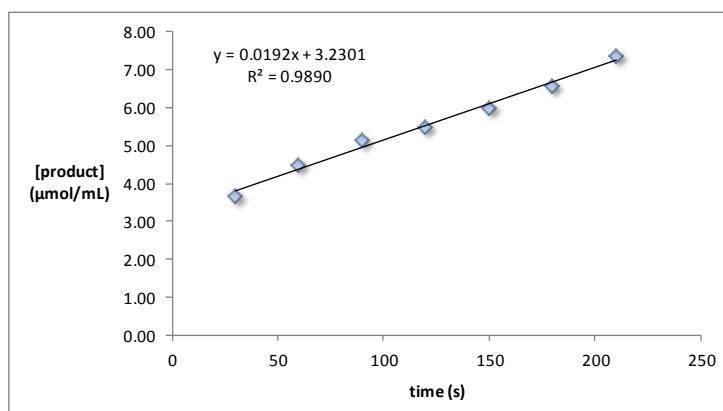


$$\text{avg. rate} = 1.98 \cdot 10^{-2} \pm 6.24 \cdot 10^{-4} \text{ mM/s}$$

7.5 equivalents (Table 8, entry 6)

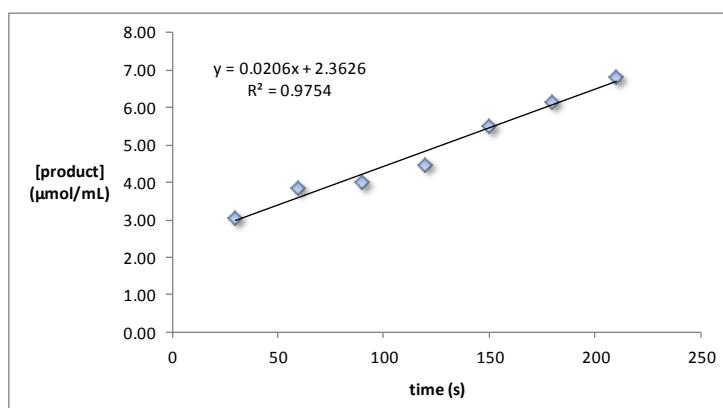
Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of Cs⁺**5**⁻ (0.131 mmol, 7 equiv) and 1,4-difluorobenzene (0.035 mmol, 2 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = 1.92 · 10⁻² mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
30	559.3	15283.0	3.66
60	683.9	15208.7	4.50
90	773.6	15084.2	5.13
120	818.2	14867.6	5.50
150	897.3	14950.9	6.00
180	979.0	14933.0	6.56
210	1103.2	14986.0	7.36



Following General Procedure VIII, **17** (8.5 mg, 0.0175 mmol) was dissolved in toluene (0.1 mL). A stock solution of Cs⁺**5**⁻ (0.131 mmol, 7 equiv) and 1,4-difluorobenzene (0.035 mmol, 2 equiv) in toluene (0.6 mL) was added. The tube was inserted into a preheated 25 °C NMR probe and the reaction progress was monitored via ¹⁹F NMR spectroscopy; nt = 8, sampling every 30 s. **rate = 2.06 · 10⁻² mM/s**

time (s)	product integral	standard integral	[product] (μmol/mL)
30	380.6	12445.8	3.06
60	478.4	12374.0	3.87
90	493.9	12358.5	4.00
120	554.9	12368.0	4.49
150	680.7	12384.7	5.50
180	760.3	12372.1	6.15
210	847.7	12442.2	6.81



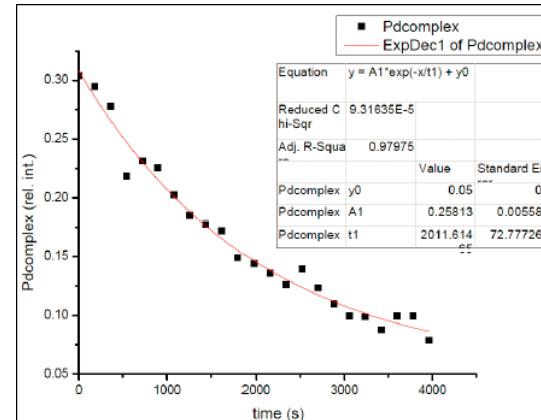
$$\text{avg. rate} = 1.99 \cdot 10^{-2} \pm 9.90 \cdot 10^{-4} \text{ mM/s}$$

General Procedure IX: Kinetic Measurements for Thermal Transmetalation Employing Cs⁺5⁻ at 50 °C Removing the Inorganic Salts

Run 1

The following modifications to General Procedure VII were made: the mixture was prepared in an oven-dried, 3-mL vial and the resulting reaction solution was transferred to the NMR tube using a 22G needle. A mixture of **17** (13.7 mg, 0.0283 mmol, 1.0 equiv) and 1,4-difluorobenzene (capillary) were dissolved in toluene (0.4 mL), followed by addition of Cs⁺5⁻ (8.7 mg, 0.0278 mmol, 0.98 equiv) and toluene (0.4 mL) to afford an orange solution. The tube was placed into a preheated 50 °C NMR probe and the reaction progress was monitored by ¹⁹F NMR spectroscopy, sampling every 180 s. **rate = 4.97 x 10⁻⁴ s⁻¹**.

Time (s)	Integral IS (-119.8 ppm)	Integral 18 (-121.7 ppm)	Normalized (mM)
0	1.252	0.381	35.38
180	1.251	0.369	34.33
360	1.192	0.331	32.34
540	1.202	0.263	25.42
720	1.381	0.320	26.93
900	1.299	0.293	26.24
1080	1.309	0.265	23.55
1260	1.333	0.246	21.51
1440	1.352	0.240	20.65
1620	1.291	0.222	20.04
1800	1.305	0.195	17.37
1980	1.363	0.197	16.80
2160	1.326	0.180	15.82
2340	1.306	0.165	14.68
2520	1.172	0.064	6.33
2700	1.211	0.049	4.73
2880	1.317	0.144	12.74
3060	1.313	0.130	11.56
3240	1.288	0.127	11.49
3420	1.433	0.125	10.17
3600	1.313	0.130	11.55
3780	1.333	0.133	11.59
3960	1.312	0.103	9.14

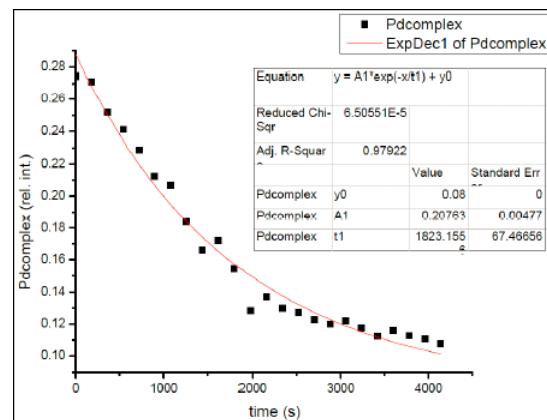


Run 2

Following General Procedure IX, the mixture was prepared in an oven-dried, 3-mL vial and the resulting reaction solution was transferred to the NMR tube using a 22G needle. A mixture of **17** (12.7 mg, 0.0263 mmol, 1.0 equiv) and 1,4-difluorobenzene (capillary) were dissolved in toluene (0.4 mL), followed by addition of Cs⁺5⁻ (8.1 mg, 0.0257 mmol, 0.98 equiv)

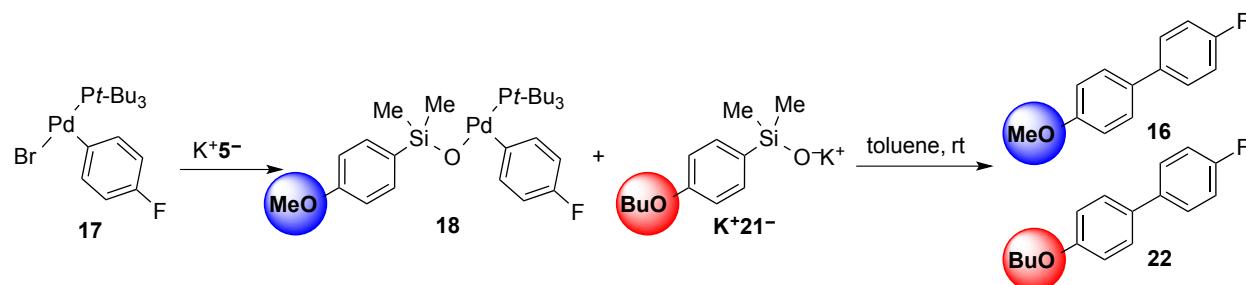
and toluene (0.4 mL) to afford an orange solution. The tube was placed into a preheated 50 °C NMR probe and the reaction progress was monitored by ^{19}F NMR spectroscopy, sampling every 180 s. **rate = $5.29 \times 10^{-4} \text{ s}^{-1}$** .

Time (s)	Integral IS (-119.8 ppm)	Integral 18 (-121.7 ppm)	Normalized (mM)
0	1.431	0.393	32.83
180	1.455	0.395	32.45
360	1.438	0.362	30.09
540	1.503	0.362	28.80
720	1.554	0.356	27.35
900	1.502	0.319	25.42
1080	1.452	0.300	24.74
1260	1.321	0.243	21.98
1440	1.349	0.224	19.85
1620	1.434	0.246	20.54
1800	1.368	0.211	18.47
1980	1.333	0.171	15.33
2160	1.402	0.192	16.40
2340	1.532	0.199	15.52
2520	1.496	0.190	15.22
2700	1.519	0.187	14.68
2880	1.485	0.178	14.35
3060	1.473	0.180	14.60
3240	1.467	0.172	14.03
3420	1.479	0.166	13.43
3600	1.475	0.171	13.87
3780	1.499	0.169	13.46
3960	1.471	0.163	13.25
4140	1.475	0.159	12.88
4320	1.491	0.161	12.93
4500	1.474	0.154	12.52
4680	1.489	0.157	12.60
4860	1.469	0.155	12.58
5040	1.391	0.147	12.60



Analysis of Product Distribution for Cross-Coupling of **18 with Exogenous $\text{K}^+ \text{21}^-$ as Activator (Scheme 11)**

1.02 equiv $\text{K}^+ \text{21}^-$



To an oven dried 4-mL scintillation vial contained in a drybox was charged **17** (8.2 mg, 0.017 mmol) as an orange solid and biphenyl (6.8 mg, 0.044 mmol) as a white solid. Dry toluene (0.5 mL) was added affording an orange solution upon mixing. $\text{K}^+ \text{5}^-$ (3.7 mg, 0.017 mmol) was added as a solid and stirring resulted in complete dissolution. $\text{K}^+ \text{21}^-$ (4.5 mg, 0.017 mmol) was added in one portion as a solid and the sides of the vial were rinsed with additional toluene (0.2 mL). The resulting reaction mixture was allowed to stir in a drybox for 3 h. The reaction was then sampled by removing 100 μL of the reaction mixture and quenching onto water (0.5 mL). The organics were extracted with ethyl acetate (1.2 mL) and then passed through a pipet plug of SiO_2 to remove any insoluble material. The filtrate was analyzed by GC: Biphenyl, 3.84 min; t_{R} 7.38 min, **16**, 0.0064 mmol (38%); t_{R} 9.29 min, **23**, 0.0079 mmol (47%). Ratio of **16**:**22** based on biaryl product formation is 45/55.

3.02 equiv $\text{K}^+ \text{21}^-$

To an oven dried 4-mL scintillation vial contained in a drybox was charged **17** (8.8 mg, 0.018 mmol) as an orange solid and biphenyl (4.8 mg, 0.031 mmol) as a white solid. Dry toluene (0.5 mL) was added affording an orange solution upon stirring. $\text{K}^+ \text{5}^-$ (3.9 mg, 0.018 mmol) was added as a solid and stirring resulted in complete dissolution. $\text{K}^+ \text{21}^-$ (14.4 mg, 0.055 mmol) was added in one portion as a solid and the sides of the vial were rinsed with additional toluene (0.2 mL). The resulting reaction mixture was allowed to stir in a drybox for 3h. The reaction was then sampled by removing 100 μL of the reaction mixture and quenching onto water (0.5 mL). The organics were extracted with ethyl acetate (1.2 mL) and then passed through a pipet plug of SiO_2 to remove any insoluble material. The filtrate was analyzed by GC:

Biphenyl, 3.84 min; t_R 7.38 min, **16**, 0.0035 mmol (19%); t_R 9.29 min, **22**, 0.015 mmol (83%). Ratio of **16:22** based on biaryl product formation is 19/81.

GC Methods and Response Factors

GC Method: Injections were made onto a Hewlett-Packard HP-1 50-m cross-linked 1%-phenyl methyl silicone gum phase column. The injector and detector temperature were 250 °C. The column oven temperature was as follows: GC: HP-1(manual) $T_0 = 150$ °C (5 min) Rate = 25 °C/min $T_f = 250$ °C (5 min), total run time was 14 min. Retention times (t_R) and integrated ratios were obtained from reporting integrators.

Response factors (R_f) for quantitative GC analysis were obtained by the equation below:

Eq1: Response factor for A = (mmol A * area IS)/(area A * mmol IS)

mmol of 22	Area of 22	mmol of IS	Area of IS	Response Factor
0.0266	132408	0.0318	121727	0.769
0.0266	134283	0.0318	119941	0.747
0.0266	131848	0.0318	126715	0.804
0.027	150262	0.0441	178980	0.729
0.027	143526	0.0441	178802	0.763
0.027	151484	0.0441	183067	0.740
0.0221	123757	0.0447	189287	0.756
0.0221	121477	0.0447	186280	0.758
0.0221	118669	0.0447	182711	0.761
AVG R_f				0.759
mmol of 16	Area of 16	mmol of IS	Area of IS	Response Factor
0.0321	108288	0.0422	129211	0.908
0.0321	105832	0.0422	129643	0.932
0.0321	104049	0.0422	130802	0.956
0.0371	154127	0.0363	145311	0.964
0.0371	154953	0.0363	149878	0.989
0.0371	153441	0.0363	145696	0.970
0.0277	123525	0.0455	185229	0.913
0.0277	124056	0.0455	181869	0.893
0.0277	120450	0.0455	176473	0.892
AVG R_f				0.935

REFERENCES

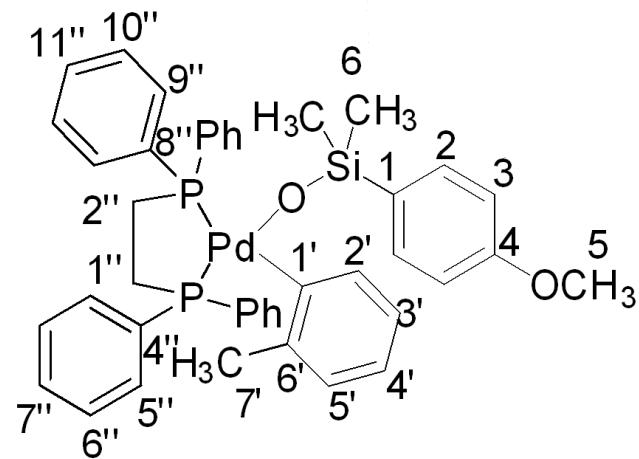
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- (16) Procedure based upon: Yamashita, M.; Vicario, J.; Hartwig, J. J. *Am. Chem. Soc.* **2003**, *125*, 16347-16360.
- (17) The time for temperature equilibration was independently confirmed by determining the time for 1 mL of ethylene glycol to reach a temperature that fluctuated by less than 0.2 °C
- (18) The reaction progress was monitored through > 3 half lives to ensure consistency throughout the course of the reaction.
- (19) The initial data points were normalized to the starting concentration of aryl halide. No product formation was observed at T = 0 min.

- (20) The arylsilanolate $\text{K}^+ \mathbf{5}^-$ proved to be insoluble at concentrations > 150 mM at room temperature. However independent solution studies showed that the silanolate was completely soluble up to a concentration of 600 mM at 90 °C.
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- (22) An independent experiment using 0.5 equiv of $\text{Cs}^+ \mathbf{5}^-$ at the onset gave similar results whereby the peak at 77.0 ppm was the predominant species with small amount of **17** present. Attempts to isolate and fully characterize the proposed dimeric complex **19** were unsuccessful.

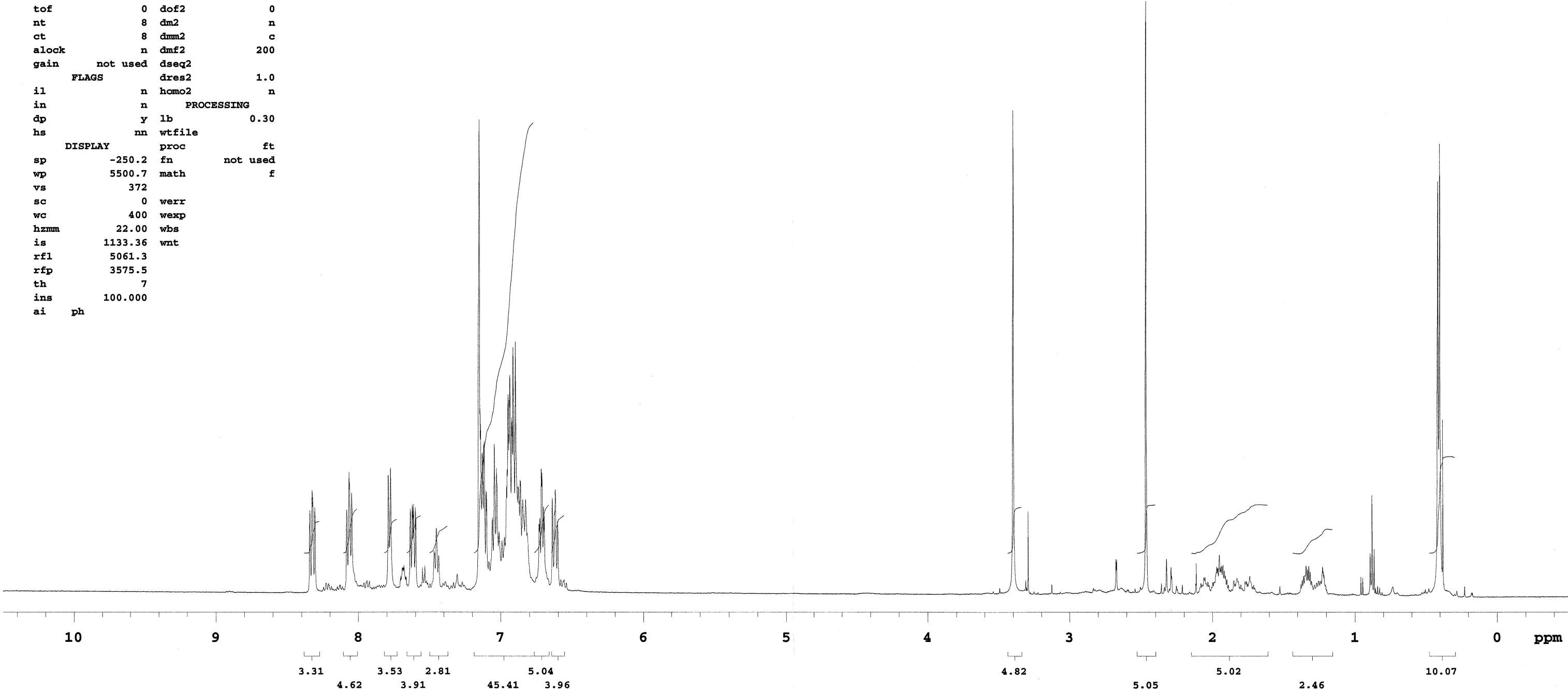
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ins 100.000
ai ph



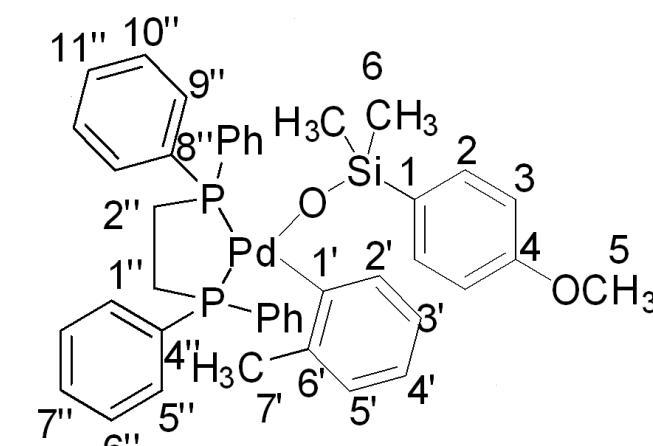
7e



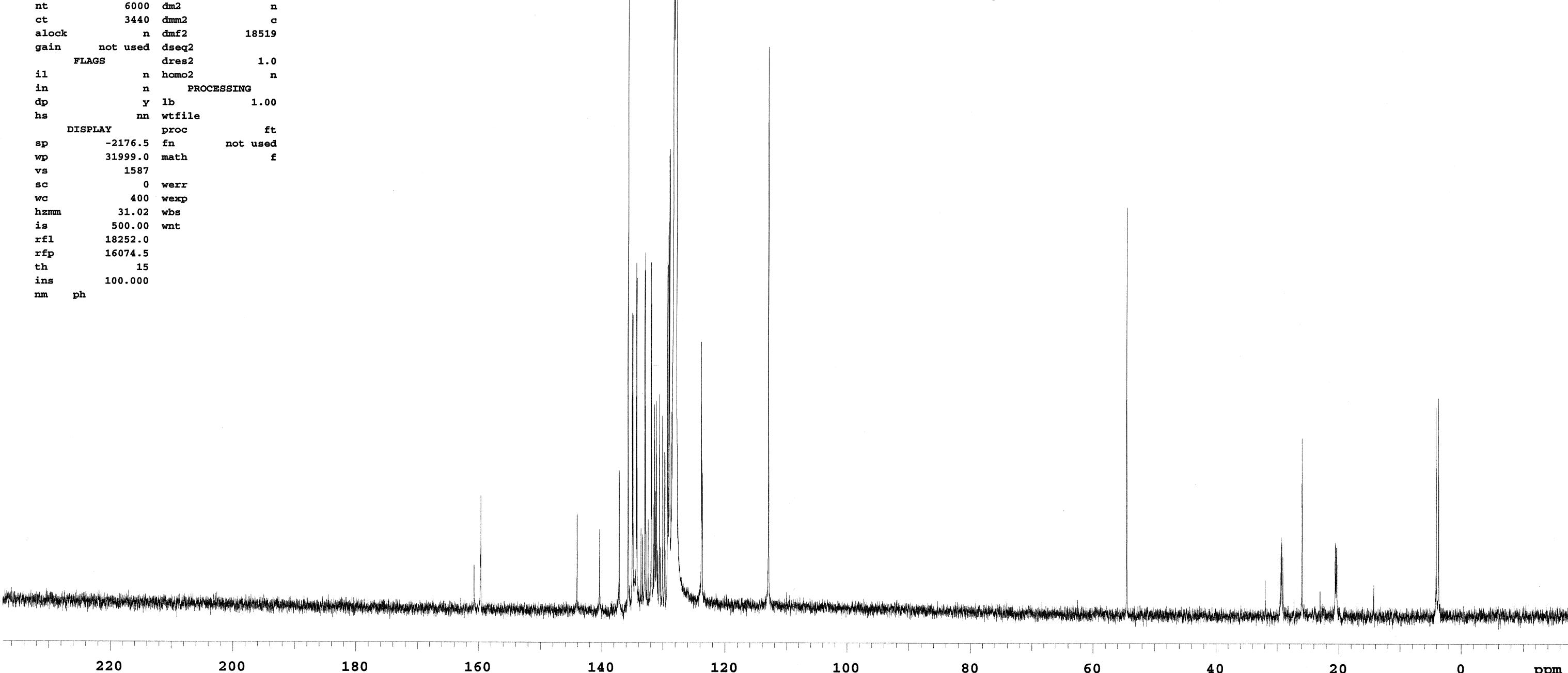
dppe complex

exp1 s2pul

SAMPLE DEC. & VT
date May 9 2007 dfrq 499.432
solvent Benzene dn H1
file exp dpwr 44
ACQUISITION dof -827.0
sfrq 125.596 dm YYY
tn C13 dmm w
at 1.024 dmf 19000
np 65536 dseq
sw 32000.0 dres 1.0
fb 18000 homo n
bs 16 DEC2
tpwr 63 dfrq2 0
pw 4.0 dn2
d1 1.000 dpwr2 1
tof 1880.0 dof2 0
nt 6000 dm2 n
ct 3440 dmm2 c
alock n dmf2 18519
gain not used dseq2
FLAGS dres2 1.0
il n homo2 n
in n PROCESSING
dp y lb 1.00
hs nm wtfile
DISPLAY proc ft
sp -2176.5 fn not used
wp 31999.0 math f
vs 1587
sc 0 werr
wc 400 wexp
hzmm 31.02 wbs
is 500.00 wnt
rfl 18252.0
rfp 16074.5
th 15
ins 100.000
nm ph



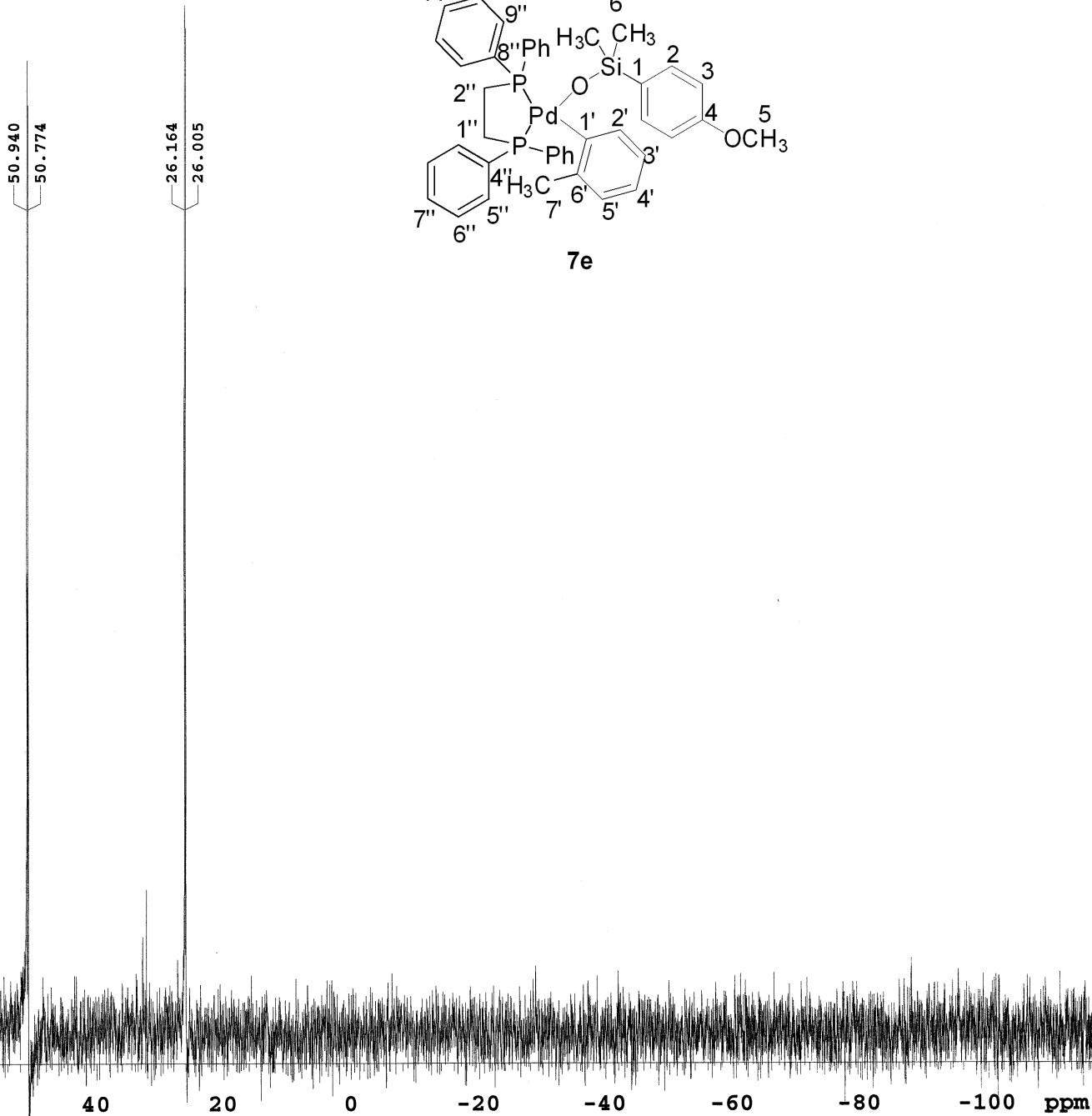
7e



sat-xvii-19-31p

exp1 s2pul

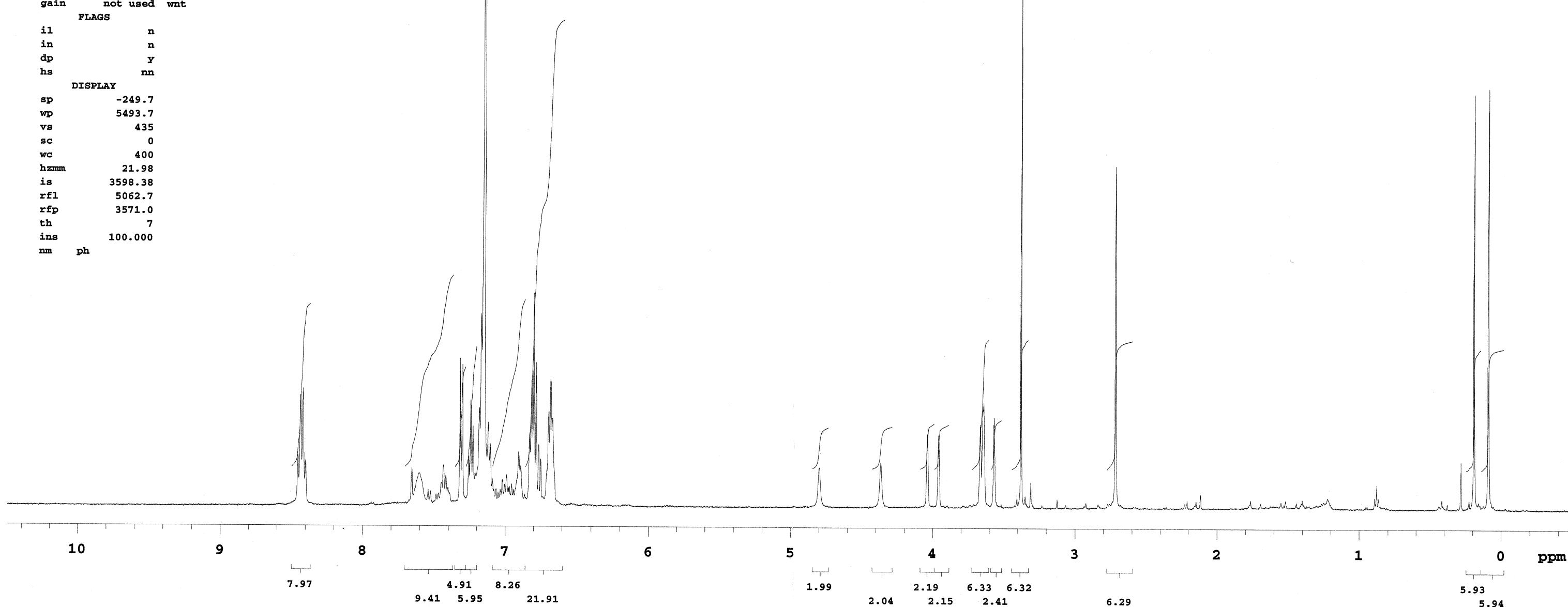
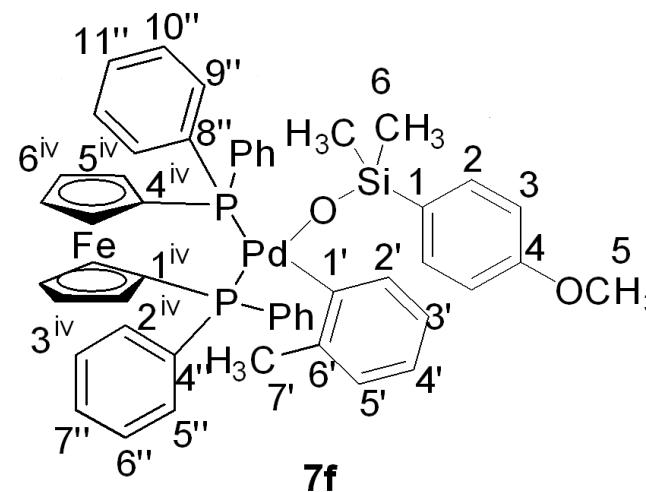
SAMPLE DEC. & VT
date Aug 14 2006 dfrq 399.950
solvent Benzene dn H1
file exp dpwr 48
ACQUISITION dof -1092.3
sfrq 161.903 dm nny
tn P31 dmnm w
at 0.819 dmf 13889
np 65536 dseq
sw 40000.0 dres 1.0
fb 22000 homo n
bs 16 PROCESSING
tpwr 50 lb 3.00
pw 6.4 wtfile
dl 1.000 proc ft
tof 0 fn not used
nt 256 math f
ct 224
alock n werr
gain not used wexp
FLAGS wbs
il n wnt
in n
dp y
hs nn
DISPLAY
sp -19014.3
wp 40000.0
vs 162
sc 0
wc 250
hzmm 6.04
is 354.40
rf1 19014.3
rfp 0
th 76
ins 100.000
nm no ph



dppf complex

expl s2pul

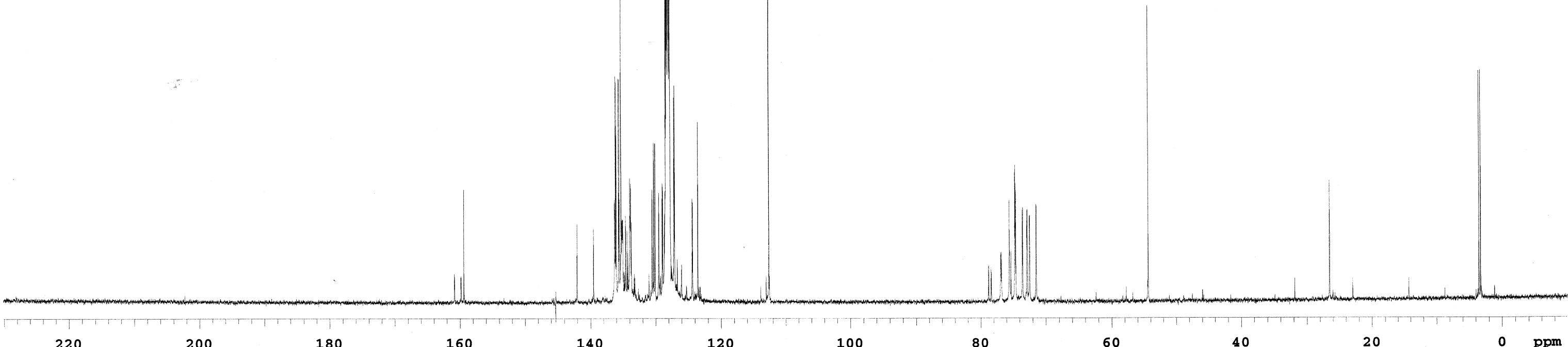
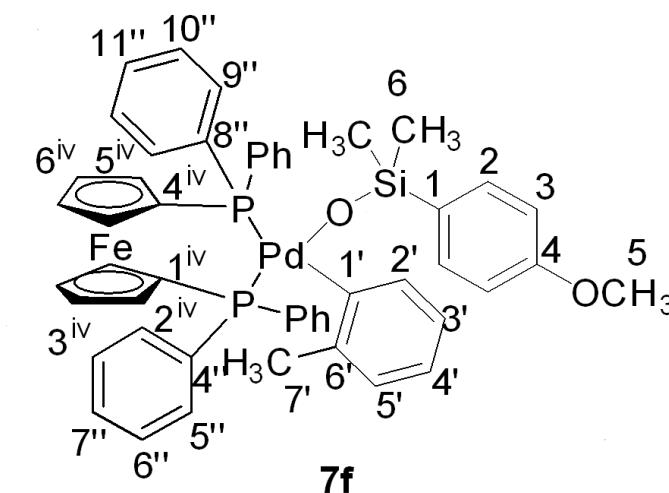
SAMPLE DEC. & VT
date Aug 29 2006 dfrq 499.438
solvent Benzene dn H1
file /export/home/~/dpwr 25
data/vxr500/tymonk dof 0
o/satxii31x.fid dm nnn
ACQUISITION dmm c
sfrq 499.438 dmf 200
tn H1 dseq
at 4.096 dres 1.0
np 65536 homo n
sw 8000.0 PROCESSING
fb 4400 lb 0.30
bs 4 wfile
tpwr 62 proc ft
pw 5.0 fn not used
dl 4.000 math f
tof 0
nt 8 werr
ct 8 wexp
alock n wbs
gain not used wnt
FLAGS
il n
in n
dp y
hs nn
DISPLAY
sp -249.7
wp 5493.7
vs 435
sc 0
wc 400
hzmm 21.98
is 3598.38
rfl 5062.7
rfp 3571.0
th 7
ins 100.000
nm ph



dppf complex

exp1 s2pul

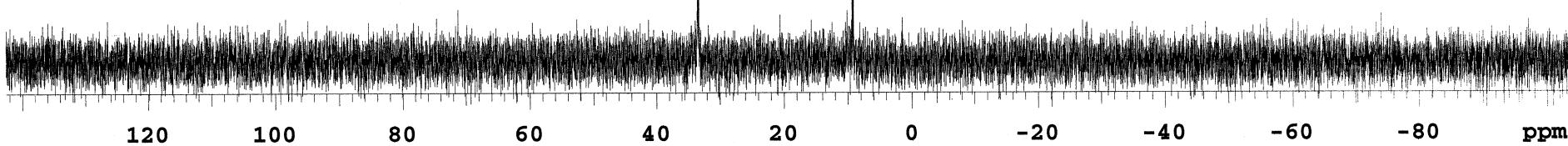
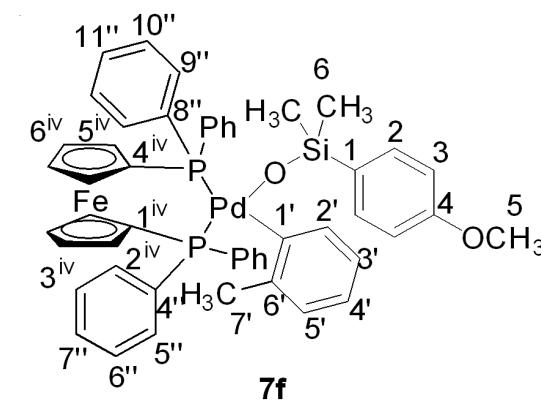
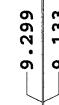
SAMPLE DEC. & VT
date May 8 2007 dfrq 499.698
solvent Benzene dn H1
file exp dpwr 44
ACQUISITION dof -827.6
sfrq 125.663 dm YYY
tn C13 dmm w
at 1.086 dmf 12000
np 65536 dseq
sw 30165.9 dres 90.0
fb 16600 homo n
bs 16 PROCESSING
ss 1 lb 1.00
tpwr 53 wtfile
pw 6.0 proc ft
d1 1.000 fn not used
t0f 1884.7 math f
nt 6000
ct 2912 werr
alock n wexp
gain not used wbs
FLAGS wnt
il n
in n
dp y
hs nn
DISPLAY
sp -1251.3
wp 30165.9
vs 941
sc 0
wc 400
hzmm 5.94
is 500.00
rfl 17334.3
rfp 16083.0
th 68
ins 100.000
nm ph



dppf complex

exp1 s2pul

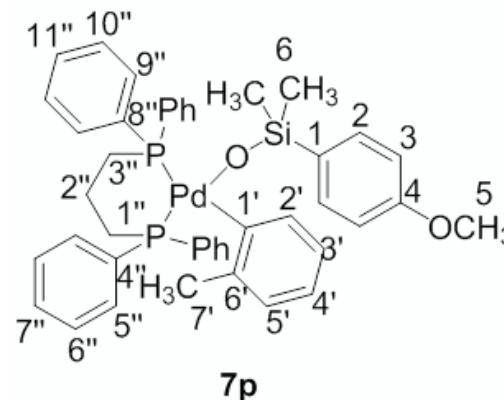
SAMPLE DEC. & VT
date Aug 29 2006 dfrq 499.438
solvent Benzene dn H1
file exp dpwr 44
ACQUISITION dof -820.0
sfreq 202.179 dm nny
tn P31 dmm w
at 0.655 dmf 18519
np 65536 dseq
sw 50000.0 dres 1.0
fb 27600 homo n
bs 16 PROCESSING
tpwr 63 lb 1.00
pw 4.0 wtfile
dl 1.000 proc ft
tof 8700.0 fn not used
nt 512 math f
ct 432
alock n werr
gain not used wexp
FLAGS wbs
il n wnt
in n
dp y
hs n
DISPLAY
sp -21172.2
wp 50000.0
vs 2185
sc 0
wc 250
hzmm 320.00
is 500.00
rfl 21172.2
rfp 0
th 65
ins 100.000
ai cdc ph



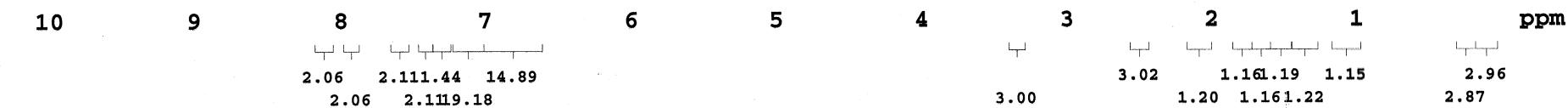
sat-xv-94-1H

exp1 s2pul

SAMPLE DEC. & VT
date Jun 7 2006 dfrq 499.438
solvent Benzene dn H1
file exp dpwr 25
ACQUISITION dof 0
sfrq 499.438 dm nnn
tn H1 dmnm c
at 4.096 dmf 200
np 65536 dseq
sw 8000.0 dres 1.0
fb 4400 homo n
bs 4 PROCESSING
tpwr 63 1b 0.30
pw 5.0 wtfile
dl 6.000 proc ft
tof 0 fn not used
nt 8 math f
ct 8
alock n werr
gain not used wexp
FLAGS wbs
il n wnt
in n
dp y
hs nn
DISPLAY
sp -249.9
wp 5493.6
vs 324
sc 0
wc 250
hzmm 21.98
is 1926.03
rfl 1507.3
rfp 0
th 7
ins 3.000
nm ph



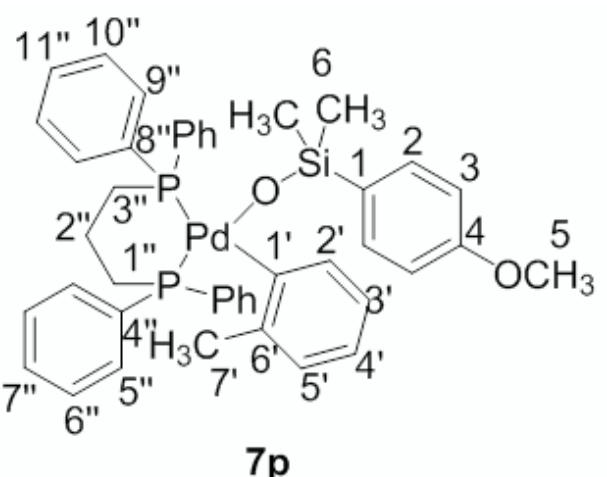
7p



sat-xv-94-13C

exp1 s2pul

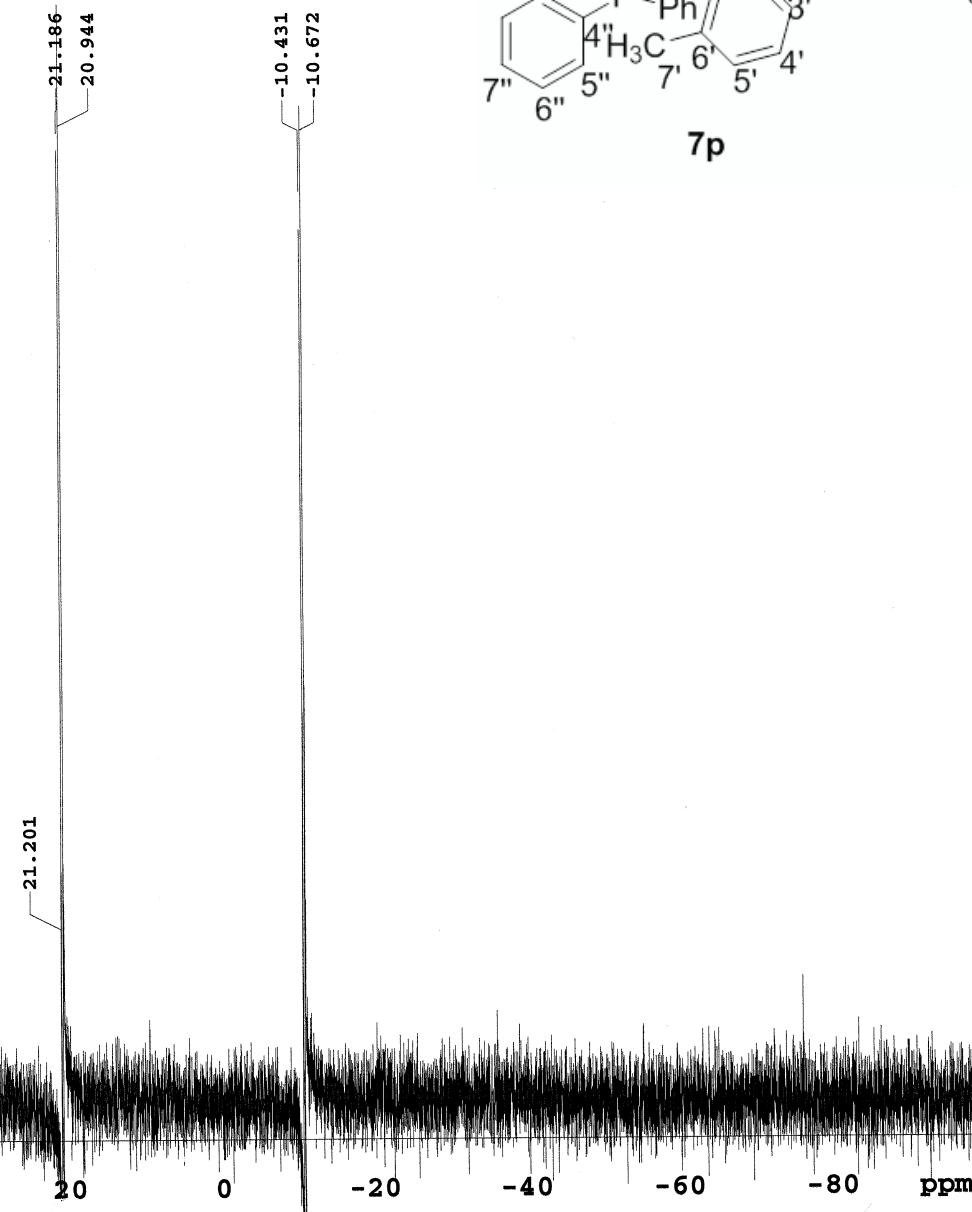
SAMPLE DEC. & VT
date Jun 12 2006 dfrq 499.439
solvent THF dn H1
file exp dpwr 44
ACQUISITION dof -827.0
sfrq 125.597 dm YYY
tn C13 dmm w
at 1.024 dmf 18519
np 65536 dseq
sw 32000.0 dres 1.0
fb 17600 homo n
bs 16 PROCESSING
tpwr 61 lb 1.00
pw 4.0 wtfile
d1 1.000 proc ft
tof 1880.0 fn not used
nt 4000 math f
ct 2208
alock n werr
gain not used wexp
FLAGS wbs
il n wnt
in n
dp y
hs nn
DISPLAY
sp -2090.7
wp 32000.0
vs 1146
sc 0
wc 400
hzmm 19.50
is 500.00
rf1 10555.0
rfp 8464.3
th 47
ins 100.000
nm ph



PdIIIsilanolate

exp1 s2pul

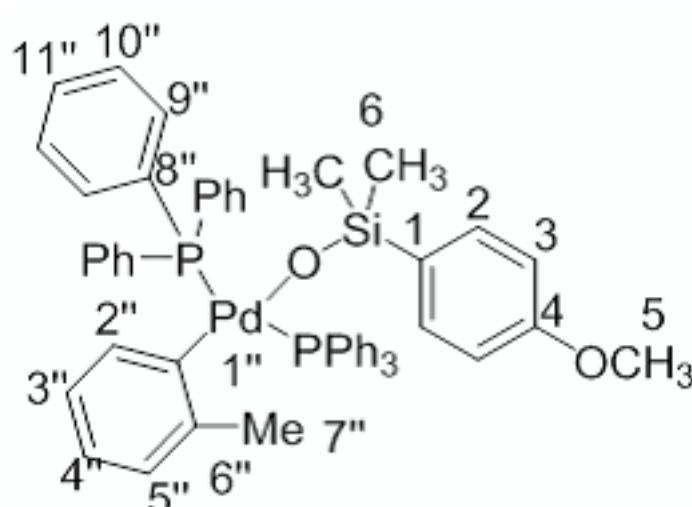
SAMPLE DEC. & VT
date Jun 12 2006 dfrq 499.440
solvent Toluene dn H1
file exp dpwr 44
ACQUISITION dof -820.0
sfrq 202.180 dm nny
tn P31 dmm w
at 0.655 dmf 18519
np 65536 dseq
sw 50000.0 dres 1.0
fb 27600 homo n
bs 16 PROCESSING
tpwr 63 lb 1.00
pw 4.0 wtfile
d1 1.000 proc ft
tof 8700.0 fn not used
nt 512 math f
ct 160
alock n werr
gain not used wexp
FLAGS wbs
il n wnt
in n
dp y
hs n
DISPLAY
sp -20171.2
wp 50000.0
vs 2033
sc 0
wc 250
hzmm 19.73
is 500.00
rf1 20171.2
rfp 0
th 102
ins 100.000
ai cdc ph



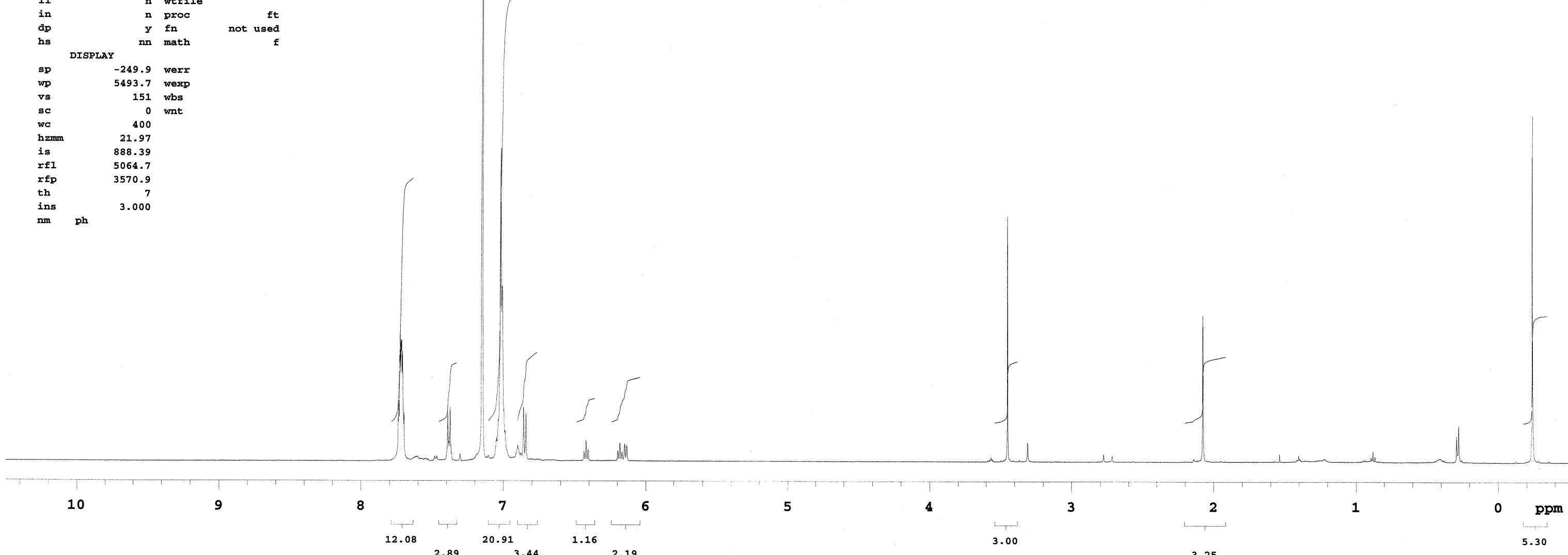
sat-xviii89x

exp1 s2pul

SAMPLE DEC. & VT
date Feb 1 2007 dfrq 499.432
solvent Benzene dn H1
file /export/home/~/dpwr 25
data/vxr500/Demmar~ dof 0
k/tymonko/satxviii~ dm nnn
89x.fid dmm c
ACQUISITION dmf 200
sfrq 499.432 dseq
tn H1 dres 1.0
at 4.096 homo n
np 65536 DEC2
sw 8000.0 dfrq2 0
fb 4000 dn2
bs 4 dpwr2 1
tpwr 62 dof2 0
pw 5.0 dm2 n
d1 6.000 dmm2 c
t0f 0 dm2 200
nt 8 dseq2
ct 8 dres2 1.0
alock n homo2 n
gain not used PROCESSING
FLAGS 1b 0.30
il n wfile
in n proc ft
dp y fn not used
hs nn math f
DISPLAY
sp -249.9 werr
wp 5493.7 wexp
vs 151 wbs
sc 0 wnt
wc 400
hzmm 21.97
is 888.39
rfl 5064.7
rfp 3570.9
th 7
ins 3.000
nm ph



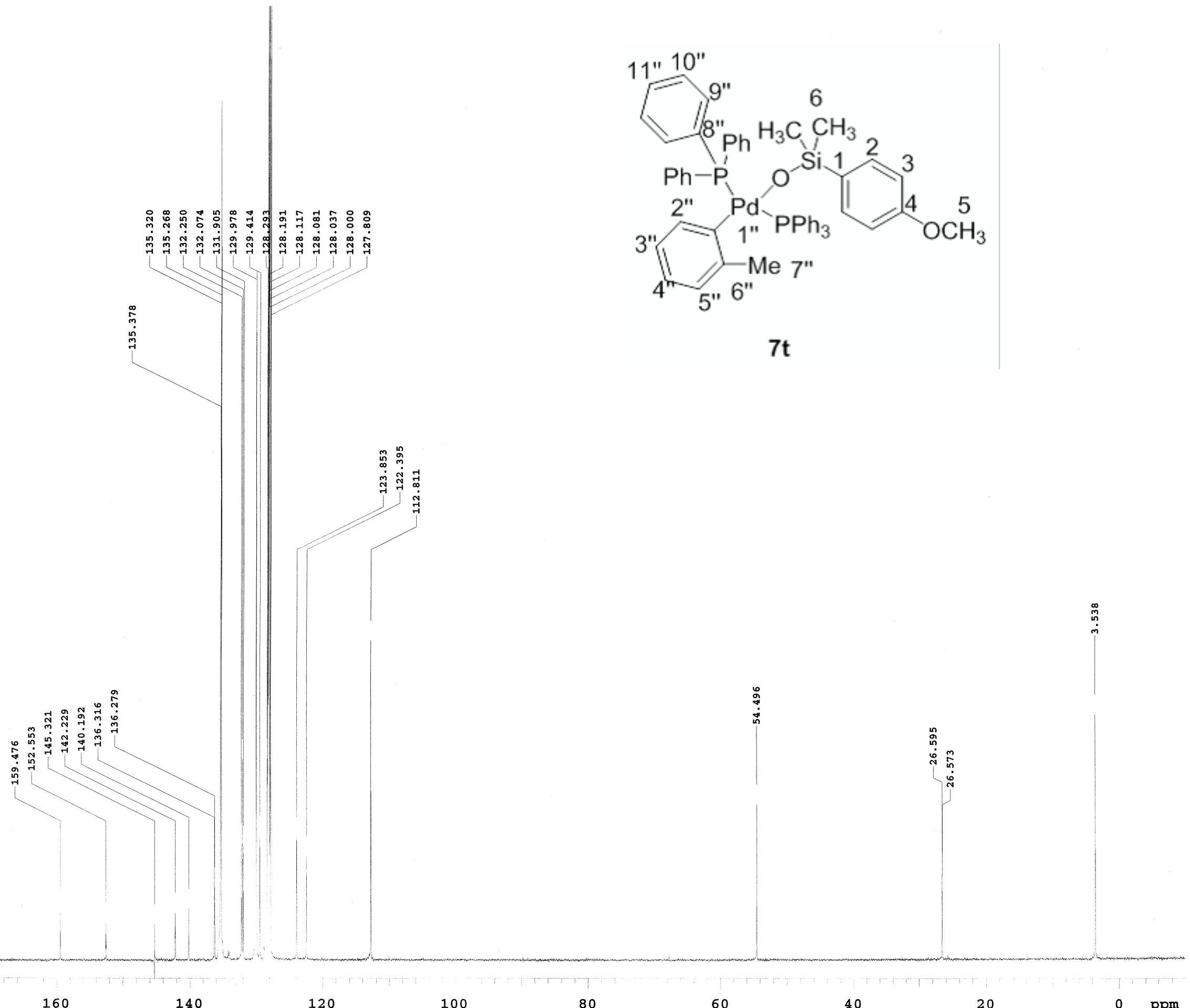
7t



PPh₃ complex

exp1 s2pul

SAMPLE DEC. & VT
date May 3 2007 dfrq 499.698
solvent Benzene dn H1
file exp dpwr 44
ACQUISITION dof -827.6
sfrq 125.663 dm YYY
tn C13 dmm w
at 1.086 dmf 12000
np 65536 dseq
sw 30165.9 dres 90.0
fb 16600 homo n
bs 16 PROCESSING
ss 1 lb 1.00
tpwr 53 wtfile
pw 6.0 proc ft
d1 1.000 fn not used
tof 1884.7 math f
nt 4000
ct 2384 werr
alock n wexp
gain not used wbs
FLAGS wnt
il n
in n
dp y
hs nn
DISPLAY
sp -1250.4
wp 30165.9
vs 280
sc 0
wc 400
hzmm 17.76
is 500.00
rfl 17333.4
rfp 16083.0
th 8
ins 100.000
nm ph

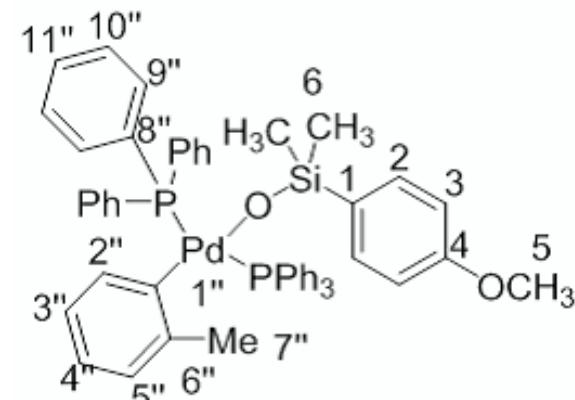


sat-xviii-89x

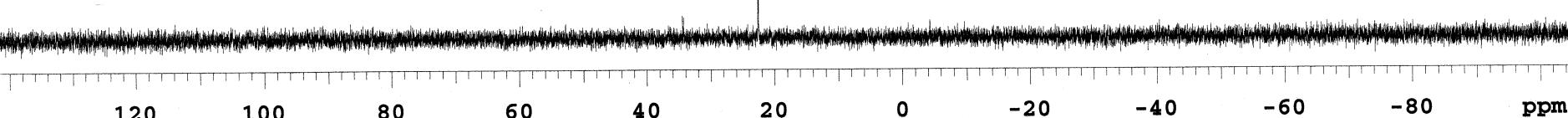
exp1 s2pul

SAMPLE	DEC. & VT
date Feb 1 2007	dfrq 499.432
solvent Benzene	dn H1
file exp	dpwr 44
ACQUISITION	dof -820.0
sfrq 202.176	dm nny
tn P31	dmmm w
at 0.655	dmf 19000
np 65536	dseq
sw 50000.0	dres 1.0
fb 28000	homo n
bs 16	DEC2
tpwr 63	dfrq2 0
pw 4.0	dn2
d1 1.000	dpwr2 1
tof 8700.0	dof2 0
nt 512	dm2 n
ct 80	dmmm2 c
alock n	dmf2 18519
gain not used	dseq2
FLAGS	dres2 1.0
il n	homo2 n
in n	PROCESSING
dp y	lb 1.00
hs n	wtfile
DISPLAY	proc ft
sp -21364.2	fn not used
wp 49998.5	math f
vs 410	
sc 0	werr
wc 250	wexp
hzmm 0.40	wbs
is 500.00	wnt
rfl 21365.7	
rfp 0	
th 131	
ins 100.000	
ai cdc ph	

22.557



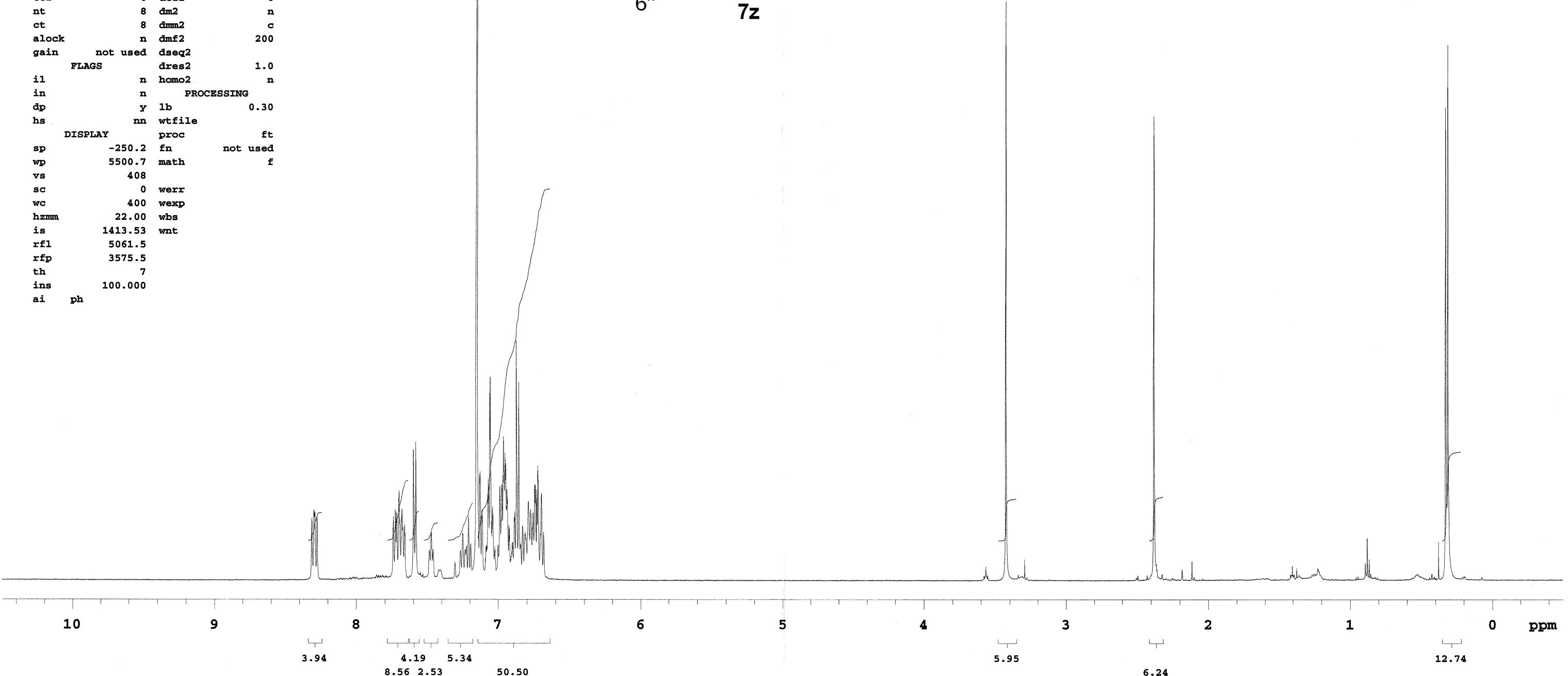
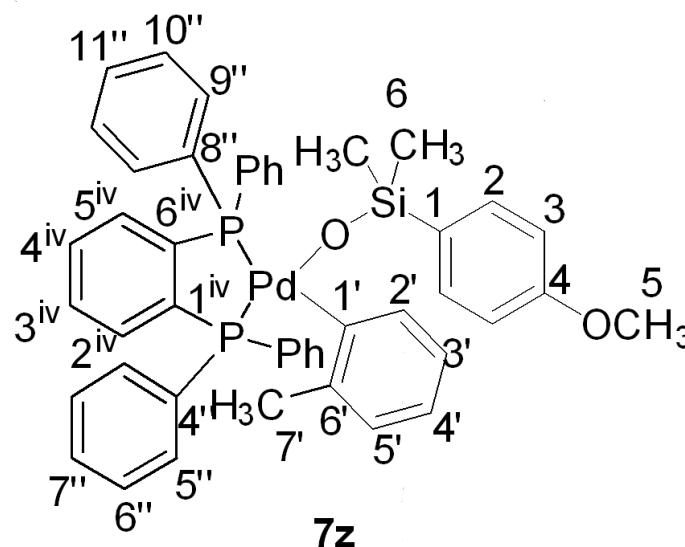
7t



sat-xvii-33x

exp1 s2pul

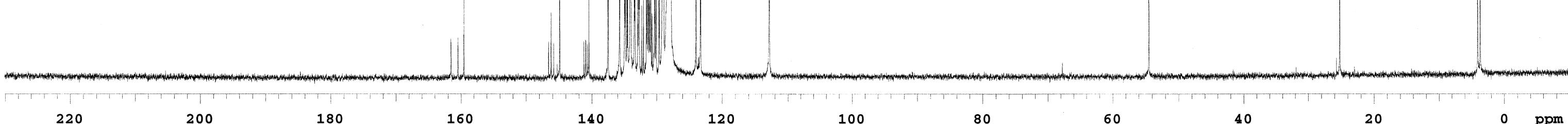
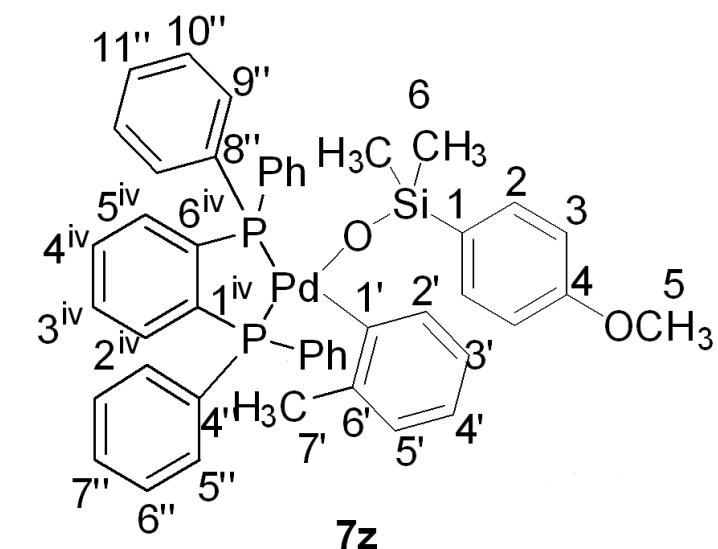
SAMPLE DEC. & VT
date Aug 30 2006 dfrq 500.076
solvent Benzene dn H1
file exp dpwr 18
ACQUISITION dof 0
sfrq 500.076 dm nnn
tn H1 dmm c
at 4.096 dmf 200
np 65536 dseq
sw 8000.0 dres 1.0
fb not used homo n
bs 16 DEC2
tpwr 55 dfrq2 0
pw 7.7 dn2
d1 5.000 dpwr2 1
tof 0 dof2 0
nt 8 dm2 n
ct 8 dmm2 c
alock n dmf2 200
gain not used dseq2
FLAGS dres2 1.0
il n homo2 n
in n PROCESSING
dp y lb 0.30
hs nn wtfile
DISPLAY proc ft
sp -250.2 fn not used
wp 5500.7 math f
vs 408
sc 0 werr
wc 400 wexp
hzmm 22.00 wbs
is 1413.53 wnt
rfl 5061.5
rfp 3575.5
th 7
ins 100.000
ai ph



1,2phenyl complex

exp1 s2pul

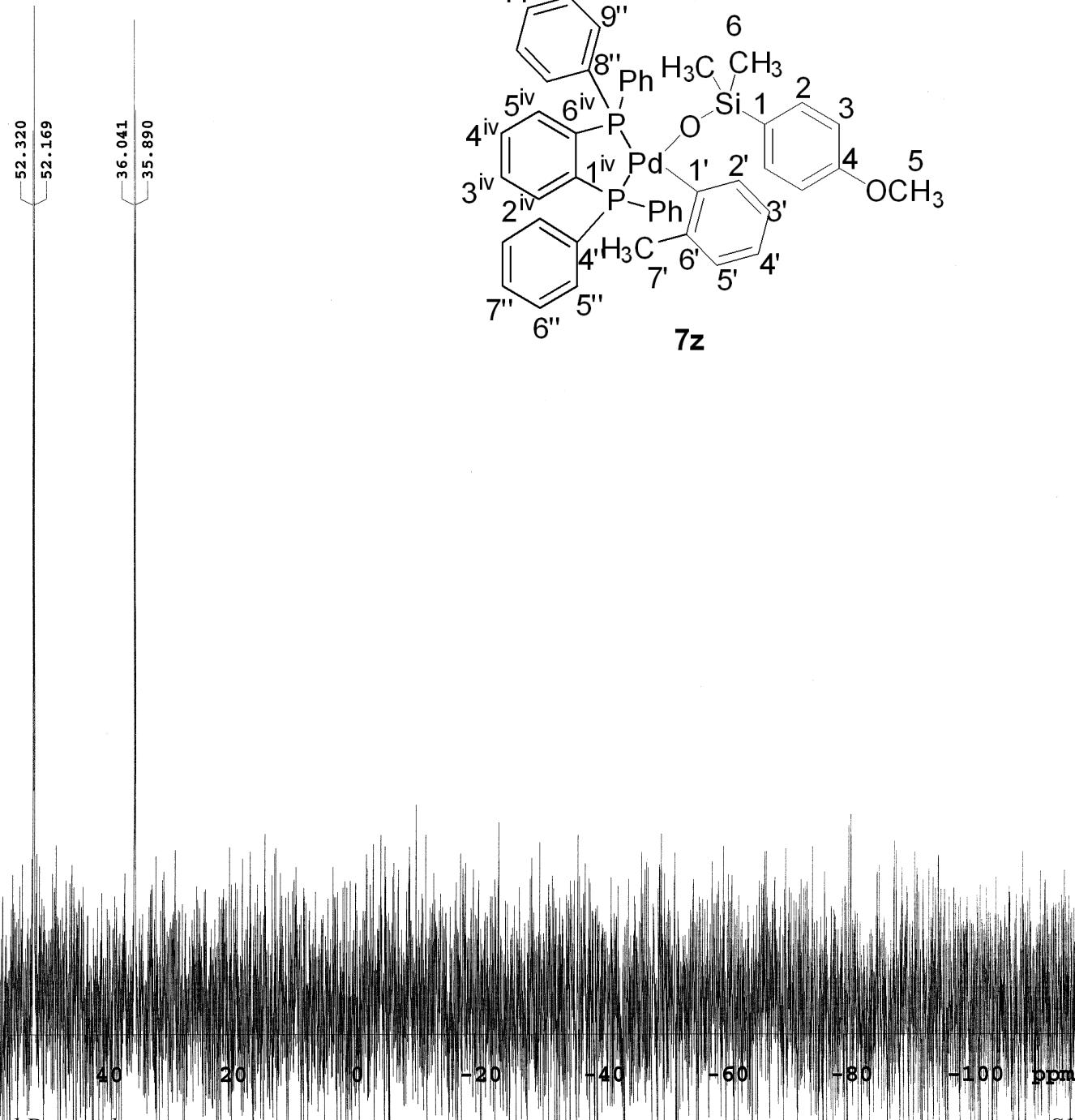
SAMPLE DEC. & VT
date May 6 2007 dfrq 499.698
solvent Benzene dn H1
file exp dpwr 44
ACQUISITION dof -827.6
sfrq 125.663 dm YYY
tn C13 dmm w
at 1.086 dmf 12000
np 65536 dseq
sw 30165.9 dres 90.0
fb 16600 homo n
bs 16 PROCESSING
ss 1 lb 1.00
tpwr 53 wtfile
pw 6.0 proc ft
di 1.000 fn not used
tof 1884.7 math f
nt 9000
ct 4304 werr
alock n wexp
gain not used wbs
FLAGS wnt
il n
in n
dp y
hs nn
DISPLAY
sp -1251.3
wp 30165.9
vs 867
sc 0
wc 400
hzmm 15.04
is 500.00
rfl 17334.3
rfp 16083.0
th 68
ins 100.000
nm ph



satxvii33-31p

exp1 s2pul

SAMPLE DEC. & VT
date Aug 30 2006 dfrq 399.950
solvent Benzene dn H1
file exp dpwr 48
ACQUISITION dof -1092.3
sfrq 161.903 dm nny
tn P31 dmm w
at 0.819 dmf 13889
np 65536 dseq
sw 40000.0 dres 1.0
fb 22000 homo n
bs 16 PROCESSING
tpwr 50 lb 3.00
pw 6.4 wtfile
d1 1.000 proc ft
tof 0 fn not used
nt 512 math f
ct 512
alock n werr
gain not used wexp
FLAGS wbs
il n wnt
in n
dp y
hs nn
DISPLAY
sp -19005.7
wp 40000.0
vs 162
sc 0
wc 250
hzmm 12.77
is 354.40
rf1 19005.7
rfp 0
th 123
ins 100.000
nm no ph



STANDARD PROTON PARAMETERS

Pulse Sequence: s2pul

Solvent: THF

Ambient temperature

UNITY-500 "u500"

Pulse 45.0 degrees

Acq. time 4.665 sec

Width 7024.9 Hz

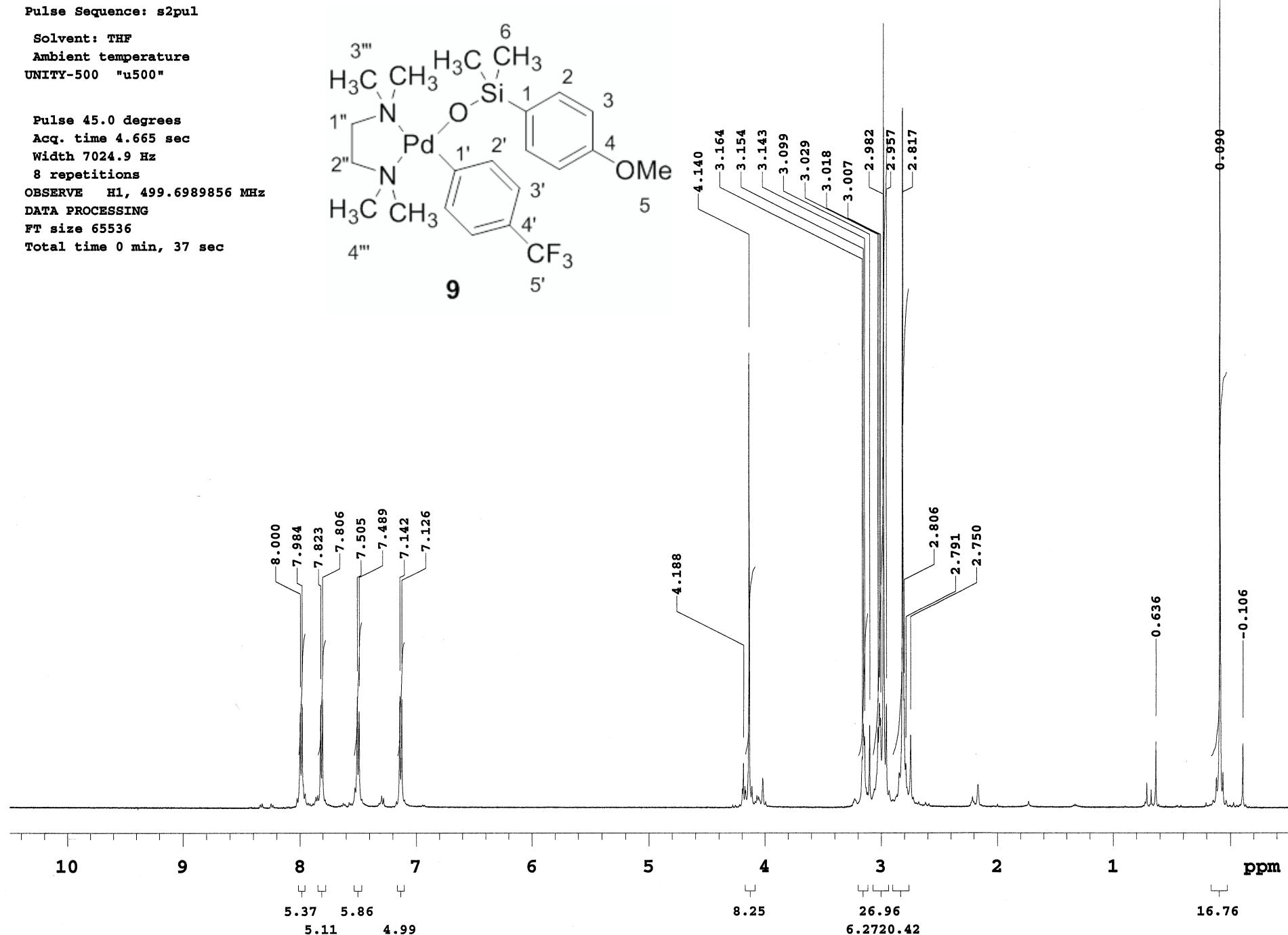
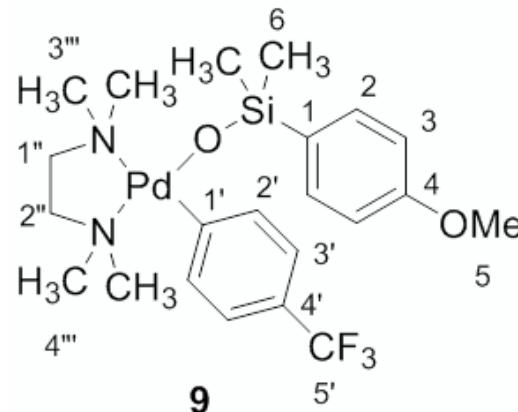
8 repetitions

OBSERVE H1, 499.6989856 MHz

DATA PROCESSING

FT size 65536

Total time 0 min, 37 sec



TMEDA-CF₃-OMe-C

Pulse Sequence: s2pul

Solvent: Benzene

Ambient temperature

UNITY-400 "u400"

Relax. delay 1.000 sec

Pulse 45.0 degrees

Acq. time 1.311 sec

Width 25000.0 Hz

459 repetitions

OBSERVE C13, 100.5671603 MHz

DECOUPLE H1, 399.9500478 MHz

Power 48 dB

continuously on

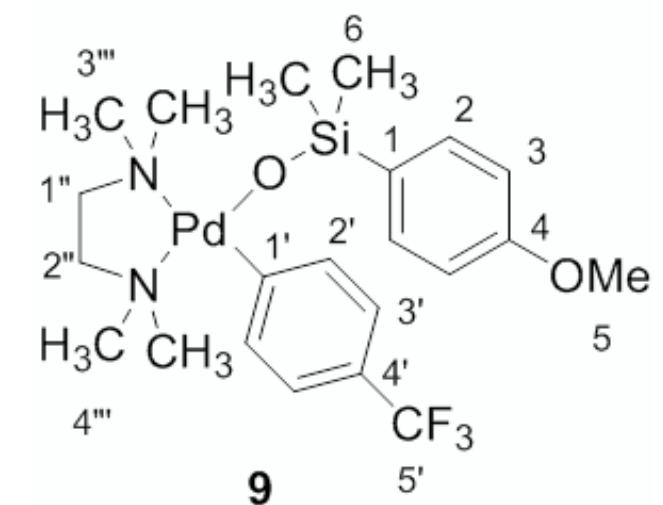
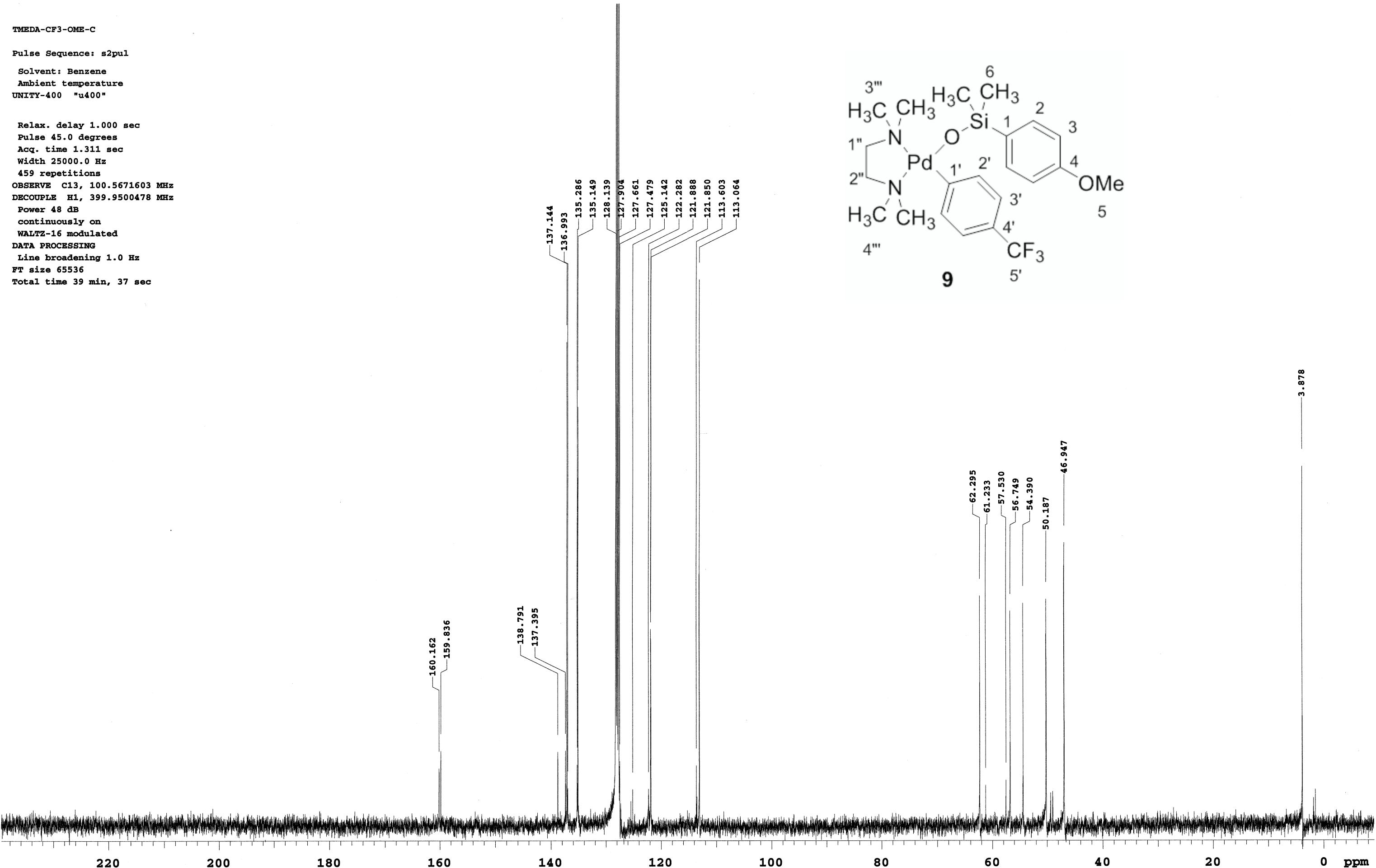
WALTZ-16 modulated

DATA PROCESSING

Line broadening 1.0 Hz

FT size 65536

Total time 39 min, 37 sec



TMEDA-CF₃-OMe-F

Pulse Sequence: s2pul

Solvent: Benzene

Ambient temperature

UNITY-400 "u400"

Relax. delay 1.000 sec

Pulse 45.0 degrees

Acq. time 0.328 sec

Width 100.0 kHz

Single scan

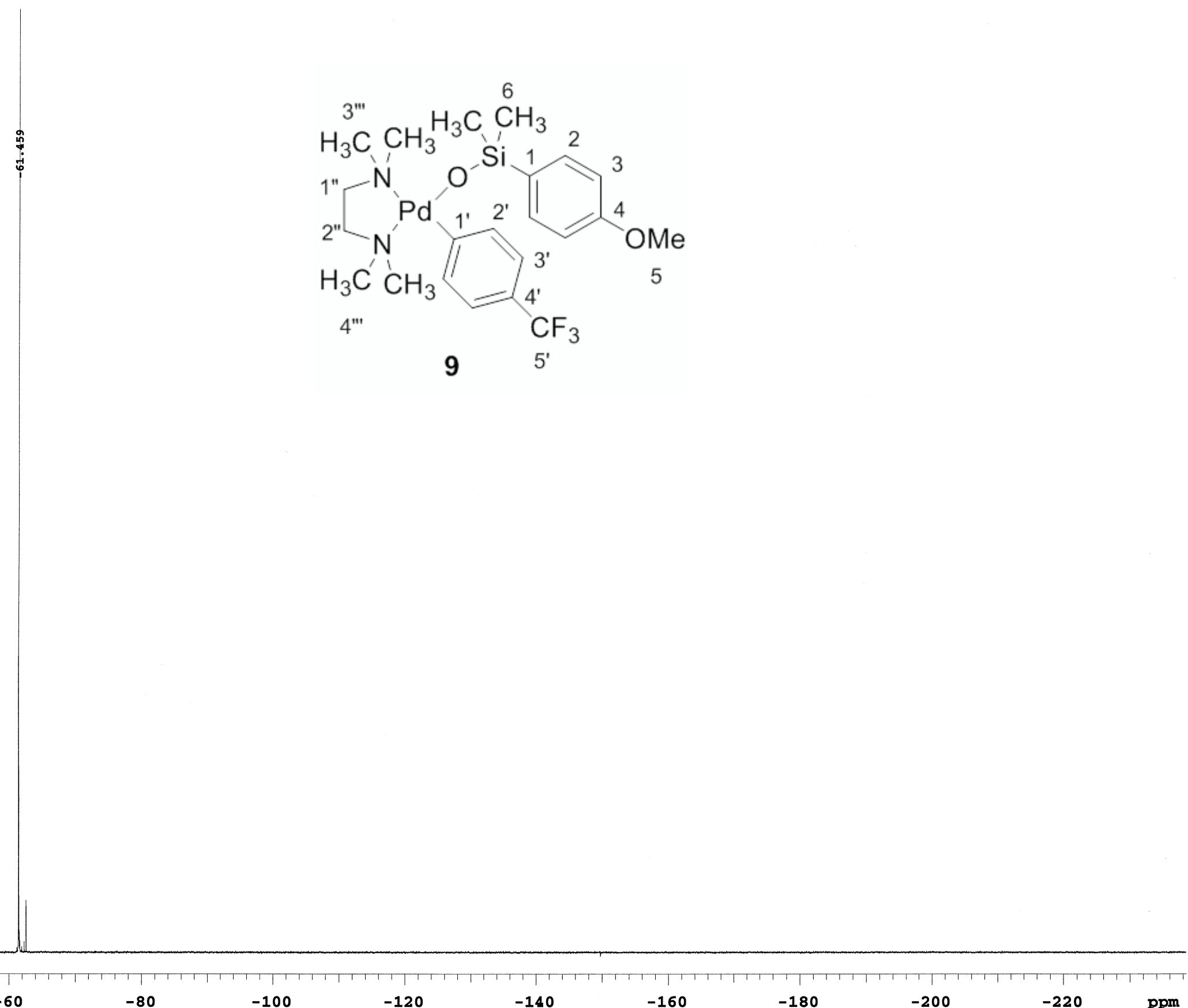
OBSERVE F19, 376.3279346 MHz

DATA PROCESSING

Line broadening 3.0 Hz

FT size 131072

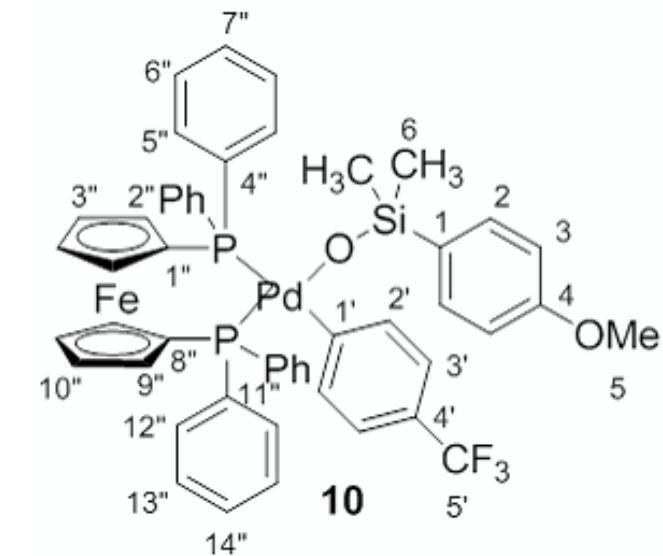
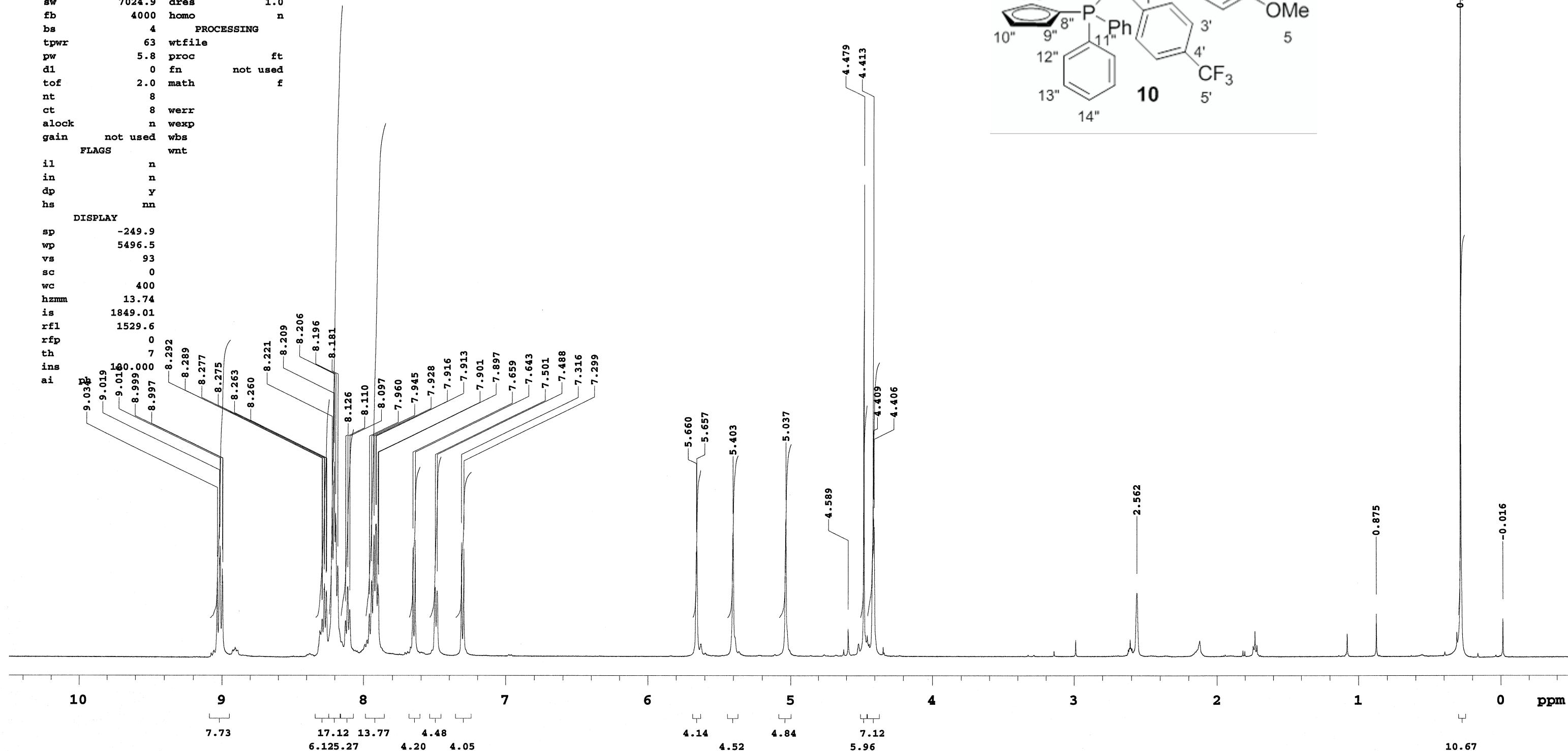
Total time 0 min, 1 sec



STANDARD PROTON PARAMETERS

exp1 s2pul

SAMPLE DEC. & VT
date Jun 22 2005 dfreq 499.701
solvent THF dn H1
file exp dpwr 20
ACQUISITION dof 0
sfrq 499.701 dm nnn
tn H1 dmm c
at 4.665 dmf 200
np 65536 dseq
sw 7024.9 dres 1.0
fb 4000 homo n
bs 4 PROCESSING
tpwr 63 wtfile
pw 5.8 proc ft
d1 0 fn not used
tof 2.0 math f
nt 8
ct 8 werr
alock n wexp
gain not used wbs
FLAGS wnt
il n
in n
dp y
hs nn
DISPLAY
sp -249.9
wp 5496.5
vs 93
sc 0
wc 400
hzmm 13.74
is 1849.01
rfl 1529.6
rfp 0
th
ins 0.000
ai



STANDARD CARBON PARAMETERS

Pulse Sequence: s2pul

Solvent: THF

Ambient temperature

User: 1-14-87

UNITY-500 "u500"

Relax. delay 1.000 sec

Pulse 45.0 degrees

Acq. time 1.086 sec

Width 30165.9 Hz

2400 repetitions

OBSERVE C13, 125.6493365 MHz

DECOUPLE H1, 499.6999412 MHz

Power 44 dB

continuously on

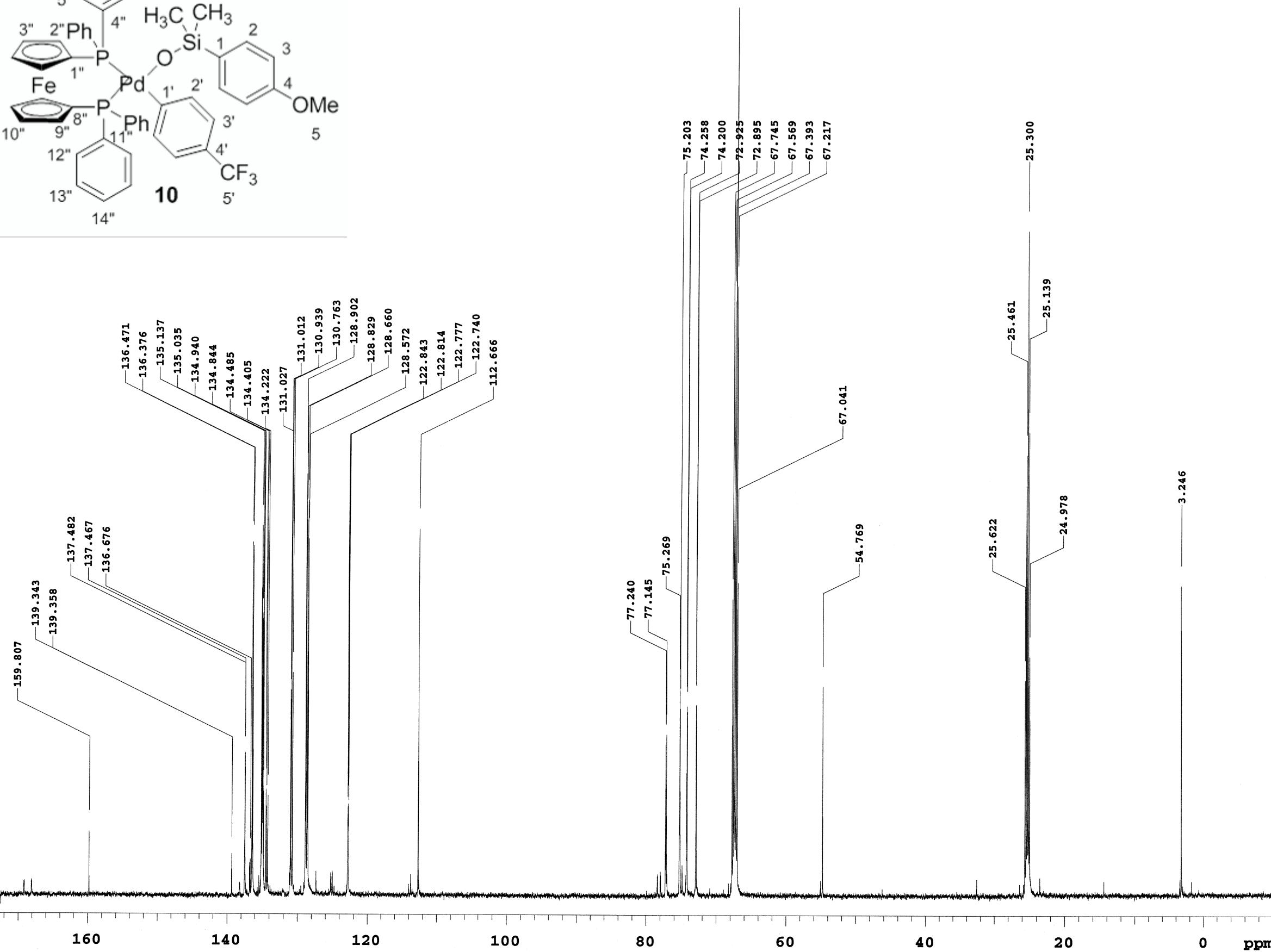
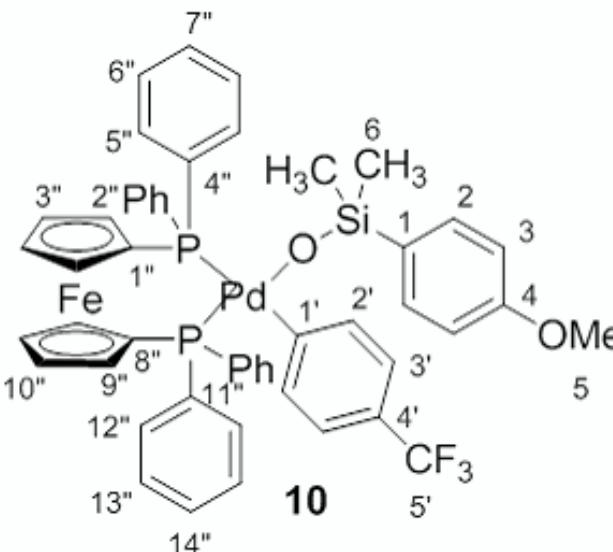
WALTZ-16 modulated

DATA PROCESSING

Line broadening 1.0 Hz

FT size 65536

Total time 1 hr, 23 min, 55 sec



P31 TRIPHENYLPHOSPHATE PARAMETERS

Pulse Sequence: s2pul

Solvent: THF

Ambient temperature
UNITY-500 "u500"

Relax. delay 1.000 sec

Pulse 45.0 degrees

Acq. time 0.655 sec

Width 50000.0 Hz

8 repetitions

OBSERVE P31, 202.2813436 MHz

DECOUPLE H1, 499.6999412 MHz

Power 44 dB

continuously on

WALTZ-16 modulated

DATA PROCESSING

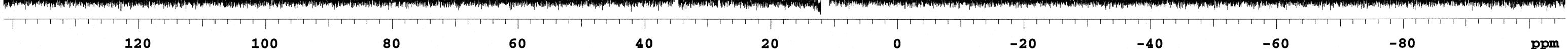
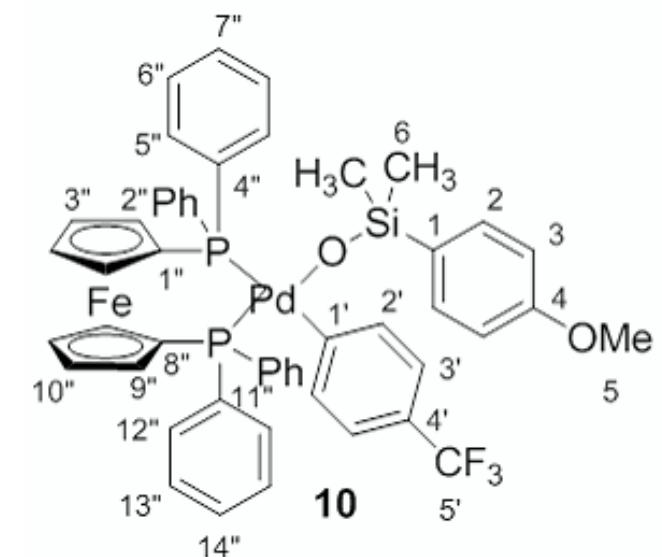
Line broadening 1.0 Hz

FT size 65536

Total time 0 min, 14 sec

35.028
34.870

12.043
11.885



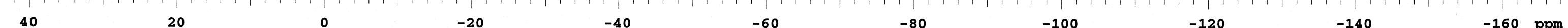
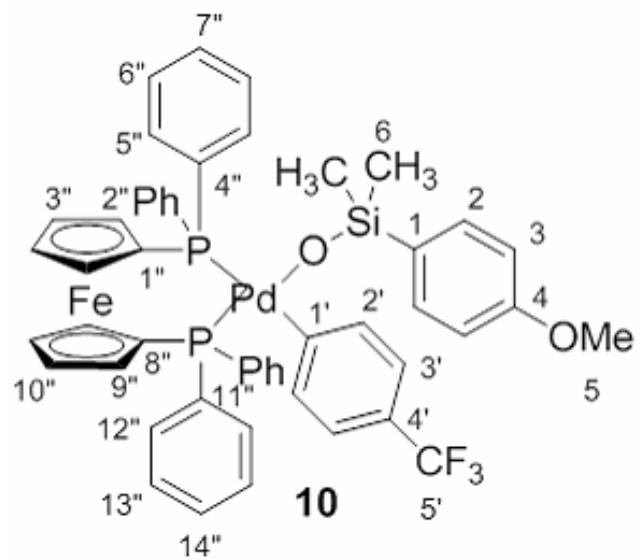
¹⁹F

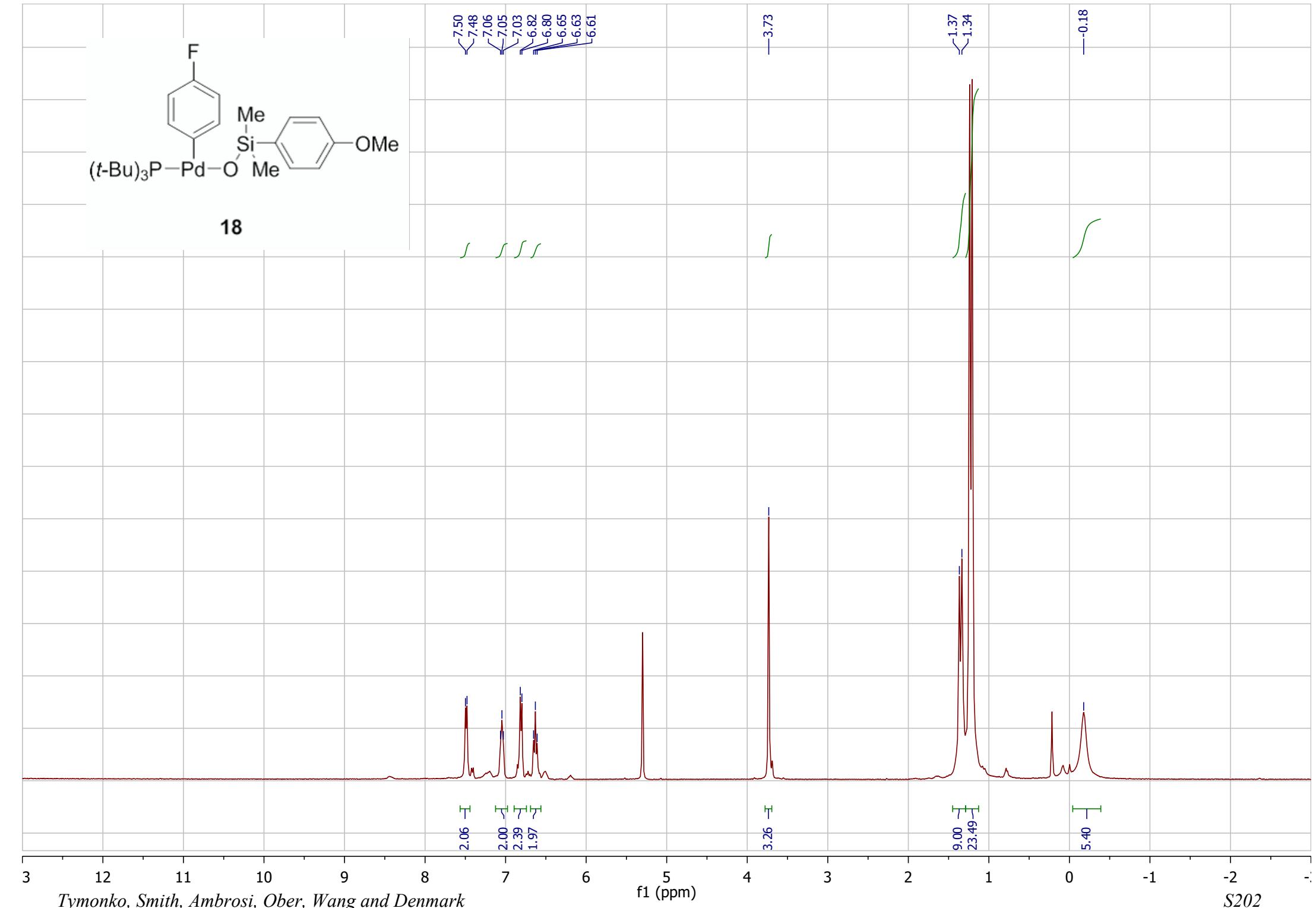
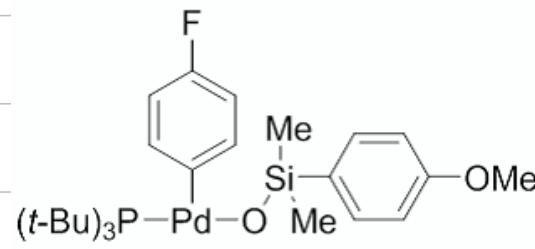
~~P31~~ TRIPHENYLPHOSPHATE PARAMETERS

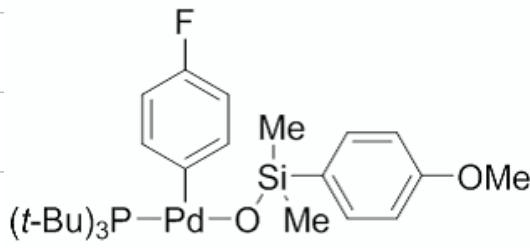
Pulse Sequence: s2pul

Solvent: THF
Ambient temperature
UNITY-500 "u500"

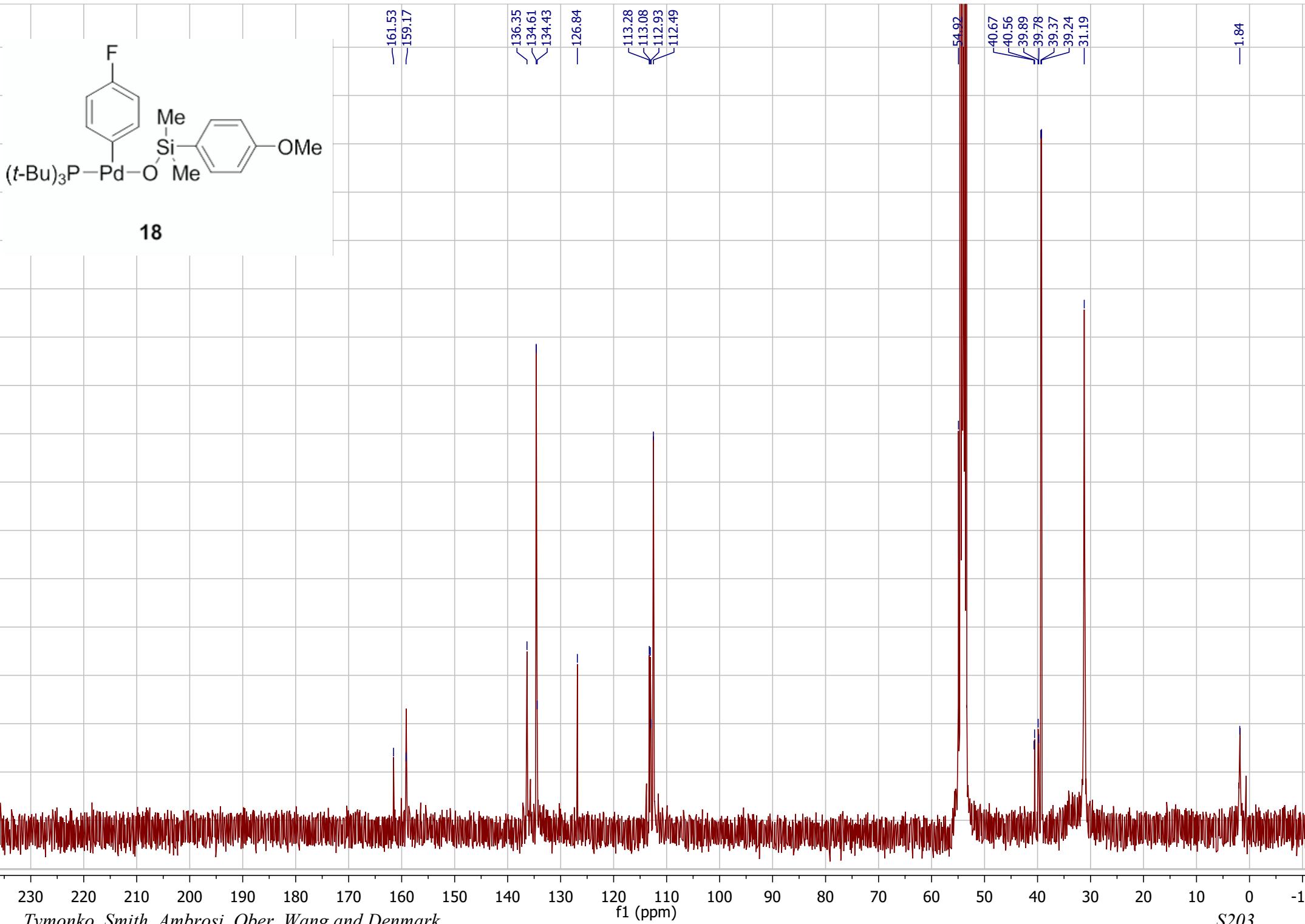
Pulse 45.0 degrees
Acq. time 0.328 sec
Width 100.0 kHz
Single scan
OBSERVE F19, 470.1863333 MHz
DATA PROCESSING
Line broadening 2.0 Hz
FT size 65536
Total time 0 min, 0 sec

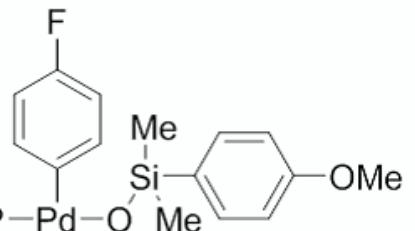




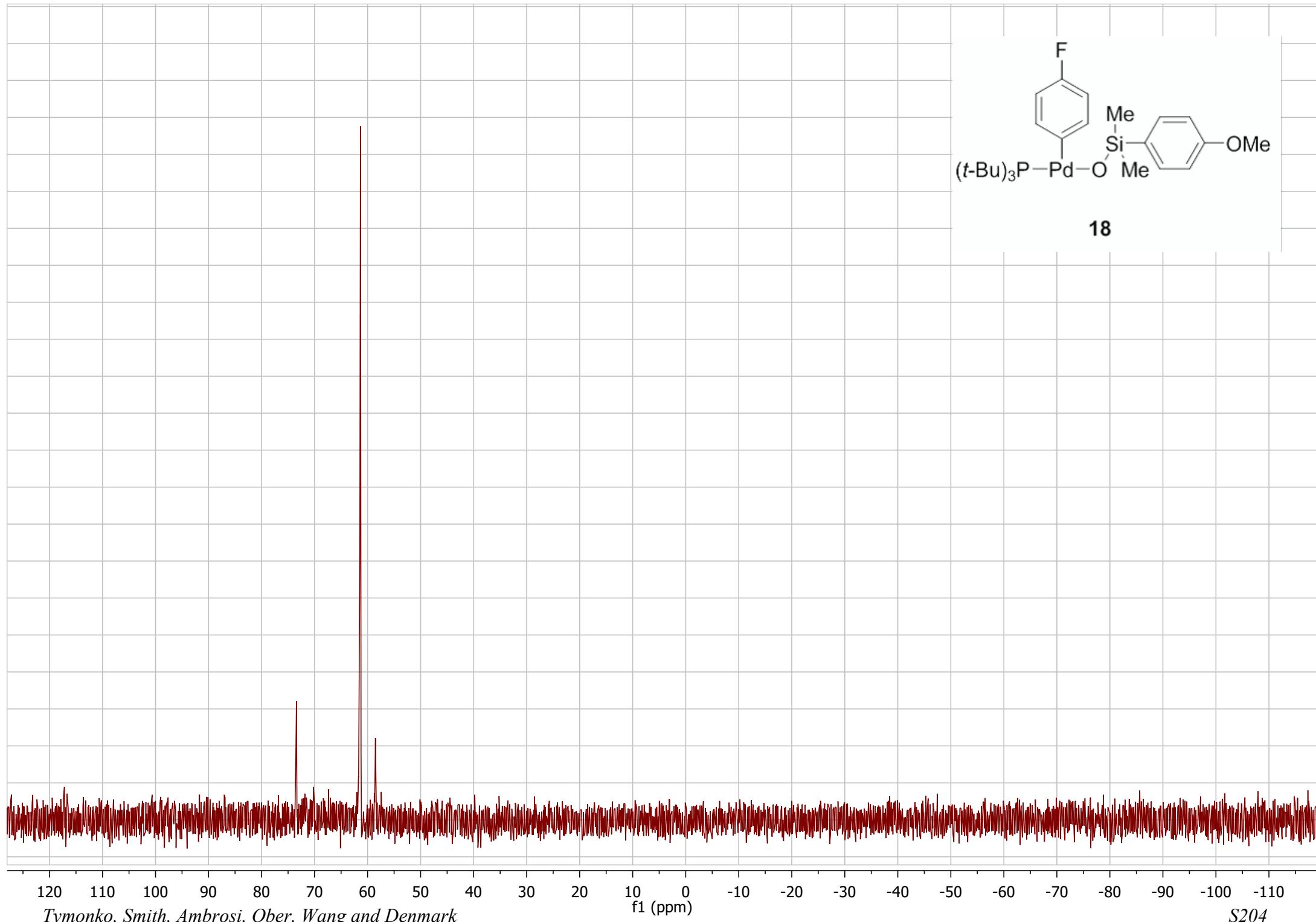


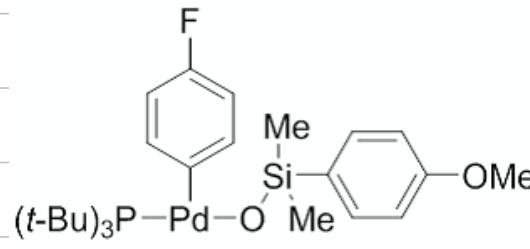
18





18





18

-123.52

111 -112 -113 -114 -115 -116 -117 -118 -119 -120 -121 -122 -123 -124 -125 -126 -127 -128 -129 -130 -131 -132 -133 -134 -135 -136 -137 -138 -139

f1 (ppm)

